### P 274 010,097

RECEIPT FOR CERTIFIED MAIL

NO INSURANCE COVERAGE PROVIDED

NOT FOR INTERNATIONAL MAIL

(See Reverse)

¤ U.S.G.P.O. 1985-480-794	Septames R. Kolanek,  Harris Semiconduct Street and No. P.O. Box 883 P.O. State and ZIP Code	
9.	Melbourne, FL 3290	)1
J.S.G.	Postage	S
æ it	Certified Fee	
	Special Delivery Fee	
ź	Restricted Delivery Fee	
	Return Receipt showing to whom and Date Delivered	
1986	Return Receipt showing to whom, Date, and Address of Delivery	
Jun	TOTAL Postage and Fees	S
3800,	Postmark or Date	
Ę	Mailed: 02/03/88	
PS Form 3800, June 1985	Permit: AC 05-138	3795

O CENTER OF THE PROPERTY OF TH	
SENDER: Complete items 1 and 2 when additional services	icec are desired
Put your address in the "RETURN TO" space on the revers card from being returned to you. The return receipt fee will delivered to and the date of delivery. For additional fees the postmaster for fees and check box(es) for additional service  1. KShow to whom delivered, date, and addressee's addressed and the company of	e side. Failure to do this will prevent this provide you the name of the person of sollowing services are available. Consult (s) requested.
Environmental Services	
Harris Semiconductor	P 274 010 097
P.O. Box 883	Type of Service:
Melbourne, FL 32901	Registered Insured COD
5. Signature Addressee	Always obtain signature of addressee or agent and DATE DELIVERED.
X	8. Addressee's Address QNLY if requested wind fee paid.
6. Signature — Agent Age	FEB TO 1988
, a	DOMESTIC RETURN RECEIPT

Jue

### STATE OF FLORIDA

### DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION
NOTICE OF PERMIT

Mr. James R. Kolanek, Manager Environmental Services Harris Semiconductor Post Office Box 883 Melbourne, Florida 32901

February 3, 1988

Enclosed is permit No. AC 05-138795, for Harris Semiconductor to install/construct an Industrial Grade Water System to provide water for the Deionized Water Plants in Buildings 52 and 59. A vacuum degasifier will be used to remove hydrogen sulfide and carbon dioxide from the raw well water and a flare will be used to oxidize the gasses. The project will occur at Harris Semiconductor's existing facility in Melbourne, Brevard County, Florida. This permit is issued pursuant to Section 403, Florida Statutes.

Any Party to this permit has the right to seek judicial review of the permit pursuant to Section 120.68, Florida Statutes, by the filing of Notice of Appeal pursuant to Rule 9.110, Florida Rules of Appellate Procedure, with the Clerk of the Department in the Office of General Counsel, 2600 Blair Stone Road, Tallahassee, Florida 32399-2400; and by filing a copy of the Notice of Appeal accompanied by the applicable filing fees with the appropriate District Court of Appeal. The Notice of Appeal must be filed within 30 days from the date this permit is filed with the Clerk of the Department.

Executed in Tallahassee, Florida.

STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

C. H. Fancy P.E.

Deputy Chief

Bureau of Air Quality Management

Copy furnished to:

T. Sawicki, CF Dist.

C. Bach, P.E.

Final Determination

Harris Semiconductor Brevard County Melbourne, Florida

Permit No. AC 05-138795

3.

Florida Department of Environmental Regulation Bureau of Air Quality Management Central Air Permitting

### Final Determination

The construction permit application has been reviewed by the Department. Public Notice of the Department's Intent to Issue was published in the Florida Today Newspaper, Cocoa, Florida, on January 6, 1988. The Technical Evaluation and Preliminary Determination were available for public inspection at the DER's Central Florida District office in Orlando and the DER's Bureau of Air Quality Management office in Tallahassee.

No comments were received during the Public Notice period. Therefore, the final action of the Department will be to issue the construction permit as drafted.

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### STATE OF FLORIDA

### DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

PERMITTEE:
Harris Semiconductor
P. O. Box 883
Melbourne, Florida 32901

Permit Number: AC 05-138795 Expiration Date: January 31, 1989 County: Brevard

Latitude/Longitude: 28° 01' 20"N/ 80° 36' 10" W

Project: Industrial Grade Water
System with Vacuum
Degasifier and Flare
System

This permit is issued under the provisions of Chapter 403, Florida Statutes, and Florida Administrative Code Rule(s) 17-2 and 17-4. The above named permittee is hereby authorized to perform the work or operate the facility shown on the application and approved drawing(s), plans, and other documents attached hereto or on file with the Department and made a part hereof and specifically described as follows:

For the construction/installation of a 600 gallon per minute Industrial Grade water system with a vacuum degasifier and flare system. The vacuum degasifier will remove hydrogen sulfide and carbon dioxide from raw well water and the removed gases will be transported to and oxidized/combusted by the flare system. The nonassisted type flare will be designed and built by the John Zink Company, which includes a self-supported flare stack, a Model EEF-U-2 flare tip (John Zink Co.), and a manual weatherproof pilot ignition panel. The construction/installation will occur at the permittee's existing facility located on Palm Bay Road, City of Palm Bay, Florida. The UTM coordinates are Zone 17, 538.7 km East and 3100.9 km North.

The source shall be in accordance with the permit application, plans, documents, amendments and drawings, except as otherwise noted in the Specific Conditions.

### Attachments to be Incorporated:

- Application to Construct Air Pollution Source, DER Form 17-1.202(1), and Mr. James R. Kolanek's cover letter dated August 28, 1987, and received August 31, 1987.
- 2. Mr. C. H. Fancy's letter dated September 25, 1987.
- Mr. J. R. Kolanek's letter with enclosures dated October 16, 1987, and received October 23, 1987.
- 4. Copy of 40 CFR 60.18, as revised July 1, 1986.
- 5. Technical Evaluation and Preliminary Determination dated December 4, 1987.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### GENERAL CONDITIONS:

- 1. The terms, conditions, requirements, limitations, and restrictions set forth herein are "Permit Conditions" and as such are binding upon the permittee and enforceable pursuant to the authority of Sections 403.161, 403.727, or 403.859 through 403.861, Florida Statutes. The permittee is hereby placed on notice that the Department will review this permit periodically and may initiate enforcement action for any violation of the "Permit Conditions" by the permittee, its agents, employees, servants or representatives.
- 2. This permit is valid only for the specific processes and operations applied for and indicated in the approved drawings or exhibits. Any unauthorized deviation from the approved drawings, exhibits, specifications, or conditions of this permit may constitute grounds for revocation and enforcement action by the Department.
- 3. As provided in Subsections 403.087(6) and 403.722(5), Florida Statutes, the issuance of this permit does not convey any vested rights or any exclusive privileges. Nor does it authorize any injury to public or private property or any invasion of personal rights, nor any infringement of federal, state or local laws or regulations. This permit does not constitute a waiver of or approval of any other Department permit that may be required for other aspects of the total project which are not addressed in the permit.
- 4. This permit conveys no title to land or water, does not constitute state recognition or acknowledgement of title, and does not constitute authority for the use of submerged lands unless herein provided and the necessary title or leasehold interests have been obtained from the state. Only the Trustees of the Internal Improvement Trust Fund may express state opinion as to title.
- 5. This permit does not relieve the permittee from liability for harm or injury to human health or welfare, animal, plant or aquatic life or property and penalties therefore caused by the construction or operation of this permitted source, nor does it allow the permittee to cause pollution in contravention of Florida Statutes and Department rules, unless specifically authorized by an order from the Department.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### **GENERAL CONDITIONS:**

- 6. The permittee shall at all times properly operate and maintain the facility and systems of treatment and control (and related appurtenances) that are installed or used by the permittee to achieve compliance with the conditions of this permit, as required by Department rules. This provision includes the operation of backup or auxiliary facilities or similar systems when necessary to achieve compliance with the conditions of the permit and when required by Department rules.
- 7. The permittee, by accepting this permit, specifically agrees to allow authorized Department personnel, upon presentation of credentials or other documents as may be required by law, access to the premises, at reasonable times, where the permitted activity is located or conducted for the purpose of:
  - a. Having access to and copying any records that must be kept under the conditions of the permit;
  - Inspecting the facility, equipment, practices, or operations regulated or required under this permit; and
  - c. Sampling or monitoring any substances or parameters at any location reasonably necessary to assure compliance with this permit or Department rules.

Reasonable time may depend on the nature of the concern being investigated.

- 8.—If, for any reason, the permittee does not comply with or will be unable to comply with any condition or limitation specified in this permit, the permittee shall immediately notify and provide the Department with the following information:
  - a. a description of and cause of non-compliance; and
  - b. the period of noncompliance, including exact dates and times; or, if not corrected, the anticipated time the noncompliance is expected to continue, and steps being taken to reduce, eliminate, and prevent recurrence of the noncompliance.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### **GENERAL CONDITIONS:**

The permittee shall be responsible for any and all damages which may result and may be subject to enforcement action by the Department for penalties or revocation of this permit.

- 9. In accepting this permit, the permittee understands and agrees that all records, notes, monitoring data and other information relating to the construction or operation of this permitted source, which are submitted to the Department, may be used by the department as evidence in any enforcement case arising under the Florida Statutes or Department rules, except where such use is proscribed by Sections 403.73 and 403.111, Florida Statutes.
- 10. The permittee agrees to comply with changes in Department rules and Florida Statutes after a reasonable time for compliance, provided however, the permittee does not waive any other rights granted by Florida Statutes or Department rules.
- 11. This permit is transferable only upon department approval in accordance with Florida Administrative Code Rules 17-4.12 and 17-30.30, as applicable. The permittee shall be liable for any noncompliance of the permitted activity until the transfer is approved by the Department.
- 12. This permit is required to be kept at the work site of the permitted activity during the entire period of construction or operation.
- 13. This permit also constitutes:

- ( ) Determination of Best Available Control Technology (BACT)
- ( ) Determination of Prevention of Significant Deterioration (PSD)
- ( ) Compliance with New Source Performance Standards
- 14. The permittee shall comply with the following monitoring and record keeping requirements:
  - a. Upon request, the permittee shall furnish all records and plans required under Department rules. The retention period for all records will be extended automatically, unless otherwise stipulated by the Department, during the course of any unresolved enforcement action.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### **GENERAL CONDITIONS:**

- b. The permittee shall retain at the facility or other location designated by this permit records of all monitoring information (including all calibration and maintenance records and all original strip chart recordings for continuous monitoring instrumentation), copies of all reports required by this permit, and records of all data used to complete the application for this permit. The time period of retention shall be at least three years from the date of the sample, measurement, report or application unless otherwise specified by Department rule.
- c. Records of monitoring information shall include:
  - the date, exact place, and time of sampling or measurements;
  - the person responsible for performing the sampling or measurements;
  - the date(s) analyses were performed;
  - the person responsible for performing the analyses;
  - the analytical techniques or methods used; and
  - the results of such analyses.
- 15. When requested by the Department, the permittee shall within a reasonable time furnish any information required by law which is needed to determine compliance with the permit. If the permittee becomes aware that relevant facts were not submitted or were incorrect in the permit application or in any report to the Department, such facts or information shall be submitted or corrected promptly.

### SPECIFIC CONDITIONS:

- 1. Annual hours of operation are 8760.
- 2. The maximum potential sulfur dioxide  $(SO_2)$  emissions are 7.0 pounds per hour and 30.7 tons per year.
- 3. The maximum potential hydrogen sulfide (H<sub>2</sub>S) emissions are 493 pounds per year, which is based on a flare efficiency of 98.5%.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### SPECIFIC CONDITIONS:

- 4. The permittee shall comply with the conditions of 40 CFR 60.18(c) thru (f).
- 5. No visible emissions (5% opacity) shall be allowed, except for a total period of 5 minutes during any consecutive 2 hours, pursuant to 40 CFR 60.18(c). Compliance shall be demonstrated annually using EPA Method 22 pursuant to 40 CFR 60.18(f)(1).
- 6. EPA Method 15 shall be performed annually to determine the maximum concentration of the  $\rm H_2S$  prior to being flared and the result should be in terms of dry standard conditions (14.7 psia and 680 F). A retest shall be required if the concentration of  $\rm H_2S$  is to be increased.
- 7. The exit velocity of the flare shall be determined using the procedure in 40 CFR 60.18(f)(4) and either EPA Method 2, 2A, 2C or 2D, as appropriate.
- 8. EPA Methods shall be as described in 40 CFR 60, Appendix A.
- 9. The Central Florida District shall be notified in writing 15 days in advance of any compliance testing and the test reports shall be submitted within 45 days after the last test run.
- 10. Objectionable odors shall not be allowed off plant property pursuant to FAC Rule 17-2.620(2).
- 11. The construction shall reasonably conform to the plans and schedule submitted in the application. If the permittee is unable to complete construction on schedule, he must notify the Department in writing 60 days prior to the expiration date of the construction permit and submit a new schedule and request for an extension of the construction permit. (FAC Rule 17-4.09)

To obtain a permit to operate, the permittee must demonstrate compliance with the conditions of the construction permit and submit a complete application for an operating permit, including the application fee, along with compliance test results and Certificate of Completion, to the Department's District office 90 days prior to the expiration date of the construction permit. The permittee may continue to operate in compliance with all terms of the construction permit until its expiration date. Operation beyond the construction permit expiration date requires a valid permit to operate. (FAC Rules 17-4.22 and 17-4.23)

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

### SPECIFIC CONDITIONS:

If the construction permit expires prior to the permittee requesting an extension or obtaining a permit to operate, then all activities at the project must cease and the permittee must apply for a new permit to construct which can take up to 90 days to process a complete application. (FAC Rule 17-4.10)

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Issued this day of feltward

STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

Dale Twachtmann, Secretary

O

# State of Florida DEPARTMENT OF ENVIRONMENTAL REGULATION





	For Routing To Other Than The Addressee	
To	Location	
То	Location	_
To:	Location:	-
Erom:	Onto:	

SUBJECT:

Approval of Harris Semiconductor

State Air Construction Permit

No. AC 05-138795

Dale Twachtmann

Howard L. Rhodes

FEB 2

DATE:

FROM:

January 28, 1988

BAQM

Attached for your approval and signature is a permit prepared by the Central Air Permitting staff for the above mentioned company to install/construct an Industrial Grade Water System to provide water for the Deionized Water Plants in Buildings 52 and 59. A vacuum degasifier will be used to remove hydrogen sulfide and carbon dioxide from the raw well water and a flare will be used to oxidize the gases. The project will occur at Harris Semiconductor's existing facility in Melbourne, Brevard County, Florida. No comments were received during the public notice period.

Day 90 after which the permit will be issued by default is March 5, 1988.

I recommend your approval and signature.

HLR/agm/bm

attachment



## Check Sheet

Company Name: HARRIS SEMICONDUC	TOR
Permit Number: 12 05-138795	
PSD Number:	
Permit Engineer:	
Application:  Initial Application  Incompleteness Letters  Responses  Waiver of Department Action  Department Response  Other	Cross References:
Intent:	
Intent to Issue	
Notice of Intent to Issue  Technical Evaluation	
BACT Determination	
Unsigned Permit	
Correspondence with:	
EPA	
Park Services	
Other	
Proof of Publication	
Petitions - (Related to extensions, hearings, etc.)	
Waiver of Department Action	
Other	t .
Final Determination:  Final Determination	
Signed Permit	
BACT Determination	
Other	
Post Permit Correspondence:	
Extensions/Amendments/Modifications Other	
- Other	



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RECEIVED

JUN 3 1992

Resources Management

May 29, 1992

R-9250 0090-92

Mr. John Turner
Engineer
State of Florida
Department of Environmental Regulation
3319 Maguire Blvd., Suite 232
Orlando, FL. 32803

Dear Mr. Turner:

Enclosed is the 1991 calendar year annual operating report for the vacuum degasified system, Florida operating permit number A005-688383. This operating report was inadvertently omitted from the March 26, 1992 submittal.

If you have any questions regarding this matter please contact me at (407)729-4076.

Sincerely,

John W. Widell

Safety Environmental Engineer

Environmental Services

Harris Semiconductor

ca: D. Bock

R. Sands

B. Mitchell, DER Tal.

C. Collins, DER Orl.

C. Fancy, DER Tal.

# STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

Twin Towers Office Building 2500 Blair Stone Road Tallahasses, Florida 22307



AMARD BOB DAMPOD EXMINDED A AIRCTOIV RATBROBE

### AUMUAL OPERATION REPORT FORM FOR AIR CHISSIONS SOURCES

1.	Saurce Name:	HARRIS SEMICONDUCTOR
	Permit Numbers	
		Palm Bay Road
		Palm Bay, Florida 32902-0883
4.	Description of Sources and Flare System	Industrial Grade Water System with Vacuum Degasif
ACT	UAL OPERATING HOURS:	24 hes/day 7 deys/wk 52 wks/ye
<u>.</u>	Raw Material	Input Process Weight
_		
<del>-</del>		

DER Form 17-1.202(6) Effective Navember 30, 1982

	106 cubic feet Natural Gas	10 <sup>3</sup> Kerasene
	10 <sup>3</sup> gallons 0il, %S	
	10 <sup>3</sup> gallone Propene	tone Carboneceous
	106 Black Liquor Solide	tans Refuse
	Other (Specify type and units)	A SECURIO DE LA CONTRACTOR DE LA CONTRAC
VI	EMISSION RATE(S) (tons/yr)	
	Perticulates 17.60 Sulfur Dioxide	Total Reduced Sulfur
	Nitrogen Oxide Carbon Monoxide	Fluoride
	Hydrocerbon Other (Specify type and un.	ita)
VII	Hydrocarbon Other (Spacify type and un. METHOD OF CALCULATING EMISSION RATES (a.g., use of fue emission factors drawn from AP 42, etc.)	
AII	METHOD OF CALCULATING EMISSION RATES (a.g., use of fu	
. *	METHOD OF CALCULATING EMISSION RATES (s.g., use of fur emission factors drawn from AP 42, etc.)	
VIII I he	METHOD OF CALCULATING EMISSION RATES (e.g., use of fue emission factors drawn from AP 42, etc.)  EPA Method 15	el and materials belance,
VIII I he	METHOD OF CALCULATING EMISSION RATES (e.g., use of fue emission factors drawn from AP 42, etc.)  EPA Method 15  CERTIFICATION:  reby certify that the information given in this report ledge.  Adding C. Allines	el and materials belance,

DATE

## HARRIS

RECEIVED

JUL 1 5 1991

Division of Air Resources Management

July 9, 1991

Mr. Pius Sanabani Air Permitting Engineer Air Resource Management Florida Department of Environmental Regulation 3319 Maguire Boulevard, Suite 232 Orlando, FL 32803-3767

SUBJECT: Source Test Report, Permit Number A005-188383

Vacuum Degasifier with Flare System - HARRIS SEMICONDUCTOR

Dear Mr. Sanabani:

Enclosed are two copies of the test report for the vacuum degasifier system as required by specific conditions 7-10 of the referenced operating permit. The testing indicates that the system operates within the permissible range and operating standards.

If you should have any questions, please call me at (407) 729-5301.

Sincerely,

Constantine Mantaglistis
Constantine Triantafyllidis

Environmental Engineer

cc: C. Fancy, Tall. DER

B. Mitchell, Tall. DER

A. Zahm, P.E., Orlando DER

G. Kuberski, Orlando DER

K. Smith

## HARRIS

August 29, 1989

Alan Zahm, P.E. Air Permitting Engineer Air Resource Management Florida Department of Environmental Regulation 3319 Maguire Boulevard, Suite 232 Orlando, FL 32803-3767

SUBJECT:

Completeness Review of AO 05-167380

·

Vacuum Degasifier with Flare System - HARRIS SEMICONDUCTOR

Dear Mr. Zahm:

This letter is written on the behalf of Harris Semiconductor ("Semiconductor"), in reply to your letter of July 25, 1989. Your letter requested additional information on three (3) items.

- Item 1. A statement from the visible emission test observer Sidney J. Carter is attached to this response (see attachment I.) The letter verifies that no objectionable odors were detected during the flare monitoring on June 16, 1989.
- Item 2. The completion of construction date for the Industrial Grade Water system that the vacuum degasifier services is April 1, 1989.
- Item 3. Attachment II. contains a letter from Air Consulting & Engineering, Inc. explaining the 0.013 factor mentioned in the test report of June 16, 1989. The original report contained a decimal point error which resulted in an incorrect calculation of the BTU value of the gas being combusted. The revised report is included with this package. Because this mathematical error resulted in a change to section H. of the operating permit application form, a corrected application sheet is also enclosed (see attachment III.) Please note that the source meets all 40 CFR 60.18 emission criteria.

If you should have any questions, please call me at (407) 729-4061.

## PFCEIVED

AUG 31 1989

Sincerely,

**DER-BAOM** 

Marcy Baldisserotte

Nancy Baldisserotto Environmental Engineer Harris Semiconductor, Melbourne

## ATTACHMENT I.



August 4, 1989 187 88 04

Mr. Allen Zahm, P.E.
Florida Department of
Environmental Regulation
3319 Maguire Blvd., Suite 232
Orlando, Florida 32803

Dear Mr. Zahm:

On June 16, 1989, I performed visible emission tests on the Vacuum Degasifier Flare System (FDER Permit Number ACO5138795) at the Harris Corporation facility in Palm Bay, Florida. During that time I detected no objectionable oders.

Please contact me with any questions or if further information is needed.

Sincerely,

AIR CONSULTING AND ENGINEERING, INC.

Sidney J./Carter

SJC:ggp

ATTACHMENT II.



August 22, 1989 187 88 04

Mr. Alan Zahn, P.E.
Air Permitting Engineer
Florida Department of
Environmental Regulation
3319 Maguire Blvd., Suite 232
Orlando, Florida 32803

RE: HARRIS SEMICONDUCTOR FLARE (PERMIT NUMBER A005-167380)

Dear Mr. Zahn:

Enclosed are two copies of a revised test report for the referenced vacuum degasifier system. A decimal point error concerning propane boost fuel usage was made in the previous submittal.

Please note that this source meets all CFR 60.16 emission criteria and that there were no objectionable odors observed during the test period. As test team leader, I attest to these facts.

The 0.013 factor mentioned in your July 25, 1989, "Completeness" inquiry is the dry gas fraction of  $\rm H_2O$  present in the degasifier off-gases prior to propane enrichment. The only combustible components of the flow to the flare are propane fuel and  $\rm H_2S$ . 60.18(F) was modified to allow a dry gas measurement as it was necessary to remove the high volume of moisture from the gas stream before measurement of  $\rm H_2S$ . This was done using a citrate buffer ice bath cooled impinger system in accordance with EPA Method 15.

If you have further questions, please contact me.

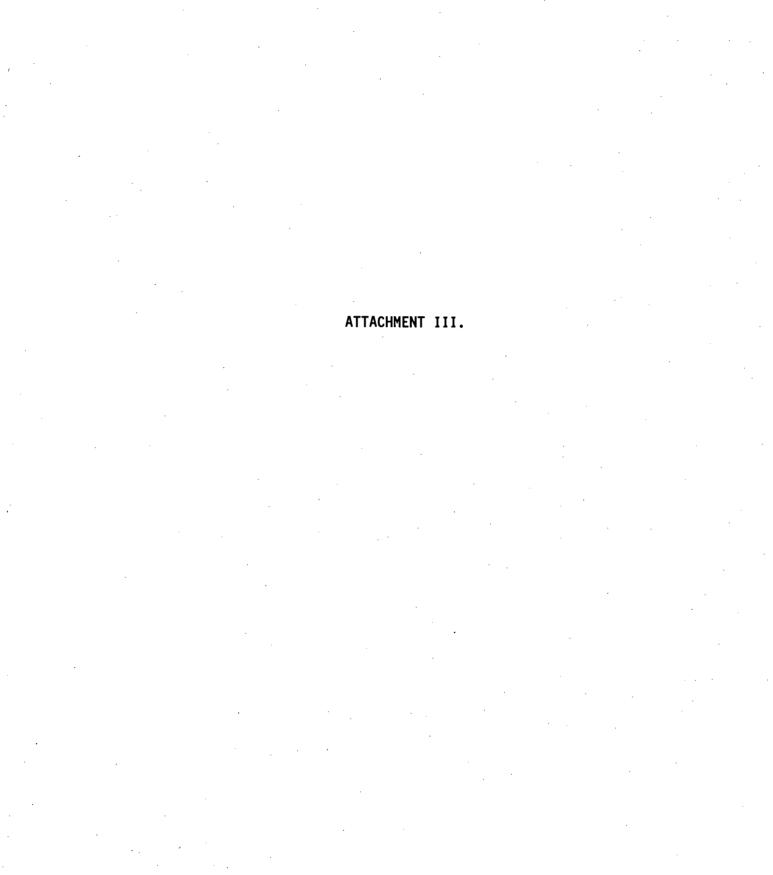
Respectfully,

AIR CONSULTING & ENGINEERING, INC.

Stephen L. Neck, P.E.

SLN:ggp

Enclosures



## DEPARTMENT OF ENVIRONMENTAL REGULATION

WIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32301-8241



BOB GRAHAM GOVERNOR VICTORIA J. TSCHINKEL SECRETARY

APPLICATION TO OPERATE/CONSTRUC	r Air	POLLUTION	SOURCES
---------------------------------	-------	-----------	---------

SOURCE TYPE: Stationary	[X] Ne	ew <sup>l</sup> [ ] Existing <sup>l</sup>
APPLICATION TYPE: [ ] Construction [X]	Operation [	[ ] Modification
COMPANY NAME: Harris Semiconductor	<u> </u>	COUNTY: Brevard
Identify the specific emission point sou Kiln No. 4 with Venturi Scrubber; Peakin		Vacuum Degasifier with
SOURCE LOCATION: Street Palm Ray Road		•
•		
Latitude 28 1	20 "N	North 17-3100900  Longitude 80 36 10 W
APPLICANT NAME AND TITLE: N. A. Baldiss	<u>erotto, Enviro</u>	conmental Engineer. Environmental Services
APPLICANT ADDRESS: P.O. Box 883, Melbou	rne, FL 32901	01
I agree to maintain and operate the facilities in such a manner as to statutes, and all the rules and regulation understand that a permit if great and the such as	n this applicate to the best ne pollution comply with tolations of the ranted by the trant upon sa	cation for a <u>operating</u> t of my knowledge and belief. Further, control source and pollution control the provision of Chapter 403, Florida he department and revisions thereof. I e department, will be non-transferable ale or legal transfer of the permitted
-Attach letter of authorization		R. Hutker Director Facilities Dept
	Name	e and Title (Please Type)
	Date:	Telephone No. (407) 729-4655
B. PROFESSIONAL ENGINEER REGISTERED IN	FLORIDA (where	re required by Chapter 471, F.S.)
been designed/examined by me and f principles applicable to the treatme	ound to be in ent and dispos	of this pollution control project have in conformity with modern engineering sal of pollutants characterized in the nce. in my professional judgment, that

DER Form 17-1.202(1) Effective October 31, 1982

See Florida Administrative Code Rule 17-2.100(57) and (104)

rules and regulations of the furnish, if authorized by the	ties, when properly maintained and operated, will discharge the all applicable statutes of the State of Florida and the department. It is also agreed that the undersigned will element, the applicant a set of instructions for the proper the pollution control facilities and, if applicable,
	Signed Rouineuce R Huther
	Name (Please Type)
·	Harris Semiconductor
	Company Name (Please Type)
	P.O. Box 883, Melbourne, FL 32901
	Mailing Address (Please Type)
ida Registration No. 35972	Date: Telephone No. 407-729-4655
1	II: GENERAL PROJECT INFORMATION
necessary.	ult in full compliance. Attach additional sheet if
•	[ndustrial Grade Water System to provide water for the deioniz
water plants in bldgs. 52 a	and 59. The system includes a vacuum degasifier to remove H <sub>2</sub> S
and COo from raw well water	r. The removed gases are flared to oxidize the products and
control H <sub>2</sub> S emissions. (see	
	in this application (Construction Permit Application Only)
Start of Construction	Completion of Construction
for individual components/un. Information on actual costs opermit.)	ystem(s): (Note: Show breakdown of estimated costs only its of the project serving pollution control purposes. shall be furnished with the application for operation
(see attachment B.)	
•	
Indicate any previous DER per point, including permit issue	rmits, orders and notices associated with the emission ance and expiration dates.
point, including permit issue	
point, including permit issue	ance and expiration dates.
point, including permit issue	ance and expiration dates.

E. Requested permitted equipment operating time:

hrs/day 24 : days/wk 7

_		
<u>-</u>		
	this is a new source or major modification, answer the following quest es or No)	ions.
1.	Is this source in a non-attainment area for a particular pollutant?	No
	a. If yes, has "offset" been applied?	
	b. If yes, has "Lowest Achievable Emission Rate" been applied?	
	c. If yes, list non-attainment pollutants.	
2.	Odes best available control technology (SACT) apply to this source? If yes, see Section VI.	No
3.	Does the State "Pravention of Significant Deterioristion" (PSD) requirement apply to this source? If yes, see Sections VI and VII.	No
٠.	Do "Standards of Performance for New Stationary Sources" (NSPS) apply to this source?	No
5.	On "National Emission Standarde for Hazardous Air Pollutants" (NESHAP) apply to this source?	No.
	"Reasonably Available Control Technology" (RACT) requirements apply this source?	No
	a. If yes, for what pollutants?	

cation for any enswer of "No" that might be considered questionable.

SECTION III: AIR POLLUTION SOURCES & CONTROL DEVICES (Other them Incinerators)

A. Raw Materials and Chemicals Used in your Process, if applicables

	Contam	inents	Utilization	
Description	Type	% Ht	Rate - lbe/hr	Relate to Flow Diagram
		,		· ·
	,			

а.	Process	Rate.	if	applicable:	(See Section	V. Item 1)

				~ ~ ~ ~	16/6- 45	$\sim$
1 Total Process	Ignut Rate (lbs/br):	5.29 lb/hr of h	1,5;	66.6/	ID/Nr of	CU

2 .	Product	Weight	(lhe/h	-).

C. Airborne Contaminants Emitted: (Information in this table must be submitted for each emission point, use additional sheets as necessary)

Name of	Emis	ssion <sup>1</sup>	Allowed 2 Emission Rate per	Allowable <sup>3</sup> Emission	I	itial <sup>4</sup>	Relate to Flow
Contaminant	ent Maximum Actual	inant Maximum Actual Rule lba/hr		lbs/yr	T/yr	Diagram	
S <b>O</b> 2	9.96	5.39			87249.6	43.6	
•							
H <sub>2</sub> S	0.0794	0.04			46355	23.2	
co,	126.6	333.8			126.6	554	

<sup>1</sup>See Section V, Itam 2.

<sup>&</sup>lt;sup>2</sup>Reference applicable emission standards and units (a.g. Rule 17-2.600(5)(b)2. Table II, (1) = 0.1 pounds per million STU heat input)

Calculated from operating rate and applicable standard.

<sup>45</sup>mission, if source operated without control (See Section V, Item 3).

Name and Type (Model & Serial No.)	Contaminant	Efficiency	Range of Particles Size Collected (in microns) (If moplicable)	Basis for Efficiency (Section V Item 5)
Stack (Donnted ni are	H <sub>2</sub> S	98.5%	N/A	See attachmt
w/model EEF-U-2 flare tip and manual				
weatherproof pilot				

	Consumot	tion•	T
Type (3e Specific)	avg/hr	max./hr	Meximum Heat Input (MM8TU/hr)
propane (pilot)	22.0 SCFH		54193.4 BTU/hr
propane (enrichment)	152 6 SCFH		375683.4 BTU/hr
Units: Natural GasMMCF/h	Ar: Fuel Gilsgalle	ns/hr: Cosl. wood	. refuse. otherlbs/hr.

STU/ge

uel Analysis:			
ercent Sulfur: 0	·	Percent Ash: 0	
ensity: 1.8324 g/L at STP	lbs/gsl	Typical Percent Nitrogen:	0
eat Capacity: 21591	BTU/16		
ther Fuel Contaminants (which may	cause air p	ollution):	
. If applicable, indicate the pe	scent of fue	l used for space heating.	N/A
nnual Average		ximum mumix	
. Indicate liquid or solid waste	a generated	and method of disposal.	
		•	
N/A			,

DER Form 17-1.202(1) Effective November 30, 1982

							168
Water Vapo	r Content:	40		*	Velocity: _	9.2	F
		SECI	TION IV:	INCINERA	LTOR INFORMA	TION	
Type of Weste	Type Q (Plastics	Type I (Rubbish)	Type II (Refuse)	Type I (Garbag	(II Type IV (Patholo ical)	g- (Liq.& G	Type VI (Solid By-prod.
Actual 1b/hr Inciner- ated							
Uncon-							
·	n of Waste						
(lbs/hr)  Description  Total Weight	n of Waste ht Inciner s Number o	ted (lbs/h	Operation		·		s/hr)wks/yr
(lbs/hr)  Description  Total Weig  Approximates  Sanufactur	n of Waste ht Inciner s Number o	sted (lbs/h	Operation	per day	da		
(lbs/hr)  Description  Total Weig  Approximates  Sanufactur	n of Waste ht Inciner s Number o	sted (lbs/h	Operation	per day	da		
(lbs/hr)  Pescription  Primary C	n of Waste ht Inciner s Number o er ructed	ted (lbs/h f Hours of Volume	Operation	Made	da d	y/wk	Temperature
(lbs/hr)  Description  Total Weig  Approximate  Anufactur  Bata Const  Primary C	n of Waste ht Inciner s Number o er ructed hamber	Volume	Operation  Heat R  (BTU	Made	Type	el Blu/ne	Temperature (*F)
(lbs/hr)  Description  Total Weign  Approximate  Approximate  Anufactur  Data Const   Primary C  Secondary  tack Heign	n of Waste ht Inciner s Number o er ructed hamber Chamber	Volume (ft)	Heat R (BTU	Made Made Made Made Made Made Made Made	Type	y/wk	Temperature (*F)

	•				
	·				
Brief de	escription of ope	rating characte	eristics of con	trol devices:	
				·	***
					· · · · · · · · · · · · · · · · · · ·
Ultimate ash, etc		effluent other	r than that emit	ited from the stac	k (scrubber weter
			<del> </del>		
-		-			
		·			
NOTE: I	tems 2, 3, 4, 6,	7, 8, and 10 i	.n Section V mus	it be included whe	re applicable.
		SECTION V:	SUPPLEMENTAL RE	QUIREMENTS	
Please pr	ravide the falla	wing supplement	s where require	ed for this applic	ation.
				w derivation [Rul	
				.ssion estimate (s	

- 2. To a construction application, attach basis of emission estimate (e.g., design calcultions, design drawings, pertinent manufacturer's test data, etc.) and attach propositions (e.g., FR Part 60 Methods 1, 2, 3, 4, 5) to show proof of compliance with a plicable standards. To an operation application, attach test results or methods us to show proof of compliance. Information provided when applying for an operation permit from a construction permit shall be indicative of the time at which the test whade.
- 3. Attach basis of potential discharge (e.g., emission factor, that is, AP42 test).
- 4. With construction permit application, include design details for all air pollution control systems (e.g., for baghouse include cloth to air ratio; for scrubber include cross-section sketch, design pressure drop, etc.)
- 5. With construction permit application, attach derivation of control device(s) efficiency. Include test or design data. Items 2, 3 and 5 should be consistent: actual emissions a potential (1-efficiency).
- 6. An 8 1/2" x 11" flow diagram which will, without revealing trade secrets, identify to individual operations and/or processes. Indicate where raw materials enter, where so idend liquid waste exit, where gaseous emissions and/or airborne particles are evolved and where finished products are obtained.
- 7. An 3 1/2" x 11" plot plan showing the location of the establishment, and points of a borne emissions, in relation to the surrounding area, residences and other permans structures and roadways (Example: Copy of relevant portion of USGS topographic map).
- 3. An 3  $1/2^{\rm m}$  x  $11^{\rm m}$  plot plan of facility showing the location of manufacturing process and outlets for airborne emissions. Relate all flows to the flow diagram.

ER Form 17-1.202(1) Effective November 30, 1982

9.	The appropriate application fee in accormade payable to the Department of Environ	dance with Rule 17-4.05. The check should mental Regulation.
10.	With an application for operation permit struction indicating that the source w permit.	e, attach a Certificate of Completion of Co as constructed as shown in the constructi
	SECTION VI: BEST AVAIL	ABLE CONTROL TECHNOLOGY
A.	Are standards of performance for new sta applicable to the source?	tionary sources pursuant to 40 C.F.R. Part
	[ ] Yes [ ] No .	
	Conteminant	Rate or Concentration
		<u></u>
		·
8.	Has EPA declared the best available cont yes, attach copy) [ ] Yes [ ] No	rol technology for this class of sources (
	Contaminant	Rate or Concentration
	<del></del>	
С.	What emission levels do you propose as be	st available control technology?
	Conteminant	Rate or Concentration
	· ·	
		· · · · · · · · · · · · · · · · · · ·
٥.	Describe the existing control and treatme	int technology (if any).
	1. Control Device/System:	2. Operating Principles:

4. Capital Costs:

3. Efficiency:\*

DER Form 17-1.202(1)

Explain method of determining

Effective November 30, 1982

5. Useful Life:  7. Energy:  8. Maintenance Cost:  9. Emissions:  Contaminent  Rate or Concentrat	ion
9. Emissions:	ion
	ion
Contaminant Rate or Concentrat	ion
	4
10. Stack Parameters	
a. Height: ft. b. Diameter:	re
c. Flow Rate: ACFM d. Temperatura:	<b>a</b> F
e. Velocity: FPS	
E. Describe the control and treatment technology available (As many types use additional pages if necessary).	as applicabl
1.	•
a. Control Device: b. Operating Principles:	
c. Efficiency: 1 d. Capital Cost:	
e. Useful Life: f. Operating Cost:	
g. Energy <sup>2</sup> h. Maintenance Coat:	
i. Availability of construction materials and process chemicals:	
j. Applicability to manufacturing processes:	
k. Ability to construct with control device, install in available spec within proposed levels:	ce, and opera
2.	
a. Control Oevice: b. Operating Principles:	
c. Efficiency: 1 d. Capital Cost:	
e. Useful Life: f. Operating Cost:	
g. Energy: <sup>2</sup> h. Maintenance Cost:	
i. Availability of construction materials and process chemicals:	

DER Form 17-1.202(1) Effective November 30, 1982

 $^{1}\mbox{Explain}$  method of determining efficiency.  $^{2}\mbox{Energy}$  to be reported in units of electrical power - XWH design rate.

within proposed levels: 3. Control Device: Operating Principles: Efficiency: 1 Capital Cost: Useful Life: Operating Coat: Energy: 2 Maintenance Coat: Availability of construction materials and process chemicals: Applicability to manufacturing processes: Ability to construct with control device, install in available space, and operate within proposed levels: 4. Control Device: Operating Principles: Efficiency: 1 Capital Costs: c. Useful Life: Operating Cost: Energy: 2 α. Maintenance Cost: Availability of construction materials and process chemicals: Applicability to manufacturing processes: Ability to construct with control device, install in available space, and operate within proposed levels: Describe the control technology selected: 1. Control Device: 2. Efficiency: 1 3. Capital Cost: Useful Life: Operating Coat: Maintenance Cost: Manufacturer: Other locations where employed on similar processes:

(4) State:

Ability to construct with control device, install in available space, and operate

Applicability to manufacturing processes:

a. (1) Company:(2) Mailing Address:

Explain method of determining efficiency.

Energy to be reported in units of electrical power - KWH design rate.

(3) City:

DER Form 17-1.202(1)

					•
(5) Environmental Manager:				:	
(6) Telephone No.:		·			
(7) Emissions: 1					
Contaminant			Rate or	Concentrat	lion
<del></del>					
:					
(8) Process Rate: 1					
b. (1) Company:		٠	•	•	٠
(2) Meiling Address:					
(3) City:		(4) State:			
(5) Environmental Manager:		•		•	
(6) Telephone No.:				,	
(7) Emissions: 1					
Contaminant			Rate or	Concentrat	tion
			_		
		<u>·</u>			
(3) Process Rate: 1					
10. Reason for selection a	nd description	of systems:			•
Applicant must provide this is available, applicant must stat			Should	d this inf	ormation not
SECTION VII	- PREVENTION OF	F SIGNIFICAN	T DETERI	ORATION	
A. Company Monitored Data					
1na. sites	TSP _	()	_ so <sup>2</sup> * _		Wind spd/diz
Period of Monitoring	month d	year t	nontn	day year	•
Other data recorded			<u> </u>		
Attach all data or statisti	cal summaries	to this appl	ication.		
Specify bubbler (8) or continu	ous (C).	18 - 422 -			
OER Form 17-1.202(1)			. •	•	

Page 11 of 12

Effective November 30, 1982

	Z. Instrumentation, rigid and Laboratory
	a. Was instrumentation EPA referenced or its equivalent? [ ] Yes [ ] No
	b. Was instrumentation calibrated in accordance with Department procedures?
	[ ] Yes [ ] No [ ] Unknown
s.	Meteorological Data Used for Air Quality Modeling
	1. Year(s) of data from / / to // month day year month day year
	2. Surface data obtained from (location)
	3. Upper air (mixing height) data obtained from (location)
	4. Stability wind rose (STAR) data obtained from (location)
c.	Computer Models Used
	1 Modified? If yes, attach description.
	Z Modified? If yes, attach description.
	3 Modified? If yes, attach description.
	4 Modified? If yes, attach description.
	Attach copies of all final model runs showing input data, receptor locations, and principle output tables.
σ.	Applicants Maximum Allowable Emission Data
	Pollutant Emission Rate
	TSP gramma/sec
	SO <sup>2</sup> grame/sec
ε.	Emission Data Used in Modeling
	Attach list of emission sources. Emission data required is source name, description of point source (on NEDS point number), UTM coordinates, stack data, allowable emissions, and normal operating time.
۶.	Attach all other information supportive to the PSD review.

G. Discuss the social and economic impact of the selected technology versus other applica-

H. Attach scientific, engineering, and technical material, reports, publications, journals, and other competent relevant information describing the theory and application of

Include

ble technologies (i.e., jobs, payroll, production, taxes, energy, 'etc.). assessment of the environmental impact of the sources.

the requested best available control technology.

### SOURCE TEST REPORT

BUILDING 52 AND 59 WATER TREATMENT PLANT VACUUM DEGASIFIER FLARE SYSTEM EPA 15 H<sub>2</sub>S TEST EPA 22 VISIBLE EMISSIONS

FDER PERMIT NUMBER: AC05-138795

JUNE 16, 1989

Prepared for:

HARRIS CORPORATION
SEMICONDUCTOR DIVISION
PALM BAY ROAD-BUILDING 61
PALM BAY, FLORIDA 32905

Prepared by:

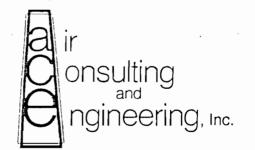
AIR CONSULTING AND ENGINEERING, INC. 2106 N.W. 67th PLACE, SUITE 4 GAINESVILLE, FLORIDA 32606 (904) 335-1889

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	APPENDIX	CEPA METHOD	15 H <sub>2</sub> S					
	APPENDIX	DTEST CALCU	LATIONS	, .				
	APPENDIX	EVOLUMETRIC VISIBLE	FLOW AND EMISSION DAT	ТА				
	APPENDIX	FFLARE SPEC	IFICATIONS					
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1	FLARE SYSTEM SCHEMATIC	. 5



# REPORT CERTIFICATION

To the best of my knowledge, all applicable field and analytical procedures comply with Florida Department of Environmental Regulation requirements and all test data and plant operating data are true and correct.

Stephen L. Neck, P.E.

State of Florida Registration No. 20020

August 21, 1989

STATE OF CONTINUES OF CONTINUES

SEAL

# 1.0 INTRODUCTION

Air Consulting and Engineering, Inc. (ACE), performed testing as required in the "Specific Conditions" of Florida Department of Environmental Regulation (FDER) permit ACO5-138795 on the Vacuum Degasifier Flare System at the Harris Corporation—Semiconductor Division facility in Palm Bay, Florida. Testing was performed on June 16, 1989.

Mr. Harvey Gray of Technical Services, Inc., performed H<sub>2</sub>S United States Environmental Protection Agency (EPA) Method 15 testing.

The flare system by John Zink Company included a calibrated thin plate orifice for purposes of monitoring degasifier outlet flows.

# 2.0 SUMMARY AND DISCUSSION OF RESULTS

Results of the testing in terms of <u>Code of Federal Regulations Title 40</u>

<u>Part 60.18</u> (see Appendix B) and permit specific conditions (see Appendix A)

are provided in Table 1.

As shown in Figure 1, flow volume determinations and H<sub>2</sub>S testing were conducted in the two inch outlet line coming from the degasifier system. Propane gas at a rate of 154 cubic feet per hour is added to the gas stream downstream of this point and prior to flaring. Calculated flare velocities include both flow streams and assume that none of the observed water content condenses prior to flaring.

Three test runs were conducted during water production periods of 600 gallons per minute (maximum rate). The average observed orifice pressure drop (P) for each run was reported to the John Zink Company who then reported the calibrated flow rates for those P values (see Appendix C).

SO<sub>2</sub> emission rates assume total H<sub>2</sub>S oxidation.

EPA Method 15 data is provided in Appendix C. An ice trap citrate buffer system was used to remove the considerable moisture content and provide a dry gas analysis.

Pertinent calculations are provided in Appendix D.

Table 1 Emission Summary
Vacuum Degreaser Flare System
Harris Corporation-Semiconduction Division
June 16. 1989

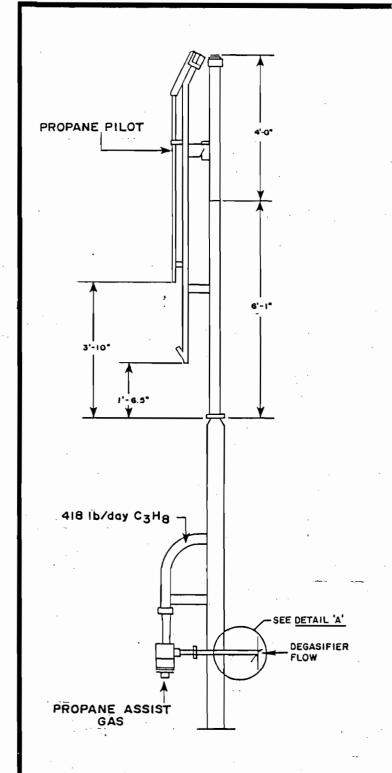
Inlet Average Composition	on (Wet Basis)
H <sub>2</sub> S %	1.3
H <sub>2</sub> O %	40.0
CO <sub>2</sub> %	24.6
02 %	3.0
N <sub>2</sub> % (by difference)	31.1
Total Degasifier Flow	578 CFH
H <sub>2</sub> S Flow	7.5 CFH
Propane Assist Flow	154 CFH
Propane Pilot Flow	22 CFH
Total Heat Value	520.2 BTU/cubic foot = 19.39 MJ/SCM(1)
Flare Velocity	$9.2 \text{ FPS} = 2.8 \text{ M/SEC}^{(2)}$
SO <sub>2</sub> Emission	1.23 lb/Hr
Visible Emission	0% opacity

<sup>(1)</sup> See Appendix D

<sup>(2)</sup> Maximum Permitted Velocity =  $10^{(Ht + 28.8)/31.7} = V_{max} = 38.6 \text{ M/SEC}$ where Ht = MJ/SCM

# 3.0 PROCESS OPERATION AND SAMPLING POINT LOCATION

The new water plant used the degasifier system to remove CO<sub>2</sub> and H<sub>2</sub>S gases prior to cleansing by reverse osmosis. The vacuum systems delivers these gases and considerable water vapor to a flame system which adds additional propane fuel prior to flaring. A propane pilot is always on.



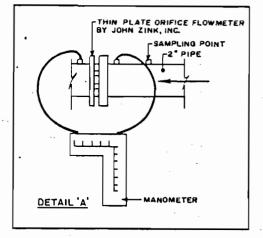


FIGURE I.
FLARE SYSTEM SCHEMATIC
HARRIS CORPORATION-SEMICONDUCTOR
DIVISION
PALM BAY, FLORIDA

AIR CONSULTING and ENGINEERING

# APPENDIX A FDER PERMIT NUMBER AC05-138795

# STATE OF FLORIDA

# DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION
NOTICE OF PERMIT

Mr. James R. Kolanek, Manager Environmental Services Harris Semiconductor Post Office Box 883 Melbourne, Florida 32901

February 3, 1988

Enclosed is permit No. AC 05-138795, for Harris Semiconductor to install/construct an Industrial Grade Water System to provide water for the Deionized Water Plants in Buildings 52 and 59. A vacuum degasifier will be used to remove hydrogen sulfide and carbon dioxide from the raw well water and a flare will be used to oxidize the gasses. The project will occur at Harris Semiconductor's existing facility in Melbourne, Brevard County, Florida. This permit is issued pursuant to Section 403, Florida Statutes.

Any Party to this permit has the right to seek judicial review of the permit pursuant to Section 120.68, Florida Statutes, by the filing of a Notice of Appeal pursuant to Rule 9.110, Florida Rules of Appellate Procedure, with the Clerk of the Department in the Office of General Counsel, 2600 Blair Stone Road, Tallahassee, Florida 32399-2400; and by filing a copy of the Notice of Appeal accompanied by the applicable filing fees with the appropriate District Court of Appeal. The Notice of Appeal must be filed within 30 days from the date this permit is filed with the Clerk of the Department.

Executed in Tallahassee, Florida.

RECEIVED

FEB 10 1988

J. R. KOLANEK

STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

C. H. Fancy P.E.

Deputy Chief

Bureau of Air Quality Management

Copy furnished to:

T. Sawicki, CF Dist.

C. Bach, P.E.

# Final Determination

Harris Semiconductor Brevard County Melbourne, Florida

Permit No. AC 05-138795

Florida Department of Environmental Regulation Bureau of Air Quality Management Central Air Permitting

# Final Determination

The construction permit application has been reviewed by the Department. Public Notice of the Department's Intent to Issue was published in the Florida Today Newspaper, Cocoa, Florida, on January 6, 1988. The Technical Evaluation and Preliminary Determination were available for public inspection at the DER's Central Florida District office in Orlando and the DER's Bureau of Air Quality Management office in Tallahassee.

No comments were received during the Public Notice period. Therefore, the final action of the Department will be to issue the construction permit as drafted.

# APE PUBLICATIONS, . C. 6 Jon. 1788

The Times

Published Weekly on Wednesday

THE TRIBUNE

STAR-ADVOCATE

Published Weekls on Wednesday

DER

38P1 80 NAL

Before the undersigned authority personally appeared.

My Commission Expires July 20, 1990



Published Daily

Linda L. Spicer

BAOM STATE OF FLORIDA COUNTY OF BREVARD

oath says that he/she is	Legal Adve	ertising Clerk
of the FLOR	RIDA TODAY	, a newspaper published in Brevard County,
	d copy of advertising being a Notice	· · · ·
	in the matte	er of
Permit to Harr	is Semiconductor	
	in the	Court
was published in the	FLORIDA TO	DAY NEWSPAPER
in the issues of	January 6, 1	988
,		
said Brevard County, F attached copy of adve	ortisement; and affiant furthe	the post office in
securing this advertisem	ent for publication in said nev	wspaper.
d: Brun Mitel	ull ichi-cfoiot}1.12.	88@
Linda	L Spicer	
Sworn and subscribed t	o before me this	
Mand	January A.D., 19_	88
Notary Public, St	ote of Florida at Large	

State of Florida
Department of Environmental
Regulation
Notics of Intent
The Department of Environmental Regulation hereby gives
indice of its intent to issue a permit to Harris Simiconductor to
install/construct an industrial
Grade Water System to provide
water for line Decinized Water
Plants in Buildings 32 and 59. The
system will include a vacuum dequasifier to remove nydrogen sulfide and carbon dioxide from the
raw wat water. The removed
gases will be designed and built
by the John Zink Cambany. Model
et EEF-U-2 Flare Tip with a
manual/weather-prode pilot ignilion panel. The pilot and enrichment fuel will be propane. The
construction/installation will take
place at the permittee's existing
facility located in Paim Bay. Brevard County. Florida.
Persons whose substantial inharsets are affected by the Department's proposed permitting
decision may petifion for an administrative determination
(hearing) in accordance with
Section 120.57. Florida: Statutes.
The petition must conform to the
requirements of Chapters 17-103
and 18-5. Florida Administrative
Code, and must be filad (recode, and must be filad (recode, and must be filad (recode) in the Department's Office of General Counsel, 240
Blatr Stone Road. Twin Towers
Office Building, Talianssee,
Florida 12379-2400, within foorteen (14)-days of outbication of
this notice. Failure to file a petition within this time period constitutes a walver of any right such
person has to request an administrative determination (hearing)
under Section 120.57, Florida side
Startes.

- if a petition is filed, the administrative
determination (hearing)
under Section 120.57, Florida
21379-2400, Failure to petition to
read the petition for intervenread to determination (hearing)
under Section 120.57, Florida
21379-2400, Failure to petition in
rem constitutes a walver of any
right such person has to request a
hearing under Section 120.57,
Forida Statutes.

The application to available for
public inspection during normal
business hours, £00 ann, to

Management — 00 Blair Stone Road bliahessee, Florida 32379-2400

Dept. of Environmental
Regulation
Central Florida Obstrict
319 Mageure Brvd., Suite 232
Oriente, Florida 1230-2767
Any person may send written
comments on the processed actions
to Air, Bill Themas at the Decentment's Tallahassee eddress. All
Comment's mailed within 14 days
of the posticiption of this notice
will be considered in the Depenment's finel determination.
TOBAUSE—IT—1/4, 1998.
Wednesday

# STATE OF FLORIDA

# DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

PERMITTEE:
Harris Semiconductor
P. O. Box 883
Melbourne, Florida 32901

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

County: Brevard

Latitude/Longitude: 28° 01' 20"N/

80° 36' 10" W

Project: Industrial Grade Water

System with Vacuum Degasifier and Flare

System

This permit is issued under the provisions of Chapter  $\underline{403}$ , Florida Statutes, and Florida Administrative Code Rule(s)  $\underline{17-2}$  and  $\underline{17-4}$ . The above named permittee is hereby authorized to perform the work or operate the facility shown on the application and approved drawing(s), plans, and other documents attached hereto or on file with the Department and made a part hereof and specifically described as follows:

For the construction/installation of a 600 gallon per minute Industrial Grade water system with a vacuum degasifier and flare system. The vacuum degasifier will remove hydrogen sulfide and carbon dioxide from raw well water and the removed gases will be transported to and oxidized/combusted by the flare system. The nonassisted type flare will be designed and built by the John Zink Company, which includes a self-supported flare stack, a Model EEF-U-2 flare tip (John Zink Co.), and a manual weatherproof pilot ignition panel. The construction/installation will occur at the permittee's existing facility located on Palm Bay Road, City of Palm Bay, Florida. The UTM coordinates are Zone 17, 538.7 km East and 3100.9 km North.

The source shall be in accordance with the permit application, plans, documents, amendments and drawings, except as otherwise noted in the Specific Conditions.

# Attachments to be Incorporated:

- 1. Application to Construct Air Pollution Source, DER Form 17-1.202(1), and Mr. James R. Kolanek's cover letter dated August 28, 1987, and received August 31, 1987.
- 2. Mr. C. H. Fancy's letter dated September 25, 1987.
- 3. Mr. J. R. Kolanek's letter with enclosures dated October 16, 1987, and received October 23, 1987.
- 4. Copy of 40 CFR 60.18, as revised July 1, 1986.
- 5. Technical Evaluation and Preliminary Determination dated December 4, 1987.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

# GENERAL CONDITIONS:

- 1. The terms, conditions, requirements, limitations, and restrictions set forth herein are "Permit Conditions" and as such are binding upon the permittee and enforceable pursuant to the authority of Sections 403.161, 403.727, or 403.859 through 403.861, Florida Statutes. The permittee is hereby placed on notice that the Department will review this permit periodically and may initiate enforcement action for any violation of the "Permit Conditions" by the permittee, its agents, employees, servants or representatives.
- 2. This permit is valid only for the specific processes and operations applied for and indicated in the approved drawings or exhibits. Any unauthorized deviation from the approved drawings, exhibits, specifications, or conditions of this permit may constitute grounds for revocation and enforcement action by the Department.
- 3. As provided in Subsections 403.087(6) and 403.722(5), Florida Statutes, the issuance of this permit does not convey any vested rights or any exclusive privileges. Nor does it authorize any injury to public or private property or any invasion of personal rights, nor any infringement of federal, state or local laws or regulations. This permit does not constitute a waiver of or approval of any other Department permit that may be required for other aspects of the total project which are not addressed in the permit.
- 4. This permit conveys no title to land or water, does not constitute state recognition or acknowledgement of title, and does not constitute authority for the use of submerged lands unless herein provided and the necessary title or leasehold interests have been obtained from the state. Only the Trustees of the Internal Improvement Trust Fund may express state opinion as to title.
- 5. This permit does not relieve the permittee from liability for harm or injury to human health or welfare, animal, plant or aquatic life or property and penalties therefore caused by the construction or operation of this permitted source, nor does it allow the permittee to cause pollution in contravention of Florida Statutes and Department rules, unless specifically authorized by an order from the Department.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

# GENERAL CONDITIONS:

- 6. The permittee shall at all times properly operate and maintain the facility and systems of treatment and control (and related appurtenances) that are installed or used by the permittee to achieve compliance with the conditions of this permit, as required by Department rules. This provision includes the operation of backup or auxiliary facilities or similar systems when necessary to achieve compliance with the conditions of the permit and when required by Department rules.
- 7. The permittee, by accepting this permit, specifically agrees to allow authorized Department personnel, upon presentation of credentials or other documents as may be required by law, access to the premises, at reasonable times, where the permitted activity is located or conducted for the purpose of:
  - a. Having access to and copying any records that must be kept under the conditions of the permit;
  - Inspecting the facility, equipment, practices, or operations regulated or required under this permit; and
  - c. Sampling or monitoring any substances or parameters at any location reasonably necessary to assure compliance with this permit or Department rules.

Reasonable time may depend on the nature of the concern being investigated.

- 8. If, for any reason, the permittee does not comply with or will be unable to comply with any condition or limitation specified in this permit, the permittee shall immediately notify and provide the Department with the following information:
  - a. a description of and cause of non-compliance; and
  - b. the period of noncompliance, including exact dates and times; or, if not corrected, the anticipated time the noncompliance is expected to continue, and steps being taken to reduce, eliminate, and prevent recurrence of the noncompliance.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

# GENERAL CONDITIONS:

The permittee shall be responsible for any and all damages which may result and may be subject to enforcement action by the Department for penalties or revocation of this permit.

- 9. In accepting this permit, the permittee understands and agrees that all records, notes, monitoring data and other information relating to the construction or operation of this permitted source, which are submitted to the Department, may be used by the department as evidence in any enforcement case arising under the Florida Statutes or Department rules, except where such use is proscribed by Sections 403.73 and 403.111, Florida Statutes.
- 10. The permittee agrees to comply with changes in Department rules and Florida Statutes after a reasonable time for compliance, provided however, the permittee does not waive any other rights granted by Florida Statutes or Department rules.
- 11. This permit is transferable only upon department approval in accordance with Florida Administrative Code Rules 17-4.12 and 17-30.30, as applicable. The permittee shall be liable for any noncompliance of the permitted activity until the transfer is approved by the Department.
- 12. This permit is required to be kept at the work site of the permitted activity during the entire period of construction or operation.
- 13. This permit also constitutes:
  - ( ) Determination of Best Available Control Technology (BACT)
  - ( ) Determination of Prevention of Significant Deterioration (PSD)
  - ( ) Compliance with New Source Performance Standards
- 14. The permittee shall comply with the following monitoring and record keeping requirements:
  - a. Upon request, the permittee shall furnish all records and plans required under Department rules. The retention period for all records will be extended automatically, unless otherwise stipulated by the Department, during the course of any unresolved enforcement action.

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# GENERAL CONDITIONS:

- b. The permittee shall retain at the facility or other location designated by this permit records of all monitoring information (including all calibration and maintenance records and all original strip chart recordings for continuous monitoring instrumentation), copies of all reports required by this permit, and records of all data used to complete the application for this permit. The time period of retention shall be at least three years from the date of the sample, measurement, report or application unless otherwise specified by Department rule.
- c. Records of monitoring information shall include:
  - the date, exact place, and time of sampling or measurements;
  - the person responsible for performing the sampling or measurements;
  - the date(s) analyses were performed;
  - the person responsible for performing the analyses;
  - the analytical techniques or methods used; and
  - the results of such analyses.
- 15. When requested by the Department, the permittee shall within a reasonable time furnish any information required by law which is needed to determine compliance with the permit. If the permittee becomes aware that relevant facts were not submitted or were incorrect in the permit application or in any report to the Department, such facts or information shall be submitted or corrected promptly.

# SPECIFIC CONDITIONS:

- 1. Annual hours of operation are 8760.
- 2. The maximum potential sulfur dioxide  $(SO_2)$  emissions are 7.0 pounds per hour and 30.7 tons per year.
- 3. The maximum potential hydrogen sulfide (H<sub>2</sub>S) emissions are 493 pounds per year, which is based on a flare efficiency of 98.5%.

Page 5 of 7

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

# SPECIFIC CONDITIONS:

- 4. The permittee shall comply with the conditions of 40 CFR 60.18(c) thru (f).
- 5. No visible emissions (5% opacity) shall be allowed, except for a total period of 5 minutes during any consecutive 2 hours, pursuant to 40 CFR 60.18(c). Compliance shall be demonstrated annually using EPA Method 22 pursuant to 40 CFR 60.18(f)(1).
- 6. EPA Method 15 shall be performed annually to determine the maximum concentration of the  $\rm H_2S$  prior to being flared and the result should be in terms of dry standard conditions (14.7 psia and 680 F). A retest shall be required if the concentration of  $\rm H_2S$  is to be increased.
- 7. The exit velocity of the flare shall be determined using the procedure in 40 CFR 60.18(f)(4) and either EPA Method 2, 2A, 2C or 2D, as appropriate.
- 8. EPA Methods shall be as described in 40 CFR 60, Appendix A.
- 9. The Central Florida District shall be notified in writing 15 days in advance of any compliance testing and the test reports shall be submitted within 45 days after the last test run.
- 10. Objectionable odors shall not be allowed off plant property pursuant to FAC Rule 17-2.620(2).
- 11. The construction shall reasonably conform to the plans and schedule submitted in the application. If the permittee is unable to complete construction on schedule, he must notify the Department in writing 60 days prior to the expiration date of the construction permit and submit a new schedule and request for an extension of the construction permit. (FAC Rule 17-4.09)

To obtain a permit to operate, the permittee must demonstrate compliance with the conditions of the construction permit and submit a complete application for an operating permit, including the application fee, along with compliance test results and Certificate of Completion, to the Department's District office 90 days prior to the expiration date of the construction permit. The permittee may continue to operate in compliance with all terms of the construction permit until its expiration date. Operation beyond the construction permit expiration date requires a valid permit to operate. (FAC Rules 17-4.22 and 17-4.23)

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# SPECIFIC CONDITIONS:

If the construction permit expires prior to the permittee requesting an extension or obtaining a permit to operate, then all activities at the project must cease and the permittee must apply for a new permit to construct which can take up to 90 days to process a complete application. (FAC Rule 17-4.10)

Issued this / day of \left\uasq

STATE OF ELORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

Dale Twachtmann, Secretary

APPENDIX B

CODE OF FEDERAL REGULATIONS 60.18

§ 60.18 General control device requirements.

- (a) Introduction. This section contains requirements for control devices used to comply with applicable subparts of Part 60 and Part 61. The requirements are placed here for administrative convenience and only apply to facilities covered by subparts referring to this section.
- (b) Flares. Paragraphs (c) through (f) apply to flares.
- (c)(1) Flares shall be designed for and operated with no visible emissions

as determined by the methods specified in paragraph (f), except for periods not to exceed a total of 5 minutes during any-2 consecutive hours.

(2) Flares shall be operated with a flame present at all times, as determined by the methods specified in paragraph (f).

(3) Flares shall be used only with the net heating value of the gas being combusted being 11.2 MJ/scm (300 Btu/scf) or greater if the flare is steam-assisted or air-assisted; or with the net heating value of the gas being combusted being 7.45 MJ/scm (200 Btu/scf) or greater if the flare is non-assisted. The net heating value of the gas being combusted shall be determined by the methods specified in paragraph (f).

(4)(i) Steam-assisted and nonassisted flares shall be designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), less than 19.3 m/sec (60 ft/sec), except as provided in paragraphs (b)(4) (ii) and (iii).

- (ii) Steam-assisted and nonassisted flares designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), equal to or greater than 18.3 m/sec (60 ft/sec) but less than 122 m/sec (400 ft/sec) are allowed if the net heating value of the gas being combusted is greater than 37.3 MJ/scm (1,000 Btu/scf).
- (iii) Steam-assisted and nonassisted flares designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), less than the velocity,  $V_{max}$ , as determined by the method specified in paragraph (f)(5), and less than 122 m/sec (400 ft/sec) are allowed.
- (5) Air-assisted flares shall be designed and operated with an exist velocity less than the velocity,  $V_{max}$ , as determined by the method specified in paragraph (f)(6).
- (6) Flares used to comply with this section shall be steam-assisted, air-assisted, or nonassisted.
- (d) Owners or operators of flares used to comply with the provisions of this subpart shall monitor these control devices to ensure that they are operated and maintained in conformance with their designs. Applicable subparts

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will provide provisions stating how owners or operators of flares shall monitor these control devices.

(e) Flares used to comply with provisions of this subpart shall be operated at all times when emissions may be vented to them.

(f)(1) Reference Method 22 shall be used to determine the compliance of flares with the visible emission provisions of this subpart. The observation period is 2 hours and shall be used according to Method 22.

(2) The presence of a flare pilot flame shall be monitored using a thermocouple or any other equivalent device to detect the presence of a flame.

(3) The net heating value of the gas being combusted in a flare shall be calculated using the following equation:

$$H_{\mathsf{T}} = \mathsf{K} \sum_{i=1}^{\mathsf{p}} \mathsf{C}_{i}\mathsf{H}_{i}$$

where:

H<sub>T</sub>=Net heating value of the sample, MJ/scm; where the net enthalpy per mole of offgas is based on combustion at 25 °C and 760 mm Hg, but the standard temperature for determining the volume corresponding to one mole is 20 °C;

$$K = Constant, \frac{1}{1.740 \times 10^{-7}} \left(\frac{1}{ppm}\right) \left(\frac{g \text{ mole}}{scm}\right) \left(\frac{MJ}{kcal}\right)$$

where the standard temperature for  $(\frac{g \text{ mole}}{scm})$  is  $20^{\circ}\text{C}$ ;

C<sub>i</sub>=Concentration of sample component I in ppm on a wet basis, as measured for organics by Reference Method 18 and measured for hydrogen and carbon monoxide by ASTM D1946-77 (Incorporated by reference as specified in § 60.17); and H.= Net heat of combustion of sample com-

H<sub>1</sub>=Net heat of combustion of sample component i, kcal/g mole at 25 °C and 760 mm Hg. The heats of combustion may be determined using ASTM D2382-76 (incorporated by reference as specified in § 60.17) if published values are not available or cannot be calculated.

- (4) The actual exist velocity of a flare shall be determined by dividing the volumetric flowrate (in units of standard temperature and pressure), as determined by Reference Methods 2, 2A, 2C, or 2D as appropriate; by the unobstructed (free) cross sectional area of the flare tip.
- (5) The maximum permitted velocity, V<sub>max</sub>, for flares complying with paragraph (c)(4)(iii) shall be determined by the following equation.

$$Log_{10}(V_{max}) = (H_T + 28.8)/31.7$$

 $V_{max}$ = Maximum permitted velocity, M/sec 28.8 = Constant

31.7 = Constant

- $H_{\tau}$ =The net heating value as determined in paragraph (f)(3).
- (6) The maximum permitted velocity.  $V_{max}$ , for air-assisted flares shall be determined by the following equation.

$$V_{max} = 8.706 + 0.7034 (H_T)$$

V<sub>max</sub>=Maximum permitted velocity, m/sec 8.706=Constant

0.7084 ... Constant

 $H_T$ =The net heating value as determined in paragraph (f)(3).

[51 FR 2701, Jan.21, 1986]

APPENDIX C
EPA METHOD 15 H2S

# CALIBRATION DATA

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COMPO	UND	· · · · ·	H <sub>2</sub> S	N	NeSH	DMS	DMDS	AMB TEMP
PERME	ATION RATE in)	105	0					Trailer 75°E
TUBE	NUMBER	303	94					BAR PRES 30, L
TIME	FLOW ml/min	CMPD		DONC opm	1	AREA C	COUNTS 3	AVG.
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1036	621	1425	. /,	21	64957	64523	64645	64707
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# CALIBRATION DATA SUMMARY

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1036	1.21	64707	1.17	3.3						
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## STATIONARY SOURCES

- 6.1.2 Using Equation 14-1, calculate the expected average velocity  $(v_4)$  in the sampling duct, corresponding to each value of  $v_m$  obtained under Section 6.1.1.
- 6.1.3 Calculate the actual average velocity (v<sub>s</sub>) in the sampling duct for each run or sub-run, according to Equation 2-9 of Method 2, and using data obtained from Method 13.
- 6.1.4 Express each value  $v_a$  from Section 6.1.3 as a percentage of the corresponding  $v_a$  value from Section 6.1.2.
- 6.1.4.1 If  $v_*$  is less than or equal to 120 percent of  $v_4$ , the results are acceptable (note that in cases where the above calculations have been performed for each sub-run, the results are acceptable if the average percentage for all sub-runs is less than or equal to 120 percent).
- 6.1.4.2 If  $v_{\bullet}$  is more than 120 percent of  $v_{\bullet}$ , multiply the reported emission rate by the following factor.

$$1 + \frac{(100 \ v_e/v_d) - 120}{200}$$

- 6.2 Average velocity of roof monitor gases. Calculate the average roof monitor velocity using all the velocity or volumetric flow readings from Section 5.1.2.
- 6.3 Roof monitor temperature. Calculate the mean value of the temperatures recorded in Section 5.2.
- 6.4 Concentration of fluorides in roof monitor air.

[6.4 corrected by 52 FR 34639, September 14, 1987]

- 6.4.1 If a single sampling train was used throughout the run, calculate the average fluoride concentration for the roof monitor using Equation 13A-2 of Method 13A.
- 6.4.2 If two or more sampling trains were used (i.e., one per sub-run), calculate the average fluoride concentration for the run, as follows:

$$\sum_{i=1}^{n} C = \frac{(F_i)_i}{(V_{m(aud)})_i}$$
Eq. 14-2

# Where:

- C.=Average fluoride concentration in roof monitor air, mg F/dscm (mgF/dscf).
- F<sub>i</sub>=Total fluoride mass collected during a particular sub-run, mg F (from Equation 13A-1 of Method 13A) or Equation 13B-1 of Method 13B).
- $V_{\text{m(ata)}}\!\!=\!\!\text{Total}$  volume of sample gas passing through the dry gas meter during a par-

ticular sub-run, dscm (dscf) (see Equation 5-1 of Method 5).

n=Total number of sub-runs.

[6.4.2 corrected by 52 FR 34639, September 14, 1987]

6.5 Average volumetric flow from the roof monitor of the potroom(s) (or potroom segment(s)) containing the anemometers is given in Equation 14-3.

$$Q_{ad} = \frac{V_{mt}M_dP_m(293 \text{ '}K)A}{(t_m + 273 \text{ '})(780 \text{ mm Hg})}$$

Eq. 14-3

Where:

Q<sub>ad</sub>::Average volumetric flow from roof monitor at standard conditions on a dry basis, m³/mln.

A = Roof monitor open area, m2.

v<sub>mt</sub>= Average velocity of air in the roof monitor, m/min, from Section 6.2.

P<sub>m</sub>= Pressure in the roof monitor; equal to barometric pressure for this application, mm Hg.

 $t_m = \text{Roof monitor temperature, }^{\circ}\text{C, from Section 6.3.}$ 

M<sub>d</sub>=Mole fraction of dry gas, which is given by:

$$M_{a}=(1-B_{-a})$$

Note:  $B_{\text{ws}}$  is the proportion by volume of water vapor in the gas stream, from Equation 5-3, Method 5.

[6.5 equation corrected by 52 FR 34639, September 14, 1987]

6.6 Conversion Factors.

1  $ft^3 = 0.02832 \text{ m}^3$ 

1 hr=60 min

[6.6 added by 52 FR 34639, September 14, 1987]

# $7.\ Bibliography.$

- 1. Shigehara, R. T., A guideline for Evaluating Compliance Test Results (Isokinetic Sampling Rate Criterion). U.S. Environmental Protection Agency, Emission Measurement Branch. Research Triangle Park, North Carolina, August 1977.
- METHOD 15—DETERMINATION OF HYDROGEN SULFIDE, CARBONYL SULFIDE, AND CARBON DISULFIDE EMISSIONS FROM STATIONARY SOURCES

[Method 15 amended and corrected by 52 FR 34639, September 14, 1987]

introduction. The method described below uses the principle of gas chromatographic separation and flame photometric detection (FPD). Since there are many systems or sets of operating conditions that represent usable methods of determining sulfur emissions, all systems which employ this principle, but differ only in details of equipment and operation, may be used as alternative methods, provided that the criteria set below are met.

# 1. Principle and Applicability

- 1.1 Principle. A gas sample is extracted from the emission source and diluted with clean dry air. An aliquot of the diluted sample is then analyzed for hydrogen sulfide (H<sub>2</sub>S), carbonyl sulfide (COS), and earbon disulfide (CS<sub>3</sub>) by gas chromatographic (GC) separation and flame photometric detection (FPD).
- 1.2 Applicability. This method is applicable for determination of the above sulfur compounds from tail gas control units of sulfur recovery plants.

### 2, Range and sensitivity

- 2.1 Range. Coupled with a gas chromtographic system utilizing a 1-milliliter sample size, the maximum limit of the FPD for each sulfur compound is approximately 10 ppm. It may be necessary to dilute gas samples from sulfur recovery plants hundredfold (99:1) resulting in an upper limit of about 1000 ppm for each compound.
- 2.2 The minimum detectable concentration of the FPD is also dependent on sample size and would be about 0.5 ppm for a 1 ml sample.

### 3. Interferences

- 3.1 Moisture Condensation. Moisture condensation in the sample delivery system, the analytical column, or the FPD burner block can cause losses or interferences. This potential is eliminated by heating the sample line, and by conditioning the sample with dry dilution air to lower its dew point below the operating temperature of the GC/FPD analytical system prior to analysis.
- 3.2 Carbon Monoxide and Carbon Dioxide. CO and CO<sub>2</sub> have substantial desensitizing effects on the flame photometric detector even after 9:1 dilution. (Acceptable systems must demonstrate that they have eliminated this interference by some procedure such as eluding CO and CO, before any of the sulfur compounds to be measured.) Compliance with this requirement can be demonstrated by submitting chromatograms of calibration gases with and without CO<sub>2</sub> in the diluent gas. The CO, level should be approximately 10 percent for the case with CO, present. The two chromatograms should show agreement within the precision limits of section 4.1.
- [3.2 corrected by 52 FR 34639 September 14, 1987]
- 3.3 Elemental Sulfur. The condensation of sulfur vapor in the sampling line can lead to eventual coating and even blockage of the sample line. This problem can be eliminated along with the moisture problem by heating the sample line.

# 4. Precision

4.1 Calibration Precision. A series of three consecutive injections of the same calibration ges, at any dilution, shall produce re-

sults which do not vary by more than  $\pm 13$  percent from the mean of the three injections.

4.2 Calibration Drift. The calibration drift determined from the mean of three injections made at the beginning and end of any 8-hour period shall not exceed ±5 percent.

## 5. Apparatus

5.1 Sampling.

[5.1 added by 52 FR 34639, September 14, 1987]

5.1.1 Probe. The probe must be made of inert material such as stainless steel or glass. It should be designed to incorporate a filter and to allow calibration gas to enter the probe at or near the sample entry point. Any portion of the probe not exposed to the stack gas must be heated to prevent moisture condensation.

5.1.2 Sample Line. The sample line must be made of Teflon, no greater than 1.3 cm (½ in) inside diameter. All parts from the probe to the dilution system must be thermostatically heated to 120° C.

[5.1.2 amended by 52 FR 34639, September 14, 1987]

5.1.3 Sample Pump. The sample pump shall be a leakless Teflon coated diaphragm type or equivalent. If the pump is upstream of the dilution system, the pump head must be heated to 120° C.

5.2 Dilution System. The dilution system must be constructed such that all sample contacts are made of inert material (e.g. stainless steel or Teflon). It must be heated to 120° C and be capable of approximately a 9:1 dilution of the sample.

5.3 Gas Chromatograph. The gas chromatograph must have at least the following

components:

5.3.1 Oven. Capable of maintaining the separation column at the proper operating temperature ±1°C.

5.3.2 Temperature Gauge. To monitor column oven, detector, and exhaust temperature  $\pm 1^{\circ}$  C.

5.3.3 Flow System. Gas metering system to measure sample, fuel, combustion gas, and carrier gas flows.

5.3.4 Flame Photometric Detector.

5.3.4.1 Electrometer. Capable of full scale amplification of linear ranges of 10<sup>-9</sup> to 10<sup>-4</sup> amperes full scale.

5.3.4.2 Power Supply. Capable of delivering up to 750 volts.

5.3.4.3 Recorder. Compatible with the output voitage range of the electrometer.

5.4 Gas Chromatograph Columns. The column system must be demonstrated to be capable of resolving three major reduced sulfur compounds: H<sub>2</sub>S, COS, and CS<sub>1</sub>.

To demonstrate that adequate resolution has been achieved the tester must submit a chromatogram of a calibration gas containing all three reduced sultur compounds in the concentration range of the applicable standard. Adequate resolution will be defined as base line separation of adjacent peaks when the amplifier attenuation is set so that the smaller peak is at least 50 percent of full scale. Base line separation is defined as a return to zero ±5 percent in the interval between peaks. Systems not meeting this criteria may be considered alternate methods subject to the approval of the Administrator.

[5.4 corrected 52 FR 34639, September 14, 1987]

5.5.1 Calibration System. The calibration system must contain the following components.

5.5.2 Flow System. To measure air flow over permeation tubes at  $\pm 2$  percent. Each flowmeter shall be calibrated after a complete test series with a wet test meter. If the flow measuring device differs from the wet test meter by 5 percent, the completed test shall be discarded. Alternatively, the tester may elect to use the flow data that would yield the lowest flow measurement. Calibration with a wet test meter before a test is optional.

5.5.3 Constant Temperature Bath. Device capable of maintaining the permeation tubes at the calibration temperature within  $\pm 0.1^{\circ}$  C.

[5.5.3 corrected by 52 FR 34639 September 14, 1987]

5.5.4 Temperature Gauge. Thermometer or equivalent to monitor bath temperature within  $\pm\,1^{\circ}$  C.

### 6. Reagents

6.1 Fuel. Hydrogen ( $H_{\mbox{\scriptsize 1}}$ ) prepurified grade or better.

6.2 Combustion Gas. Oxygen (O $_{\text{1}}$ ) or air, research purity or better.

6.3 Carrier Gas. Prepurified grade or better.

6.4 Diluent. Air containing less than 0.5 ppm total sulfur compounds and less than 10 ppm each of moisture and total hydrocarbons.

6.5 Calibration Gases. Permeation tubes, one each of H.S. COS, and CS, gravimetrically calibrated and certified at some convenient operating temperature. These tubes consist of hermetically sealed FEP Teflon tubing in which a liquified gaseous substance is enclosed. The enclosed gas permeates through the tubing wail at a constant rate. When the temperature is constant, calibration gases covering a wide range of known concentrations can be generated by varying and accurately measuring the flow rate of diluent gas passing over the tubes. These calibration gases are used to calibrate the GC/FPD system and the dilution system.

# 7. Pretest Procedures

The following procedures are optional but would be helpful in preventing any problem

which might occur later and invalidate the entire test.

7.1 After the complete measurement system has been set up at the site and deemed to be operational, the following procedures should be completed before sampling is initiated.

7.1.1 Leak Test, Appropriate leak test procedures should be employed to verify the integrity of all components, sample lines, and connections. The following leak test procedure is suggested: For components upstream of the sample pump, attach the probe end of the sample line to a manometer or vacuum gauge, start the nump and pull greater than 50 mm (2 in.) Hg vacuum, close off the pump outlet, and then stop the pump and ascertain that there is no leak for I minute. For components after the pump, apply a slight positive pressure and check for leaks by applying a liquid (detergent in water, for example) at each joint. Bubbling indicates the presence of a leak,

7.1.2 System Performance. Since the complete system is calibrated following each test, the precise calibration of each component is not critical. However, these components should be verified to be operating properly. This verification can be performed by observing the response of flowmeters or of the GC output to changes in flow rates or calibration gas concentrations and ascertaining the response to be within predicted limits. If any component or the complete system fails to respond in a normal and predictable manner, the source of the discrepancy should be identifed and corrected before proceeding.

### 8. Calibration

Prior to any sampling run, calibrate the system using the following procedures. (If more than one run is performed during any 24-hour period, a calibration need not be performed prior to the second and any subsequent runs. The calibration must, however, be verified as prescribed in section 10, after the last run made within the 24-hour period.)

8.1 General Considerations. This section outlines steps to be followed for use of the GC/FPD and the dilution system. The procedure does not include détailed instructions because the operation of these systems is complex, and it requires an understanding of the individual system being used, Each system should include a written operating manual describing in detail the operating procedures associated with each component in the measurement system. In addition, the operator shuld be familiar with the operating principles of the components; particularly the GC/FPD. The citations in the Bibilography at the end of this method are recommended for review for this purpose.

8.2 Calibration Procedure. Insert the permeation tubes into the tube chamber. Check the bath temperature to assure agreement with the calibration temperature of the tubes within ±0.1°C. Allow 24 heurs for the tubes to equilibrate. Alternatively equilibration may be verified by injecting samples of

<sup>&#</sup>x27;Mention of trade names or specific products does not constitute an endorsement by the Environmental Protection Agency.

calibration gas at 1-hour intervals. The per-ternatively, the GC/FPD system may be the procedures in section 8. Only H.S (or meation tubes can be assumed to have reached equilibrium when consecutive hourly samples agree within the precision limits of section 4.1.

Vary the amount of air flowing over the tubes to produce the desired concentrations for calibrating the analytical and dilution systems. The air flow across the tubes must at all times exceed the flow requirement of the analytical systems. The concentration in parts per million generated by a bube containing a specific permeant can be calculated as follows:

### $C = K \times P_r / ML$

Equation 15-1

where:

C=Concentration of permeant produced in ppm.

 $P_r$ =Permeation rate of the tube in  $\mu g/min$ . M = Molecular weight of the permeant: g/gmole.

L=Flow rate, I/min, of air over permeant @ 20°C, 760 mm Hg.

K=Gas constant at 20°C and 760 mm Hg = 24.04 l/g mole.

8.3 Calibration of analysis system. Generate a series of three or more known concentrations spanning the linear range of the FPD (approximately 0.05 to 1.0 ppm) for each of the four major sulfur compounds. Bypassing the dilution system, inject these standards in to the GC/FPD analyzers and monitor the responses. Three injects for each concentration must yield the precision described in section 4.1. Failure to attain this precision is an indication of a problem in the calibration or analytical system. Any such problem must be identified and corrected before proceeding.

8.4 Calibration Curves. Plot the GC/FPD response in current (amperes) versus their causative concentrations in upm on log-log coordinate graph paper for each sulfur compound. Alternatively, a least squares equation may be generated from the calibration

8.5 Calibration of Dilution System, Generate a know concentration of hydrogen sulfied using the permeation tube system. Adjust the flow rate of diluent air for the first dilution stage so that the desired level of dilution is approximated. Inject the diluted calibration gas into the GC/FPD system and monitor its response. Three injections for each dilution must yield the precision described in section 4.1. Failure to attain this precision in this step is an indication of a problem in the dilution system. Any such problem must be identified and corrected before proceeding. Using the calibration data for H<sub>2</sub>S (developed under 8.3) determine the diluted calibration gas concentration in ppm. Then calculate the dilution factor as the ratio of the calibration gas concentration before dilution to the diluted calibration gas concentration determined under this paragraph. Repeat this procedure for each stage of dilution required, Al-

calibrated by generating a series of three or more concentrations of each sulfur compound and diluting these samples before inecting them into the GC/FPD system. This data will then serve as the callbration data for the unknown samples and a separate determination of the dilution factor will not be necessary. However, the precision requirements of section 4.1 are still applicable.

# 9. Sampling and Analysis Procedure

9.1 Sampling. Insert the sampling probe into the test port making certain that no dilution air enters the stack through the port. Begin sampling and dilute the sample approximately 9:1 using the dilution system. Note that the precise dilution factor is that which is determined in paragraph 8.5, Condition the entire system with sample for a minimum of 15 minutes prior to commencing analysis.

9.2 Analysis. Aliquots of diluted sample are injected into the GC/FPD analyzer for analysis.

9.2.1 Sample Run. A sample run is composed of 16 individual analyses (injects) performed over a period of not less than 3 hours or more than 6 hours.

9.2.2 Observation for Clogging of Probe. If reductions in sample concentrations are observed during a sample run that cannot be explained by process conditions, the sampling must be interrupted to determine if the sample probe is clogged with particulate matter. If the probe is found to be clogged, the test must be stopped and the results up to that point discarded. Testing may resume after cleaning the probe or replacing it with a clean one. After each run, the sample probe must be inspected and, if necessary, dismantled and cleaned.

### 10. Post-Test Procedures

10.1 Sample Line Loss, A known concentration of hydrogen sulfide at the level of the applicable standard, ±20 percent, must be introduced into the sampling system at the opening of the probe in sufficient quantities to ensure that there is an excess of sample which must be vented to the atmosphere. The sample must be transported through the entire sampling system to the measurement system in the normal manner. The resulting measured concentration should be compared to the known value to determine the sampling system loss. A sampling system loss of more than 20 percent is unacceptable. Sampling losses of 0-20 percent must be corrected by dividing the resulting sample concentration by the fraction of recovery. The known gas sample may be generated using permeation tubes. Alternatively, cylinders of hydrogen sulfide mixed in air may be used provided they are traceable to permeation tubes. The optional pretest procedures provide a good guideline for determining If there are leaks in the sampling system.

10.2 Recalibration. After each run, or after a series of runs made within a 24-hour period, perform a partial recalibration using

other permeant) need be used to recalibrate the GC/FPD analysis system (8.3) and the dilution system (8.5).

10.3 Determination of Calibration Drift. Compare the calibration curves obtained prior to the runs, to the calibration curves obtained under Section 10.2. The calibration drift should not exceed the limits set forth in Section 4.2. If the drift exceeds this limit, the intervening run or runs should be considered not valid. The tester, however, may instead have the option of choosing the calibration data set which would give the highest sample values. [10.3 amended by 52 FR 34639, September 14,

## 11. Calculations

11.1 Determine the concentrations of each reduced sulfur compound detected directly from the calibration curves. Alternatively, the concentrations may be calculated using the equation for the least squares line.

11.2 Calculation of SO, Equivalent, SO, equivalent will be determined for each analysis made by summing the concentrations of each reduced sulfur compound resolved during the given analysis.

## SO, equivalent = $\Sigma(H_1S, COS, 2CS_1)d$

Equation 15-2

where:

SO, equivalent=The sum of the concentration of each of the measured compounds (COS, H,S, CS,) expressed as sulfur dioxide in ppm.

H.S = Hydrogen sulfide, ppm. COS=Carbonyl sulfide, ppm.  $CS_1 = Carbon disulfide, ppm.$ d=Dilution factor, dimensionless.

11.3 Average SO, equivalent will be determined as follows:

Average 
$$SO_2$$
 equivalenc =  $\frac{1}{N}$   $\frac{1}{N}$   $\frac{1}{N}$   $\frac{1}{N}$   $\frac{1}{N}$ 

Equation 15-3

where:

Average SO, equivalent = Average equivalent in ppin, dry basis.  $SO_2$  equivalent<sub>1</sub> =  $SO_2$  in ppm as determined by Equation 15-2.

N=Number of analyses performed.

Bwo = Praction of volume of water vapor in the gas stream as determined by Method 4-Determination of Moisture in Stack Gases (36 FT: 24887).

[11.3 corrected by 52 FR 34639, September 14, 19871

12.1.4.1 Tube Chamber, Glass chamber of sufficient dimensions to house permeation tubes.

12.1.4.2 Mass Flowmeters. Two mass flowmeters in the range 0-3 1/min. and 0-10 1/ min. to measure air flow over permeation tubes at  $\pm 2$  percent. These flowmeters shall be cross-calibrated at the beginning of each test. Using a convenient flow rate in the measuring range of both flowmeters, set and monitor the flow rate of gas over the permeation tubes. Injection of calibration gas generated at this flow rate as measured by one flowmeter followed by injection of calibration gas at the same flow rate as measured by the other flowmeter should agree within the specified precision limits. If they do not, then there is a problem with the mass flow measurement. Each mass flowmeter shall be calibrated prior to the first test with a wet test meter and thereafter at least once each year.

12.1.4.3 Constant Temperature Bath. Capable of maintaining permeation tubes at certification temperature of 30°C within ±0.1°C.

12.2 Reagents.

12.2.1 Fuel. Hydrogen (H.) prepurified grade or better.

12.2.2 Combustion Gas. Oxygen (O<sub>1</sub>) research purity or better.

12.2.3 Carrier Gas. Nitrogen (N<sub>2</sub>) prepurified grade or better.

12.2.4 Diluent. Air containing less than 0.5 ppm total sulfur compounds and less than 10 ppm each of moisture and total hydrocarbons, and filtered using MSA filters 46727 and 79030, or equivalent. Removal of sulfur compounds can be verified by injecting dilution air only, described in section 8.3.

12.2.5 Compressed Air. 60 psig for GC valve actuation.

12.2.6 Calibration Gases. Permeation tubes gravimetrically calibrated and certified at 30.0 °C.

12.3 Operating Parameters. The operating parameters for the GC/FPD system are as follows: nitrogen carrier gas flow rate of 100 cc/min, exhaust temperature of 110° C, detector temperature 105° C, oven temperature of 40° C, hydrogen flow rate of 80 cc/minute, oxygen flow rate of 20 cc/minute, and sample flow rate of 80 cc/minute.

12.4 Analysis. The sample valve is actuated for 1 minute in which time an aliquot of diluted sample is injected onto the separation column. The valve is then deactivated for the remainder of analysis cycle in which time the sample loop is refilled and the separation column continues to be foreflushed. The elution time for each compound will be determined during calibration.

13. Bibliography.

13.1 O'Keeffe, A. E. and G. C. Ortman. "Primary Standards for Trace Gas Analysis." Anal. Chem. 38,760 (1966).

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13.3 Mulick, J. D., R. K. Stevens, and R. Baumgardner. "An Analytical System Designed to Measure Multiple Malodorous Compounds Related to Kraft Mill Activities." Presented at the 12th Conference on Methods in Air Pollution and Industrial Hygiene Studies, University of Southern California, Los Angeles, Calif. April 6-8, 1971.

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Canada, 73,3 (March, 1972).

13.5 Grimley, K. W., W. S. Smith, and R. M. Martin. "The Use of a Dynamic Dilution System in the Conditioning of Stack Gases for Automated Analysis by a Mobile Sampling Van." Presented at the 63rd Annual APCA Meeting in St. Louis, Mo. June 14-19, 1970.

13.6 General Reference. Standard Methods of Chemical Analysis Volume III A and B Instrumental Methods. Sixth Edition. Van Nostrand Reinhold Co.

Method 15A — Determination of Total Reduced Sulfur Emissions From Sulfur Recovery Plants in Petroleum Refineries

[Method 15A added by 52 FR 20392, June 1, 1987]

1. Applicability, Principle, Interferences, Precision, and Bias

1.1 Applicability. This method is applicable to the determination of total reduced sulfur [TRS] emissions from sulfur recovery plants where the emissions are in a reducing atmosphere, such as in Stretford units. The lower detectable limit is 0.1 ppm of sulfur dioxide (SO<sub>1</sub>) when sampling at 2 liters/min for 3 hours or 0.3 ppm when sampling at 2 liters/min for 1 hour. The upper concentration limit of the method exceeds TRS levels generally encountered in sulfur recovery plants.

1.2 Principle. An integrated gas sample is extracted from the stack, and combustion air is added to the oxygen (O<sub>3</sub>-deficient gas at a known rate. The TRS compounds (hydrogen sulfide, carbonyl sulfide, and carbon disulfide) are thermally oxidized to sulfur dioxide, collected in hydrogen peroxide as sulfate ion, and then analyzed according to the Method 6 barium-thorin titration procedure.

1.3 Interferences. Reduced sulfur compounds, other than TRS, that are present in the emissions will also be oxidized to SO<sub>2</sub>. For example, thiophene has been identified in emissions from a Stretford unit and produced a positive bias of 30 percent in the Method 15A result. However, these biases may not affect the outcome of the test at uoits where emissions are low relative to the standard.

Calcium and aluminum have been shown to interfere in the Method 6 titration procedure. Since these metals have been identified in particulate matter emissions from Stretford units, a Teflon filter is required to remove this interference.

Note. — Mention of trade name or commercial products in this publication does not constitute the endorsement or recommendation for use by the Environmental Protection Agency.

When used to sample emissions containing 7 percent moisture or less, the midget impingers have sufficient volume to contain the condensate collected during sampling. Dilution of the H<sub>2</sub>O<sub>1</sub> does not affect the collection of SO<sub>1</sub>. At higher moisture contents, the potassium citratecitric acid buffer system used with Method 16A should be used to collect the condensate.

1.4 Precision and bias. Relative standard deviations of 2.8 and 6.9 percent at 41 ppm TRS have been obtained when sampling for 1 and 3 hours, respectively. Results obtained with this method are likely to contain a positive bias due to the presence of nonregulated sulfur compounds (that are present in petroleum) in the emissions.

2. Apparatus

2.1 Sampling. The sampling train is shown in Figure 15A-1, and component parts are discussed below. Modifications to this sampling train are acceptable provided that the system performance check is met.

### 12. Example System

Described below is a system utilized by EPA in gathering NSPS data. This system does not now reflect all the latest developments in equipment and column technology, but it does represent one system that has been demonstrated to work.

12.1 Apparatus.

12.1.1 Sample System.

12.1.1.1 Probe. Stainless steel tubing, 6.35 mm (4 in.) outside diameter, packed with glass wool.

12.1.1.2 Sample Line. % inch inside diameter Teflon tubing heated to 120° C. This temperature is controlled by a thermostatic heater.

12.1.1.3 Sample Pump. Leakless Teflon coated diaphragm type or equivalent. The pump head is heated to 120°C by enclosing it in the sample dilution box (12.2.4 below).

12.1.2 Dilution System. A schematic diagram of the dynamic dilution system is given in Figure 15-2. The dilution system is constructed such that all sample contacts are made of inert materials. The dilution system which is heated to 120° C must be capable of a minimum of 9:1 dilution of

sample. Equipment used in the dilution system is listed below:

12.1.2.1 Dilution Pump. Model A-150 Kohmyhr Teflon positive displacement type, nonadjustable 150 cc/min. ±2.0 percent, or equivalent, per dilution stage, A 9:1 dilution of sample is accomplished by combining 150 cc of sample with 1350 cc of clean dry air as shown in Figure 15-2.

12.1.2.2 Valves. Three-way Teflon solenoid or manual type.

12.1.2.3 Tubing. Teflon tubing and fittings are used throughout from the sample probe to the GC/FPD to present an inert surface for sample gas.

12.1.2.4 Box. Insulated box, heated and maintained at 120°C, of sufficient dimensions to house dilution apparatus.

12.1.2.5 Flowmeters. Rotameters or equivalent to measure flow from 0 to 1500 mi/min. ±1 percent per dilution stage.

12.1.3.0 Gas Chromatograph.

12.1.3.1 Column—1.83 m (6 ft.) length of Teflon tubing, 2.16 mm (0.085 in.) inside diameter, packed with deactivated silica gel, or equivalent.

12.1.3.2 Sample Valve. Teflon six port gas

sampling valve, equipped with a 1 ml sample loop, actuated by compressed air (Figure 15-1)

12.1.3.3 Oven. For containing sample valve, stripper column and separation column. The oven should be capable of maintaining an elevated temperature ranging from ambient to  $100^{\circ}$  C, constant within  $\pm 1^{\circ}$  C.

12.1.3.4 Temperature Monitor. Thermocouple pyrometer to measure column oven, detector, and exhaust temperature ±1°C.

12.1.3.5 Flow System. Gas metering system to measure sample flow, hydrogen flow, oxygen flow and nitrogen carrier gas flow.

12.1.3.6 Detector. Flame photometric detector.

12.1.3.7 Electrometer. Capable of full scale amplification of linear ranges of 10<sup>-9</sup> to 10<sup>-4</sup> amperes full scale.

12.1.3.8 Power Supply. Capable of delivering up to 750 volts.

12.1.3.9 Recorder. Compatible with the

output voltage range of the electrometer. 12.1.4 Chibration. Permeation tube system (Figure 15-3).

STACK WALL TO GC/FPD ANALYZERS 100:1 10:1 PROBE CTV. VI DILUENT AIR 1350 cc/min POSITIVE FILTER (GLASS WOOL) DISPLACEMENT PUMPS FLOW (150 cc/min) METERS HEATED PERMEATION SAMPLE LINE TUBE CALIBRATION GAS CLEAN. 3-WAY DRY AIR VALVE (25 psi) **HEATED DILUTION BOX** (100 °C) DIAPHRAGM PUMP (HEATED) VENT

Figure 15-1. Sampling and dilution apparatus.

(Figure 15-1 added by 52 FR 34639, September 14, 1987)

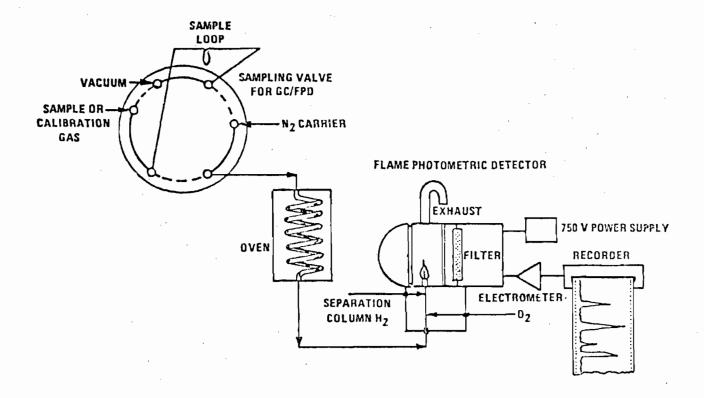


Figure 15-2. Gas chromatographic flame photometric analyzer.

[Figure 15-2 added by 52 FR 34639, September 14, 1987]

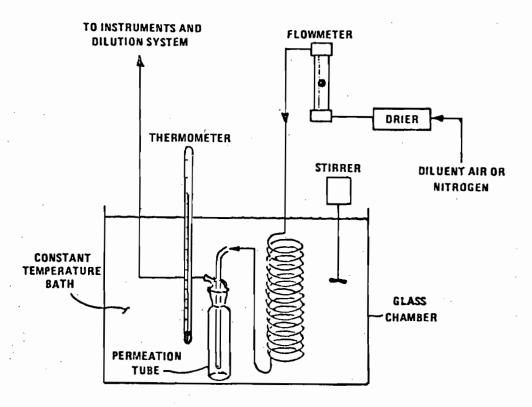


Figure 15-3. Apparatus for field calibration.

[Figure 15-3 added by 52 FR 34639, September 14, 1987]

```
CHANNEL A
            INJECT 06/16/89 10:20:01
                                     1. 25
                          06/16/89 10:20:01 CH= "A" PS= 1.
FILE 1.
          METHOD 0. RUN 22
                                   INDEX 22
PEAK#
                   RT
       AREA%
                          AREA BC
   1
          1.24
                   0.3
                           616 02
                   0.35
          1.028
                           511 02
         32.642
                   Ø. 49
                          16221 03
   4
        · 65.09
                   1.25
                          32346 01
TOTAL 190.
                          49694
CHANNEL A INJECT 06/16/89 10:24:01
                         06/16/89 10:24:01 · CH= "A" PS= 1.
         METHOD 0. RUN 23
FILE 1.
                                   INDEX
                                         23
PEAK#
        AREA%
                  RT AREA BC
          2.821
                   0.3
   1
                           596 02
                  0.35
          2.5
                           528 02
         30.16
                   0.49
                          6371 03
         64.519
                   1.25
                          13629 01
TOTAL
     199.
                          21124
CHANNEL A
            INJECT
                   06/16/89 10:28:01
                          06/16/89 10:28:01 CH= "A" PS= 1.
FILE 1.
         METHOD 0. RUN 24 INDEX 24
                  RT AREA BC
PEAK#
        AREA%
          2.967
                   0.3
                           610 02
   2
          2.335
                   0.35
                            480 02
         30.18
                   0.49
                           6204 03
         64.518
                   1.25
                          13263 01
TOTAL
        100.
                          20557
CHANNEL A INJECT 06/16/89 10:32:01
```

```
CHANNEL A
              INJECT
                      86/16/89 09:48:00
         2.22
                              96/16/89 09:48:00
                                                     CH= "A"
                                                               PS=
                                                                    1.
FILE
             METHOD
                     Ē.
                             RUN 14
                                          INDEX
                                                14
PEAK#
          AREA%
                        RT
                                AREA BC
    1
            9.971
                       0.3
                                 533 02
                       0.35
            0.068
                                 511 02
          ·32.856
                       0.49
                              245623 03
    4
           66.966
                       1.26
                              500626 01
    5
            0.038
                       2. 22
                                 286 01
TOTAL
          100.
                              747579
CHANNEL A
              INJECT
                       06/16/89 09:52:00
                                                                    1,25
           2.25
                                                     CH= "A" PS=
                               06/16/89 09:52:00
             METHOD
FILE
      1.
                     0.
                             RUN 15
                                          INDEX
                                                 15
PEAK#
          AREA%
                        RT
                                AREA BC:
    1
            0.272
                       0.3
                                 562 02
           9.224
                       0.35
                                 463 02
           33.74
                              69764 03
                       9.49
    4
           65.662
                       1.26
                              135767 01
            0.103
                       2, 25
                                  212 01
TOTAL .
          100.
                              206768
CHANNEL A
              INJECT
                       06/16/89 09:56:00
                                                       CH= "A" PS= 1.
                               06/16/89 09:56:00
FILE
             METHOD
                     Ø.
                             RUN 16
                                          INDEX
                                                 16
PEAK#
          AREA%
                        RT
                                 AREA BC
            0.284
                       0.3
    1
                                 584 02
            0.23
                       0.35
                                 472 02
           33.617
                       0.49
                              · 69112 03
           65.87
                       1.25
                              135420 01
TOTAL
                               205588
          199.
CHANNEL A
              INJECT
                       96/16/89 10:00:00
```

```
ILE 1. METHOD 0. RUN 40 INDEX 40
PEAK# AREA% RT AREA BC
  1... 100. .... 0.49 ... 317436 01
EOTAL 100. 317436
CHANNEL A INJECT 06/16/89 11:40:02
                                           . 49
                    96/16/89 11:40:02 CH= "A" PS≈ 1.
ILE 1. METHOD 0. RUN 41 INDEX 41
PEAK# AREA% RT AREA BC
 1
      100.
               0.49 241497 01
TOTAL 100.
                    241497
CHANNEL A INJECT 06/16/89 11:44:02
               06/16/89 11:44:02 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 42 INDEX 42
PEAK# AREA% RT
                     AREA BC
       0.104 0.33
99.896 0.49
  1
                      323 02
       99.896
                     311005 08
       9.091
               0.8
                     2 05
TOTAL 100.
                    311330
CHANNEL A INJECT 06/16/89 11:48:02 -
                    06/16/89 11:48:02 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 43 INDEX 43
PEAK# AREA% RT AREA BC
                              29N TNSD
 1
       100. 0.49 437669 01
TOTAL 100.
                 437669
CHANNEL A INJECT 06/16/89 11:52:02
```

```
06/16/89 13:25:04     CH= "A" PS=    
FILE 1.
         METHOD 0.
                       RUN 57 INDEX 57
PEAK#
      AREAX
                   RT
                        AREA BC
   1
          1.631
                   0.23
                          43974 02
         28.914
                   Ø. 44
                         779673 02
         28.654
                   0. 49
                         772665 02
         38.408
                   0.56
                         1035690 02
          2.394
                   1.02
                          64551 03
TOTAL
     100.
                         2696553
CHANNEL A INJECT 06/16/89 13:28:04
       . 1411.929
                          06/16/89 13:28:04 CH= "A" PS=
FILE 1. METHOD 0. RUN 58
                                    INDEX 58
PEAK# AREAZ
                  RT
                          AREA BC
   1
         4.782
                   1.08
                            91 02
         42.144
                   1.14
                           -802 02
         53.074
                   1.2
                           1010 03
    100.
TOTAL
                          1903
CHANNEL A INJECT 06/16/89 13:31:04
          1.02
     <u>2.</u> 44
                          06/16/89 13:31:04 CH= "A" PS= 1.
FILE 1.
           METHOD 0. RUN 59 INDEX 59
                  RT AREA BC
PEAK#
      AREA%
                            58 01
         14.286
                   1.02
         85.714
                    2.44
                            348 01
TOTAL 100.
                            496
CHANNEL A INJECT 06/16/89 13:34:04
           .19
                                           CH= "A" PS=
                           06/16/89 13:34:04
           METHOD. 0. RUN 60 INDEX 60
FILE 1.
```

```
CHANNEL A INJECT 06/16/89 18:36:06
      1.02
                       06/16/89 18:36:06 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 131 INDEX 131
PEAK# AREA% RT AREA BC
       99.977
                0.43 998743 08
  1
        0.023
                1.02
                        230 05
                    998973
TOTAL 190.
CHANNEL A INJECT 06/16/89 18:39:06
      1.02
                      06/16/89 18:39:06 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 132 INDEX 132
PEAK# AREA% RT AREA BC
                 0.43 1102023 08
       99.978
        0.022
                 1.02
                       239 05
                      1102262
TOTAL 100.
CHANNEL A INJECT 06/16/89 18:42:06
      T. 03
                      06/16/89 18:42:06 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 133 INDEX 133
PEAK#
    AREA%
                 RT
                     AREA BC
                     1037564 08
       99.981
                0.43
        0.019
                1.03
                         198 05
                      1037762
TOTAL 100.
CHANNEL A INJECT 06/16/89 18:45:06
      1.92 42
                       96/16/89 13:45:86 CH= "A" PS= 1.
FILE 1. METHOD 0. RUN 134 INDEX 134
PEAK# AREAM RT
                        ARFA BC
```

TABLE I

## TOTAL REDUCED SULFUR EMISSIONS $H_2S$ FLARE - INLET GASES

## HARRIS CORPORATION, SEMICONDUCTOR DIVISION MELBOURNE, FLORIDA

DATE	RUN NO.	TIME PERIOD	LEVEL	H <sub>2</sub> s <sup>2</sup>	CONÇENTRATION,	ppm <sub>4</sub> * CS <sub>2</sub>	so <sub>2</sub> EQ. <sup>5</sup>
06/16/89	1	1337-1421	MAX MIN	23700 19400	ND <sup>6</sup>	ND	23700 19400
			AVG	21500			21500
06/16/89	2	1724-1809	MAX	23200	ND	ND	23200
			MIN AVG	17800 21000			17800 21000
05/15/00	2	1015 0000					
06/16/89	. 3	1815-2200	MAX MIN	24500 20300	ND	ND	24500 20300
**MEAN FOR	THREE RUN	19	AVG	22600 21700			2260 <u>0</u> 21700

<sup>\*</sup>Concentration Corrected for Recovery

<sup>\*\*</sup>Mean determined as arithmetic average of the average results for each of the three runs.

<sup>1</sup>ppm-----Parts per million by volume, corrected for recovery.

 $<sup>^2</sup>$ H $_2$ S-----Hydrogen Sulfide

<sup>&</sup>lt;sup>3</sup>COS-----Carbonyl Sulfide

 $<sup>^4\</sup>text{CS}_2$ -----Carbon Disulfide

<sup>&</sup>lt;sup>5</sup>SO<sub>2</sub>EQ.----Sulfur Dioxide Equivalent

 $<sup>^6\</sup>mathrm{ND}\text{----}\mathrm{Not}$  detected, i.e., below detection limit of the system.

Company: Harris Corp., Semiconductor Div.

Date: 6-16-89 Melbourne, Fla.

Source: H, 5 Flare - Inlet Gases R.In

SHEET \_\_ OF \_\_\_\_

Run 1

TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
13.37	Compo			1/25		987796	4.34	5000	2.17
1340				H2S		986022	4,34		2.17
								·	
1343			· ·	H25	·	928158	4.21		2.11
1346				HLS		877835	4.10		0.05
			·						
1349				H25		934272	4,23		2.11
1352				425		854426	4.05		2.02
12 -0				1/ 6		835144	// 0.0		
1355				1/25		033144	4.00		2.00
1358				11.5	· · · · · · · · · · · · · · · · · · ·	741377	3.78	· · · · · · · · · · · · · · · · · · ·	1.89
1.25				117-0		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			, , , , , , , , , , , , , , , , , , , ,
1401	·			1/25		1120992	4.62		2,3/
1404	***	V		t/2-S		892.046	4.13	V	2.07

Company: Harris Corp., Semiconductor Div,

Date: 6-16-89 Melbourne, Fla.

Source: H. S. Flore - Inlet Gases

TIME COMMON 1995

SHEET 2 OF \_\_\_\_

(Val. %)

2041	·				,				(101: 10)
TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
1407	Combo	ري .		11.5		960607	4.29	5000	2.14
				• •					
1410				H25	·	929915	4.22		2.11
					·			į	
1413				H25	· · ·	970705	4.31		2.15
1416				H25		963907	4.29		2.15
			<u> </u>						
1419				H25		948381	4,26		2.13
1421		_		4,5		851673	4.04	V	2.02
								·	· · · · · · · · · · · · · · · · · · ·
			,				· .		
						, ·		·	· · · · · · · · · · · · · · · · · · ·
						· · · · · ·			

GLC DATA SHEET

Company: Harris Corp. Semiconductor Div.

Date: 6-16-89 Molhourne, Fla.

Source: H<sub>2</sub>S Flare - Inlet Gases

SHEET 3 OF \_\_\_\_

Dun 1

(12/ %)

Sour	ce. [1] 3		(Vol. le)						
TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
1724	Combo	,5		H25		874092	4.09	5000	2.05
1727				H25		1034095	4.44		2.22
1730				H25		1069632	4,52		2.26
1733				H2S		953259	4.27		2.13
			-				<i>(</i> 1 0 5		1.00
1736				Has		827039	3,98		1.99
1739		1		H25		781636	3.88		1.94
1.1.2.1				1/20		101000	0.000		// /
1742				H15		623673	3.47		1.74
1745				1/25		910910	4.18		2.09
1748			`	H15		899542	4.15		2.08
		_							
1751		V		H25		1026635	4.43		2.21
			ł						

#### GLC DATA SHEET

Company: Harris Corp., Semiconductor Div.

Date: 6-16-89 Melbourne, Fla.

Source: HzS Flare - Inlet Gases

TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
1154	Combo	٠, ১		H25		820152	3.97	5000	1,98
1757				H2-5		911237	4,18		2.09
1800				H25		890618	4, 13		2.07
70.00				- 11 <del>1 ~</del>		0,02,			
1803				1/25		728637	3.75		1.87
1806				11,5		780971	3.87		1.94
1809				11,5		932483	4.22		2.11
			·						
									·
						· .			
					•				
1			`.						

Company: Harris Corp., Semiconductor Div

Date: 6-16-89 Melhourne, Ela.

Source: Hg.S Flare - Inlet Gases Run 3

Run 3

SHEET 5 OF \_\_\_\_

(b) %)

2011	ce <i>[1]</i>	<u> </u>	1.5	LY IT	Wases	1,0	,, _		(00/ 20)
TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
1815	Combo	.5		H25		926746	4.21	5000	3.11
						. '			
1818				1/25		966877	4.30		2.15
1821				1-1,5		111 4543	4.61		2.30
1824				H25		969336	4.30		2.15
1827				H25		1067884	4,51		2.26
1830				H25		1068868	4,51		2.26
				·					
1833				1125		1030164	4,43		2.22
							· .		
1836				H25		998743	4.37		2.18
						<i>:</i>			
1839				Has	. ,	1102023	4,5-8		2.29
				·					
1842	V			H35		1037564	4.45	<u> </u>	2.2.2

Company: Harris Porp., Semiconductor Viv.

Date: 6-16-89 Melbourne, Fla

Source: Has Flare - Inlet Gases

SHEET 6 OF

(vol. %)

TIME	COLUMN	LOOP	ATTN.	CMPD.	PEAK HT.	PEAK AREA	CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
1845	Combo	.5		H25		1073042	4.52	5000	2.26
				·					
2142				Has		815950	3.96		1.98
2145				H25		917713	4.19		2.10
2148				H25		921277	4,20		2.10
					: 				
3151				11-5		854248	4,05		2.02
					·				
2154				1/25		1130307	4.64		2,32
2157				H25		1120459	4.62		2.3/
		·							
2200		<u> </u>		1/25		1208851	4,79	· •	2.40
							:		
								·	
									,

+ Note. RO Unit shut down @ 1430

			,			2		, GL	C DATA SHE	ET		
	Comp	pany	: <i>Ha</i>	rr	<u>15</u>	Corp.	Semic	onductor	DATA SHE		SHEET	OF
	Date	:	6-1	<u> 16 ·</u>	- 8º	7	10/00	urne, Fla Gases	<b>.</b>	•		(vol. %)
. 1	TIME	COLU					CMPD.		PEAK AREA	Run 2 CONC. (PPM)	DIL. FACTOR	TRS CONC. (PPM)
	1428						H25		1049534		5000	2,24
			1.00									
	1431		_				1/25		1226755	4.83		2.41
*]	1434				·		H25		388350	2.76		1.38
			_	-		<u> </u>						
	1437						H25		391874	2.77		1.39
-					,		,					
-	1440						H25		36 945-1	2.69		1.35
-							// -					
}	1443						1725		212831	2.06		/,63
ľ	1446						Has		132423	1,63		0.82
							1/12	-	100.125	7,02		0.02
Ī	1449						Hzs		60605	1.12		0.56
								<u>.</u>		·		
	11/52						H25		73401	1,23		0.61
			-									
ļ	1455		.		/		HOS		41180	0.93		0.46
									2 N - 2			

#### RECOVERY DATA

CLIENT ACE-Horris	Orp. IS	I PROJECT NO	060389
PLANT Harris Corp.,	Semiconductor	Div.	
LOCATION Melbourne.			
SOURCE Has Flore - In		TE 6-16-0	89

		St			luent	Total	Time
TLOW (ml/min	n)	F	rom	Pe	rmea	tion sy.	tem
IME	IME PEAK AREA		AVC	vg. conc. remark		S	
2245	2613	397					
<del> </del>	2603	820	2651	1/34	7.02	Direc	4
	2731	6173				٠.	<del> </del>
					· · · · · · · · · · · · · · · · · · ·		
2315	2599	4838					· · · · · · · · · · · · · · · · · · ·
<del></del>	25-2	1639	251	8908	6.85		
	7-44	0247		<u> </u>			·
· · · · · · · · · · · · · · · · · · ·		·		<del></del>			·
•		-		<u>-</u>			· · · · · · · · · · · · · · · · · · ·
<u>.</u>	,			<i>.</i>	-		
	_						

To Recovery = 6.85 x100 = 97.6%

#### CALIBRATION DATA

CLIENT ALE-Harris Corp. TSI PROJECT NO. 060389

PLANT Harris Corp., Semiconductor Line Holbourne, Fla.

LOCATION Melbourne, Florida

SOURCE Has Flare - Inlet Gases DATE 6-16-89

COMPOUND	H <sub>2</sub> S	MeSH	DMS	DMDS	AMB TEMP
PERMEATION RATE (ng/min)	1050	2620	·		75° F
TUBE NUMBER	30394				BAR PRES 30.0

	FLOW		CONC	AREA COUNTS						
TIME	ml/min	CMPD	ppm	1	2	3	AVG.			
1104	329	H25	2.29	270571	270011	279597	213415			
		MeSH	4,05	521198	532400	538389	527296			
	·		,							
0956	. 621	H25	1.21	69/12	67739	66548	67,800			
		MeSH	2.14	135420	137938	125/03	136154			
10/2	1276	H25	0.59	16805	16264	14221	16430			
		MOSH	1.04	33845	32984	32:346	3 3058			
1032		1/25	0,35	5817	5582	55/6	5639			
		MOSH	0.62	12606	1347	11952	12302			

#### CALIBRATION DATA SUMMARY

CLIEN	ACI	E- Hor	ric.	Con	7 <u>.</u>	 TSI	PROJE	CT NO. 06	0389	}
		is Porp		- /						_
		elhourn			,					_
			,				E 6	-16-8	9	_
					100	 <u></u> DA1.	L	- 16 - 0	<del>Z</del>	-
								•		
CMPD_	Hz	S_Curve	<u> </u>			CMPD_	Hes	#_Curve	/	_
<del></del>				-	·	l			1	· ·
PIME	CONC ppn	RESPONSE	CALC CONC ppm	DIFF		TIME	CONC	RESPONSE	CALC CONC ppm	% DIFF
1104	2,29	273415	2,32	1.3	]	1104	4.05	527296	4.12	1.7
0952	1,21	67800	1.18	2.5	.	0956	3.14	136154	2.09	2.3
1012	0.59	16430	0.59	0.0				33058		
1032	0.35	5639	0.35	0.0				12302	_	
SLOPE	2.05	<u>57                                    </u>	nt 48	23/	•	SLOPE	1.99	790 Y i	nt 7/	108
CONC (							<u> </u>		··· <u>····</u>	,
CMPD		Curve							7 <del>0</del>	
			·		.					
TIME		RESPONSE	CALC			TIME	CONC			_ I
	ppm		DONC ppm	DIFF			ppn	·	NOO	1
_			-		-					
			-		.					
			-		+	-				
				-	1		· · · · · ·			
					1	1				
SLOPE		Y :	int		+	SLOPE	:	Y	int	
CONC	<u>a</u>					CONC	<u>a</u>	•		

,9998

#### INSTRUMENT DATA

CLIENT ACE - Harris Corp.	TSI_PROJECT_NO
LOCATION Horris Port Seminoraques	tor Div. Holhowing Fla.
SOURCE Has Flare - Inlet Gases	DATE 6-16-89
INSTRUMENT TRAPOR 250 HC	·

Gas Pressures	Temperatures	Columns
H <sub>3</sub> : 32psi	TRAILER: 15°F	PPE-1: 0,5 m
01: 26 psi	COLUMN: 55°C	BHT-100: 1,5 m
Na: 60 psi	VALVE: 55°C	PPE-2: 2.0 m
Air:	DETECTOR: /15°C	

TIMER PROGRAM	INTEGRATOR PROGRAM
EVENT / 2 3 PROGRAM / 2 3 CIRCUIT / 3 44 ON (min) 0.0 0.0 7.0 ON (sec) 0.0 30.0 VARY (min) 5.0 5.0 5.0 OFF (min) 1.0 0.0 31.0 VARY (min) 5.0 5.0 5.0 VARY (min) 5.0 5.0 5.0  VARY (min) 5.0 5.0 5.0  VARY (min) 5.0 5.0 5.0	Auto Manual Peak Height Peak Area
Event 1: Valve switch to inject sample  Event 2: Integrator on simultaneously w/ Event 1.  Event 3: Integrator off Generate report,	

## CERTIFICATE

The permeation rate of the DYNACAL® PERMEATION DEVICE listed below is certified traceable to N.B.S. standards.

CHEMICAL FILL : HYDROGEN SULFIDE

DEVICE TYPE : EXTENDED LIFE TUBE-STD

LENGTH/GEOMETRY : 1.2 CM.

PART NUMBER: 177-012-0110-10

METHOD OF CERTIFICATION: GRAVIMETRIC CERTIFICATION NUMBER: 10-30394

RATE: 1050 NG/MIN +/- 2% AT 35 DEG C

NOTE:

DATE: 13 APRIL 1989 BY: BYIANA QUAN

W.G

**VICI Metronics** 

2991 Corvin Drive Santa Clara, California 95051 U.S.A.

(408) 737-0550 Telex: 35-2129

#### INDIVIDUAL DEVICE CERTIFICATION

The gravimetric method measures the weight loss per unit of time at the certification temperature. Traceability is thus established by the use of temperature and weight standards traceable to N.B.S. standards.

Individual certification is accomplished by: (1) maintaining the device in a constant temperature chamber with a purge flow of dry nitrogen, and (2); weighing periodically on a semi-microanalytical balance, accurate to the nearest 0.01 mg, until a steady weight loss per unit time has been achieved. Jemperature control and accuracy are better than  $\pm$  0.05°C, referenced against temperature stundards traceable to the National Bureau of Standards. The semi-microanalytical balances are routinely serviced and calibrated by an independent service organization using N.B.S. traceable weight standards. Gravimetric permeation rate-determinations are continued until the standard error of the permeation rate meets the required accuracy at the 95% contidence level.

Validation of the certification procedures and standards at VICI Metronics is accomplished by routine certification of Standard Reference Material (SRM) permeation devices obtained from the National Bureau of Standards.

# METRONICS DYNACALIBRATOR ROTOMETER CALIBRATION DATA - READ AT CENTER OF FLOAT -

340-31	SERIAL NO:	-846	DATE CALIBRATED: 11/17/86 BY: TKP							
		DYNACALIBRATOR SERIES 450								
SPAN CH	ANNEL	SPAN	SPAN 1 CHANNEL SPAN 2 CHANNEL							
	•	TUBE TYPE:		TUBE TYPE:						
TOP FLOAT MATERIAL: GLASS	BOITOM FLOAT MATERIAL:	TOP FLO <b>AT</b>	BOTTOM FLOAT MATERIAL:	TOP FLOAT MATERIAL	BOTTOM FLOAT MATERIAL:					
FLOW RATE -LPM	FLOW RATE -LPM-	FLOW RATE -LPM-	FLOW RATE -LPM-	FLOW RAT	E FLOW RATE -LPM-					
.292			·	1						
.947										
1.81				1						
2.60										
3.44			· ·							
4.26										
5.16	•									
5.03			·							
6.92										
7.79										
8.70										
9.63										
10.63		5. 7								
11.59					<u> </u>					
12.50										
	DYNACAL SERIES 23 SPAN CH TUBE TYPE: B-2 SERIAL NO. 643 TOP FLOAT MATERIAL: GLASS FLOW RATE -LPM .292 .947 1.81 2.60 3.44 4.26 5.16 6.03 6.92 7.79 8.70 9.63 10.63 11.59	DYNACALIBRATOR SERIES 230 & 340  SPAN CHANNEL  TUBE TYPE: B-250-4  SERIAL NO. 643943  TOP FLOAT MATERIAL: MATERIAL: GLASS  FLOW RATE -LPM -1PM292 .947 1.81 2.60 3.44 4.26 5.16 6.03 6.92 7.79 8.70 9.63 10.63 11.59	DYNACALIBRATOR SERIES 230 & 340  SPAN CHANNEL  TUBE TYPE: B-250-4  SERIAL NO. 643943  TOP FLOAT MATERIAL: MATERIAL: MATERIAL:  FLOW RATE -LPM  -LPM  1.81  2.60  3.44  4.26  5.16  6.03  6.92  7.79  8.70  9.63  10.63  11.59	DYNACALIBRATOR SERIES 230 & 340  SPAN CHANNEL  SPAN I CHANNEL  TUBE TYPE: B-250-4  SERIAL NO. 643943  TOP FLOAT MATERIAL: MATERIAL: MATERIAL: MATERIAL:  FLOW RATE -LPM -LPM -LPM	DYNACALIBRATOR SERIES 230 & 340  SPAN CHANNEL  SPAN I CHANNEL  SPAN I CHANNEL  SPAN I CHANNEL  SPAN TUBE TYPE: SERIAL NO.  TOP FLOAT MATERIAL: MATERIAL: MATERIAL: MATERIAL: FLOW RATE -LPM  -LPM  -LPM  1.81  2.60  3.44  4.26  5.16  6.03  6.92  7.79  8.70  9.63  10.63  11.59					

All flow rate measurements are corrected to 25°C and 760 mm Hg.



#### VICI METRONICS

DYNACALIBRATOR MODEL 1348 -31 SERIAL NUMBER 1 M-846

DATE CALIBRATED 111-18-86 BY KAR

#### STANDARD TEMPERATURE RANGE

TEMPERATURE	DIGITAL				
-DEG C-	SETTING				
25	332				
26	361				
27	389				
28	415				
29	441				
30	465				
31	489				
32	511				
33	533				
34	553				
35	573				
36	592				
37	610				
38	627				
39	644				
48	660				
41	676				
42	690				
43	705				
44	718				
. 45	731				
46	744				
47	756				
• -	768				
48	778				
50	798				

APPENDIX D
TEST CALCULATIONS

#### TEST CALCULATIONS

ACE Supplied Values	John Zink Mass Flowlbs/Hr
Run 1 = 0.12 "H <sub>2</sub> O Average Orifice P	32.66
Run 2 = 0.16 "H <sub>2</sub> O Average Orifice P	37.88
Run 3 = 0.147 "H <sub>2</sub> O Average Orifice P	36.57
Average Molecular Weight = 28.13	
Overall Density = $\frac{28.13 \text{ gm}}{22.4 \text{ liters}} \times \frac{492^{\circ} \text{R}}{628^{\circ} \text{R}} \times \frac{1\text{b}}{453.6 \text{ gm}}$	x <u>28.32 liters</u> ft <sup>3</sup>
= 0.062 lb/SCF	
Specific Volume = 1/density = 16.18 ft <sup>3</sup> /lb	
TOTAL FLOW	
Degasifier Flow $(35.70 \text{ lb/Hr}) (16.18 \text{ ft}^3/\text{lb}) = 578.0 \text{ CFH}$	
Propane Assist Flow 154.0 CFH	
Total Flow 732.0 CFH	
Flare Velocity = 732.0 CFH/0.0222 ft <sup>2</sup> flare (2.0 = 32973.0 FPH = 550 FPM = 9.2 FPS	16 inches ID)
BTU Content = <u>154 (2440 BTU/CF Propane) + 578 (0.0</u> 732 Total CF	
$H_2S$ Concentration = $(H_2S Dry) (FDA) = (0.0217) (0.$	60) $\times$ 100 = 1.3%

 $SO_2$  Emission = (7.5 CFH H<sub>2</sub>S) (0.087 lb/CF)  $\frac{64.064}{34.064}$  = 1.23 lb/Hr

APPENDIX E

VOLUMETRIC FLOW AND VISIBLE EMISSION DATA

I 1/21:	SMOKE EMISSION INSPECTION TDOOR LOCATION
COMPANY HARRIS SEMICON QUETOR LOCATION PALM BAY R. P. PALM BAY FIA COMPANY REPRESENTATIVE	OBSERVER Sid CARTER AFFILIATION AIR CONSULTING TENGINEERING T. DATE 6-16-89
sky conditions <u>Clear</u> precipitation	wind direction <u>NE</u> wind speed <u>7</u>
INDUSTRY <u>SEMICON LIVETORS</u>	PROCESS UNIT
Sketch process unit; Indicate observer position relative points and/or actual emission points.  Emission Points  TROPANE  PROPANE	e to source and sun; indicate potential emission
	CLOCK OBSERVATION ACCUMULATED TIME PERIOD EMISSION TIME (min:sec)  THER SHEETS 120

1822

END OBSERVATION

#### **BEST AVAILABLE COPY**

al ir Consulting VISI	OBSERVATION DATE START TIME STOP TIME											
ngineering OBSERVATION FORM					on dat 6 – 8			T TIME 740	Ξ	STOP TIME		
SOURCE NAME HARRIS CORP				0	15	30	45	SEC	0	15	30	45
ADDRESS				0	0	0	0	MIN 31	0	R)	0	0
CITY STATE ZIP			2	0	0	O	0	32	7)	0	D	0
	STATE			Ð	$\mathcal{O}$ .	0	0	33	0	0	D	0
PHONE	SOURCE I.D NUMBE		3 4 **	Ð	£)	0	0	34	(A)	0	<u>)</u>	0
PROCESS EQUIPMENT	OPERATING M	ODE ·	75	Đ	Đ	0	0	35	0	0	0	0
		· 	MACE	0	0	O.	0	36	0	Ø	0	0
CONTROL EQUIPMENT	OPERATING M	ODE	7	0	0	Ð	0	37	0	0	0	Ð
DESCRIBE EMISSION POINT			8	0	D	D	0	38	$\mathcal{D}$	0	$\mathcal D$	O·
START HEIGHT ABOVE GROUND LEVE	STOP	DESERVED	9	ව	0	Ð	0	39	0	0	0	0
START STOP	START STO		10	0	0	D	0	40	0	7	Ŏ	Ŏ
DISTANCE FROM OBSERVER	DIRECTION FROM	*	11	0	0	<u>n</u>	0	41		D	2	0
START STOP	START STO	)P						-	0			_
DESCRIBE EMISSIONS			12	0	0	0	0	42	0	0	2	0
START	STOP	NIT C	13	0	0	0	0	43	0	0	0	0
EMISSION COLOR START STOP	FUGITIVE   IN	TER.	14	Ø	Ð	0	0	44	0	0	D	0
WATER DROPLETS PRESENT:			15		D	D	Ď	45	0	0	0	0
NO D YES D	ATTACHED DETA		16	D N		_			_		<del></del>	<del></del>
POINT IN PLUME AT WHICH OPACITY WAS DETERMINED				1/	0	0	0	46	0	0	0	0
START	STOP		17	0	0	0	0	47	0	0	0	0
DESCRIBE BACKGROUND			IB.	0	0	1	D	48	· D	0	0	0
START				0	0	0	D	49	0	0	ñ	0
BACKGROUND COLOR START STOP	SKY CONDITIONS START - STO	· 20	20	0	<u>n</u>	0	0	50		0	$\mathcal{D}$	+
WIND SPEED	START STO	20		<u> </u>		1		0	-	U	0	
START STOP	STÁRT STO		21	0	0	D	0	51	0	0	0	0
AMBIENT TEMP	WET BULB TEMP.	RH %	22	0	0	0	0	52	0	0	0	0
START STOP			23	9	0	0	0	53	0	0	0	0
SOURCE LAYOUT SKETCH	DRAW NORTH	ARROW	24	0	0	0	0	54	V	0	13	0
.*	\$ <sup>1</sup> 8,	(F)	25	0	0	0	n	55	0	12	0	0
	//		26	0	0	0	0	56	0	0	1	0
. "	X EMISSION POINT		27	8)	0	Ď.	0	57		0	0	0
sun - 💠			28	0	0	0	0	58	0	0	0	0
wind - —>			29	0	0	0	0	59	0	8	0	0
PLUME & STACK-			30	$\Omega$	0	0	0	60	0	0	0	0
	OBSERVER'S PO	SITION	AVERAGE OPACITY FOR NUMBER OF READINGS ABOVE									
	40°		RANGE OF OPACITY READINGS									
SUN LOCA	TION LINE	<u> </u>	MINIMUM O MAXIMUM O  OBSERVER'S NAME (PRINT)									
COMMENTS	•			.56	S \$16	· ( '×	12 ls	21	,	DATE		
Unit WENT LOWN	AT 1445	<u> </u>	<u> </u>	50/2		04/	15. 12.		•	G	16	89
RESTARTED At 17:	2.2			NIŽA.								
I HAVE RECEIVED A COPY OF		ERVATIONS		(IFIE)						DATE	_	
TITLE	DATE		VER	FIED	BY	,				DATE		<del></del>
				·								

STATE OF ELORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION
THIS IS TO CERTIFY THAT

STATE OF FLORIDA visible emissions evaluation training and is a qualified observer of visible emissions as specified by EPA reference method 9.

THIS CERTIFICATE EXPIRES

BEARER'S SIGNATURE

# VISIBLE EMISSIONS EVALUATOR

This is to certify that,

Ned G Carter

met the specifications of Federal Reference Method 9 and qualified as a visible emissions evaluator. Maximum deviation on white and black smoke did not exceed 7.5% opacity and no single error exceeding 15% opacity was incurred during the certification test conducted by Eastern Technical Associates of Raleigh, North Carolina. This certificate is valid for six months from date of issue.

Vice PAsident

Certificate Numb

Certificate Numb

Location

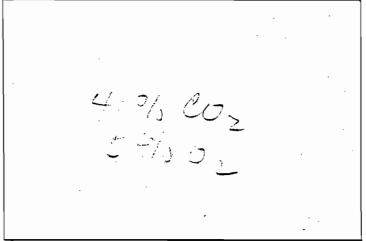
Control 8, 1989

Date I Issue

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Consulling								
Ingineering HARP	15							
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PORT DIAM.  PITOT TUBE NO. ORIF	CE TYPE.	E LENGTH						
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	LOCITY	STACK	DISTANCE	TRAV	ERSE	VELOCITY	STACK	

TRAVERSE VELOCITY POINT HEAD TE NUMBER ( $\Delta P_s$ ) in H <sub>2</sub> 0	STACK EMPERATURE (T <sub>s</sub> ,°F)	DISTANCE FROM INSIDE 7 STACK WALL	TRAVERSE S POINT	VELOCITY HEAD	STACK TEMPERATURE	
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1340 .115	176		1210	1205	146	/
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	ici ir —	LIMINARY V	ELOCITY TRA	<u>VERSE</u>			
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	TRAVERSE	VELOCITY	STACK	DISTANO	E	TRAVERSE	
	POINT	HEAD	TEMPERATURE	FROM INS	SIDE	POINT	
	NUMBER	$(\Delta P_s)$ in $H_2O$	(Tg,°F)	STACK W	ALL '	NUMBER	ľ
	2140	,27	1731				
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SCHEMATIC OF TRAVERSE POINT LAYOUT

TRAVERSE	VELOCITY	STACK	DISTANCE	TRAVERSE	VELOCITY	STACK
POINT	HEAD	TEMPERATURE	FROM INSIDE	POINT	HEAD	TEMPERATURE
NUMBER	(ΔP <sub>s</sub> ) in. H <sub>2</sub> O	(Tg ,°F)	STACK WALL	NUMBER	(ΔP <sub>s</sub> in. H <sub>2</sub> 0	(T <sub>s</sub> ,°F)
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2145	· KT	172	165.3	83		
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AVERAGE				AVERAGE		

APPENDIX F
FLARE SPECIFICATIONS

UTILITY STYLE FLARE BURNER

DATA SHEET

1. CUSTOMER: REFERENCE NO .: 2. PLANT LOCATION: PALM BAY FIA PROPOSAL NO.: ENGINEER: WS MODEL: EEF-U- & U DATE: ( -16 OVERALL LENGTH 10'-1" NO. OF PILOTS 5. FIRING POSITION: VERTICAL TYPE OF PILOTS: EEP-20 FUEL Ca DESIGN CONDITION FUEL CONSUMPTION 22 SCEH @ 60 C FLOW RATE: SEE Below MV: SP. GR. THERMOCOUPLE: VES No. Required 8. PRESS. DROP @ DESIGN: K (Chromel-Alumel) 10. MATERIAL OF CONSTRUCTION 11. SECTION MATERIAL Flame Retention Ring 310 or equal Upper Sect. FRR to 4'-0" 3!6ss Lower Sect. 4'-0" to 10'-1" c. stl. Flange A-105 Pilot(s) 309 SS WELDING PROCEDURES: 18. 19. 20. 21. FINISH: High temperature aluminum on carbon steel 22. NOZZLE NO. SIZE IN. TYPE 2 ANSI 150# RF Inlet N1 Pilot Plain End 3/4 25. C4 Plain End Ignitor C5 1 27. 28. 'H' PATTERN TID 31. 127 DAY 1600 4 DAL 34. 16002/21 666 CFH \* Assist GAS 37. \_510 FP:N 40. 42. 43. 44. 45. 46. 47. 49. 50.

52.

2440 BUYE

3	CUSTOMER:				REFERENCE NO.:
	PLANT LOCATION: PALM	BAY FIA	ENGINEER:	<del></del>	PROPOSAL NO.:
_	OVERALL HEIGHT: 20'0" (1				DATE:
	EARTHQUAKE ZONE: 2	WIND VELOCITY:	IOMPH	STRUCTURAL COOE:	
	STRUCTURAL TYPE: SELF				
	DESIGN PRESSURE:	OPERATING PRESS	TRE:		
	DESIGN TEMPERATURE:	OPEPATING TEMPER		-	CORROSION ALLOVANCE:
8.				· ·	
9.	MATERIALS OF CONSTRUCTION:				
10.	FABRICATION: AISC	WELD PROCEDURE:	AWS	CODE STAMP:	NA
11.	IMPROTEST: NA	RADIOGRAPH: /	IA		AT TREAT: NA
12.	OTHER:				
<u>13.</u>	LADOFRS:	·	PLATFO	RMS:	
14.	AIRCRAFT WARNING LIGHTS:	<u> </u>			
<u> 15.</u>	SURFACE PREPARATION: 5P	-6	·		<u> </u>
	PAINT: CZ-11	·····			·
<u>17.</u>					·
	NOZZLE SIZE IN. NO.	MATERIAL TYPE CONNEC	CTION		
19.		ISO#RF ANSI	<del></del>		•
		ISOURF ANSI		· <b>=</b>	ELIIO-C
	Inlet 2"	120 K.F ANJL	<del> </del>	γ:	}
23.	Crain		<del></del>		
24.				1	
	Inlet Elevation:	·	·	1:	A-255-C
26.	THIEC LIEVACION:				1
	REMARKS:	· · · · · · · · · · · · · · · · · · ·		: این د	6.00 x 257 A51M
28.			<del></del>		A-536 WELDED
29.			<del></del>	1 0 81.	
30.					
	NOTE: ALL DIMENSIONS ARE PR	EI.IMINARY			
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33.				1'- 0"	E Charles
34.	·	·			
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36.					July cess
37.					John Mr. A.
38.	<u> </u>	· .			w \ \all
<u> 39.</u>					1
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41.					
42.					/ KMINE /
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46.				/	<b></b>
47.			<u> </u>	• • •	O DE
48.			٢/	Ropane	:. <b>k</b>
49.			<del></del>	455157	i .
50.	<del></del>	· · · · · · · · · · · · · · · · · · ·			
51.			<del></del>	• ;	•

#### AUTOMATIC-MANUAL FLAME FRONT GENERATOR

DATA SHEET

1.	CUSTOMER: UNIT	<u>-ed  </u>	ENGR				<u></u>	REFERE	NCE NO.:		
2.	PLANT LOCATION: P	ALN	N BAY					PROPOS	SAL NO.:		
3.	DESCRIPTION: A/M	<u>. w</u>	11th Plot	Reic	W. ENGINEER	R: WS		DATE:			
4.	VALVES TYPE	2	SIZE	MATER	TAL		SOLENOID VALVES		SIZE	TAM.	ERIAL
5.	GAS: NEET	TE	1/2 Inch	Carbon	Steel		Gas		1/2 Inch	E	Brass
6.	AIR NEEL	ЛE	3/4 Inch	Carbon	Steel		Air		3/4 Inch	E	Brass
7.	IGNITION LINES: .		<u>.                                      </u>								
8.	PRESSURE GAUGES:	Press	sure Range		Connection		PRESSURE	REGULAT	ORS: Fis	sher 621	3/4 incir
9.	GAS	0-30	Psig		k Inch MNPT	·	Gas:	3/4	Inch Inlet-	-Outlet	
10.	AIR	0-30	Psig		k Inch MNPT		Air:	3/4	Inch Inlet-	Outlet	
11.								Max.	Inlet Pre	essure 20	00 Psig
12.	ELECTRICAL TRANSFOR	MER:	120 VAC	/50 H	12/10						
13.	HOUSING CLASSIFICAT	MICN:	NEMA (	4							
14.	SURFACE PREPARATION	√: SP-1	. Handelean, 7	ink Chr	omate Epoxy	Primer, P	olyurethane Enamel	Finish	Coat		
15.										<u> </u>	
16.	UTLITIES: Pressure		SP.GR	Flow !	Rate		Connections	No.	Size		Туре
17.	Gas 15.0 Psig	<u> .                                    </u>	1.5	150	SCEH		Transformer	1	3/4 Inct	<u>a</u>	FNPT
18.	Air 15.0 Psig	<b>3</b> •	1.0	1500	SCEH		Ignition Gas	1	1/2 Inch	nn	FNPT
19.			<u> </u>		,		Air	1	3/4 Inch	n	FNPT
20.	Electrical 3.0 Am	nps					Ignition Lines		1 Inch	<u>n</u>	FNPT
21.							Pilot Gas		3/4 Inct	<u>a</u>	FNPT
22.	Temp. Switches: Ac	tionpa	ick Type K Cal	ibratic	n		Thermocouple	1	3/4 Inc	Ω	FNPT

23.24.25.26.27.

28. 29.

30. 31. 32. 33.

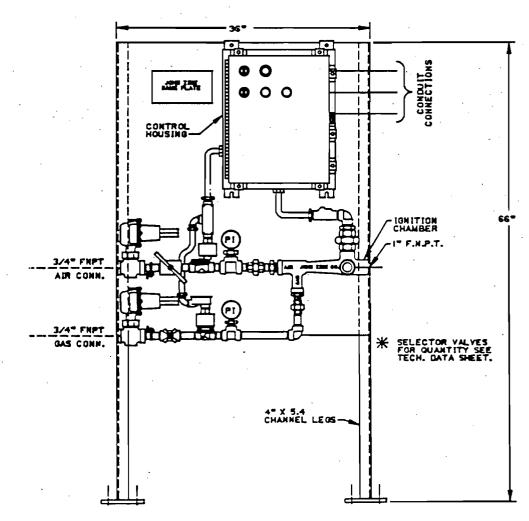
35. 36.

37.

38. 39. 40.

41. 42. 43.

45. 46. 47. 48. 49. 50. 51.



APPENDIX G
PROJECT PARTICIPANTS

#### PROJECT PARTICIPANTS

	~	
4		м
~	·	_

Stephen L. Neck, P.E.

Sidney J. Carter

J. Colleen Hodge

Karie L. Philman

TECHNICAL SERVICES, INC.

Harvey Gray

HARRIS CORPORATION

Nancy Baldisserotto

**FDER** 

Pius Sanabani

Field Testing Report Preparation

**VE** Testing

Report Graphics

Document Production

EPA Method 15 Testing EPA Method 15 Reporting

Project Coordinator

Test Observer

### ₩ P. 274 007 535 RECEIPT FOR CERTIFIED MAIL NO INSURANCE COVERAGE PROVIDED NOT FOR INTERNATIONAL MAIL (See Reverse) James R. Kolanek Harris-Gorp.-Semiconductor-Piret and No. 883 P.O., State and ZIP Code Melbourne, FL Postage Certified Fee Special Delivery Fee Restricted Delivery Fee Return Receipt showing to whom and Date Delivered Return Receipt showing to whom. Date, and Address of Delivery TOTAL Postage and Fees Postmark or Date AC 05-138795 (amendment) mailed: 12/15/88

SENDER: Complete items 1 and 2 when additional 3 and 4.  Put your address in the "RETURN TO" Space on the rever card from being returned to you. The return receipt fee will p to and the date of delivery. For additional fees the following for fees and check box(es) for additional service(s) request 1.   Show to whom delivered, date, and addressee's ad (Extra charge)	se side. Failure to do this will prevent this rovide you the name of the person delivered services are available. Consult postmaster ted.
3. Article Addressed to: James P. Kolanek Harris CorpSemiconductor Sector P.O. Box 883 Melbourne, FL 32901	4. Article Number  P 274 007 535  Type of Service:  Registered Insured COD Express Mail Return Receipt for Merchandise  Always obtain signature of addressee
5. Signature - Address  X  6. Signature - Agent Coniscomic Military  7. Date of Delivery	or agent and DATE DELIVERED.  8. Addressee's Address (ONLY if requested and fee paid)



## Florida Department of Environmental Regulation

Twin Towers Office Bldg. ● 2600 Blair Stone Road ● Tallahassee, Florida 32399-2400 Bob Martinez, Governor

Dale Twachtmann, Secretary

John Shearer, Assistant Secretary

December 13, 1988

CERTIFIED MAIL - RETURN RECEIPT REQUESTED

Mr. James R. Kolanek, Manager Environmental Services Harris Semiconductor Post Office Box 883 Melbourne, Florida 32901

Dear Mr. Kolanek:

Amendment to Air Construction Permit No. AC 05-138795 Re:

The Department is in receipt of your letter dated November 16, 1988, which requested an extension of the expiration date for the above referenced permit. The following shall be changed and added:

#### Expiration Date:

From: January 31, 1989 October 1, 1989 To:

#### Attachment to be Incorporated:

6. Mr. James R. Kolanek's letter dated November 16, 1988, and received November 28, 1988.

This letter must be attached to the construction permit, No. AC 05-138795, and shall become a part of the permit.

Singérely,

Dale Twachtmann

Secretary

DT/ks

C. Collins, CFD

B. Hewitt, Esq., DER

C. Bach, P.E.



## State of Florida DEPARTMENT OF ENVIRONMENTAL REGULATION

	For Routing To Other Than The Addressee
То:	Location:
То:	Location:
То:	Location:
From:	Oate:

## Interoffice Memorandum<sub>RECEIVED</sub>

**DEC 1 4 1988** 

DER - BAQM

TO: Dale Twachtmann

SUBJ:

FROM:

Approval of an Amendment to the Construction Permit

No. AC 05-138795 Harris Corporation

Steve Smallwood

DATE: December 13, 1988

Attached for your approval and signature is an amendment prepared by Central Air Permitting for the above mentioned company to extend the expiration date of the construction permit, No. AC 05-138795.

I recommend your approval and signature.

SS/BM/s

attachment



November 16, 1988

Mr. Claire Fancy Deputy Chief Bureau of Air Quality Management Florida Department of Environmental Regulation Twin Towers Office Building 2600 Blair Stone Road Tallahassee, FL 32399-2400

Extension of Construction Permit No. AC 05-138795 Subject:

Dear Mr. Fancy:

In accordance with FAC Rule 17-4.09 and Specific Condition No. 11, the purpose of this letter is to request an extension of construction permit No. AC 05-138795 until October 1, 1989. date will grant our facility adequate time to complete the delayed installation of the Industrial Grade water system with a vacuum degasifier and flare system, and to perform the compliance testing necessary to accompany the operating permit application.

file copy

Enclosed is a copy of the updated schedule for the Industrial Water Plant project.

Please feel free to phone me at (407) 724-7467 if you have any questions.

グ.R. Kolanek, Manager Environmental Services

/nab

Enclosures

NOV 28 1988

DER-BAQM



J. R. Kolanek, MS59-006

SEMICONDUCTOR SECTOR P.O. BOX 883 MELBOURNE, FLORIDA 32901



Mr. Claire Fancy
Deputy Chief
Bureau of Air Quality Management
Florida Department of Environmental Regulation
Twin Towers Office Building
2600 Blair Stone Road
Tallahassee, Florida 32399-2400

- Inflationthibidathalidaithallachaid

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RACK STRU. STEEL	1	+	+	1		<del>-  </del>	<b>⊣</b>		<del> </del>	_						_		1	(X)	(XX)	XXX	(XXX	XXX						1		1	7		1		1	T	1		T	T	1		T -	T	Т 1-	1
TXFRM & MCC DELIVERY	<del>   </del>	<del> </del>		7	<del>-</del> †	<b>⊣</b>	_		1	1		<del> </del>		_		<del> </del>	XX	XXX	XXX	(XX)	XX)	(XXX	XXX									<u> </u>	1	<u> </u>		1	T	<del>-</del>	T	1	1			Τ.	1	T	T
ELECTRICAL INSTL.	1	1	<del>-</del> †	1	_	_			<del> </del>	1	_	<del></del>		<del> </del>			Ι.	T	<del> </del>	<del> </del>	†	+		Ι.	<del> </del>											<del>-</del>	<del> </del>	1								Τ	1
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#### The Times

Published Weekly on Wednesday

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Notary Public, State of Florida at Large My Commission Expires July 20, 1990

#### THE TRIBUNE

Published Weekly on Wednesday



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STATE OF FLORIDA **COUNTY OF BREVARD** 

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STAR-ADVOCATE

Published Weekly on Wednesday

State of Florida
Department of Environmental
Regulation
Notice of Intent
The Department of Environmental Regulation
Herby gives
notice of its intent to Issue a permit to Harris Semiconductor to
Install/construct an Industrial
Grade Water System to provide
water for the Delonized Water
Plants in Buildings 22 and 39. The
system will include a vacuum degasifier to remove hydrogen sulfide and carbon dioxide from the
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manual/weatherproof bilot ignition panel. The pilot and enrichment fuet will be propane. The
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facility located in Palm Bay, Brevard County, Fiorida.
Persons-whose substantial Interests are affected by the Department's proposed permitting
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(hearing) in accordance with
Section 120.57, Florida Statutes.
The petition must conform to the
requirements of Chapters 17-103
and 28-5, Florida Administrative
Code, and must-be-filed (received) in the Department's Office of General Counsel, 2600
Blair Stone Road, Twin Towers
Office Building, Tailahassee,
Florida 23399-2400, within fourteen (14) days of publication of
this notice, Failure to file a petition within this time period constilutes a waiver of any right such
person has to request an administrative determination (hearing)
under Section 120.57, Florida
Statutes.

If a bettition is filed, the administrative hearing process is designed to formulate agency action. Accordingly, the
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who may not wish to file a petition
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Florida 23301: the
hearing officer has been assigned, the petition is to be filed
with the Depart

management 2600 Blair Stone Road Tallahassee, Florida 32399-2400

Dept. of Environmental Regulation Central Florida District 319 Maguire Blvd., Suite 232 Orlando, Florida 32803-3767 Any person may send written comments on the proposed action to Mr. Bill Thomas at the Department's Talahassee address. All comments mailed within 14-days of the publication of this notice will be considered in the Department's final determination. T084595—11—176, 1988. Wednesday





FLORIDA TODAY/USA TODAY GANNETT PLAZA P.O. BOX 363000 MELBOURNE, FL 32936







STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION
TWIN TOWERS OFFICE BUILDING
2600 BLAIR STONE ROAD
TALLAHASSEE, FLORIDA 32301-8241

Attn: Mr Bill Thomas
Bureau of Air Quality Management



PS Form 3511, July 1983 447-845	SENDER: Complete item Put your address in the "RET reverse side. Failure to do this being returned to you. The re	URN TO" space on the s will prevent this card from turn receipt fee will provide
11, July 1	you the name of the person delivery. For additional fees available. Consult postmaster for service(s) requested.	he following services are
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RECEIPT FOR CERTIFIED MAIL

NO INSURANCE COVERAGE PROVIDED

NOT FOR INTERNATIONAL MAIL

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# State of Florida Department of Environmental Regulation Notice of Intent

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## State of Florida Department of Environmental Regulation Notice of Intent

The Department of Environmental Regulation hereby gives notice of its intent to issue a permit to Harris Semiconductor to install/construct an Industrial Grade Water System to provide water for the Deionized Water Plants in Buildings 52 and 59. The system will include a vacuum degasifier to remove hydrogen sulfide and carbon dioxide from the raw well water. The removed gases will be oxidized in a flare, which will be designed and built by the John Zink Company, Model EEF-U-2 Flare Tip with a manual/weatherproof pilot ignition panel. The pilot and enrichment fuel will be propane. The construction/installation will take place at the permittee's existing facility located in Palm Bay, Brevard County, Florida.

Persons whose substantial interests are affected by the Department's proposed permitting decision may petition for an administrative determination (hearing) in accordance with Section 120.57, Florida Statutes. The petition must conform to the requirements of Chapters 17-103 and 28-5, Florida Administrative Code, and must be filed (received) in the Department's Office of General Counsel, 2600 Blair Stone Road, Twin Towers Office Building, Tallahassee, Florida 32399-2400, within fourteen (14) days of publication of this notice. Failure to file a petition within this time period constitutes a waiver of any right such person has to request an administrative determination (hearing) under Section 120.57, Florida Statutes.

If a petition is filed, the administrative hearing process is designed to formulate agency action. Accordingly, the Department's final action may be different from the proposed agency action. Therefore, persons who may not wish to file a petition may wish to intervene in the proceeding. A petition for intervention must be filed pursuant to Rule 28-5.207, Florida Administrative Code, at least five (5) days before the final hearing and be filed with the hearing officer if one has been assigned at the Division of Administrative Hearings, Department of Administration, 2009, Apalachee Parkway, Tallahassee, Florida If no hearing officer has been assigned, the petition is to be filed with the Department's Office of General Counsel, 2600 Blair Stone Road, Tallahassee, Florida 32399-2400. Failure to petition to intervene within the allowed time frame constitutes a waiver of any right such person has to request a hearing under Section 120.57, Florida Statutes.

The application is available for public inspection during normal business hours, 8:00 a.m. to 5:00 p.m., Monday through Friday, except legal holidays, at:

340

Dept. of Environmental Regulation Bureau of Air Quality Management 2600 Blair Stone Road Tallahassee, Florida 32399-2400

Dept. of Environmental Regulation Central Florida District 3319 Maguire Blvd., Suite 232 Orlando, Florida 32803-3767

Any person may send written comments on the proposed action to Mr. Bill Thomas at the Department's Tallahassee address. All comments mailed within 14 days of the publication of this notice will be considered in the Department's final determination.

PS Form	SENDER: Complete items 1, 2, 3 and 4.  Put your address in the "RETURN TO" space on the reverse side. Failure to do this will prevent this card from												
PS Form 3611, July 1983 447-845	being returned to you. The return receipt fee will provide you the name of the person delivered to and the date of delivery. For additional fees the following services are available. Consult postmaster for fees and check box(es) for service(s) requested.												
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447-8	2. Restricted Delivery.												
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	Environmental Services												
	Harris Semiconductor												
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	Melbourne, FL 32901												
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RECEIPT FOR CERTIFIED MAIL

NO INSURANCE COVERAGE PROVIDED

NOT FOR INTERNATIONAL MAIL

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#### STATE OF FLORIDA

#### DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

December 4, 1987

CERTIFIED MAIL-RETURN RECEIPT REQUESTED

Mr. James R. Kolanek, Manager Environmental Services Harris Semiconductor Post Office Box 883 Melbourne, Florida 32901

Dear Mr. Kolanek:

Attached is one copy of the Technical Evaluation and Preliminary Determination and proposed permit for Harris Semiconductor to install/construct and Industrial Grade Water System (IGWS) to provide water for the Deionized Water Plants in Buildings 52 and 59. The system will include a vacuum degasifier to remove hydrogen sulfide (H<sub>2</sub>S) and carbon dioxide from the raw well water. The removed gases will be oxidized in flare, which will be designed and built by the John Zink Company, Model EEF-U-2 Flare Tip with manual weatherproof pilot ignition panel. The pilot and enrichment fuel will be propane.

Please submit, in writing, any comments which you wish to have considered concerning the Department's proposed action to Mr. Bill Thomas of the Bureau of Air Quality Management.

Sincerely,

C. H. Fancy, P.E.

Deputy Chief

Bureau of Air Quality

Management

CHF/bm

Attachments

cc: T. Sawicki, CF Dist.

C. Bach, P.E.

## State of Florida Department of Environmental Regulation Notice of Intent

The Department of Environmental Regulation hereby gives notice of its intent to issue a permit to Harris Semiconductor to install/construct an Industrial Grade Water System (IGWS) to provide water for the Deionized Water Plants in Building 52 and 59. The system will include a vacuum degasifier to remove hydrogen sulfide (H2S) and carbon dioxide from the raw well water. The removed gases will be oxidized in a flare, which will be designed and built by the John Zink Company, Model EEF-U-2 Flare Tip with manual/weatherproof pilot ignition panel. The pilot and enrichment fuel will be propane. The Department is issuing this Intent to Issue for the reasons stated in the attached Technical Evaluation and Preliminary Determination.

Persons whose substantial interests are affected by the Department's proposed permitting decision may petition for an administrative determination (hearing) in accordance with Section 120.57, Florida Statutes. The petition must conform to the requirements of Chapters 17-103 and 28-5, Florida Administrative Code, and must be filed (received) in the Department's Office of General Counsel, 2600 Blair Stone Road, Twin Towers Office Building, Tallahassee, Florida 32399-2400, within fourteen (14) days of publication of this notice. Failure to file a petition within this time period constitutes a waiver of any right such person has to request an administrative determination (hearing) under Section 120.57, Florida Statutes.

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Dept. of Environmental Regulation Central Florida District 3319 Maguire Blvd., Suite 232 Orlando, Florida 32803-3767

Any person may send written comments on the proposed action to Mr. Bill Thomas at the Department's Tallahassee address. All comments mailed within 14 days of the publication of this notice will be considered in the Department's final determination.

# RULES OF THE ADMINISTRATIVE COMMISSION MODEL RULES OF PROCEDURE CHAPTER 28-5 DECISIONS DETERMINING SUBSTANTIAL INTERESTS

#### 28-5.15 Requests for Formal and Informal Proceedings

- (1) Requests for proceedings shall be made by petition to the agency involved. Each petition shall be printed, typewritten or otherwise duplicated in legible form on white paper of standard legal size. Unless printed, the impression shall be on one side of the paper only and lines shall be double spaced and indented.
- (2) All petitions filed under these rules should contain:
  - (a) The name and address of each agency affected and each agency's file or identification number, if known;
  - (b) The name and address of the petitioner or petitioners;
  - (c) All disputed issues of material fact. If there are none, the petition must so indicate;
  - (d) A concise statement of the ultimate facts alleged, and the rules, regulations and constitutional provisions which entitle the petitioner to relief;
  - (e) A statement summarizing any informal action taken to resolve the issues, and the results of that action;
  - (f) A demand for the relief to which the petitioner deems himself entitled; and
  - (g) Such other information which the petitioner contends is material.

## BEFORE THE STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

In the Matter of Application for Permit by:

Harris Semiconductor Palm Bay Road Palm Bay, Florida 32901 DER File No. AC 05-138795

#### INTENT TO ISSUE

The Department of Environmental Regulation hereby gives notice of its intent to issue a permit (copy attached) to Harris Semiconductor to install/construct an Industrial Grade Water System (IGWS) to provide water for the Deionized Water Plants in Buildings 52 and 59. The system will include a vacuum degasifier to remove hydrogen sulfide (H2S) and carbon dioxide from the raw well water. The removed gases will be oxidized in a flare, which will be designed and built by the John Zink Company, Model EEF-U-2 Flare Tip with manual/weatherproof pilot ignition panel. The pilot and enrichment fuel will be propane. The proposed installation/ construction will occur at the applicant's existing facility located on Palm Bay Road, in Palm Bay, Brevard County, The Department is issuing this Intent to Issue for the Florida. reasons stated in the attached Technical Evaluation and Preliminary Determination.

The applicant, Harris Semiconductor, applied on August 3, 1987, to the Department of Environmental Regulation for a construction permit.

The Department has permitting jurisdiction under Chapter 403, Florida Statutes (F.S.), and Florida Administrative Code (FAC) Rules 17-2 and 17-4. The project is not exempt from permitting procedures. The Department has determined that an air construction permit was needed for the proposed work.

Pursuant to Section 403.815, F.S., and FAC Rule 17-103.150, you (the applicant) are required to publish at your own expense the enclosed Notice of Proposed Agency Action on permit application. The notice must be published one time only in a

section of a major local newspaper of general circulation in the county in which the project is located and within thirty (30) days from receipt of this intent. Proof of publication must be provided to the Department within seven days of publication of the notice. Failure to publish the notice and provide proof of publication within the allotted time may result in the denial of the permit.

The Department will issue the permit with the attached conditions unless petition for an administrative proceeding (hearing) is filed pursuant to the provisions of Section 120.57, F.S. A person whose substantial interests are affected by the Department's proposed permitting decision may petition for an administrative proceeding (hearing) in accordance with Section 120.57, F.S. Petitions must comply with the requirement of FAC Rules 17-103.155 and 28-5.201 (copies enclosed) and be filed with (received by) the Office of General Counsel of the Department at 2600 Blair Stone Road, Tallahassee, Florida 32399-2400. Petitions filed by the permit applicant must be filed within fourteen (14) days of receipt of this intent. Petitions filed by other persons must be filed within fourteen (14) days of publication of the public notice or within fourteen (14) days of receipt of this intent, whichever first occurs. Failure to file a petition within this time period shall constitute a waiver of any right such person may have to request an administrative determination (hearing) under Section 120.57, F.S., concerning the subject permit application. Petitions which are not filed in accordance with the above provisions will be dismissed.

Executed in Tallahassee, Florida.

STATE OF FLORIDA DEPARTMENT OF ENVIRONMENTAL REGULATION

C. H. Fancy, P.E.

Deputy Chief

Bureau of Air Quality

Management

Copies furnished to:

T. Sawiski, CF Dist.

C. Baen, P.E.

#### CERTIFICATE OF SERVICE

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FILING AND ACKNOWLEDGEMENT FILED, on this date, pursuant to \$120.52(9), Florida Statutes, with the designated Department Clerk, receipt of which is hereby acknowledged.

Martha Wise

12-7-87 Date Technical Evaluation and Preliminary Determination

Harris Corporation Semiconductor Sector Brevard County Melbourne, Florida

Permit Number: AC 05-138795

Florida Department of Environmental Regulation Bureau of Air Quality Management Central Air Permitting

#### I. Project Description

#### A. Applicant

Harris Corporation Semiconductor Sector Post Office Box 883 Melbourne, Florida 32901

B. Project Description and Location

The applicant proposes to install/construct an Industrial Grade Water System (IGWS) to provide water for the Deionized Water Plants in Buildings 52 and 59. The system will include a vacuum degasifier to remove hydrogen sulfide (H2S) and carbon dioxide from the raw well water. The removed gases will be oxidized in a flare, which will be designed and built by the John Zink Company, Model EEF-U-2 Flare Tip with manual/weatherproof pilot ignition panel. The pilot and enrichment fuel will be propane.

The proposed installation/construction will occur at the applicant's existing facility located on Palm Bay Road in Palm Bay City, Brevard County, Florida. The UTM coordinates are Zone 17, 538.7 km East and 3100.9 km North.

The Standard Industrial Code is 3674, Semiconductors.

The Standard Classification Codes are:

o Miscellaneous Manufacturing Industries
Major Group 39; Flares - Natural Gas: 3-99-900-23 (106 cubic feet burned)

#### C. Controls

Due to the foul smelling of  $H_2S$  extracted with the IGWS, the control measure is to capture, transport, and oxidize the gas in a flare. The resultant pollutant emissions will be sulfur dioxide ( $SO_2$ ).

#### II. Rule Applicability

The applicant's intent is subject to preconstruction review pursuant to Florida Administrative Code (FAC) Rules 17-2 and 17-4, in accordance with Section 403, Florida Statutes.

The application package was deemed complete on October 23, 1987.

The proposed project and existing facility is located in an area designated attainment for all pollutants. Therefore, review of the potential pollutant emissions shall be in accordance with FAC Rule 17-2.500, Prevention of Significant Deterioration (PSD).

Based on the 1984 emissions inventory and subsequent modifications, the estimated facility's VOC and organic solvent emissions are greater than 100 tons per year (TPY), but less than 250 TPY. Therefore, the facility is a major facility in accordance with FAC Rule 17-2.100(111). The facility does not belong to any of the major facility categories listed in FAC Rule 17-2, Table 500-1.

The following Table 1 will exhibit the projected potential pollutant emissions for the proposed project:

#### Table 1

Source	Potential	Pollutant	Emissions
		H <sub>2</sub> S	so <sub>2</sub>
Flare System		0.22 TPY	27.61 TPY

Note: Based on 98.5% efficiency of the flare to oxidize H<sub>2</sub>S to SO<sub>2</sub>

Based on the table, the proposed increase in potential H<sub>2</sub>S and SO<sub>2</sub> emissions would be a minor modification to a major facility and exempt from new source review requirements pursuant to FAC Rule 17-2.500, PSD, and is the first permitting activity of these specific pollutants at this facility. Therefore, the potential pollutant emissions will be reviewed in accordance with FAC Rule 17-2.520, Sources Not Subject to PSD or Nonattainment Requirements.

Since there is no specific emission limiting standard contained in FAC Rule 17-2.600 nor is there any standards of performance for new stationary sources contained in FAC Rule 17-2.660 for the proposed project, the modification will be permitted in accordance with FAC Rules 17-2.250, Excess Emissions, and 17-2.620, General Pollutant Emission Limiting Standards. Also, the source will be subject to the conditions of 40 CFR 60.18(c) thru (f), General Control Device Requirements.

FAC Rule 17-2.620(2) states that no person shall cause, suffer, allow or permit the discharge of air pollutants which cause or contribute to an objectionable odor. Objectionable odor is defined as any odor present in the outdoor atmosphere, which by itself or in combination with other odors, is or may be harmful or injurious to human health or welfare, which unreasonably interferes with the comfortable use and enjoyment of

life or property, or which creates a nuisance according to FAC Rule 17-2.100(131).

In accordance with FAC Rule 17-2.250(4), sources are to be properly operated and maintained so that excess emissions are minimized. FAC Rule 17-2.250(6), requires that the Department be notified in the case of excess emissions, and the Department, in this case, is the DER's Central Florida District office.

Pursuant to 40 CFR 60.18, the flare will be subject to no visible emissions (5% opacity), except for a total period of 5 minutes during any 2 consecutive hours. Compliance tests shall be performed using EPA Methods 2, 2A, 2C or 2D, as appropriate, 15 and 22, pursuant to FAC Rule 17-2.700 and 40 CFR 60, Appendix A.

#### III. Summary of Emissions

#### A. Emission Limitations

The regulated pollutant emissions from this modification are visible emissions, which is "no visible emissions (5% opacity), except for periods not to exceed a total of 5 minutes during any 2 consecutive hours".

Table 2 reflects the projected maximum potential pollutant emissions from the proposed modification for the purpose of tracking for PSD Rule applicability pursuant to FAC Rule 17-2.500.

Table 2

Source	Maximum	Potential	Pollutant F	missions
			H <sub>2</sub> S	so <sub>2</sub>
Flare System			0.06 lbs/h (0.22 TPY)	7.06 lbs/hr (27.61 TPY)

<sup>\*</sup>Based on 8760 hours per year operations.

The permitted emissions are in compliance with all requirements of Chapter 17-2, FAC.

#### B. Air Quality Impacts

From a technical review of the application and supplementary material, an air quality analysis is not required.

#### IV. Conclusion

The emission limitations to be imposed have been determined to be in compliance with all applicable requirements of Chapter

17-2, FAC. The permitted allowable emissions should not cause any violations of Florida's ambient air quality standards.

The General and Specific Conditions listed in the proposed permit will assure compliance with all applicable air pollution regulations..

#### STATE OF FLORIDA

#### DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

PERMITTEE:
Harris Semiconductor
P. O. Box 883
Melbourne, Florida 32901

Permit Number: AC 05-138795 Expiration Date: January 31, 1989 County: Brevard

Latitude/Longitude: 28° 01' 20"N/ 80° 36' 10" W

Project: Industrial Grade Water
System with Vacuum
Degasifier and Flare
System

This permit is issued under the provisions of Chapter 403, Florida Statutes, and Florida Administrative Code Rule(s)  $\overline{17-2}$  and 17-4. The above named permittee is hereby authorized to perform the work or operate the facility shown on the application and approved drawing(s), plans, and other documents attached hereto or on file with the Department and made a part hereof and specifically described as follows:

For the construction/installation of a 600 gallon per minute Industrial Grade water system with a vacuum degasifier and flare system. The vacuum degasifier will remove hydrogen sulfide and carbon dioxide from raw well water and the removed gases will be transported to and oxidized/combusted by the flare system. The nonassisted type flare will be designed and built by the John Zink Company, which includes a self-supported flare stack, a Model EEF-U-2 flare tip (John Zink Co.), and a manual weatherproof pilot ignition panel. The construction/installation will occur at the permittee's existing facility located on Palm Bay Road, City of Palm Bay, Florida. The UTM coordinates are Zone 17, 538.7 km East and 3100.9 km North.

The source shall be in accordance with the permit application, plans, documents, amendments and drawings, except as otherwise noted in the Specific Conditions.

#### Attachments to be 'Incorporated:

- 1. Application to Construct Air Pollution Source, DER Form 17-1.202(1), and Mr. James R. Kolanek's cover letter dated August 28, 1987, and received August 31, 1987.
- 2. Mr. C. H. Fancy's letter dated September 25, 1987.
- Mr. J. R. Kolanek's letter with enclosures dated October 16, 1987, and received October 23, 1987.
- 4. Copy of 40 CFR 60.18, as revised July 1, 1986.
- 5. Technical Evaluation and Preliminary Determination dated December 4, 1987.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### **GENERAL CONDITIONS:**

- 1. The terms, conditions, requirements, limitations, and restrictions set forth herein are "Permit Conditions" and as such are binding upon the permittee and enforceable pursuant to the authority of Sections 403.161, 403.727, or 403.859 through 403.861, Florida Statutes. The permittee is hereby placed on notice that the Department will review this permit periodically and may initiate enforcement action for any violation of the "Permit Conditions" by the permittee, its agents, employees, servants or representatives.
- 2. This permit is valid only for the specific processes and operations applied for and indicated in the approved drawings or exhibits. Any unauthorized deviation from the approved drawings, exhibits, specifications, or conditions of this permit may constitute grounds for revocation and enforcement action by the Department.
- 3. As provided in Subsections 403.087(6) and 403.722(5), Florida Statutes, the issuance of this permit does not convey any vested rights or any exclusive privileges. Nor does it authorize any injury to public or private property or any invasion of personal rights, nor any infringement of federal, state or local laws or regulations. This permit does not constitute a waiver of or approval of any other Department permit that may be required for other aspects of the total project which are not addressed in the permit.
- 4. This permit conveys no title to land or water, does not constitute state recognition or acknowledgement of title, and does not constitute authority for the use of submerged lands unless herein provided and the necessary title or leasehold interests have been obtained from the state. Only the Trustees of the Internal Improvement Trust Fund may express state opinion as to title.
- 5. This permit does not relieve the permittee from liability for harm or injury to human health or welfare, animal, plant or aquatic life or property and penalties therefore caused by the construction or operation of this permitted source, nor does it allow the permittee to cause pollution in contravention of Florida Statutes and Department rules, unless specifically authorized by an order from the Department.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### GENERAL CONDITIONS:

- 6. The permittee shall at all times properly operate and maintain the facility and systems of treatment and control (and related appurtenances) that are installed or used by the permittee to achieve compliance with the conditions of this permit, as required by Department rules. This provision includes the operation of backup or auxiliary facilities or similar systems when necessary to achieve compliance with the conditions of the permit and when required by Department rules.
- 7. The permittee, by accepting this permit, specifically agrees to allow authorized Department personnel, upon presentation of credentials or other documents as may be required by law, access to the premises, at reasonable times, where the permitted activity is located or conducted for the purpose of:
  - a. Having access to and copying any records that must be kept under the conditions of the permit;
  - Inspecting the facility, equipment, practices, or operations regulated or required under this permit;
     and
  - c. Sampling or monitoring any substances or parameters at any location reasonably necessary to assure compliance with this permit or Department rules.

Reasonable time may depend on the nature of the concern being investigated.

- 8. If, for any reason, the permittee does not comply with or will be unable to comply with any condition or limitation specified in this permit, the permittee shall immediately notify and provide the Department with the following information:
  - a. a description of and cause of non-compliance; and
  - b. the period of noncompliance, including exact dates and times; or, if not corrected, the anticipated time the noncompliance is expected to continue, and steps being taken to reduce, eliminate, and prevent recurrence of the noncompliance.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### GENERAL CONDITIONS:

The permittee shall be responsible for any and all damages which may result and may be subject to enforcement action by the Department for penalties or revocation of this permit.

- 9. In accepting this permit, the permittee understands and agrees that all records, notes, monitoring data and other information relating to the construction or operation of this permitted source, which are submitted to the Department, may be used by the department as evidence in any enforcement case arising under the Florida Statutes or Department rules, except where such use is proscribed by Sections 403.73 and 403.111, Florida Statutes.
- 10. The permittee agrees to comply with changes in Department rules and Florida Statutes after a reasonable time for compliance, provided however, the permittee does not waive any other rights granted by Florida Statutes or Department rules.
- 11. This permit is transferable only upon department approval in accordance with Florida Administrative Code Rules 17-4.12 and 17-30.30, as applicable. The permittee shall be liable for any noncompliance of the permitted activity until the transfer is approved by the Department.
- 12. This permit is required to be kept at the work site of the permitted activity during the entire period of construction or operation.
- 13. This permit also constitutes:
  - ( ) Determination of Best Available Control Technology (BACT)
  - ( ) Determination of Prevention of Significant Deterioration (PSD)
  - ( ) Compliance with New Source Performance Standards
- 14. The permittee shall comply with the following monitoring and record keeping requirements:
  - a. Upon request, the permittee shall furnish all records and plans required under Department rules. The retention period for all records will be extended automatically, unless otherwise stipulated by the Department, during the course of any unresolved enforcement action.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### GENERAL CONDITIONS:

- b. The permittee shall retain at the facility or other location designated by this permit records of all monitoring information (including all calibration and maintenance records and all original strip chart recordings for continuous monitoring instrumentation), copies of all reports required by this permit, and records of all data used to complete the application for this permit. The time period of retention shall be at least three years from the date of the sample, measurement, report or application unless otherwise specified by Department rule.
- c. Records of monitoring information shall include:
  - the date, exact place, and time of sampling or measurements;
  - the person responsible for performing the sampling or measurements;
  - the date(s) analyses were performed;
  - the person responsible for performing the analyses;
  - the analytical techniques or methods used; and
  - the results of such analyses.
- 15. When requested by the Department, the permittee shall within a reasonable time furnish any information required by law which is needed to determine compliance with the permit. If the permittee becomes aware that relevant facts were not submitted or were incorrect in the permit application or in any report to the Department, such facts or information shall be submitted or corrected promptly.

#### SPECIFIC CONDITIONS:

- 1. Annual hours of operation are 8760.
- 2. The maximum potential sulfur dioxide ( $SO_2$ ) emissions are 7.0 pounds per hour and 30.7 tons per year.
- 3. The maximum potential hydrogen sulfide (H<sub>2</sub>S) emissions are 493 pounds per year, which is based on a flare efficiency of 98.5%.

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### SPECIFIC CONDITIONS:

- 4. The permittee shall comply with the conditions of 40 CFR 60.18(c) thru (f).
- 5. No visible emissions (5% opacity) shall be allowed, except for a total period of 5 minutes during any consecutive 2 hours, pursuant to 40 CFR 60.18(c). Compliance shall be demonstrated annually using EPA Method 22 pursuant to 40 CFR 60.18(f)(l).
- 6. EPA Method 15 shall be performed annually to determine the maximum concentration of the  $\rm H_2S$  prior to being flared and the result should be in terms of dry standard conditions (14.7 psia and 680 F). A retest shall be required if the concentration of  $\rm H_2S$  is to be increased.
- 7. The exit velocity of the flare shall be determined using the procedure in 40 CFR 60.18(f)(4) and either EPA Method 2, 2A, 2C or 2D, as appropriate.
- 8. EPA Methods shall be as described in 40 CFR 60, Appendix A.
- 9. The Central Florida District shall be notified in writing 15 days in advance of any compliance testing and the test reports shall be submitted within 45 days after the last test run.
- 10. Objectionable odors shall not be allowed off plant property pursuant to FAC Rule 17-2.620(2).
- 11. The construction shall reasonably conform to the plans and schedule submitted in the application. If the permittee is unable to complete construction on schedule, he must notify the Department in writing 60 days prior to the expiration date of the construction permit and submit a new schedule and request for an extension of the construction permit. (FAC Rule 17-4.09)
- To obtain a permit to operate, the permittee must demonstrate compliance with the conditions of the construction permit and submit a complete application for an operating permit, including the application fee, along with compliance test results and Certificate of Completion, to the Department's District office 90 days prior to the expiration date of the construction permit. The permittee may continue to operate in compliance with all terms of the construction permit until its expiration date. Operation beyond the construction permit expiration date requires a valid permit to operate. (FAC Rules 17-4.22 and 17-4.23)

Permit Number: AC 05-138795 Expiration Date: January 31, 1989

#### SPECIFIC CONDITIONS:

If the construction permit expires prior to the permittee requesting an extension or obtaining a permit to operate, then all activities at the project must cease and the permittee must apply for a new permit to construct which can take up to 90 days to process a complete application. (FAC Rule 17-4.10)

19			day or	
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Dale Twachtmann, Secretary

## § 60.18 General control device requirements.

- (a) Introduction. This section contains requirements for control devices used to comply with applicable subparts of Part 60 and Part 61. The requirements are placed here for administrative convenience and only apply to facilities covered by subparts referring to this section.
- (b) Flares. Paragraphs (c) through (f) apply to flares.
- (c)(1) Flares shall be designed for and operated with no visible emissions

as determined by the methods specified in paragraph (f), except for periods not to exceed a total of 5 minutes during any 2 consecutive hours.

(2) Flares shall be operated with a flame present at all times, as determined by the methods specified in paragraph (f).

(3) Flares shall be used only with the net heating value of the gas being combusted being 11.2 MJ/scm (300 Btu/scf) or greater if the flare is steam-assisted or air-assisted; or with the net heating value of the gas being combusted being 7.45 MJ/scm (200 Btu/scf) or greater if the flare is non-assisted. The net heating value of the gas being combusted shall be determined by the methods specified in paragraph (f).

(4)(i) Steam-assisted and nonassisted flares shall be designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), less than 18.3 m/sec (60 ft/sec), except as provided in paragraphs (b)(4) (ii) and (iii).

(ii) Steam-assisted and nonassisted flares designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), equal to or greater than 18.3 m/sec (60 ft/sec) but less than 122 m/sec (400 ft/sec) are allowed if the net heating value of the gas being combusted is greater than 37.3 MJ/scm (1,000 Btu/scf).

(iii) Steam-assisted and nonassisted flares designed for and operated with an exit velocity, as determined by the methods specified in paragraph (f)(4), less than the velocity,  $V_{max}$ , as determined by the method specified in paragraph (f)(5), and less than 122 m/sec (400 ft/sec) are allowed.

(5) Air-assisted flares shall be designed and operated with an exist velocity less than the velocity,  $V_{max}$ , as determined by the method specified in paragraph (f)(6).

(6) Flares used to comply with this section shall be steam-assisted, air-assisted, or nonassisted.

(d) Owners or operators of flares used to comply with the provisions of this subpart shall monitor these control devices to ensure that they are operated and maintained in conformance with their designs. Applicable subparts

#### **Environmental Protection Agency**

will provide provisions stating how owners or operators of flares shall monitor these control devices.

(e) Flares used to comply with provisions of this subpart shall be operated at all times when emissions may be vented to them.

(f)(1) Reference Method 22 shall be used to determine the compliance of flares with the visible emission provisions of this subpart. The observation period is 2 hours and shall be used according to Method 22.

(2) The presence of a flare pilot flame shall be monitored using a thermocouple or any other equivalent device to detect the presence of a flame.

(3) The net heating value of the gas being combusted in a flare shall be calculated using the following equation:

$$H_{T} = K \sum_{i=1}^{\infty} C_{i}H_{i}$$

where:

H<sub>T</sub>=Net heating value of the sample, MJ/ scm; where the net enthalpy per mole of offgas is based on combustion at 25 °C and 760 mm Hg, but the standard temperature for determining the volume corresponding to one mole is 20 °C;

$$\begin{array}{cccc} & = & \text{Constant,} \\ & 1.740 \times 10^{-7} & (\frac{1}{\text{ppm}}) & (\frac{g \text{ mole}}{\text{scm}}) & (\frac{\text{MJ}}{\text{kcal}}) \end{array}$$

where the standard temperature for  $(\frac{g \text{ mole}}{\text{scm}})$  is  $20^{\circ}\text{C}$ ;

C<sub>1</sub>=Concentration of sample component I in ppm on a wet basis, as measured for organics by Reference Method 18 and measured for hydrogen and carbon monoxide by ASTM D1946-77 (Incorporated by reference as specified in § 60.17); and

H<sub>1</sub>=Net heat of combustion of sample component i, keal/g mole at 25 °C and 760 mm Hg. The heats of combustion may be determined using ASTM D2382-76 (incorporated by reference as specified in § 60.17) if published values are not available or cannot be calculated.

(4) The actual exist velocity of a flare shall be determined by dividing the volumetric flowrate (in units of standard temperature and pressure), as determined by Reference Methods 2, 2A, 2C, or 2D as appropriate; by the unobstructed (free) cross sectional area of the flare tip.

(5) The maximum permitted velocity,  $V_{max}$ , for flares complying with paragraph (c)(4)(iii) shall be determined by the following equation.

$$Log_{10} (V_{max}) = (H_T + 28.8)/31.7$$

V<sub>max</sub> = Maximum permitted velocity, M/sec 28.8 = Constant 31.7 = Constant

 $H_T$ =The net heating value as determined in paragraph (f)(3).

(6) The maximum permitted velocity,  $V_{\text{max}}$ , for air-assisted flares shall be determined by the following equation.

$$V_{max} = 8.706 + 0.7084 (H_T)$$

 $V_{max} = Maximum$  permitted velocity, m/sec 8.706 = Constant

0.7084 = Constant

 $H_r$ =The net heating value as determined in paragraph (f)(3).

[51 FR 2701, Jan.21, 1986]



November 4, 1987

Mr. C. H. Fancy, P.E. Deputy Chief, Bureau of Air Quality Mgt. State of Florida, DER Twin Towers Office Building 2600 Blair Stone Road Tallahassee, Florida 32399

Dear Mr. Fancy:

Enclosed is a check for \$150.00 to cover the deficit amount of the construction permit application fee for Permit No. AC 05-138795.

Sincerely,

J.R. Kolanek, Manager Environmental Services

Enclosure

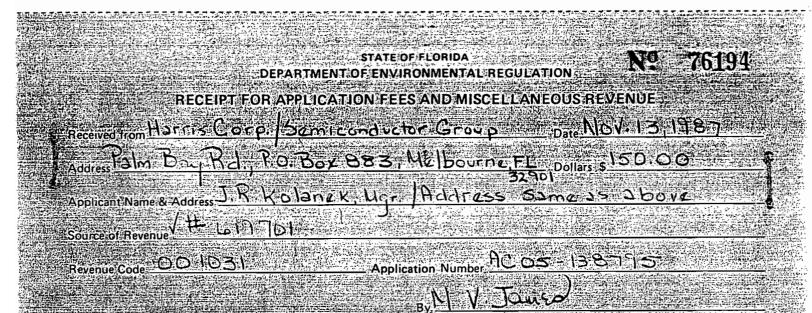
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FS-NAB-71-88

October 16, 1987

Mr. C. H. Fancy, P.E. Deputy Chief, Bureau of Air Quality Mgt. State of Florida, DER Twin Towers Office Building 2600 Blair Stone Road Tallahassee, Florida 32399

Re: Completeness Review of Construction Applications

Permit No.: AC 05-138794 AC 05-138795

Dear Mr. Fancy:

This letter is in reply to your request of September 25, 1987, for further information regarding permits AC 05-138794 and 138795. The following information is pertinent to Permit No. AC 05-138794:

- 1. The only piece of equipment containing organics that is proposed to be attached to the exhaust fan F54E17 is a gas cabinet containing two K-gas cylinders of halocarbon-23 (trifluoromethane). Under normal operating conditions, no emissions of this fluorocarbon will occur. In the unlikely event that an entire cylinder should be lost, a maximum of 70.12 lbs. of organic compounds would be released into the exhaust system. The rate of release would depend on the nature of the leak.
- 2. The facility designated identification numbers for the scrubber systems that will handle gas cylinder purges are F54SO3 and F54SO4.
- 3 & 4.

  The scrubber medium for F54SO3 and F54SO4 is water. After collection occurs, the water drains to the T112 sump where it is pumped to the wastewater treatment facility located on the Harris Semiconductor plant

The following information is pertinent to Permit No. AC 05-138795:

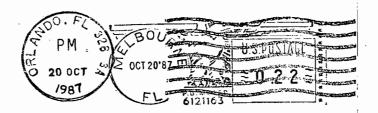
site. F54E17 is an exhaust fan with no control device.

- 1. The heat capacity of propane is 330.39 BTU/gal.
- 2. In order to determine pollutant emissions generated by the firing of propane, design engineer Brian Duck of John Zink Co. was contacted. He explained that the flare pilot is a pre-mixed burner. The propane in the system is mixed with the process gases under the proper stoichiometric concentrations to allow for complete, smokeless combustion at the flare tip. This is performed under high pressure and high velocity conditions at the point of combustion. Consequently, propane burns clean and emissions attributed to propane are negligible.

DER OCT 23 1987 BAOM

### HARRIS

J. R. Kolanek, MS58-055
HARRIS CORPORATION
SEMICONDUCTOR SECTOR
P.O. BOX 883
MELBOURNE, FLORIDA 32901



Mr. C. H. Fancy, P.E.
Deputy Chief, Bureau of Air Quality Management
- State of Florida, DER
- Twin Towers Office Building
- 2600 Blair Stone Road
Tallahassee, Florida 32399

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Fancy/Kolanek Letter of 10/15/87 Page 2

- 3. Because the amount of both pilot and enrichment propane utilized will remain constant on the flare system, the maximum consumption of propane is equal to the average consumption stated in the permit application (43 SCFH for the pilot gas, and 21.37 SCFH for the enrichment gas).
- 4. We are currently processing the internal paper work for the deficient amount of \$150.00. We shall forward the check under separate cover.

We trust the above information addresses all outstanding information. If you should have any questions, please feel free to contact me at (305) 724-7467.

Sincerely,

James K. Kolanek by S J. R. Kolanek, Manager

Environmental Services

/pgc

**Enclosures** 

Copiel: BTICHE

Bruse Mitchell & 10/26/87 000

Tom Sauncki

#### ATTACHMENT A

Calculations:

#### AC 05-138794

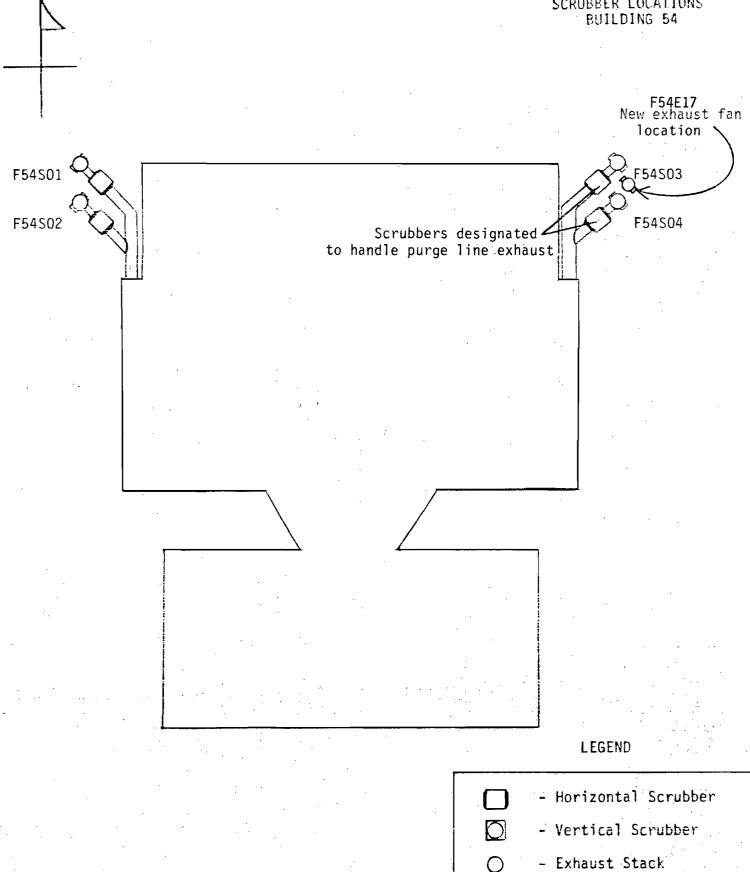
1. 1 lb. propane = 21,591 BTU/lb
@ STP, 1 liter propane = 1.8324g propane
(1 liter = 0.2642 gallons)
(1 lb. = 453.59 grams)

= <u>330.39 BTU/gal</u>

- Exhaust Fan

- Stack mounted on fan

- Epitaxial Scrubber



_									
PS Form 3841, JULY 1	Put your address in the "RETURN TO" space on the reverse side. Failure to do this will prevent this card from being returned to you. The return receipt fee will provide you the name of the person delivered to and the date of delivery. For additional fees the following services are available. Consult postmaster for fees and check box(es) for service(s) requested.								
983, 447-8	1. Show to whom, date and address of delivery.  2 Restricted Delivery.								
45	Article Addressed to:  Mr. James R. Kolanek  Harris Semiconductor  P 0 Box 883  Melbourne, F1 32901								
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## P 274 007 683 AC 05-138794, -138795 RECEIPT FOR CERTIFIED MAIL NO INSURANCE COVERAGE PROVIDED NOT FOR INTERNATIONAL MAIL

(See Reverse)

Sent to J. Kolanek, Harris Semicond. Box 883

P.O. State and ZIP Code
and Elbourne, F1 32901

Postage
Certification Special Delivery Fee Restricted Delivery Fee Return Receipt showing to whom and Date Delivered Return Receipt showing to whom, Date, and Address of Delivery TOTAL Postage and Fees PS Form 3800, Postmark or Date

#### STATE OF FLORIDA

#### DEPARTMENT OF ENVIRONMENTAL REGULATION

TWIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32399-2400



BOB MARTINEZ GOVERNOR DALE TWACHTMANN SECRETARY

September 25, 1987

CERTIFIED MAIL - RETURN RECEIPT REQUESTED

Mr. James R. Kolanek Manager, Environmental Services Harris Semiconductor P. O. Box 883 Melbourne, Florida 32901

Dear Mr. Kolanek:

Re: Completeness Review on the Applications to Construct Air Pollution Sources
Permit Nos. AC 05-138794 and -138795

The Department received your cover letter dated August 28, 1987, and the above reference applications on August 31, 1987. Based on a review of these applications, they have been deemed incomplete. The following information, including all reference material, calculations and assumptions, will have to be submitted to the Department's Bureau of Air Quality Management before the status can, again, be ascertained.

#### AC 05-138794

- For the volatile organic compounds and organic solvents, quantify the potential emissions per chemical in pounds per hour, month, and year.
- What is the facility designated identification for the scrubber to be used to handle the gas cyclinder purges?
- 3. Where will the scrubber medium be discharged after collection occurs?
- 4. What is the scrubber's medium?

Mr. James R. Kolanek Page 2 September 25, 1987

#### AC 05-138795

- 1. What is the heat capacity in Btu per gallon of the propane?
- 2. Quantify the potential pollutant emissions in pounds per hour and annually from the firing of the propane.
- 3. What is the maximum consumption of propane per hour?
- 4. Based on the submitted potential pollutant emissions, the appropriate processing fee, pursuant to FAC Rule 17-4.05, is \$250.00. Therefore, remit to the Department of Environmental Regulation the amount deficient, which is \$150.00.

If there are any questions, please call Bruce Mitchell at (904)488-1344 or write to me at the above address.

Sincerely,

C. H. Fancy, P.E.

Deputy Chief

Bureau of Air Quality

Management

CHF/BM/s

cc: T. Sawicki

Subcode 05

PM 8-28-87 Purolator Caurier Ticht # 880612765

AC 05-138795 . Ric'd : 8-31-87 Pd. \$100.00 Ricipt: 76178 MR:9-1-87

HARRIS

FS-JRK-030-88

August 28, 1987

DER

AUG 31 1987

BAQM

Mr. C. H. Fancy Deputy, Bureau Chief Department of Environmental Regulation Bureau of Air Quality Management 2600 Blair Stone Road Tallahassee, Florida 32301

SUBJECT: Air Permit Application - Vacuum Degasifier with

Flare System - Harris Semiconductor

Dear Mr. Fancy:

Enclosed please find the original and three copies for the subject air permit application for Harris Semiconductor's facility in Palm Bay, Florida. Also enclosed is the construction permit application fee.

The subject source is a part of our Industrial Water Project. This project will utilize Floridan water as the source of industrial water which is currently being met by potable water. Potential emissions from this source are hydrogen sulfide, carbon dioxide, and sulfur dioxide. The flare is designed to control odors caused by the removal of hydrogen sulfide from the water. The projected sulfur dioxide emissions are 27.61 tons/year. This is less than the amount considered a significant emission rate as specified by 17-2.500(2)(e)2.

If you should have any questions, please feel free to contact me at (305) 724-7467.

Sincerely,

James R. Kolanek, Manager Environmental Services

/pgc 🕜

enclosures

cc: FDER - Orlando

# **BEST AVAILABLE COPY**

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!	MARRIS SEMICONDUCTOR	Company Name  FDFR  Company Name  Company Name  Company Name
1	PALH BAY RU BLOG 58	Street Address (P.O. Box numbers not deliverable)
Ì	PALM SAY F L 3 2 9 0 5	TALLAMSSES FL 32301
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FS-JRK-030-88

August 28, 1987

00/03)

Mr. C. H. Fancy Deputy, Bureau Chief Department of Environmental Regulation Bureau of Air Quality Management 2600 Blair Stone Road Tallahassee, Florida 32301

SUBJECT: Air Permit Application - Vacuum Degasifier with

Flare System - Harris Semiconductor

Dear Mr. Fancy:

Enclosed please find the original and three copies for the subject air permit application for Harris Semiconductor's facility in Palm Bay, Florida. Also enclosed is the construction permit application fee.

The subject source is a part of our Industrial Water Project. This project will utilize Floridan water as the source of industrial water which is currently being met by potable water. Potential emissions from this source are hydrogen sulfide, carbon dioxide, and sulfur dioxide. The flare is designed to control odors caused by the removal of hydrogen sulfide from the water. The projected sulfur dioxide emissions are 27.61 tons/year. This is less than the amount considered a significant emission rate as specified by 17-2.500(2)(e)2.

If you should have any questions, please feel free to contact me at (305) 724-7467.

Holanett

Sincerely,

James R. Kolanek, Manager

Environmental Services

/pgc

enclosures

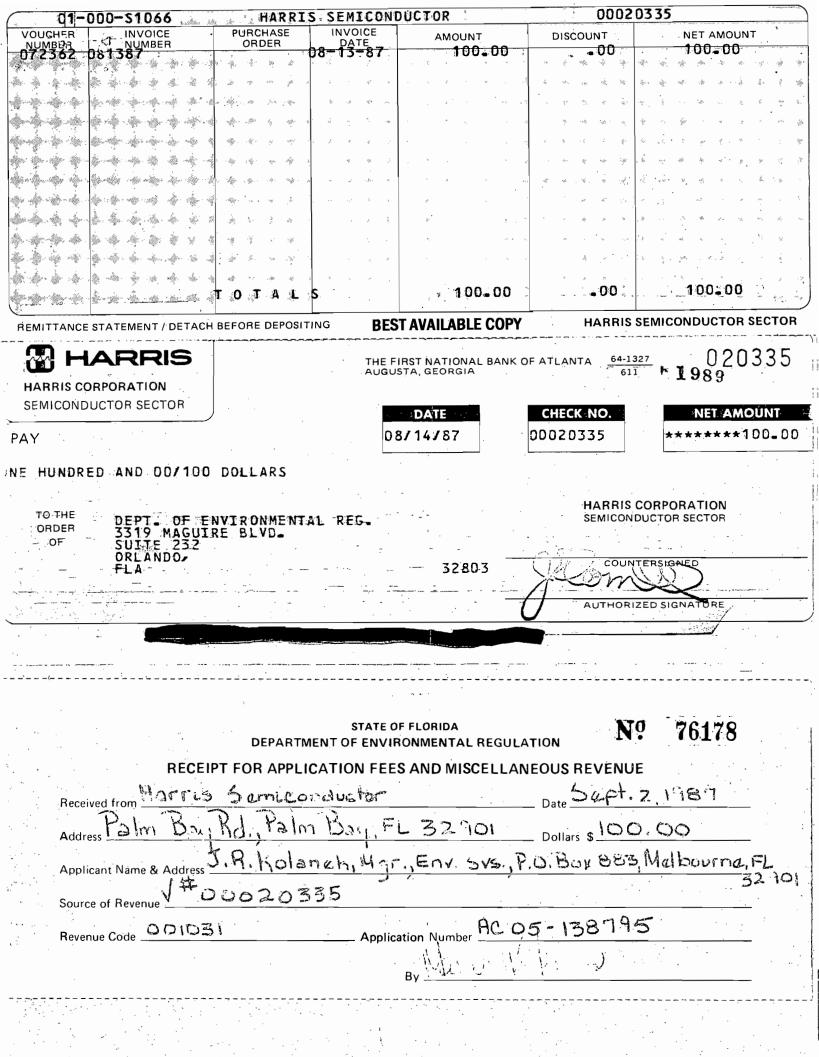
cc: FDER - Orlando

DER

SEP 01 1987

BAQM

1987 SEP - 1 M ID: 31



# DEPARTMENT OF ENVIRONMENTAL REGULATION

Recd: 8-31-87 Pd.\$100.00

MR: 9-1-87

WIN TOWERS OFFICE BUILDING 2600 BLAIR STONE ROAD TALLAHASSEE, FLORIDA 32301-8241



DER

BOB GRAHAM GOVERNOR

AUG 3 1 1987 VICTORIA J. TSCHINKEL

# APPLICATION TO OPERATE/CONSTRUCT AIR POLLUTION SOURCES

SOURCE TYPE: Stationary [X] N	lew <sup>l</sup> [ ] Existing <sup>l</sup>
APPLICATION TYPE: [X] Construction [] Operation	[ ] Modification
COMPANY NAME: Harris Semiconductor	COUNTY: Brevard
Identify the specific emission point source(s) addressed in No. 4 with Venturi Scrubber; Peaking Unit No. 2	Vacuum Dogasifion
SOURCE LOCATION: Street Palm Bay Road	City Palm Bay
UTM: East 17-538700	North 17-3100900
Latitude 28 1' 20 "N	Longitude 80 36' 10 "W
APPLICANT NAME AND TITLE: N. A. Baldisserotto Enviro	nmental Engineer, Environ, Science
APPLICANT ADDRESS: P.O. Box 883 Melbourne, FL 32901	·
Nan	ication for a construction st of my knowledge and belief. Further, a control source and pollution control the provision of Chapter 403, Florida the department and revisions thereof. I he? department, will be non-transferable sale of legal transfer of the permitted

B. PROFESSIONAL ENGINEER REGISTERED IN FLORIDA (where required by Chapter 471, F.S.)

This is to certify that the engineering features of this pollution control project have been designed/examined by me and found to be in conformity with modern engineering principles applicable to the treatment and disposal of pollutants characterized in the permit application. There is reasonable assurance, in my professional judgment, that

1 See Florida Administrative Code Rule 17-2.100(57) and (104)

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pollution sources.	Signed Chistot Back
	Chet Bach
1 South	Name (Please Type)
Just 2007	Harris Semiconductor  Company Name (Please Type)
(19)25101	P.O. Box 883, Melbourne, FL 32901
	Mailing Address (Please Type)
orida Registration No. 191	10 Date: 8/25/87 Telephone No. 305-724-7324
SECT	ION II: GENERAL PROJECT INFORMATION
and expected improvements	xtent of the project. Refer to pollution control equipment, in acurce performance as a result of installation. State result in full compliance. Attach additional sheet if
Semiconductor will be u	tilizing an Industrial Grade Water System to provide water for
the DeionizedWater Plan	its in Buildings 52 and 59. The system includes a vacuum
degasifier to remove H <sub>2</sub>	S and CO <sub>2</sub> from the Raw Well water. The removed gases will be
degasifier to remove H <sub>2</sub>	oroducts and control H <sub>2</sub> S emissions. (See Attachment A)
degasifier to remove H <sub>2</sub> flared to oxidize the p	S and CO <sub>2</sub> from the Raw Well water. The removed gases will be
degasifier to remove H <sub>2</sub> flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components	es and CO2 from the Raw Well water. The removed gases will be products and control H2S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only)  12787 Completion of Construction 1/89  I system(s): (Note: Show breakdown of estimated costs only Junits of the project serving pollution control purposes. ts shall be furnished with the application for operation
degasifier to remove H <sub>2</sub> flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components  Information on actual cos permit.)	es and CO2 from the Raw Well water. The removed gases will be products and control H2S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only)  12787 Completion of Construction 1/89  I system(s): (Note: Show breakdown of estimated costs only Junits of the project serving pollution control purposes. ts shall be furnished with the application for operation
degasifier to remove H <sub>2</sub> flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components Information on actual cos permit.)  One (1) self-supported	estimated costs only funits of the project serving pollution control purposes.  The removed gases will be project serving pollution control purposes.  Flare Stack; One (1) Model EEF-U-2 flare tip; One (1) manual
degasifier to remove H <sub>2</sub> flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components Information on actual cos permit.)  One (1) self-supported	or oducts and control H <sub>2</sub> S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only  12/87 Completion of Construction 1/89  I system(s): (Note: Show Breakdown of estimated costs only  /units of the project serving pollution control purposes.  ts shall be furnished with the application for operation  Flare Stack; One (1) Model EEF-U-2 flare tip; One (1) manual  ition panel: \$6,000.00
degasifier to remove H <sub>2</sub> flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components Information on actual cos permit.)  One (1) self-supported  weatherproof pilot igni	es and CO2 from the Raw Well water. The removed gases will be products and control H2S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only 12787 Completion of Construction 1/89  I system(s): (Note: Show breakdown of estimated costs only Junits of the project serving pollution control purposes. ts shall be furnished with the application for operation  Flare Stack; One (1) Model EEF-U-2 flare tip; One (1) manual ition panel: \$6,000.00
degasifier to remove Hz  flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components  Information on actual cos permit.)  One (1) self-supported  weatherproof pilot igni	or oducts and control H <sub>2</sub> S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only  12/87 Completion of Construction 1/89  I system(s): (Note: Show Breakdown of estimated costs only  /units of the project serving pollution control purposes.  ts shall be furnished with the application for operation  Flare Stack; One (1) Model EEF-U-2 flare tip; One (1) manual  ition panel: \$6,000.00
degasifier to remove Hz  flared to oxidize the p  Schedule of project cover  Start of Construction  Costs of pollution contro for individual components  Information on actual cos permit.)  One (1) self-supported  weatherproof pilot igni	es and CO2 from the Raw Well water. The removed gases will be products and control H2S emissions. (See Attachment A)  ed in this application (Construction Permit Application Only)  12/87 Completion of Construction 1/89  I system(s): (Note: Show breakdown of estimated costs only funits of the project serving pollution control purposes. Its shall be furnished with the application for operation  Flare Stack; One (1) Model EEF-U-2 flare tip; One (1) manual ition panel: \$6,000.00

Page 2 of 12

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с.	
If this is a new source or major modification, answer the following quest: (Yes or No)	ions.
l. Is this source in a non-attainment area for a particular pollutant?	No.
a. If yes, has "offset" been applied?	
b. If yes, has "Lowest Achievable Emission Rate" been applied?	
c. If yes, list non-attainment pollutants.	
<ol> <li>Does best available control technology (SACT) apply to this source?</li> <li>If yes, see Section VI.</li> </ol>	No
<ol> <li>Ones the State "Prevention of Significant Deterioriation" (PSD) requirement apply to this source? If yes, see Sections VI and VII.</li> </ol>	No
4. Do "Standards of Performance for New Stationary Sources" (NSPS) apply to this source?	No
5. Do "National Emission Standards for Hazardous Air Pollutants"  (NESHAP) apply to this source?	No
Do "Reasonably Available Control Technology" (RACT) requirements apply to this source?	No
a. If yes, for what pollutants?	

Attach all supportive information related to any answer of "Yes". Attach any justifi-

cation for any answer of "No" that might be considered questionable.

#### SECTION III: AIR POLLUTION SOURCES & CONTROL DEVICES (Other than Incinerators)

A. Raw Materials and Chemicals Used in your Process, if applicable:

	Contaminants		Utilization			
Description	Туре	% Wt	Rate - lbs/hr	Relate to Flow Diagram		
-						
				· · · · · · · · · · · · · · · · · · ·		
				,		

B. Process Rate, if applicable: (See Section V, Item	в.	cocess Rate.	1.5	applicable:	(See	Section	٧.	Item	]
--	----	--------------	-----	-------------	------	---------	----	------	---

Product Weight (lbs/hr):\_

1.	Total Process	Input Rate	(lbs/hr):	1200 1b/hr of CO2; 90 1b/hr of H2S

Airborne Contaminants Emitted:	(Information in this tab	le must be submitted for each
emission point, use additional	sheets as necessary)	· · · · · · · · · · · · · · · · · · ·

Name of	Emiss	ionl	Allowed <sup>2</sup> Emission Rate per	Allowable <sup>3</sup> Emission	Potent Emiss		Relate to Flow
Contaminant	Maximum lbs/hr	Actual T/yr	Rule 17-2	lbs/hr	lbs/yr	T/yr	Diagram
\$0 <sub>2</sub> ,	7.059	27 . <del>6</del> 1	Model No.	on a second of the second	61836.84	27.61	
H <sub>2</sub> S	0.05625	0.220			32,850	14.67	·
CO <sub>2</sub>	0.75	2.933	(x,y) = (x,y) = (x,y)	ं .	438,000	195.54	
	. :					- 1	•

<sup>1</sup> See Section V, Item 2.

C.

 $<sup>^2</sup>$  Reference applicable emission standards and units (e.g. Rule 17+2.600(5)(b)2. Table II, E. (1) = 0.1 pounds per million BTU heat input)

<sup>3</sup>Calculated from operating rate and applicable standard.

 $<sup>^4</sup>$ Emission, if source operated without control (See Section V, Item 3).

J. Control Devices: (See Section V, Item 4)

Name and Type (Model & Serial No.)	Contaminant	Efficiency	Renge of Particles Size Collected (in microns) (If applicable)	Basis for Efficiency (Section V Item 5)
Self-supported flare Stack (John Zink Co.)	H <sub>2</sub> S	98.5%	N/A	See Attachment
w/Model EEF-U-2 Flare   tip (John Zink Co.)	CO <sub>2</sub>	98.5%	N/A	. "
and Manual/Weatherpro	of			
	_			

#### E. Fuels

	Consump	tion*		
Type (3e Specific)	avq/hr	max./hr	Maximum Heat Input (MMBTU/hr)	
propane (pilot)	43 SCFH		108624.32 BTU/hr	
propane (enrichment)	21.37 SCFH		53977.5 BTU/hr	

\*Units: Natural Gas -- MMCF/hr; Fuel Oils -- gallons/hr; Ceal, wood, refuse, other -- lbs/hr.

Fuel Analysis:		· . · ·	ار ایستان کار ایستان استان ایستان استان ایستان کار ایستان کار ایستان کار ایستان کار ایستان کار ایستان کار داده مارک میران کار ایستان			
Percent Sulfur:	0		Percent Ash:	0	·	
Density: 1.8324x10 <sup>-3</sup>						
Heat Capacity: 21,5	91ause include					STU/gal
Other Fuel Contaminants	(which may cau	use air po	llution):			-
·			Section 18 to 18			
F. If applicable, indi	date the percer	nt of fuel	used for sp	pace heating.	N/A	
Annual Average	्र विक्रिक्तिसम्बद्ध	Max	imum		-	
G. Indicate liquid or					•	
	N/A					
•						

	t:	20'		ft. Si	tack Diamete	r:6"	ft
as Flow Ra	te:7.9	985_ACFM	7.985	_DSCFM Ga	s Exit Temp	erature:	75•F
ater Vapor	Content:		_	% Ve	locity:	5.7	<u> 12</u> FP
		SECT	ION IY1	INCINERATO	OR INFORMATI	0 N	
Type of Waste	Type O (Plastics)				Type IV (Patholog- ical)		Type VI (Solid By-prod.)
Actual lb/hr Inciner- ated						-	-
Uncon- trolled (lbs/hr)							
	•	1		<u> </u>	<u> </u>		
otal Weigh	t Incinera	Hours of	r) Operation	per day _	day/	wk	hr)wks/yr
otal Weigh oproximate anufacture	t Incinera Number of	ted (lbs/h	r)	per day _	Design Cap	wk	wks/yr.
otal Weigh oproximate anufacture	t Incinera Number of	ted (lbs/h	r)	per day _	_ Design Cap	wk	wks/yr.
otal Weigh oproximate anufacture	t Incinera Number of	ted (lbs/h	Cperation	per dayModel	_ Design Cap day/	wk	wks/yr.
otal Weigh oproximate anufacture ate Constr	t Incinera Number of r ucted	ted (lbs/h Hours of  Volume (ft) <sup>3</sup>	Gperation  Heat R  (BTU	per dayModel	Design Cap	wk	Temperature
otal Weigh oproximate anufacture ate Constr	t Incinera Number of r ucted	ted (lbs/h Hours of	Gperation  Heat R  (BTU	model  elease /hr)	Design Cap	BIU/hs.	Temperature
otal Weigh  pproximate  anufacture  ate Constr  Primary Ch  Secondary	t Incinera Number of r ucted amber	ted (lbs/h Hours of  Volume (ft) <sup>3</sup>	Gperation  Heat R  (BTU	Model elease /hr)	Design Cap	BTU/hs.	Temperature
otal Weigh oproximate anufacture ate Constr  Primary Ch Secondary tack Heigh	Number of r ucted amber Chamber	Volume (ft)3	Heat R (BTU	per dayModel elease /hr)	Design Cap	BTU/hr	Temperature (°F)
otal Weigh oproximate anufacture ate Constr  Primary Ch Secondary tack Heigh as Flow Ra If 50 or m	t Incinera Number of r ucted amber Chamber t:	Volume (ft)	Heat R (BTU	mtar:	Design Cap  day/  No.  Fuel  IYP9.  DSCFM+	BTU/hr.  Stack T	Temperature (°F)

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Brief description	ar ape	rating ch	aracte	risti(	2 <b>3</b> or	control	devi	: 8 3 :			· · · · · · · · · · · · · · · · · · ·
						•			· · · · · · · · · · · · · · · · · · ·		
								•			
Iltimate disposal ash, etc.):	of any	effluent	other	than	that	emitted	from	the	stack	(scrubber	water,
		, ···			<u> </u>						
				· · -							

NOTE: Items 2, 3, 4, 6, 7, 8, and 10 in Section V must be included where applicable.

#### SECTION V: SUPPLEMENTAL REQUIREMENTS

Please provide the following supplements where required for this application.

- 1. Total process input rate and product weight -- show derivation [Rule 17-2.100(127)]
- ?. To a construction application, attach basis of emission estimate (e.g., design calculations, design drawings, pertinent manufacturer's test data, etc.) and attach proposed methods (e.g., FR Part 60 Methods 1, 2, 3, 4, 5) to show proof of compliance with applicable standards. To an operation application, attach test results or methods used to show proof of compliance. Information provided when applying for an operation permit from a construction permit shall be indicative of the time at which the test was made.
- 3. Attach basis of potential discharge (e.g., emission factor, that is, AP42 test).
- 4. With construction permit application, include design details for all air pollution control systems: (e.g., for baghouse include cloth to air ratio; for scrubber include cross-section sketch, design pressure drop, etc.)
- 5. With construction permit application, attach derivation of control device(s) efficiency. Include test or design data. Items 2, 3 and 5 should be consistent: actual emissions = potential (1-efficiency).
- 6. An 8 1/2" x 11" flow diagram which will, without revealing trade secrets, identify the individual operations and/or processes. Indicate where raw materials enter, where solid and liquid waste exit, where gaseous emissions and/or airborne particles are evolved and where finished products are obtained.
- 7. An 8 1/2" x 11" plot plan showing the location of the establishment, and points of airborne emissions, in relation to the surrounding area, residences and other permanent structures and roadways (Example: Copy of relevant portion of USGS topographic map).
- 3. An 3 1/2" x 11" plot plan of facility showing the location of manufacturing processes and outlets for airborne emissions. Relate all flows to the flow diagram.

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9.	The appropriate application fee in accordance payable to the Department of Environ	dance with Rule 17-4.05. The check should be mental Regulation.
10.		, attach a Certificate of Completion of Con- as constructed as shown in the construction
	SECTION VI: BEST AVAIL	ABLE CONTROL TECHNOLOGY
A.	Are standards of performance for new state applicable to the source?	tionary sources pursuant to 40 C.F.R. Part 60
	[ ] Yes [ ] No	
	Contaminant	Rate or Concentration
_	<u> </u>	· · · · · · · · · · · · · · · · · · ·
в.	Has EPA declared the best available cont yes, attach copy)	rol technology for this class of sources (If
	[ ] Yes [ ] No	
	Contaminant	Rate or Concentration
	·	
	·	e a contraction of the contracti
		3 cm j
c.	What emission levels do you propose as be	st available control technology?
	Conteminant	Rate or Concentration
		, <del></del>
		<del></del>
0.	Describe the existing control and treatme	nt technology (if any).
	1. Control Device/System:	2. Operating Principles:
	3. Efficiency:*	4. Capital Costs:
·E x	plain method of determining	
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	5.	Useful Life:			6.	Operating Costs:	•
	7.	Energy:			8.	Maintenance Cost:	
	9.	Emissions:	,				•
		Contamina	nt			Rate or Concentrati	an '
	-						
	10.	Stack Parameters					
	a.	Height:	·	ft.	<b>b</b> .	Diameter:	'ft.
	c.	Flow Rate:		AC FM	đ.	Temperature:	o.F.
	<b>9.</b> .	Velocity:		FPS			
ε.		cribe the control additional pages			olog	y available (As many types	as applicable
	1.						
	a .	Control Device:			ь.	Operating Principles:	
	с.	Efficiency: 1			đ.	Capital Cost:	
	e.	Useful Life:			f.	Operating Cost:	
	g.	Energy 2			h.	Maintenance Cost:	
	i.	•				d process chemicals:	
	j.	Applicability to	~	proces	385:		
	k.		uct with cont:			rould this information int, install in available space	and operate
	2.					•	
	a.	Control Device:			ь.	Operating Principles:	
	c.	Efficiency: 1			d.	Capital Cost:	
	e.	Useful Life:			r.	Operating Cost:	
	g.	Energy: <sup>2</sup>	÷.		h.	Maintenance Cost:	
	i.	Availability of c	onstruction ma	terial	s an	d process chemicals:	
		n method of determ					
4En	ergy	to be reported in	units of elec	trical	. раж	er - KWH design rate.	

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Applicability to manufacturing processes: Ability to construct with control device, install in available space, and operate within proposed levels: 3. Control Device: Operating Principles: Efficiency: 1 Capital Cost: Useful Life: Operating Cost: Energy: 2 Maintenance Cost: g. Availability of construction materials and process chemicals: Applicability to manufacturing processes: Ability to construct with control device, install in available space, and operate within proposed levela: 4. Control Device: Operating Principlea: Efficiency: 1 Capital Costs: Useful Life: Operating Cost: Energy: 2 Maintenance Cost: Availability of construction materials and process chemicals: Applicability to manufacturing processes: Ability to construct with control device, install in available space, and operate within proposed levels: grans & ran that a F. Describe the control technology selected: Control Device: 2. Efficiency: 1 Capital Cost: Useful Life: 3. Operating Cost: Energy: 2 7. Maintenance Cost: Manufacturer: Other locations where employed on similar processes:

(3) City: (4) State:

Explain method of determining efficiency.

Energy to be reported in units of electrical power - KWH design rate.

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a. (1) Company:

(2) Mailing Address:

•		,		
(5) Environmental Manager:				
(6) Telephone No.:				:
(7) Emissions: <sup>1</sup>				
Contaminant			Rate or Concentr	ation
•				
(8) Process Rate: 1				•
b. (1) Company:				
(2) Mailing Address:				
(3) City:		(4) State:		
(5) Environmental Manager:				•
(6) Telephone No.:			÷	
(7) Emissions: <sup>1</sup>				
Contaminant	•		Rate or Concentr	ation
		-		
(8) Process Rate: 1				
(8) Process Rate: 1	•			
	ormation whe	n available. ) why.	Should this in	formation no
IO. Reason for selection and I Applicant must provide this infavailable, applicant must state  SECTION VII -	ormation whe	n available. ) why.  F SIGNIFICANT	Should this in	
10. Reason for selection and  L'Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data	ormation whe the reason(s	n available. ) why.  F SIGNIFICANT	Should this in DETERIORATION	_ Wind spd/di
IO. Reason for selection and  I Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data  1no. sites	ormation whe the reason(s	n available. ) why.  F SIGNIFICANT	Should this in	_ Wind spd/di
IO. Reason for selection and  L'Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT	Should this in DETERIORATION  S02+  Month day yes	_ Wind spd/di
IO. Reason for selection and  L'Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT  ( )  / to	Should this in DETERIORATION  SD2+  Month day yes	_ Wind spd/d:
IO. Reason for selection and  L'Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT  ( )  / to	Should this in DETERIORATION  SD2+  Month day yes	_ Wind spd/d:
IO. Reason for selection and LApplicant must provide this infavailable, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT  ( )  / to	Should this in DETERIORATION  SD2+  Month day yes	_ Wind spd/di
IO. Reason for selection and  L'Applicant must provide this inf available, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT  ( )  / to ay year  to this applie	Should this in DETERIORATION  SD2+  Month day yes	_ Wind spd/di
IO. Reason for selection and LApplicant must provide this infavailable, applicant must state  SECTION VII -  A. Company Monitored Data  1	PREVENTION OF TSP	n available. ) why.  F SIGNIFICANT  ( )  / to ay year  to this applie	Should this in DETERIORATION  SD2+  Month day yes	_ Wind spd/di

	indication, 12016 and Education,	
	a. Was instrumentation EPA referenced or i	ts equivalent? [ ] Yes [ ] No
٠	b. Was instrumentation calibrated in accor	dance with Department procedures?
	[ ] Yes [ ] No [ ] Unknown	
a.	Meteorological Data Used for Air Quality Mo	deling
	1. Year(s) of data from // month day	year month day year
	2. Surface data obtained from (location)_	
	3. Upper air (mixing height) data obtained	from (location)
-	4. Stability wind rose (STAR) data obtaine	d from (location)
c.	Computer Models Used	
	1.	Modified? If yes, attach description.
•	2	Modified? If yes, attach description.
•	3.	Modified? If yes, attach description.
		Modified? If yes, attach description.
		ng input data, receptor locations, and prin-
0.	Applicants Maximum Allowable Emission Data	e de la companya del companya de la companya del companya de la co
	Pollutant Emission Rate	
	TSP	grams/sec
ट लग	norv stra the provision of Chapter 400,	grams/sec
SE S	This ion bate deed in Modeling	ierable sav tlen noder
	Attach list of emission sources. Emission	data required is source name, description of bordinates, stack data, allowable emissions,
F.	Attach all other information supportive to	the PSD review.
G.	Discuss the social and economic impact of t	he selected technology versus other applica- roduction, taxes, energy, etc.). Include he sources.

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the requested best available control technology.

H. Attach scientific, engineering, and technical material, reports, publications, journals, and other competent relevant information describing the theory and application of

#### ATTACHMENT A

Flow diagram dwg. P-1 illustrates the Industrial Grade water system that Harris Semiconductor will be employing. The purpose of the operation is to produce industrial water for use in Buildings 52 and 59 Deionized (DI) plants. Initial design criteria is based on a 600 GPM water flow rate.

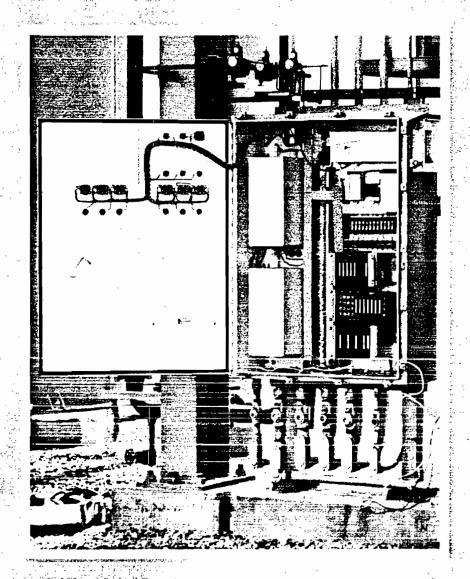
Floridian aquifer water will be drawn from two existing 1140 gpm wells and will be conveyed via a pipe rack to the facility. The system will be located between waste treatment plant and Building 58, as illustrated in Dwg. C-2.

Following chemical treatment to reduce scaling and corrosive properties, the water will be filtered and pumped into the reverse osmosis (RO) membranes. A minimum of 75% of this water will pass through the membranes as purified water and will be treated in one of two vacuum degasifiers to remove hydrogen sulfide and carbon dioxide gases. The gas streams from the degasification process will be sent to a flare system to oxidize the hydrogen sulfide. The RO reject water will go to the deep well disposal system.

ATTACHMENT B
HARRIS SEMICONDUCTOR
CONTROL EQUIPMENT



# JOHN ZINK COMPANY



**FLARE SYSTEM CONTROLS** 

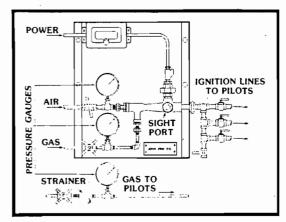
John Zink Company, the world's largest manufacturer of flare systems, offers complete flare control packages. Flare controls by John Zink ensure sole source compatibility and single point responsibility. John Zink's experience with thousands of flare systems ensures a well designed and proven control system.

**Pilot Ignition Systems** 

The single most important piece of control equipment in the flare system is the pilot ignition system. Safety of the entire plant depends upon proper operation of the ignition system. Each John Zink pilot ignition system offers the following features:

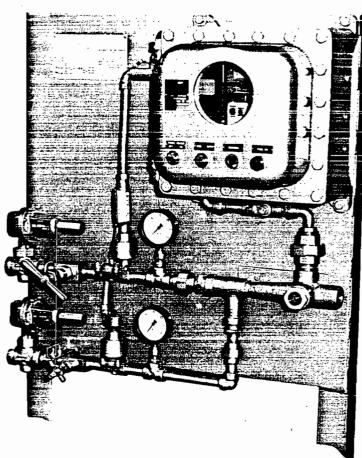
- All control components are easily accessible at grade
- · Proven reliable ignition
- Complete shop testing prior to shipment
- All systems are designed to meet the required electrical area classifications
   The following models are indicative of the variety of ignition systems available from John Zink:

Manual FFG (Flame Front Generator) The industry standard for ignition of flare systems for over 30 years, the Manual FFG provides simple, reliable ignition where compressed air is available.



Automatic FFG (Flame Front Generator) An additional level of safety is obtained by constant monitoring of the flare pilot.

In the rare event of pilot failure, automatic reignition of the flare pilot is available with this packaged ignition system. This system features solid state controls reliable monitoring and automatic reignition.



**Self Inspirating FFG** (Automatic or Manual)

In the event compressed air is not available, the self inspirating ignition system utilizes fuel gas to inspirate air into a venturi mixer on the ignition line. This gas/air mixture is ignited and travels through the ignition line to light the pilot.

Pilot Monitoring

John Zink offers PilotEye, a unique dual-waveband infrared monitor. Mounted up to 1,000 feet away, the unit monitors the flare tip and signals in case of pilot flame-out. Simple installation requires no flare shut-down. Conventional thermocouple monitors and alarm packages are also available.

#### Special Ignition Systems

In areas where no electricity is available, battery powered or piezoelectric ignition systems can be provided.

In addition to ignition systems, other control systems are necessary for proper flare operation. These control systems include:

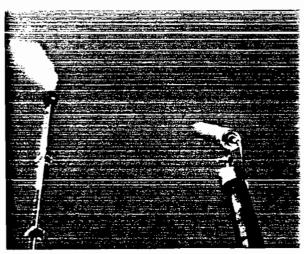
#### Purge Controls

Purge controls for molecular seals and airrestors can be provided to ensure proper flow of purge gas to the flare system and prevent air ingression into the flare header.

In the event high temperature purge gases are vented, John Zink can provide a patented *Tempurge system*. This system monitors pressure and temperature in the flare line and injects the proper amount of additional purge gas necessary to prevent the formation of a vacuum as the waste gases cool.

#### Steam Controls

John Zink offers a complete steam control package including steam control valves, restriction orifices, block valves, gauges and controllers all coordinated around and designed for use with the John Zink Zoom System.



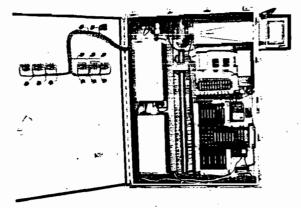
John Zink Zoom System

Details of this system are available in the John Zink Zoom System bulletin.

#### Special Flare Controls

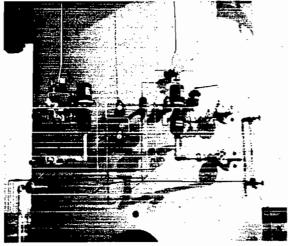
John Zink special flare controls include:

- Patented staging controls
- Knock out and liquid seal drum controls including:
  - Level controls
  - Gauge glasses
  - Condensate pumps
  - Alarms
  - Temperature controls
  - Control valves



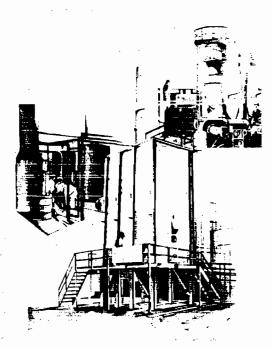
Staging Control Panel

 Blower controls for air assisted flares and other applications.



Typical Liquid Seal 'Knockout Drum Controls

- Explosion prevention systems for gases such as Hydrogen, Ethylene Oxide, Carbon disulfide, Acetylene and other explosive gases.
- Flame extinguishing systems
- Steam desuperheaters
- Skidded, assembled, completely prepiped control packages for flaring wastes



#### Rely on John Zink for:

- · Flares
- Burners
- · Packaged Burners
- · Resource Recovery
- Incinerators
- · Heat Recovery
- Heating & Air Conditioning and Associated Equipment



International Headquarters 4401 South Peoria P.O. Box 702220 Tulsa, Oklahoma 74170 (918) 747-1371

Other offices are located in major cities around the world.

World's Largest Manufacturer of Flare Systems

**Brochure 5022** 

#### **EEF Series Flares**

The EEF Series Flares are the latest development in flaring technology from the John Zink Company. More than half a century of combustion expertise has been incorporated with the latest technological advances and a stringent testing program to provide you with a dependable, efficient flaring system at substantially lower operating costs.

Normally, plant wastes are supplied to the flare system through safety relief valves. With this unpredictable source of waste, the only acceptable means of ignition is a continuously burning pilot. With today's high energy costs, utilities for a continuously burning pilot represent a substantial operating expenditure. John Zink has developed the EEF series flares, which reduce pilot gas consumption by as much as 80%.

Flare burner stability is accomplished with a combination of the flare pilot and a Flame Stability Tip. John Zink has developed a state of the art Flame Stability Tip for use specifically with the lower energy pilots of the EEF Series Flares. With smaller pilot capacities, the Flame Stability Tip assumes a critical role in the overall combustion stability of the flare.

The metallurgical selection for the EEF Series Flares is based upon actual field experience and continuing research and development efforts to ensure optimum design at lower costs.

The following series of EEF flares will have an attractive pay back when considered for your next turnaround or in new plant construction. The nomograph on the back cover will help you estimate your energy savings.

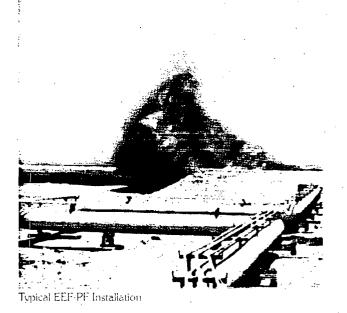
#### EEF-U

The EEF-U is an excellent choice in services which do not require smokeless

operation or where waste gases will not smoke. Where service conditions warrant, a very high temperature refractory, secured with a special stainless steel anchoring system and reinforced with high alloy needles, is installed.

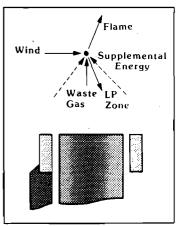
#### **EEF-PF**

The EEF-PF flare is specifically designed for horizontal firing in a pit where two phase flow will exist. Proper installation of the PF flare and the pit design are critical to flare life. John Zink Company has the experience and the expertise to assist you with the necessary engineering to ensure proper operation.



#### **EEF-LS**

John Zink has developed a special EEF-LS Flare which extends flare life.



Vector Diagram of EEF LS Hare

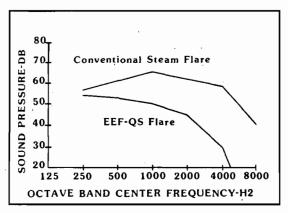
The EEF-LS
design uses energy
from a secondary
source to overcome wind effects
and move the
flame away from
the flare thus
eliminating wind
influenced flare
damage. This
flare can be

utilized in a horizontal or vertical position.

The EEF-LS Flare effectively increases flare service life, reduces down time and cuts maintenance and replacement costs.

#### **EEF-QS**

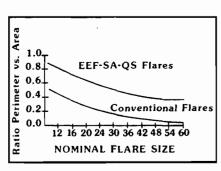
The EEF-QS Series Flares reduce noise levels up to one half that of conventional steam assisted smokeless flares. When smokeless flaring is required, the traditional solution has been steam injection. One of the most persistent problems with steam injection is the associated noise. The EEF-QS utilizes the well proven noise abatement effect of increased shear to flow area as depicted in the accompanying graph.



### **EEF-SA-QS**

The EEF-SA-QS offers the maximum possible steam efficiency in a single point flare with the lowest possible noise levels and the highest smokeless flow rates available. During normal operation, noise levels may be 1/8 those of conventional steam flares.

The perimeter of the flare, which is the

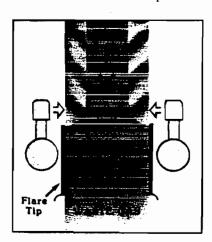


site of air and steam injection, increases linearly with the diameter, simultaneously, the waste flow area increases as a squared

function of the diameter. The net effect is a squared increase in the flow and a linear increase in the air and steam availability. The EEF-SA-QS produces an artificial increase in the critical perimeter to area ratio by a multiplicity of steam/air injection points internally and externally.

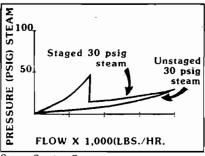
#### **EEF-SS**

The EEF-SS Flare is designed to utilize low cost, low pressure steam. Traditionally, the use of this low pressure steam has



resulted in poor steam efficiency at turndown. This phenomenon can be demonstrated by the adjacent diagram. In order to produce smokeless operation,

it is necessary to completely mix the waste gas, steam and air. Low pressure steam at turndown does not have sufficient energy, in a traditional flare, to penetrate the waste. This results in an unnecessarily high steam usage at turndown. John Zink's concept of staged



Steam Staging Curve

steam flaring produces a higher penetration of steam and air over the entire operating range.

The above steam staging curve demonstrates the improvement in energy levels. Significant cost savings can be achieved through the use of low pressure steam in the EEF-SS flare.

The new series of EEF flares has many advantages for your facility including:

- Lower utility costs
- Improved efficiency
- Lower steam noise levels
- · Increased flare burner life

#### **NOMOGRAPH** SHOWING SAVINGS FOR EEF FLARE

50

40 \_30

. 20

\_10

-0.8

0.6

\_0.5

Energy Index Scale

INSTRUCTIONS FOR NOMOGRAPH:

1. Calculate difference of Energy
Index.

2. Locate energy index difference on scale Locate energy index difference on scale labeled Energy Index Scale.
 Locate fuel cost in \$\\$/mmBTU\$ on scale labeled Fuel Cost (\$\\$/mmBTU\$).
 Use a straight edge to connect these two points.
 Read yearly savings on center scale in \$/yr.

# 438,000 400,000 300,000 200,000 100,000 80,000 \_60,000 \_60,000 \_50,000 \_40,000 30,000 20,000 Fuel Cost (\$/mmBTU) 10,000 8,000 6,000 5,000 4,000 3,000 2,000 \_1,000 \_800 \_600 \_500 \_438

# TABLE OF TYPICAL **ENERGY INDEX OF** EEF FLARES VS. IN USE FLARES

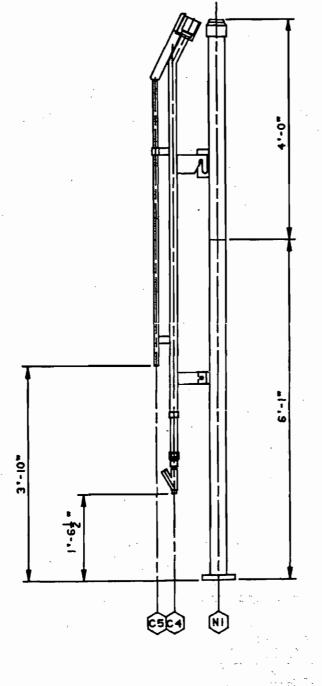
	LLAKES	•
FLARE TIP SIZE	EEF FLARE	TYPICAL EXISTING FLARE
8"	1 .	7
12"	1	10.5
18"	2	10.5
24"	3	10.5
36"	3	14
48"	3 .	14
60"	3	14



#### UTILITY STYLE FLARE BURNER

#### DATA SHEET

1. CUSTOMER:		·
2. PLANT LOCATION:		
3. MODEL: EEF-U- 2		ENGINEER:
4. OVERALL LENGTH 10'-1"		NO. OF PILOT
5. FIRING POSITION: VE	RTICAL	
6. DESIGN CONDITION		
7. FLOW RATE:	MW:	SP. GR.
8. PRESS. DROP @ DESIGN:		-
9.		· · · · · · · · · · · · · · · · · · ·
10. MATERIAL OF CONSTRUCTIO	N	
11. SECTION		MATERIAL
12. Flame Retention Ring		310 or equal
13. Upper Sect. FRR to 4'-	0"	304 SS
14. Lower Sect. 4'-0" to 1	0'-1"	C. Stl.
15. Flange		A-105
16. Pilot(s)		309 SS
17.		
18. WELDING PROCEDURES:	AWS	
19.		
20.		
21. FINISH: High temperat	ure aluminu	m on carbon steel
22.		
23. NOZZLE NO. SI	ZE IN.	TYPE
24. Inlet N1	2	ANSI 150# RF
25. Pilot C4	3/4	Plain End
26. Ignitor C5	1	Plain End
27.		
28.		
29. REMARKS		
30.		
31.		
32.		
33.		
34.		
35.		
36.		
37.	-	·
38.		
39.		
40.	······································	· .
41.	-	
42.		
43.		
44.		
	-	
45.	-	
45. 46.		
45. 46. 47.		
45. 46. 47. 48.		
45. 46. 47. 48.		
45. 46. 47. 48. <b>49</b> .		
45. 46. 47. 48.		



PROPOSAL NO.:

EEP

FUEL

SCFH

No. Required

DATE:

TYPE OF PILOTS:

THERMOCOUPLE:

FUEL CONSUMPTION

Type: K (Chromel-Alumel)

# ATTACHMENT C HARRIS SEMICONDUCTOR MANUFACTURER TEST DATA





4401 South Peoria Avenue P.O. Box 702220 Tulsa, Oklahoma 74170 918/747-1371 Telex 497414

July 17, 1987

Harris Corporation P. O. Box 883 Melbourne, FL 32901

Attention:

Nancy Bardisserotto

Mail Stop 58-55

Reference:

Flare Efficiency Study

John Zink File F609-031DL

Dear Ms. Bardisserotto:

Pursuant to your recent request, attached find a report on a flare efficiency study performed by Engineering-Science. The study was sponsored by the U.S. Environmental Protection Agency and the Chemical Manufacturers Association and took place at the John Zink International Research Center in Tulsa, Oklahoma using John Zink Flare Equipment.

You will notice on page 1-4 of the report that the combustion efficiency of the gases with a low heating value averages approximately 99.0%. We would expect the same efficiency for your application.

The attached is a condensed version of the 123 page full report which may be obtained from Chemical Manufacturers Association, 2501 M. Street N.W., Washington, D.C. 20037. If you have any questions, or require further information, please do not hesitate to contact us.

Sincerely,

JOHN ZINK COMPANY

Brian Duck

A REPORT ON

Α

# FLARE EFFICIENCY STUDY



CHEMICAL MANUFACTURERS ASSOCIATION

PREPARED BY

# **ENGINEERING-SCIENCE**

DESIGN • RESEARCH • PLANNING
3109 NORTH INTERREGIONAL AUSTIN, TEXAS 78722 • 512/477-9901
OFFICES IN PRINCIPAL CITIES



# A REPORT ON A FLARE EFFICIENCY STUDY

VOLUME I

Prepared for

CHEMICAL MANUFACTURERS ASSOCIATION Washington, D.C.

September 1982

Prepared by:

Engineering-Science Austin, Texas

#### CHAPTER 1

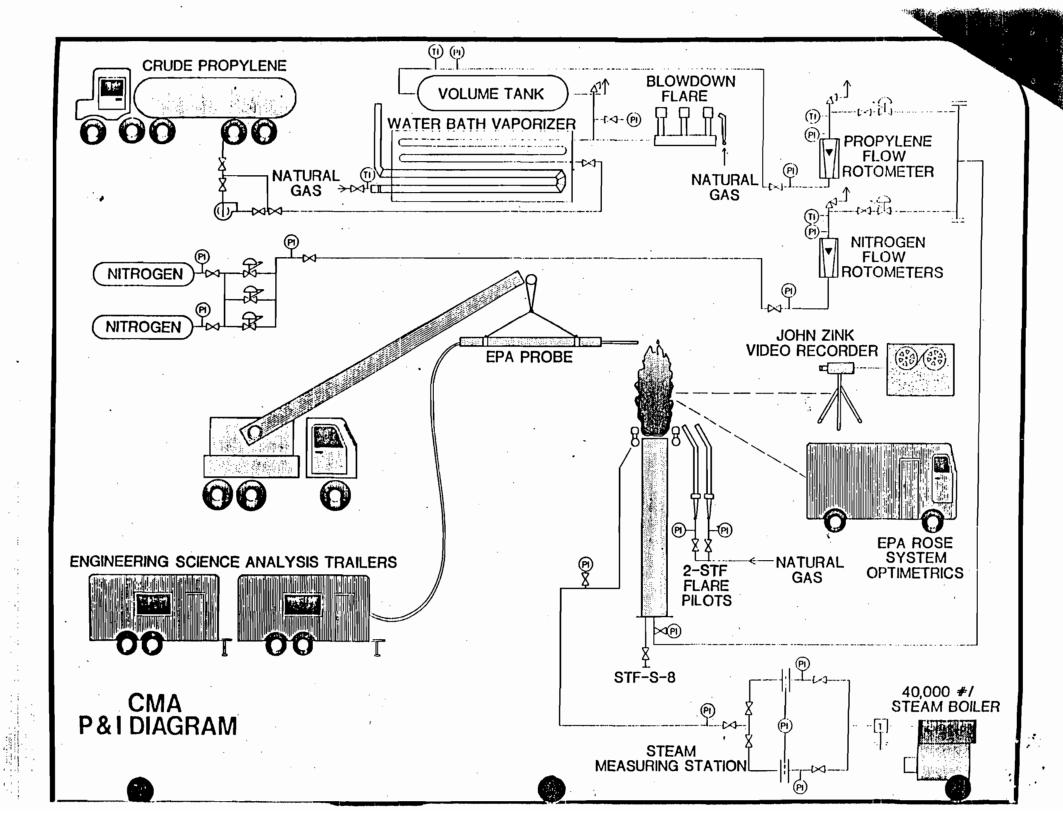
#### EXECUTIVE SUMMARY

#### PROJECT OVERVIEW

This document is a report on an experimental study to determine the efficiencies of flare burners as devices for the disposal of hydrocarbon emissions from refinery and petrochemical processes. The primary objectives of this study were to determine the combustion efficiency and hydrocarbon destruction efficiency for both air- and steam-assisted flares under a wide range of operating conditions. The test results indicate that flaring is generally an efficient means of hydrocarbon disposal.

Separate elements of this flare efficiency study were sponsored by the U.S. Environmental Protection Agency (EPA) and the Chemical Manufacturers Association (CMA). Other project participants included John Zink Company who provided the flares, test facility and flare operation, and Optimetrics, Inc. who operated the EPA's Remote Optical Sensing of Emissions (ROSE) system. Engineering-Science, Inc. operated the extractive flare sampling and analysis systems.

Figure 1.1 is an overview of the equipment used to operate and test the flares. The test methodology utilized during the study employed a specially constructed 27-foot sample probe suspended by a crane over the flare flame. The sample extracted by the probe was analyzed by continuous emission monitors to determine concentrations of carbon dioxide (CO<sub>2</sub>), carbon monoxide (CO), total hydrocarbons (THC), sulfur dioxide (SO<sub>2</sub>), oxides of nitrogen  $(NO_X)$  and oxygen  $(O_2)$ . In addition, the probe tip temperature, ambient air temperature and wind speed and direction were Integrated samples of the flare gas were collected for hydrocarbon specie analysis by gas chromatograph. Particulate matter samples were collected during the smoking flare tests. Sulfur was used as a tracer material in an effort to determine the dilution of the flare gas between the flare burner and the sampling probe location. However, the implementation of this untried sulfur balance method for determining dilution ratios encountered several difficulties. An alternate method of



determining dilution ratios using the CO<sub>2</sub> concentration data was substituted for the sulfur balance method.

The rigorous test program included flare testing under thirty-three different operating conditions during a three-week period in June 1982. Test variables included Btu content of the flare gas (propylene diluted with nitrogen), flare gas flow rates, steam flow rates and air flow rates. The range of flare gas heating values was 80 to 2,183 Btu/scf. Steam-to-flare gas ratios varied from 0:1 to 123:1. When the flares were operated under conditions which were representative of industrial operating practices, the combustion efficiencies at the sampling probe were determined to be greater than 98%. Combustion efficiencies were observed to decline under conditions of excessive steam (steam quenching) and high flow rates of low Btu gases. Table 1.1 summarizes the results of the thirty-three flare efficiency tests.

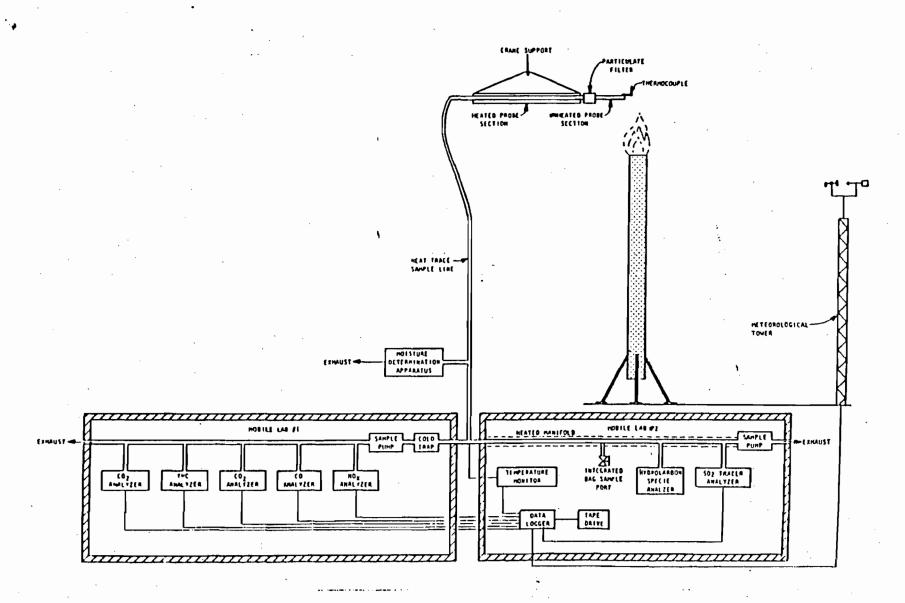
#### CONCLUSIONS AND OBSERVATIONS

- Flares are generally an efficient means of hydrocarbon disposal over a wide range of operating conditions.
- Excess steam may contribute to lower combustion efficiencies.
- Flaring high volumes of low heating value gases may result in lower combustion efficiencies.
- Smoking flares do not necessarily indicate inefficient combustion.
- Although the use of sulfur as a tracer material shows promise, further development of the techniques are required.
- When the flares were operated under conditions that represent typical industrial operations, the combustion efficiencies observed at the sampling probe were equal to or greater than those commonly found in ambient air.

TABLE 1.1
FLARE EFFICIENCY TEST RESULTS

	Fla	re Gas		· · · · · · · · · · · · · · · · · · ·		
Test Number	Flow (SCFM)	Heating Value (Btu/SCF)	Steam-to-Flare Gas Ratio (Lb/Lb)	Combustion Efficiency (%)	Comments	
STEAM-ASSIS	TED FLARE TEST	\$				
2	464	2183	0.508	99.82		
3	456	2183	0.448	99.82	Incipiant emphise floor	
2 3 1	473	2183	0.688	99.96	Incipient smoking flare	
5	149	2183	1.56	99.94		
67	148	2183	0.725	••	Sampling probe in flare flame	
7 .	154	2183	0.757	99.84	Incipient smoking flare	
17	24.5	2183	0.926	99.84	and ip reme , smoking it rare	
50	24.4	2183	3.07	99.45		
51	325	309	0.168	98.66	·	
23	0.494	267		100.01		
52 .	0.556	268	77.5	98.82	•	
53	0.356	209	123	99.40		
54	0.356	209		99.90		
4	283	2183		99.80	Smoking flare	
8	157	2183		98.81	Smoking flare	
55	24.7	2183	6.86	68.95	Steam-quenched flare	
56	24.5	2183	3.45	<b>9</b> 9.70		
lla	. <b>6</b> 60	305	,	99.79		
11b 11c	.599· 556	342		99.86		
57	703	364	2.152	99.82	£ .	
16a	320	294 339	0.150	99.90	•	
166	<b>25</b> 2		••	99.73	No smoke	
16c	194	408		99.75	No smoke	
16d	159	519		99.74	Incipient smoking flare	
59a	591	634		99.78	Smoking flare	
59b	496	192 `		97.95	• .	
60	334	232 298		99.33		
61	25.0	2183		98.92	************	
		2103	5.67	82.18	Steam-quenched flame	
	FLARE TESTS.			• .		
28	157	2183		99.94		
31	22.7	2183		99.17		
26a	481.6	2183		100.00		
26b	481.6	2183		99.95		
66	639	158 .	•-	61.94	Detached flame observed	
29a	<b>510</b> . :	168		54.13	Detached flame; no air assistance	
29ь	392	146		64.03	Detached flame; with air assistant	
33	0.714	83		98.24		
32a	0.556	294		98.94		
32b	0.537	228		98.82	•	
62	217 •	153	. ••	94.18	Flame slightly detached	
64	249	282	••	99.74	3.2.2.7	
63	121	289		99.37		
65	159	2183		99.57	Smoking flaré; no air assistance	

### FLARE SAMPLING AND ANALYSIS SYSTEM



ATTACHMENT C
HARRIS SEMICONDUCTOR
CURRENT AIR PERMITS

CURRENT AIR PERMITS--HARRIS SEMICONDUCTOR--AUGUST 21, 1987

PERMIT NO.	SCRUB #	ISSUED	EXPIRES
-	Marie Marie Server Marie serve Marie America Appete Server Appete Server Server	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	
•	•		
AC 05-104512	F63S02	01/15/86	06/30/86
AC 05-104513	F62802	01/15/86	06/30/86
AC 05-104515	F59803	01/15/86	06/30/86
AC 05-10 <b>4519</b>	F61S01	01/15/86	06/30/86
AC 05-104521	F58S01	01/15/86	06/30/86
AC 05-104522	F57S01	01/15/86	06/30/86
AC 05-104523	F55S01	01/15/86	06/30/86
AC 05-104524	F04S03	01/15/86	06/30/86
AC 05-104525	F04S01	01/15/86	06/30/86
AC 05-104527	F58S02	01/15/86	06/30/86
AD 05-108260	F63803	02/28/86	06/30/86
AD 05-109845	N/A (F04S07)	11/05/85	10/30/90
AD 05-109846	N/A (F04506)	11/05/85	10/30/90
AD 05-109850	F04S04 .	11/05/85	10/30/90
AO 05-109852	N/A (F04S08)	11/05/85	10/30/90
AD 05-109853	F51S02	11/05/85	10/30/90
	F51S03	11/05/85	10/30/90
•			
AO 05-109855	F51S04	11/05/85	10/30/90
AD 05-115803	. F04S02	05/20/86	05/22/91
AD 05-115804	F54S03	05/20/86	05/22/91
	F54S04	05/20/86	05/22/91
AD 05-117084	F60S01	05/20/86	05/22/91
AD 05-117085	F51S01	05/20/86	05/22/91
AO 05-121924	F59S01	09/17/86	09/14/91
AO 05-121927	F62S01	09/18/86	09/14/91
AB 05-121930	F63S01	09/16/86	09/14/91
AD 05-121934	F04805	09/16/86	09/14/91
AD 05-121939	F61802	09/16/86	09/14/91
AO 05-65408	F54S01	05/03/83	05/02/88
	F54S02	05/03/83	05/02/88
A0 05-71405	F51S05	09/13/83	09/12/88

ATTACHMENT D

HARRIS SEMICONDUCTOR

EMISSIONS CÁLCULATIONS

#### Calculations - Flare Permit

$$H_2S + {}^3O_2 = SO_2 + H_2O.$$
  
 $2H_2S + 3 O_2 = 2 SO_2 + 2 H_2O.$ 

$$1 \text{ mol } H_2S = 34g$$
  
 $1 \text{ mol } SO_2 = 64g$ 

grams x 
$$(2.205 \times 10^{-3}) = 1bs$$

at 100% efficiency (assuming all H<sub>2</sub>S converted:)

=  $3201.3 \text{ g/hr } S0_2 \text{ emited.}$ 

$$(3201.3 \text{ gSO}_2)$$
  $(2.205 \times 10^{-3} \text{ lb}) = 7.059 \text{ lbs}$  max rate of SO<sub>2</sub> emitted

$$(7.059 \ \frac{1\text{bs } SO_2}{\text{hr}}) \ ( \frac{1 \ \text{Ton}}{2.0 \ \text{x} \ 10^3 \ \text{lb}}) \ ( \frac{24 \ \text{hr}}{1 \ \text{day}}) \ ( \frac{365 \ \text{day}}{\text{yr}})$$

$$= 30.92 \text{ Ton/day } S02$$

H<sub>2</sub>S: : Based on 98.5% Efficiency of flare sytem:

$$(32850 \frac{1\text{bs H}_2\text{S}}{\text{year}})$$
  $(0.985) = 32357.25 \frac{1\text{bs}}{\text{yr}}$  of H<sub>2</sub>S converted

32850 - 32357.25 = 492.75 
$$\frac{1bs}{yr}$$
 potential H<sub>2</sub>S emission

$$(492.75 \frac{1bs}{yr}) (\frac{1 yr}{365 day}) (\frac{1 day}{24 hr}) = 0.05625 \frac{1b}{yr}$$

$$(492.75 \frac{1bs H_2S}{yr}) (\frac{1 Ton}{2.0 \times 10^3 1b}) = 0.246 \frac{Ton}{year}$$

$$\frac{\text{CO}_2}{\text{yr}}$$
 (438.000  $\frac{1\text{bs CO}_2}{\text{yr}}$ ) (0.985) = 431430  $\frac{1\text{bs CO}_2}{\text{yr}}$  removed

438.000 - 431430 = 6570 <u>lbs CO<sub>2</sub></u> potential emission

$$(6570 \frac{1 \text{bs } CO_2}{\text{yr}}) \quad (\frac{1 \text{ yr}}{365 \text{ day}}) \quad (\frac{1 \text{ day}}{24 \text{ hr}}) = 0.75 \frac{1 \text{b } CO_2}{\text{hr}}$$

$$(6570 \ \frac{1b}{yr}) \ ( \frac{1 \ Ton}{2.0 \ x \ 10^3 \ 1b} ) = 3.285 \ \frac{Ton}{yr} \ CO_2$$

(2.5#/hr propane)  $(\frac{\text{ft}^3}{0.117\#})$  = 21.37 SCFH Enrichment Fuels: 1# Propane = 21,591 BTU

(21591 BTU) = 53977.5 BTU/hrEnrichment gas--> (2.5 1b/hr)

Pilot gas--> (43 FT
$$^3$$
/hr) (0.11 $^7$  1b) (21591 BTU) == 108624.32 BTU hr

C<sub>3</sub>H<sub>8</sub> at STP --> 
$$(1.8324 \text{ g})$$
  $(\frac{1 \times 10^{-3} \text{ L}}{1 \text{ m}^3}) = 1.83 \times 10^{-3} \frac{\text{g}}{\text{m}^3}$ 

Gas Flow Rate: 
$$H_2S = 0.0892 \text{ lb}$$

$$(90 \text{ lb H}_2\text{S})$$
  $(1 \text{ day})$   $(1 \text{ hr})$   $(60 \text{ min})$   $(60 \text{ min})$ 

$$(\frac{1200 \text{ lb } \text{CO}_2}{\text{day}})$$
  $(\frac{1 \text{ day}}{24 \text{ hr}})$   $(\frac{1 \text{ hr}}{60 \text{ min}})$   $(\frac{\text{ft}^3}{0.1144 \text{ lb}})$  = 7.284 CFM CO<sub>2</sub>

0.7007 CFM + 7.284 CFM = 7.985 CFM Gas

Velocity:

(based on 2" Inlet from degasifier to flare) 2" Pipe area =  $0.0233 \text{ ft}^2$ 

$$\frac{7.985 \text{ ft3/min gas}}{0.0233 \text{ ft}^2}$$
 = 342.7 ft/min + 60 = 5.712 ft/sec

#### TABLE 2

Sample Designation: New Feed Water Well for 58

Sample Analyzed by: S. Slasor, K. Hanley - Harris Corporation

Sample Analyzed: 8/08/86

# Total Dissolved Solids Suspended Solids Total Dissolved Solids 1.2-1.7 mg/l

Total Hardness as CaCO3 659 mg/l
Calcium 122 mg/l
Magnesium 86 mg/l
Sodium 381 mg/l
Iron < 0.1 mg/l
Manganese < 0.1 mg/l

Manganese < 0.1 mg/l
Sulphate 117 mg/l
Alkalinity 145 mg/l
Bicarbonate Alkalinity < 0.5 mg/l
Carbonate Alkalinity < 0.5 mg/l

 Chloride
 870 mg/l

 Copper
 < 0.1 mg/l</td>

 Lead
 < 0.1 mg/l</td>

 Barium
 < 0.5 mg/l</td>

 Silica
 21.7 mg/l

Total Organic Carbon 9.4 mg/l  $\times$  1 mg/l  $\times$  2 mg/l Potassium 8.5 mg/l Hydrogen Sulfide 12.4-12.6 mg/l  $\times$  2 carbon Droxide  $\times$  0.1 mg/l

pH 7.35
Temperature 25.5°C

Conductivity, mucromho/cm 3330

# RO UNIT PERFORMANCE PROJECTION using "ROPRO" (c) v3.03U (1/16/87) Provided to the San Diego Office by Fluid Systems Division of UOP Inc.

PROJECT: PILOT DATE: 7/9/87 The unit has 32 Model 8031MP MAGNUM elements which are 1 yrs. old. 5 / 3 with 4 element tubes The Array is Permeate Flow = 288000. gpd ( 200.0 gpm) at 75.0% recovery. Feed Temp. = 25.0 C ( 77.0 F) Avg. annual unit Temp. = 25.0 C ( 77.0 F) Brine Press. = 339.5 psi Brine Osmotic Press. = 82.4 psi = 378.9 psi Feed Press. Feed Osmotic Press. = 21.2 psi The ratio of the concentration in the brine to the saturation level for CaSO4 is 1.16 SiO2 is .52 SrS04 is 2.52

If the precipitation inhibitor addition is interrupted for even a short time, the elements may become irreversibly fouled with CaSO4.

BANK	FEE	D	CONCEN	ITRATE	AVERAGE	TUBE	FINAL	ELEMENT
	TOTAL	TUBE	TOTAL	TUBE	ELEMENT	DELTA P	BETA	% RECOVERY
	gpm	gpm	gpm	gpm	gpd	psi		•
1	266.7	53.3	134.6	26.9	9510.9	22.4	1.099	18.9
2	134.6	44.9	66.8	22.3	8127.2	17.1	1.098	18.8

	RAW FEED	PRETREATED FEED	CONCENTRATE	PERMEATE
	mg/L	mg/L	mg/L	mg/L
· C -	<del>-</del>	_		<del>-</del>
Ca	310.	310.	1233.	1.4
Mg	80.	80.	318 <b>.</b>	. 4
Na	405.	405.	1551.	21,7
K	9 <b>.</b>	9.	33.	.5
NH4	o.	O	o. `	.0
CO3	ο.	0.	· o.	.0
HCO3	64.	64 <b>.</b>	253 <b>.</b>	4.9
SO4	572 <b>.</b>	572.	2281.	.2
Cl	1120.	1120.	4358 <b>.</b>	36.8
NO3	1.	1.	4.	. 1
F	2.	2.	7.	.0
Si 02	18.	18.	45 <b>.</b>	2.8
Sr	13.40	13.40	53.28	06
SUM	2594.	2594.	10156.	64.8
TDS	2562.	2562.	10029.	64.3
CO2	542.	542.	542.	539.
pН	5.3	5.3	5.9	4.2
pHs		7.3	6.7	

This projection is the anticipated performance and is based on nominal properties of the elements. No allowance was made for fouling or for pressure losses in the manifolds.

This computer printout should not be considered a guarantee of system performance unless accompanied by a statement to that effect.

# ATTACHMENT E HARRIS SEMICONDUCTOR MATERIAL SAFETY DATA SHEETS



Division of The BOC Group, Inc.

575 Mountain Avenue Murray Hill New Jersey 07974 Telephone: 201-464-8100 TWX: 710-984-7970

### MATERIAL SAFETY DATA SHEET

Welding Consumables and Related Products Conforms to OSHA 1910.1200

#### IDENTIFICATION

PRODUCT NAME: Hydrogen Sulfide

CHEMICAL FAMILY: Nonmetal Hydride

SYNONYMS: Dihydrogen Sulfide

DOT HAZARD CLASS: Flammable Gas

CAS NUMBER: 7783-06-4

DOT IDENTIFICATION NUMBER: UN 1053

FORMULA: H2S

CHEMTREC: 800-424-9300

#### HEALTH HAZARD DATA

#### TIME WEIGHTED AVERAGE EXPOSURE LIMIT:

10 Molar PPM; STEL = 15 molar PPM (ACGIH, 1984-85). 20 PPM - Ceiling, 50 PPM - Peak for 10 minutes (OSHA).

#### SYMPTOMS OF EXPOSURE:

Continuous exposure to low (15-50 PPM) concentrations will generally cause in ititation to mucous membranes and conjunctivae of the eyes. It may also cause headache, dizziness or nausea. Higher concentrations (200-300 PPM) can tesult in respiratory arrest leading to come or unconsciousness. Exposures for more than 30 minutes at concentrations of greater than 700 PPM have been fatal. Continuous inhalation of low concentrations may cause olfactory fatigue or paralysis rendering the detection of its presence by odor ineffective.

#### TOXICOLOGICAL PROPERTIES:

Inhalation of hydrogen sulfide it is highly toxic. It is also an irritant to mucous tissue, membranes and the conjunctivae of the eyes. Continued exposure renders the olfactory sensors inopertive. Toxicologically its reaction with enzymes in the bloodstream inhibit cell respiration resulting in pulmonary paralysis, sudden collapse and death. This overshadows its irritant effect on mucous membranes and tissues which at worst will cause pulmonary edema or conjunctival lesions. Repeated exposures to low concentrations is reported to cause conjunctivites, photophobea, corneal bullae, tearing, pain and blurred vision.

#### RECOMMENDED FIRST AID TREATMENT:

PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO HYDROGEN SULFIDE. RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS. RESCUE PERSONNEL SHOULD RECOGNIZE THE HAZARDS OF OVEREXPOSURE DUE TO OLFACTORY FATIGUE.

Inhalation: Extreme fire hazard when rescuing semi-conscious or unconscious persons due to flammability of hydrogen sulfide. Avoid use of rescue equipment which might contain ignition sources or cause static discharge. Move affected person to any uncontaminated area. If breathing has stopped, give assisted respiration. Oxygen or a mixture of 5% cargon dioxide in oxygen should be administered by a qualified person. Keep the victim warm and calm. Seek immediate medical assistance. Further treatment should be symptomatic and supportive.

Eye Contact: PERSONS WITH POTENTIAL EXPOSURE TO HYDROGEN SULFIDE SHOULD NOT WEAR CONTACT LENSES.

Flush contaminated eye(s) with copious quantities of water. Part eyelids with fingers to assure complete flushing. Continue for minimum of 15 minutes.

#### Hazardous Mixtures of Other Liquids, Solids, or Gases:

Hydrogen sulfide will explode or burn over a wide range or mixtures in air. It becomes dangerously reactive when mixed with nitric acid or other strong oxidizers such as sulfuric acid. Vapors will combust spontaneously when mixed with vapors of chlorine, oxygen difluoride or nitrogen trifluoride.

#### PHYSICAL DATA

Boiling Point: -76°F (-60°C)

Liquid Density @ Boiling Point: 57.1 lb/ft<sup>3</sup> (915 kg/m<sup>3</sup>)

Vapor Pressure @ 70°F (21.1°C): 267 psia (1840 kPa)

Specific Gravity @ 70°F, 1 atm (Air=1): 1.21

Solubility in Water: Soluble

Freezing Point: -117°F (-82.8°C)

Appearance and Odor: Shipped and stored as a liquid under its own vapor pressure. Vapor is colorless with a characteristic "rotten egg" odor.

#### FIRE/EXPLOSION HAZARDS DATA

Flash Point (Method Used): Gas

Auto Ignition Temperature: 554° (290°C)

LEL: 4.0

UEL: 44.0

Extinguishing Media: Carbon dioxide, dry chemical or water spray

Electrical Classification: NEC Class 1

Special Fire Fighting Procedures: Shut off flow of gas. Cool surrounding fire-exposed containers with water spray. Fire fighters should use self-contained breathing apparatus.

Unusual Fire and Explosion Hazards: Hydrogen sulfide is heavier than air so may accumulate in low spots and may "travel" a considerable distance to a flame or other source of ignition.

#### REACTIVITY DATA

Stability: Stable

Conditions to Avoid: Avoid heat, flame or other sourcs of ignition.

Incompatibility (Materials to Avoid): Concentrated nitric acid, chlorine, nitrogen trifluoride, oxygen difluoride or other strong oxidizing agents.

Hazardous Decomposition Products:

Oxides of sulfur

Hazardous Polymerization: Will not occur

Conditions to Avoid: None

#### SPILL OR LEAK PROCEDURES

#### STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED:

Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in container or container valve, contact CHEMTREC for emergency assistance or call your closest Airco location.

#### Waste Disposal Method:

Do not attempt to dispose of waste or unused quantities. Return in the shipping container properly labeled, with any valve outlet plugs or caps secured and valve protection cap in place to Airco for proper disposal.

#### SPECIAL PROTECTION INFORMATION

Respiratory Protection: Positive pressure air line with mask or self-contained breathing apparatus should be available for emergency use.

Ventilation: Hood with forced ventilation.

local Exhaust: To prevent accumulation above the TWA.

Special: None

Mechanical (Gen.): None

Other None

Protective Gloves: Neoprene or butyl rubber, PVC, polyethylene.

Eye Protection: Safety goggles or glasses.

Other Protective Equipment: Safety shoes, safety shower, eyewash "fountain".

#### SPECIAL PRECAUTIONS

#### Special Labeling Information:

DOT Shipping Name: Hydrogen sulfide (RQ-100/45.4)
DOT Shipping Label: Flammable Gas and Poison

DOT Hazard Class: Flammable Gas

1.D. No.: UN 1053

#### Special Handling Recommendations:

Use only in well-ventilated areas. Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure (<750 psig) piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder.

For additional handling recommendations, consult Compressed Gas Association Pamphlets P-1 and G-12.

#### Special Storage Recommendations:

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 130°F (54°C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in-first out" inventory system to prevent full cylinders being stored for excessive periods of time. Post "No Smoking or Open Flames" signs in the storage or use area. There should be no sources of ignition in the storage or use area.

For additional recommendations, consult Compressed Gas Association Pamphlets P-1 and G-12.

#### Special Packaging Recommendations:

Many metals corrode rapidly with wet hydrogen sulfide. Anhydrous (water content -40f or C) hydrogen sulfide can be handled in carbon steel, aluminum, Inconel®, Stellite® and 304 and 316 stainless steels. Avoid hard steels which are highly stressed since they may be susceptible to hydrogen embrittlement from hydrogen sulfide.

#### Other Recommendations or Precautions:

Earth-ground and bond all lines and equipment associated with the hydrogen sulfide system. All electrical equipment should be non-sparking or explosion proof. Do not rely on the olfactory sense to detect the presence of hydrogen selenide. Analytical devices and instumentation are readily available for this purpose. Perform frequent analytical tests to be certain that the TWA is not exceeded.

Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with his (written) consent is a violation of Federal Law (49CFR).



Division of The BOC Group, Inc.

575 Mountain Avenue Murray Hill New Jersey 07974 Telephone: 201-464-8100 TWX: 710-984-7970

### MATERIAL SAFETY DATA SHEET

Welding Consumables and Related Products Conforms to OSHA 1910.1200

#### IDENTIFICATION

PRODUCT NAME: Sulfur Dioxide

CHEMICAL FAMILY: Inorganic Acid

SYNONYMS: Sulfurous Acid Anhydride

DOT HAZARD CLASS: Nonflammable Gas

CAS NUMBER: 7446-09-5

DOT IDENTIFICATION NUMBER: UN 1079

FORMULA: SO2

CHEMTREC: 800-424-9300

#### HEALTH HAZARD DATA

#### TIME WEIGHTED AVERAGE EXPOSURE LIMIT:

2 Molar PPM (ACGIH, 1984-85). 5 Molar PPM (OSHA, 1984-85).

Note: Prior to the 1984-85 issue of ACGIH's "TLVs Threshold Limit Values for Chemical Substances and Physical Agents in the Work Environment and Biological Exposure Indices with Intended Changes", Sulfur Dioxide had a STEL of 5 Molar PPM. This STEL value is deleted in the 1984-85 issue.

#### SYMPTOMS OF EXPOSURE:

Corrosive and irritating to the upper and lower respiratory tracts, skin and eyes. Symptoms depend on the concentration and duration of exposure and vary from mild irritation to severe destruction of tissues. They may also include burning sensations, coughing, wheezing, laryngitis, shortness of breath, headache, nausea, and vomiting. If the sulfur dioxide penetrates the lower airway, it can produce bronchitis, chemical pneumonitis and pulmonary edema. Eye contact results in pain, lacrymation, inflammation, swelling of tissue and possible destruction of the eye. Skin contact causes irritation or chemical-like burns. Contact with rapidly evaporating liquid can cause cryogenic "burns" or frostbite.

#### TOXICOLOGICAL PROPERTIES:

Inhalation human  $TC_{LO} = 3 PPM/5 days$ 

Exposure to atmospheres contaminated with sulfur dioxide is extremely irritating. Its odor and prompt irritant action provide a warning of exposure to toxic conditions. High concentrations are extremely destructive to tissues of the airway, eyes and skin. Inhalation may have fatal consequences as a result of spasm, inflammation and edema of the larynx and bronchi, chemical pneumonitis and pulmonary edema. Exposure of the eyes to high concentrations may result in ulceration of the conjunctiva and cornea and destruction of all ocular tissues. Contact with the skin causes severe burns. Systemic toxicity due to sulfur dioxide is not known to occur. Frost-bite effects are a change in color of the skin to gray or white, possibly followed by blistering.

#### RECOMMENDED FIRST AID TREATMENT:

PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO SULFUR DIOXIDE. RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.

Inhalation: Conscious persons should be assisted to an uncontaminated area and inhale fresh air. Unconscious persons should be moved to an uncontaminated area, and given mouth-to-mouth resuscitation and supplemental oxygen. Keep the victim warm and quiet. Assure that mucous or vomited material does not obstruct the airway by positional drainage. The physician should be informed that the patient has inhaled acidic vapors.

Eye Contact: PERSONS WITH POTENTIAL EXPOSURE TO SULFUR DIOXIDE SHOULD NOT WEAR CONTACT LENSES.

Flush contaminated eye(s) with copious quantities of water. Part eyelids to assure complete flushing. Continue for a minimum of 15 minutes.

Skin Contact: Flush affected area with copious quantities of water. Remove affected clothing as rapidly as possible.

Dermal Contact or Frostbite: Remove contaminated clothing and flush affected areas with lukewarm water. DO NOT USE HOT WATER. A physician should see the patient promptly if the cryogenic "burn" has resulted in blistering of the dermal surface or deep tissue freezing.

#### Hazardous Mixtures of Other Liquids, Solids, or Gases:

Sulfur dioxide reacts violently with peroxides, chromates, bichromates, permanganates and oxygen difluoride. It also reacts with chlorates to form chlorine which at elevated temperatures may become an exlosive reaction.

#### PHYSICAL DATA

Boiling Point: 14.0°F (-10.0°C)

Liquid Density @ Boiling Point: 91.1 lb/ft<sup>3</sup> (1460 kg/m<sup>3</sup>)

Vapor Pressure @ 70°F (21.1°C): 49.1 psia (339 kPa)

Specific Gravity @ 70°F, 1 atm (Air=1): 2.26

Solubility in Water: Soluble

Freezing Point: Gas Mixtures: -103.9°F (-75.5°C)

Appearance and Odor: Colorless gas with highly irritating, pungent odor.

#### SPECIAL PROTECTION INFORMATION

#### Respiratory Protection:

Positive pressure air line with mask or self-contained breathing apparatus should be available for emergency use.

Ventilation: Hood with forced ventilation

Local Exhaust: To prevent accumulation above the TWA.

Special: N/A

Mechanical (Gen.): N/A

Other: N/A

Protective Gloves: Plastic or rubber

Eye Protection: Safety goggles or glasses.

Other Protective Equipment: Safety shoes, safety shower, eyewash "fountain",

face shield.

#### SPECIAL PRECAUTIONS

#### Special Labeling Information:

DOT Shipping Name: Sulfur dioxide DOT Shipping Label: Nonflammable gas DOT Hazard Class: Nonflammable gas

I.D. No.: UN 1079

#### Special Handling Recommendations:

Use only in well-ventilated areas. Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure (<150) psig) piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder.

For additional handling recommendations, consult Compressed Gas Association's Pamphlet P-1 and G-3.

#### Special Storage Recommendations:

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 130°F (54°C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in-first out" inventory system to prevent full cylinders being stored for excessive periods of time.

For additional storage recommendations, consult Compressed Gas Association's Pamphlet P-1 and G-3.

#### FIRE/EXPLOSION HAZARDS DATA

UEL:

None

Flash Point (Method Used): None

Auto Ignition Temperature: None

Extinguishing Media: Nonflammable gas

Electrical Classification: Nonhazardous

Special Fire Fighting Procedures: None

Unusual Fire and Explosion Hazards: None

#### REACTIVITY DATA

Stability: Stable

LEL: None

Conditions to Avoid: None

Incompatibility (Materials to Avoid): Oxidizing materials.

Hazardous Decomposition Products: None

Hazardous Polymerization: Will not occur

#### SPILL OR LEAK PROCEDURES

#### STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED:

Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in container or container valve, contact CHEMTREC for emergency assistance or your closest Airco location.

#### Waste Disposal Method:

Do not attempt to dispose of waste or unused quantities. Return in the shipping container properly labeled, with any valve outlet plugs or caps secured and valve protection cap in place to Airco for proper disposal.



#### Special Packaging Recommendations:

Most metals corrode rapidly with wet sulfur dioxide.

#### Other Recommendations or Precautions:

Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with his (written) consent is a violation of Federal Law (49CFR).



TITLE:

CARBON DIOXIDE

CONTROLLED

DOCUMENT

### PROPRIETARY DATA

VERTICAL BAR IN RIGHT MARGIN INDICATES REVISION.

CHEMICAL SAFETY
SPECIFICATION

SPECIFICATION NUMBER
CODE 856068

REVISION

IDENT. NO.

34371

PAGE

1 of :/

#### 1.0 PURPOSE

This specification defines the chemical safety requirements for Carbon Dioxide supplied by the applicable manufacturer(s) or distributor(s) of the product.

TITLE		REVISION	SPECIFICATION NUMBER
CARBON DIOX	IDE	1	856068
MARRIS	THIS DOCUMENT CONTAINS PROPRIETARY INFORMATION OF MARRIS SEMICONDUCTOR AND IS TENDERED SUBJECT TO THE CONDITIONS THAT THE INFORMATION (A) BE RETAINED IN CONFIDENCE, (B) NOT BE REPRODUCED OR COPIED IN WHOLE OR IN PART, AND (C) NOT BE RELEASED OUTSIDE	CODE	1-8-86
SEMICONDUCTOR	HARRIS SEMICONDUCTOR WITHOUT THE EXPRESS APPROVAL OF THE GENERAL MANAGER HARRIS SEMICONDUCTOR. ADDITIONAL RESTRICTIONS ON THE USE OF THIS INFORMA- TION MAY BE IMPOSED BY THE CONTRACT, PROPOSAL OR	34371	PAGE 3 OF 7

**L-4674-B** July 1986

#### MATERIAL SAFETY DATA SHEET

An explanation of the terms used herein may be found in OSHA 29 CFR 1910.1200, greateste from OSHA regional or area offices.

(Similar to U.S. Department of Labor Form OMB No. 1218-0072 and generally accepted in Canada for information purposes) Do Not Duplicate This Form. Request an Original.

PRODUCT	Carbon Dioxide -		
CHEMICAL NAME	Carbon Dioxide	SYNONYMS	Carbonic Anhydride, Carbonic Acid Gas
FORMULA	CO1	CHEMICAL FAMILY	Acid Anhydride
		MOLECULAR WEIGHT	44.01

والأكلف والمراج والمستعلق المعاري والأبار والمستعلق وأراد المراز والروايا

TRADE NAME Carbon Dioxide

For mixtures of this product request the respective component Material Data Safety Sheets. See Section IX.

MATERIAL (CAS NO.)	- Wt (%)	1985-1986 ACGIH TLV-TWA (OSHA-PEL	
Carbon Dioxide (124-38-9)	100	5000 ppm	(5000 ppm)
-			

SUBLIMATION POINT, 760 mm. Hg	-78.5°C (-109.3°F)	FREEZING POINT	Not applicable
SPECIFIC GRAVITY (H <sub>2</sub> O = 1)	Not applicable	VAPOR PRESSURE AT 21°C.	830 psig
VAPOR DENSITY (air = 1)	1.522 @ 21°C	SOLUBILITY IN WATER, % by wt.	Slight
PERCENT VOLATILES BY VOLUME	100	EVAPORATION RATE (Butyl Acetate = 1)	High

APPEARANCE AND ODOR Colorless gas at normal temperature and pressure; odorless.

IN CASE OF EMERGENCIES involving this material, further information is available at all times: In the USA 1-800-UCC-HELP (1-800-822-4357) In Canada 514 — 645-5311 For routine information contact your local supplier

Union Carbide requests the users of this product to study this Material Safety Data Sheet (MSDS) and become aware of product hazards and safety information. To promote safe use of this product a user should (1) notify its employees, agents and contractors of the information on this MSDS and any product hazards and safety information. (2) furnish this same information to each of its customers for the product, and (3) request such customers to notify their employees and customers for the product of the same product

hazards and safety information.

UNION CARBIDE CORPORATION . LINDE DIVISION UNION CARBIDE CANADA LIMITED . LINDE DIVISION

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Page 1 of 4

TITLE		REVISION	SPECIFICATION NUMBER
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PRODUCT: Carbon Dloxide

L-4574-B July 1986

THRESHOLD LIMIT VALUE: 5,000 ppm - ACGIH (1985-86).

#### EFFECTS OF A SINGLE (ACUTE) OVEREXPOSURE:

SWALLOWING — A highly unlikely route of exposure. Frostbite of the lips and mouth may result from contact with the liquid.

SKIN ABSORPTION — No evidence of adverse effects from available information.

بالمعوان والمستهجين والسابات أبارا المارا المارات 

INHALATION - Asphyxiant. Moderate concentrations may cause headache, drowsiness, dizziness, stinging of the nose and throat, excitation, rapid breathing, excess salivation, vomiting, and unconsciousness. Lack of oxygen can cause death.

SKIN CONTACT - No harmful effect expected from vapor. Liquid may cause frostbite.

EYE CONTACT — Vapor may cause a stinging sensation; liquid may cause trostbite.

EFFECTS OF REPEATED (CHRONIC) OVEREXPOSURE: No evidence of adverse effects from available information.

OTHER EFFECTS OF OVEREXPOSURE: Damage to retinal ganglion cells and central nervous system may occur.

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: A knowledge of the available toxicology information and of the physical and chemical properties of the material suggest that overexposure is unlikely to aggravate existing medical conditions.

SIGNIFICANT LABORATORY DATA WITH POSSIBLE RELEVANCE TO HUMAN HEALTH HAZARD EVALUATION: None currently known.

#### EMERGENCY AND FIRST AID PROCEDURES:

SWALLOWING - This product is a gas at normal temperature and pressure.

SKIN CONTACT - For exposure to liquid, immediately warm frostbite area with warm water (not to exceed 105°F). In case of massive exposure, remove clothing while showering with warm water. Call a physician.

INHALATION — Remove to fresh air. Give artificial respiration if not breathing. Give oxygen if breathing is difficult. Call a physician.

EYE CONTACT — In case of splash contamination, immediately flush eyes thoroughly with water for at least 15 minutes. See a physician, preferably an ophthalmologist, immediately.

NOTE TO PHYSICIAN: There is no specific antidote. Treatment of overexposure should be directed at the control of symptoms and the clinical condition.

Page 2 of 4

CARBON DIC	DXIDE	REVISION	SPECIFICATION NUMBER 856068
HARRIS SEMICONDUCTOR	THIS DOCUMENT CONTAINS PROPRIETARY INFORMATION OF MARRIS SEMICONDUCTOR AND IS TENDERED SUBJECT TO THE CONOLITIONS THAT THE INFORMATION IS HE RETAINED IN CONFIDENCE, IBI NOT BE REPRODUCED OR COPIED IN WHOLE OR IN PART, AND IC! NOT BE RELEASED OUTSIDE HARRIS SEMICONDUCTOR WITHOUT THE EXPRESS APPROVAL OF THE GENERAL MANAGER, HARRIS SEMICONDUCTOR, ADDITIONAL RESTRICTIONS ON THE USE OF THIS INFORMATION INAY BE IMPOSED BY THE CONTRACT, PROPOSAL OR OTHER AGREEMENT OF WHICH THIS SHEET IS A PART.	CODE IDENT. NO. 34371	PAGE 5 OF

PRODUCT:

**Carbon Dioxide** 

L-4574 B July 1986

FLASH POINT

Not applicable

AUTOIGNITION TEMPERATURE

Not applicable

FLAMMABLE LIMITS LOWER IN AIR, % by volume

Not applicable

والمحاورة والمراكبة أنجير والمحارض ويصفعوني وأواريعي يعيون

Not applicable

EXTINGUISHING MEDIA: Carbon Dioxide cannot catch fire. Use media appropriate for surrounding fire.

SPECIAL FIRE FIGHTING PROCEDURES: Evacuate all personnel from danger area. Immediately deluge containers with water spray from maximum distance until cool, then move containers away from fire area if without risk.

UNUSUAL FIRE AND EXPLOSION HAZARDS: Gas cannot catch fire. Container may rupture due to heat of fire. No part of a container should be subjected to a temperature higher than 52°C (approximately 125°F). Most containers are provided with a pressure relief device designed to vent contents when they are exposed to elevated temperature.

STABILITY | CONDITIONS TO AVOID: See Section IX.

UNSTABLE	STABLE			
٠.	X			

INCOMPATIBILITY (materials to avoid): Alkali metals, alkaline earth metals, metal acetylides, chromium, titanium above 550°C. uranium above 750°C.

HAZARDOUS DECOMPOSITION PRODUCTS: In the presence of an electrical discharge, carbon dioxide is decomposed to form carbon monoxide and oxygen.

<u> </u>	
HAZARDOUS P	OLYMERIZATION
May Occur	Will not Occur

CONDITIONS TO AVOID: None currently known.

#### STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED

Evacuate all personnel from danger area. Use self-contained breathing apparatus where needed. Shut off leak if without risk. Ventilate area of leak or move leaking container to well-ventilated area. Test area, especially confined areas, for sufficient oxygen content prior to permitting re-entry of personnel.

WASTE DISPOSAL METHOD: Slowly release into atmosphere outdoors. Discard any product, residue, disposable container or liner in an environmentally acceptable manner, in full compliance with federal, state and local regulations.

Page 3 of 4

TITLE

CARBON DIOXIDE

REVISION

SPECIFICATION NUMBER

856068

HARRIS SEMICONDUCTOR

THIS DOCUMENT CONTAINS PROPRIETARY INFORMATION OF MARRIS SEMICONDUCTOR AND IS TENDERED SUBJECT TO THE CONDITIONS THAT THE INFORMATION IA BE RETAINED IN CONFIDENCE. 181 NOT SER RELEASED OUTSIDE IN CONFIDENCE. 181 NOT SER ELEASED OUTSIDE WHOLE OR IN PART AND IT HE EXPRESS APPROVAL OF THE SEMICONDUCTOR HARRIS SEMICONDUCTOR HARRIS SEMICONDUCTOR AGE HARRIS SEMICONDUCTOR AGE TO THE USE OF THIS INFORMATION IN A CONTROL OF THE OFFICE ON THE CONTROL OF THIS INFORMATION MAY SEMICH SOME SEMICH THIS SHEET IS A PART.

CODE IDENT. NO. 34371

DATE OF REVISION PAGE

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PRODUCT.

Carbon Dioxide

L-4574-8 July 1986

RESPIRATORY PROTECTION (apecify type): Select in accordance with OSHA 29 CFR 1910.134. Respirators shall be acceptable to MSHA and NIOSH.

LOCAL EXHAUST - Preferred

MECHANICAL (general) - Acceptable

**VENTILATION** 

SPECIAL - Not applicable

OTHER - Not applicable

PROTECTIVE GLOVES: insulated Neoprene

EYE PROTECTION: Select in accordance with OSHA 29 CFR 1910.133

OTHER PROTECTIVE EQUIPMENT: Metatarsal shoes for cylinder handling. Select in accordance with OSHA 29 CFR 1910.132 and 1910.133.

CAUTION: High pressure liquefied gas. Use piping and equipment adequately designed to withstand pressures to be encountered. Can cause rapid suffocation due to oxygen deficiency. Store and use with adequate ventilation. Close valve when not in use and when empty. Carbon dioxide, being heavier than air, tends to accumulate near the floor of an enclosed space displacing the air upward and creates an oxygen-deficient atmosphere. Ventilate space before entry. Verify sufficient oxygen concentration.

MIXTURES: When two or more gases, or liquefled gases are mixed, their hazardous properties may combine to create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an Industrial Hygienist, or other trained person when you make your safety evaluation of the end product. Remember, gases and liquids have properties which can cause serious injury or death.

Be sure to-read and understand all labels and other instructions supplied with all containers of this product.

For safety information on general handling of compressed gas cylinders, obtain a copy of pamphlet P-1, "Safe Handling of Compressed Gases in Containers" from the Compressed Gas Association, Inc., 1235 Jefferson Davis Highway, Arlington, VA 22202.

OTHER HANDLING AND STORAGE CONDITIONS: Never work on a pressurized system. If there is a leak, close the cylinder valve, blow down the system by venting to a safe place, then repair the leak. Store in well ventilated, cool dark place.

The opinions expressed herein are those of qualified experts within Union Carbide. We believe that the information contained herein is current as of the date of this Material Safety Data Sheet. Since the use of this information and these opinions and the conditions of use of the product are not within the control of Union Carbide, it is the user's obligation to determine the conditions of safe use of the product.



GENERAL OFFICES

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CARBON DIOXIDE

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ATTACHMENT F
HARRIS SEMICONDUCTOR
SUPPLEMENTAL INFORMATION

