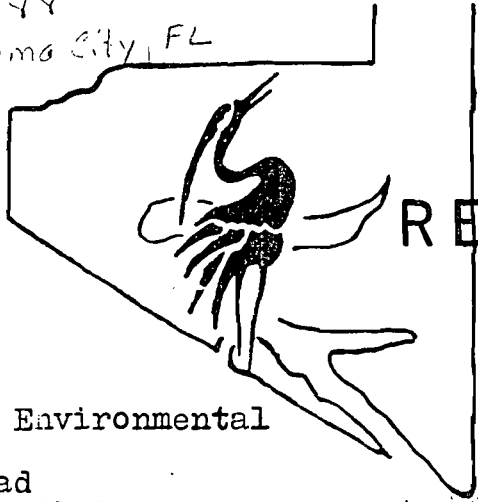


9-17-88
Panama City, FL

Barry

9/20/88

"FROM BIRDWATCHING



TO THE TOTAL ENVIRONMENT"

RECEIVED

September 17, 1988

SEP 20 1988

Mr. Clair Fancy
State Department of Environmental
Regulation
2600 Blair Stone Road
Twin Towers Office Building
Tallahassee, Florida 32301

DER-BAQM

Dear Mr. Fancy:

I am writing to outline the reasons for our objection to the granting of permits nos. AC 03-145061 and AC 03-152196. These permits involve increasing the charging rate of municipal solid waste (MSW) from 350 tons per day to 510 tons per day at the Bay County Resource Recovery Facility.

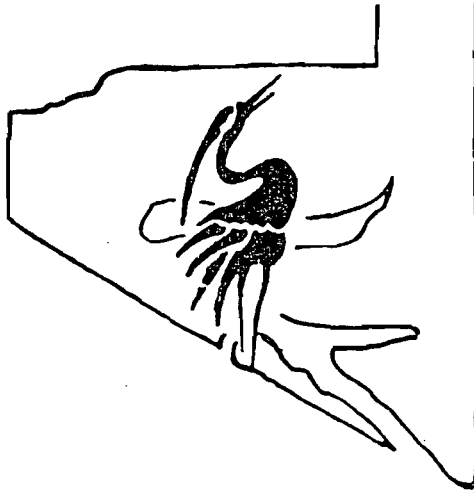
The proposed project will result in a net significant increase in SO₂ emissions as well as increases in particulate matter, carbon monoxide, nitrogen oxides, VOC, lead, mercury, beryllium, fluoride; sulfuric acid mist, and hydrogen chloride. The emission of furans, dioxins, and other carcinogenic compounds will undoubtedly also increase. A BACT review demonstrates that the cost per ton of overall pollutants controlled by dry scrubbers would be \$1863. The EPA has considered costs of up to \$2000 per ton as being reasonable. Yet your agency did not require the Bay County Resource Facility to add these scrubbers.

The Bay County Resource Recovery Facility was not initially intended to be a regional waste to energy facility. Therefore your objection that the time period needed to install additional equipment would be detrimental to attracting waste from surrounding counties is invalid.

The new Steelfield landfill is designed to handle MSW in the event that that the Bay County Resource Recovery Facility is shut down. Therefore your objection to a plant shutdown to install additional pollution control equipment based on the lack of an alternative disposal site for waste is invalid.

If the EPA is in the process of developing a policy in regard to the control equipment requirements for existing municipal waste combustors, then it follows that a decision on this permit should be delayed or denied until the EPA's final policy is released.

"FROM BIRDWATCHING



TO THE TOTAL ENVIRONMENT"

Mr. Clair Fancy
September 17, 1988
Page Two

Many of the pollutants that would increase if the permit were granted will contribute to the acid-rain problem facing our state and nation. Florida currently has 677 acid sensitive lakes, the highest number of any state in the country. The Office of Technology Assessment predicts that acid rain--causing pollutants are responsible for 50,000 premature deaths nationwide every year. Additionally there is evidence that nitrogen oxides may be contributing to the formation of low-lying ozone, which the Clean Air Act has sought to limit.

We are very disappointed with your decision to grant this permit despite the availability and economic feasibility of installing the necessary pollution control equipment. Your apparent lack of concern for the health of the environment and citizens of Bay County is disturbing. Your agency was established to regulate and control pollution, not to grant permits to increase it.

Very truly yours,

BAY COUNTY AUDUBON SOCIETY

BY:

Jeffrey D. Palgut
Jeffrey D. Palgut, Conservation Committee

cc: Robert Kriegel, DER
Nelson B. Kverno, Fish & Wildlife Service, Region 6
Bruce P. Miller, EPA Region IV

Bay County Board of Commissioners

*copied: Pradeep Raval
Barry Andrews
Tom Rogers*

K11.
8-8-88
Pittsburgh, PA



full copy

Westinghouse
Electric Corporation

Resource Energy Systems
Division

Cost Building
2400 Ardmore Boulevard
Pittsburgh Pennsylvania 15221
(412) 636 5800
W.N. 61 5800

EN1351DB

RECEIVED

AUG 12 1988

DER-BAQM

August 4, 1988

Mr. Clair H. Fancy
Deputy Chief
Bureau of Air Quality Management
Florida DER
Twin Towers Office Building
2600 Blair Stone Road
Tallahassee, FL 32399

Dear Mr. Fancy:

We would like to make the following comments and ask a few questions regarding the Technical Evaluation, Preliminary Determination, and proposed permit for the waste-to-energy facility located in Panama City, Florida.

- ✓ 1. There is an error on Page 6 of 11 of the proposed permit, Point No. 3. The flue gas emissions listed in the table are for both units combined, not from each combustor as shown.
- OK 2. The expiration date on the permit is given as December 1, 1988. This date should be changed to April 1, 1989 because the facility will need to install and certify additional continuous emission monitors and to conduct additional stack testing.
- OK 3. Specific Condition No. 1b states that each combustor can burn a maximum of 80 TPD wood waste. We would like the flexibility of being able to burn all of the allowable wood waste in one ~~combustor.~~ In addition, in the unlikely event that less than 350 TPD MSW was available (i.e., 250 TPD), can the facility make up this shortfall by burning wood waste (i.e., 260 tons of wood)?
- No 4. Condition No. 1f states that the facility must be started using fuel oil. We would like to request the flexibility of starting the facility using wood instead of fuel oil.

No

August 4, 1988

- OK
OK
5. Specific Condition No. 4 states that tests must be conducted to measure particulate matter, SO₂, NO_x, CO, VOC, lead, fluorides, mercury, and beryllium. Westinghouse contracted Entropy to conduct sampling in March, 1988 to measure many of these pollutants. Would DER accept some of this data to reduce the amount of testing required by this condition?
- OK
6. Condition No. 3 states that opacity shall not exceed 15%. Is the opacity limit of 15% for six minutes in any one hour?
- No
7. Condition No. 5 requires installing a number of continuous emission monitors. The plant currently has opacity, O₂, and CO monitors installed on each combustor/boiler train. We would like to propose using the O₂ monitor to determine the CO₂ concentration. We have made numerous simultaneous measurements for both O₂ and CO₂, approximately 400 hours of data. These data show that O₂ and CO₂ levels vary linearly. Therefore, the computer could be programmed to compute the CO₂ level from the corresponding reading of O₂. We feel that this would be adequate to determine CO₂ levels.
8. Condition No. 1b states that the plant can burn a maximum of 510 TPD MSW (255 TPD per combustor). The facility keeps records of the amount of waste that is delivered to the site and the amount burned. We would like to request that the first sentence under No. 1b be changed to read "The average throughput for each MWC shall not exceed the design average heat input rate of 95.6 MM Btu/hr, which corresponds to 255 TPD of waste having a HHV of 4500 Btu/lb.

No

The facility will be operated to maintain a steam flow rate of approximately 68,500 lb/hr. The computer control system has a setpoint that allows each combustor/boiler to maintain this value within a range of ±2000 lb/hr. When the steam flow drops below the setpoint, the rams are activated to increase the MSW feed into the combustor. When the steam flow exceeds the setpoint, less MSW is fed into the unit.

The higher heating value (HHV) of the waste varies because of the heterogeneous nature of the waste. In the spring and summer months, the waste can typically have a HHV less than 4500 because the waste is occasionally soaked by heavy rainfalls. In this case, it will take more than 255 TPD MSW to generate an average heat input of 95.6 MM Btu/hr. However, in terms of pollutant levels, the concentrations of particulate matter, NO_x, SO₂, HCl, CO, etc. in the flue gas will be approximately the same as when burning 4500 Btu/lb waste because the additional weight in the MSW is comprised of mostly water and/or inerts. On the other hand, if the waste contains more cardboard or paper (as during the holidays), the waste will have a heating value of more than 4500 Btu/lb and less waste, tonnage-wise, will be burned in the combustor.

8. (Continued)

We would like to propose using the boiler operating parameters to determine the average heat input into the combustor boiler. The computerized control system can be programmed to record various parameters including steam flow and flue gas O₂ concentration and temperature. By assuming the boiler efficiency and ash burnout remains relatively constant, the heat input from the waste can be estimated using these parameters. The facility will measure and record all of the appropriate data including the amount of waste burned and submit this data for review by DER.

- Jo
9. Specific Condition No. 1d specifies that the normal operating range of the MWC shall be 80% to 100% of the design rated capacity. The facility is designed to operate at 50% to 120% design rated capacity. The equipment is designed with the excess margin so that it can operate for short time intervals and not be subject to strain and subsequently damaged. We would like to propose that this condition read "The normal operating range of the MWC shall be 80% to 120% of the design rated capacity."

If you have any questions regarding the above items, please call me at (412) 636-5806.

Sincerely,

David S Beachler

D. S. Beachler, Manager
Environmental and Quality Engineering

cc: S. J. Brady, Westinghouse RESD
G. G. Pennington, Bay County Resource Management Center

/kjd
EN1351DB-EN09

*copied: Pradip Patel
Barry Andrews
Don Rogers
Ed Middlewart
Tom Moody
Hayne Aronson
Miguel Alarcon
CHART*



Florida Department of Environmental Regulation

Twin Towers Office Bldg. • 2600 Blair Stone Road • Tallahassee, Florida 32399-2400

Bob Martinez, Governor

Dale Twachtmann, Secretary

John Shearer, Assistant Secretary

August 2, 1988

Mr. D. S. Beachler
Westinghouse RESD, Cost Bldg.
2400 Ardmore Boulevard
Pittsburg, PA 15221

Dear Mr. Beachler:

RE: The Technical Evaluation and Preliminary Determination and proposed permit for Bay County Waste-to-Energy Facility, located in Panama City, Bay County, Florida, PSD-FL-129.

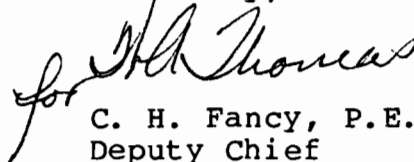
Specific Condition No. 1.b., in the above mentioned proposed permit needs to be corrected to reflect that wood waste use shall not exceed 160 TPD for the facility, and is not limited to 80 TPD per combustor.

Specific Condition No. 3 in the proposed permit needs to be corrected to reflect that the emission limits therein are for the facility and not just for one combustor.

Please find attached pages 5 and 6 as corrected.

Please submit, in writing, any comments which you wish to have considered concerning the Department's proposed action to Mr. Bill Thomas of the Bureau of Air Quality Management.

Sincerely,



C. H. Fancy, P.E.
Deputy Chief
Bureau of Air Quality
Management

CHF/pr

Attachments

cc: E. Middleswart, NW District
T. Moody, NW District
W. Aronson, EPA
M. Flores, NPS
A. Richter, P.E.
J. Kolk, Audubon Society

PERMITTEE: Bay Resource
Management Center

Permit Numbers: AC 03-145061
03-152196

GENERAL CONDITIONS:

- b. The permittee shall retain at the facility or other location designated by this permit records of all monitoring information (including all calibration and maintenance records and all original strip chart recordings for continuous monitoring instrumentation), copies of all reports required by this permit, and records of all data used to complete the application for this permit. The time period of retention shall be at least three years from the date of the sample, measurement, report or application unless otherwise specified by Department rule.
- c. Records of monitoring information shall include:
- the date, exact place, and time of sampling or measurements;
 - the person responsible for performing the sampling or measurements;
 - the date(s) analyses were performed;
 - the person responsible for performing the analyses;
 - the analytical techniques or methods used; and
 - the results of such analyses.

15. When requested by the Department, the permittee shall within a reasonable time furnish any information required by law which is needed to determine compliance with the permit. If the permittee becomes aware that relevant facts were not submitted or were incorrect in the permit application or in any report to the Department, such facts or information shall be submitted or corrected promptly.

SPECIFIC CONDITIONS:

1. Municipal Waste Combustor

- a. Each of the two municipal waste combustors (MWC) shall have a design rated capacity of 255 tons municipal solid waste (MSW) per day, 95.6 million Btu heat input per hour, assuming a heating value of 4,500 Btu per pound, and a steam production rate of 68,000 lbs/hr.

(MSW or MSWad WW)
510
191.2
b. The maximum throughput for ^{the facility} each MWC shall not exceed ~~the design charging rate of 255 TPD~~ or the heat input rate of ~~95.6 MMBtu/hr.~~ In absence of sufficient MSW to maintain a steady heat rate, wood waste may be used but not in excess of 160 TPD for the facility.

PERMITTEE: Bay Resource
Management Center

Permit Numbers: AC 03-145061
03-152196

SPECIFIC CONDITIONS:

- c. The furnace mean temperature at the fully mixed zone of the combustor shall not be less than 1,800°F.
- d. The normal operating range of the MWC shall be 80% to 100% of design rated capacity.
- e. The MWC shall be fueled with municipal solid waste and wood waste only. Other wastes shall not be burned without specific prior written approval of Florida DER.
- f. Auxiliary fuel burners shall be fueled only with distillate fuel oil or natural gas. If the annual capacity factor for oil or gas is greater than 10%, as determined by 40 CFR 60.43b(d), the facility shall be subject to 40 CFR 60.44b, standards for nitrogen oxides.
- g. Auxiliary fuel burners shall be used at start up during the introduction of MSW fuel until design furnace gas temperature is achieved.
- h. The facility may operate continuously (8760 hrs/yr).

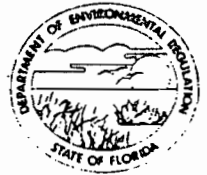
2. Each MWC shall be equipped with an electrostatic precipitator for particulate emission control.

3. Flue gas emissions from ^{each combustor} the facility shall not exceed the following:

Pollutant	Emissions	
	lbs/hr	TPY
PM	13.5/2	59/2
CO	185.6/2	813/2
NOx	53.9/2	236/2
SO ₂	71.5/2	313/2
VOC	14.2/2	62/2
Lead	0.08/2	0.36/2
Mercury	0.36/2	0.16/2
Beryllium	1 x 10 ⁻⁵ /2	4.5 x 10 ⁻⁵ /2
Hydrogen Chloride	123.3/2	540/2
Sulfuric Acid Mist	3/2	13/2
Fluoride	0.3/2	1.3/2

Visible emissions shall not exceed 15% opacity.

State of Florida
DEPARTMENT OF ENVIRONMENTAL REGULATION



Interoffice Memorandum

FOR ROUTING TO OTHER THAN THE ADDRESSEE

TO: _____	LOCTN: _____
TO: _____	LOCTN: _____
TO: _____	LOCTN: _____
FROM: _____	DATE: _____

TO: Bay County Waste-to-Energy Facility File, PSD-129
FROM: Pradeep A. Raval *PR*
DATE: July 13, 1988
SUBJ: Use of Wood Waste in Combustors.

Spencer Brady of Westinghouse called Barry Andrews to request permission to burn wood waste when municipal solid waste was not available in sufficient quantity (to sustain a steady heat rate), for the two O'Connor combustors at the Bay County Waste-to-Energy facility. The permit currently being drafted will reflect this change.

Since rule applicability for the proposed project does not change, and since the applicant accepts the restricted use of wood waste up to 160 tons per day (allowed currently), no new review will be triggered. Changes in the projected emissions will be noted as attached.

7/25
I agree -
CHJ

PR/plm

Attachment

Best Available Copy

TABLE 1

Pollutant	Annual Emissions (Tons/Year) Based on 350 TPD MSW 135 TPD Wood	Maximum Annual Emissions (tons/yr) Based on 510 TPD MSW or 350 TPD MSW & 160 TPD Wood	Difference (Tons/Year)	PSD Significant Emissions Rate (Tons/Year)
Particulate Matter	50	59	9	25
Carbon Monoxide	722	813	91	100
Nitrogen Oxides	223	236	13	40
Sulfur Dioxide	223	313	90	40
VOC	54	62	8	40
Lead	0.25	0.36	0.11	0.6
Mercury	0.11	0.16	0.05	0.1
Beryllium	0.000031	0.000045	0.0000136	0.004
Fluorides	0.9	1.30	0.40	3
Sulfuric Acid Mist	9.0	13	4	7
Hydrogen Chloride	370	540	170	-

24 Nov 1988
Pittsburgh, PA



Westinghouse
Electric Corporation
ENG/MG:DSB:88-051

Resource Energy Systems
Division

Cost Building
2400 Ardmore Boulevard
Pittsburgh, Pennsylvania 15221
(412) 638-5800
WIN 261-5300

RECEIVED

MAR 28 1988

DER-BAQM

March 23, 1988

Mr. Barry Andrews
Florida DER
Twin Towers Office Building
2600 Blair Stone Road
Tallahassee, FL 32301

Dear Barry,

Enclosed is a copy of a recent article concerning the retrofit costs of scrubbers for waste-to-energy plants. This article appeared in the March, 1988 issue of Waste Age. We thought that you would be interested in this information.

Sincerely,

David S Beachler

D. S. Beachler, Manager
Environmental and Quality Engineering

Enclosure

/kjd
0751MM-EN01:21

Copies: CHF/BT
Pradeep Patel
Tom Rogus
Barry Andrews } 3-28-88 ml

HOW MUCH WILL PLANT RETROFITS COST?

Local disposal costs may skyrocket if existing refuse-to-energy plants are required to add sophisticated emission controls.

Sometime next year the EPA is scheduled to propose new Clean Air Act regulations for control of refuse-to-energy plant emissions. One result could be a stiff increase in disposal costs for some or all of the 200-odd locales which, by then, will have refuse incineration facilities in place or under construction.

Final regulations, to take effect in 1991, apparently will require new facilities to include sophisticated air pollution control (APC) systems, probably including acid gas scrubbers. As scrubbers have not normally been included in most recently-built U.S. plants, this could mean millions of dollars each in additional costs to new plants.

Table One
Estimated APC Retrofit Costs At Three Florida Plants
(all mass-burn plants, refitted with acid-gas scrubbers)

	McKay Bay	Pinellas	Hillsborough
Plant size (tpd)	1,000	3,000	1,200
Type	Refractory	Waterwall	Waterwall
Construction Costs	\$16,306,000	\$15,000,000	\$18,000,000
Total Financing Costs	\$29,644,945	\$28,048,638	\$31,241,407
Lost Revenues and Extra Expenses	\$16,707,510	\$41,573,500	\$27,159,504
Total Retrofit Cost	\$62,658,455	\$84,622,138	\$76,400,911
Cost Per TPD of Design Capacity	\$62,658	\$28,207	\$63,666

Source: HDR Technology, Inc.

Notes: HDR Technology considered at least two different retrofit plans for each of the facilities studied. In most cases, total downtime for retrofitting of the facilities was estimated to be two years. "Total Financing Costs" amalgamates HDR estimates for bond discount/issuance, debt service reserve, surplus reserve and contingency, net interest during construction, bond issue, and annual debt service. "Lost Revenues and Extra Expenses" includes payments of operating fees, lost emergency revenues, landfilling costs.

EPA also will issue "guidelines" for existing facilities. By 1991, each state must tell the federal agency how it plans to bring existing plants into compliance with the guidelines. State actions could, it is believed, force existing facilities to add pollution control systems such as scrubbers. This will be costly as, generally speaking, adding a scrubber or other APC item to an existing plant costs much more than integrating such a system in plant design before construction starts.

EPA has proposed to control emissions from existing facilities under Section 111(d) of the Clean Air Act. This section requires EPA to issue guidelines to the states for existing sources, which they would use in developing specific emissions standards. APC requirements would then be determined on a case-by-case basis, taking into account not only health risks but the remaining useful life of the facility and costs of emission controls.

Depending on exactly what the states require after a case-by-case analysis, many plants operating today could be required to spend millions of dollars for retrofits. These costs would be passed on to their customers.

This possibility raises two key questions:

- Why are new regulations with this potential effect being proposed?
- Are the benefits produced by such retrofits (and additions to new plants) commensurate with the costs the public will have to pay to gain them?

What the EPA has said

In the July 7, 1987, *Federal Register*, EPA said refuse-to-energy plants, which it calls "municipal waste com-

bustors" (MWCs), "may be reasonably anticipated to contribute to the endangerment of public health and welfare."

Specifically, EPA fears that emissions contribute to acid rain, increased cases of cancer, and other health problems. Unless Congressional action redirects EPA's current thinking, the 1991 regulations will probably require new facilities be equipped with APC systems incorporating both acid gas scrubbers and either fabric filter baghouses or electrostatic precipitators (ESPs). The guidelines for states to follow in regulating existing plants may be made final next year or in 1990.

A key question is: What will the states do in response? One answer could come from Florida, where stricter APC regulations are under consideration. Note that Florida is ahead of the pack by a step or two because it has more existing plants than any other state.

EPA's July 7 action was precipitated by the fact that, other than for particulates, there are no federal stack emission limits for MWC facilities. Some states have imposed emissions limits on various pollutants, but many have not.

Retrofitting cost estimates vary

Just how much will it cost to retrofit existing plants with more sophisticated APC systems?

EPA estimates that it will cost 1.8 times the cost of a control system. That factor is supposed to cover all costs, including the system vendor's bill, construction, and additional ductwork in the plant. Using EPA's equation, the cost of retrofitting a 1,000-tpd waterwall mass-burn facility with acid gas scrubbers would be \$18.7 million (or \$18,700 per ton).

But an estimate for addition of acid-gas scrubbers to the existing 1,200-tpd waterwall mass-burn Hillsborough County, Fla., facility puts the total retrofit cost closer to \$76 million!

See Table One for estimates for APC retrofits at three Florida plants, from a study performed by HDR Techserv, Inc. (Tampa, Fla.).

HDR looked at costs at the Hillsborough, Tampa, and Pinellas County plants to estimate the financial effect of new regulations under consideration in Florida.

In its estimate, the consulting firm factored in construction costs and the price of the control system, as EPA says it does. HDR's estimate assumes retrofits take two years.

Plant shutdowns for retrofit purposes have a cost, which HDR includes in its estimate. In addition to the lost energy revenues, there is a high price tag on land-

filling refuse that the shut-down-for-retrofit facility would otherwise burn. HDR also computes debt service, bond issuance, and other costs into its equation.

What's interesting is that when a *Waste Age* staffer asked a U.S. EPA official for comment on the HDR estimates, the reaction was that the Florida figures might be more accurate than the federal agency's 1.8-times-system-cost factor:

"Our numbers were very preliminary," Michael Johnston, an EPA spokesperson, says. "Now that we've decided to regulate . . . we will look (more closely) at the economic effect on the facility and the community."

Is it worth the extra cost?

Potential health and welfare impacts from refuse-to-energy plants that concern EPA span a broad range. These concerns include the "relatively low risks of cancer associated with several trace constituents, as well as the significantly higher, but much more uncertain, cancer risks posed by the predicted emissions of (dioxins and furans)."

EPA is additionally concerned about possible noncarcinogenic effects related to particulates, sulfur dioxide, carbon monoxide, and nitrogen oxide emissions, and the potential health implications of hydrochloric acid emissions.

The addition of scrubbers to a plant that already has ESPs or fabric filters in place can help control emissions of sulfur dioxide, hydrochloric acid, and hydrofluoric acid, among others, to levels well within the recommended range of acceptable health standards. But how much of a problem do refuse-to-energy plant emissions of these substances really present to the public?

Based on its Florida study results, HDR Techserv maintains that the threat to public health is relatively low. For one thing, the consulting firm's report says it is unclear how significant waste combustion emissions are in the formation of acid rain.

Moreover, many of the potentially harmful compounds released from refuse-to-energy facilities are emitted in very small quantities. HDR found that close to 70% of Florida's sulfur dioxide emissions were generated by utility companies, while more than 40% of total nitrogen oxide emissions came from cars, buses, trains, and trucks. Solid waste facilities were found to contribute less than 1% of Florida's sulfur dioxide and nitrogen oxide emissions!

Says Marc Rogoff, an HDR environmental scientist and former Hillsborough County official: "I'm not sure it's worth spending \$50 million or more to retrofit a

plant to reduce something that is negligible to begin with."

Impact on disposal prices

According to HDR estimates, if the state of Florida required the three plants studied to add APC equipment, refuse disposal cost increases of from \$40 to \$80 per resident per year would result.

In other words, if Florida plants are forced to add acid gas scrubbers, citizens living in areas served by the plants could pay as much as \$6.67 per month just for the extra APC equipment! "That's a big chunk to be asking from the public," Rogoff opines. "They won't want to pay it."

Perhaps this very factor will cause states to pull back from requiring APC retrofits. Under the Clean Air Act, states will be able to consider costs and a plant's remaining useful life. EPA's Johnston concedes that "We do take cost into account, but how much of a fac-

tor it is going to be, I can't say for sure right now."

Congress may change scenarios

As matters now stand, what happens to existing plants will depend on what EPA places in its guidelines and how states respond. But there's more.

Members of Congress may take the decision away from the agency. In Senate and House hearings:

- The EPA's timetable for regulating refuse-to-energy plant emissions has been sharply criticized;
- Critics have also complained that the agency's plans for regulating specific pollutants are not broad enough (i.e., additional pollutants need to be covered); and
- Critics deem emission levels of listed pollutants that EPA will allow as "unacceptably high."

Federal legislation calling for tougher APC controls and faster regulation was under discussion in 1987 and may well be made into law this year if EPA's response does not satisfy its Congressional and other critics. ■

DRASTICALLY REDUCE YOUR HAULING COSTS AND LANDFILL TIPPING FEES!

FHE's New Tub Grinder Produces Chips From Bulky Wood Waste:

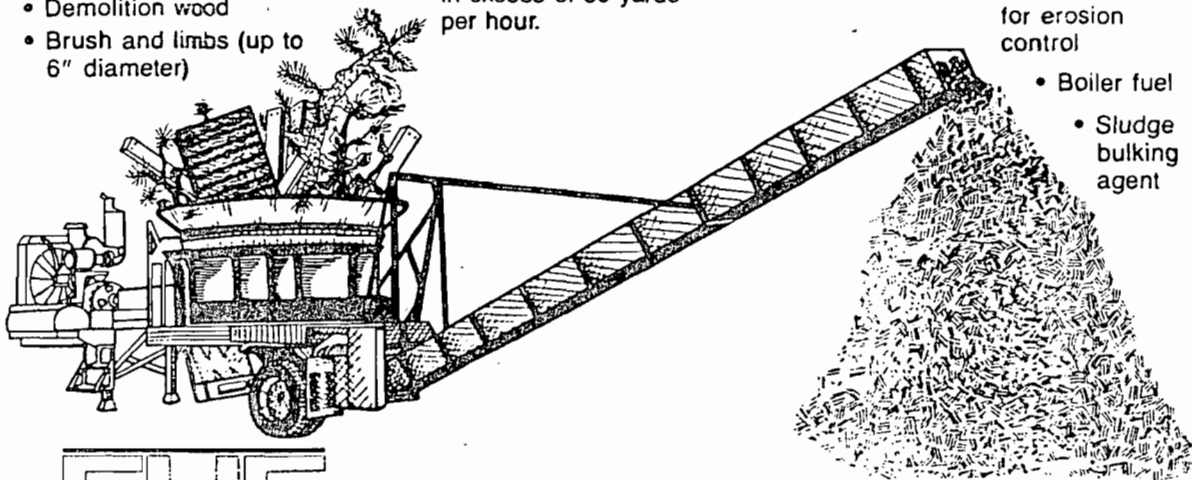
- Construction wood waste
- Pallets
- Demolition wood
- Brush and limbs (up to 6" diameter)

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FHE's Tub Grinder with Caterpillar® power processes bulky wood waste at a rate in excess of 50 yards per hour.

Markets For Recycled Bulky Wood Include:

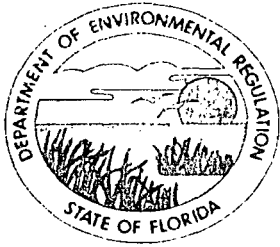
- Garden mulch
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- Boiler fuel
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Fuel Harvesters
Equipment

Fuel Harvesters Equipment Inc.
12759 Loma Rica Drive Grass Valley, CA 95945

Tel: (916) 272-7664



K-1112

Florida Department of Environmental Regulation

Twin Towers Office Bldg. • 2600 Blair Stone Road • Tallahassee, Florida 32399-2400

Bob Martinez, Governor

Dale Twachtmann, Secretary

John Shearer, Assistant Secretary

August 11, 1988

Mr. Steven A. McLellan
Water & Wastewater Systems
3400 Transmitter Road
Panama City, Florida 32404

Dear Mr. McLellan:

Re: Proposal to Incinerate Wastewater Treatment Sludge in the
Bay County Incinerator - Your Letter Dated July 12, 1988

The new draft permit to increase the MSW charging rate to the incinerators from 350 TPD to 510 TPD has been sent to Westinghouse and Bay County. The final permit needs to be issued prior to considering any amendments such as the one you proposed. Once the final permit is issued, we will consider this request, if, at that time you reactivate the request.

Sincerely,

G. H. Fancy, P.E.
Deputy Chief
Bureau of Air Quality
Management

CHF/ks

cc: T. Moody



3.22.88
Atlanta, GA

UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

REGION IV

345 COURTLAND STREET
ATLANTA, GEORGIA 30365

MAR 21 1988

4APT-APB

Margaret V. Janes, Planner
Bureau of Air Quality Management
Florida Department of Environmental Regulation
Twin Towers Office Building
2600 Blair Stone Road
Tallahassee, Florida 32399-2400

Re: Bay County Resource Recovery Facility (PSD-FL-129)

Dear Ms. Janes:

This is to acknowledge receipt of a copy of the permit application for the Bay County Resource Recovery Facility and to confirm the March 4, 1988, telephone conversation between Pradeep Raval of your staff and Gary Ng of my staff. We have reviewed the application for an increase in municipal waste throughput and have the following comments:

- 1) As was indicated in the phone conversation, we noticed that the applicant did not address the possible hazardous air pollutants (HAPs) (i.e., dioxin, HCl, H₂SO₄, etc.) from the combustion of the proposed additional municipal solid waste. As you know, EPA is now requiring all municipal waste combustors (MWCs) to take unregulated HAPs into consideration when evaluating BACT requirements for the regulated air pollutants (i.e., SO₂, TSP, etc.). Please require the applicant to provide the necessary analysis on the associated HAPs.
- 2) Also mentioned was the BACT determination for SO₂; we noticed that the application lacked an incremental cost analysis in dollars per ton of pollutant removed. Please request the applicant to provide the analysis. Also, in order to have a more complete incremental analysis, the applicant must take into consideration the amount of HAPs controlled in addition to the amount of SO₂ controlled in determining the incremental cost for each option. Then, in accordance with the December 1, 1987, Potter memorandum, we would also suggest that the applicant perform the analysis in a top-down manner.

Thank you for the opportunity to provide you with our comments. Please ask the applicant to address the above concerns before issuing your preliminary determination. If you have any additional comments or information, please contact me or Gary Ng of my staff at (404) 347-2864.

Sincerely yours,

Bruce P. Miller

Bruce P. Miller, Chief
Air Programs Branch
Air, Pesticides, and Toxics
Management Division

Copied: Pradeep Baval
Tom Rogus
CHFIBT
Barry Andrews } 3-28-28 ~~NYD~~

Barny's Copy



Florida Department of Environmental Regulation

Twin Towers Office Bldg. • 2600 Blair Stone Road • Tallahassee, Florida 32399-2400

Bob Martinez, Governor

Dale Twachtmann, Secretary

John Shearer, Assistant Secretary

May 26, 1988

CERTIFIED MAIL - RETURN RECEIPT REQUESTED

Mr. David S. Beachler
Westinghouse RESD
Cost Building
2400 Ardmore Blvd.
Pittsburgh, Pennsylvania 15221

Dear Mr. Beachler:

Re: Completeness Review of Application to Modify
Bay County Resource Recovery Facility
Permit No. AC 03-145061, PSD-FL-129

As per our discussion of May 19, 1988, please provide unique and convincing arguments to justify your position that acid gas scrubbing should not be required for the above referenced project. EPA has indicated that this justification is needed to avoid the acid gas control issue. I anticipate setting up a meeting between Westinghouse, Bay County, EPA, and myself next week.

If you have any questions please call me at (904)488-1344, or write to me at the above address.

Sincerely,

C. H. Fancy, P.E.
Deputy Chief
Bureau of Air Quality
Management

CHF/PR/s

cc: E. Middleswart, NW District
W. Aronson, EPA
M. Flores, NPS
A. Richter, P.E.

Bunny's Copy

TO ALL BCC'S OF LETTER NO. ENG:DSB:87-130 DATED MARCH 19, 1987:

Please disregard the first version of this letter that you received (to Mr. Robert Kriegel of the Florida Department of Environmental Resources). The letter had to be modified, but unfortunately it was too late to retrieve all cc and bcc addressees as the mail had already gone out.

To determine which is the first version and which is the second version--the first version only had two pages, the second version has three pages.

Thank you,

Kathie Daschke
for Mick Pompelia



Westinghouse
Electric Corporation

Resource Energy Systems
Division

ENG:DSB:87-130

Cost Building
2400 Ardmore Boulevard
Pittsburgh Pennsylvania
(412) 636 5800
WIN 261 5800

March 19, 1987

Mr. Robert Kriegel
Florida Department of
Environmental Resources
Northwest District
160 Government Center
Pensacola, FL 32501

Dear Mr. Kriegel:

I am writing you as directed by Mr. Jack Preece to request permission from your office to burn 510 TPD of Municipal Solid Waste (MSW) during the upcoming acceptance test and emission compliance tests at the Bay County Resource Recovery Facility. The acceptance test will be conducted to verify capacity, electricity generated, and ash burnout values as guaranteed by the contract between Westinghouse Electric Corporation and the Bay County Resource Authority.

The permit issued by Florida DER, AC03-84703, states that the facility can burn a maximum of 350 TPD of MSW. However, the third party engineer for the Bay County Resource Authority, Roy F. Weston, Inc., has stipulated that the facility must be operated at 100% capacity during the acceptance test. The acceptance test period will run from approximately March 25 through April 10, 1987. The emission compliance tests are to be performed in that period and are scheduled to run from April 5 through April 8, 1987. The acceptance test period includes time for combustor/boiler warm-up, combustion stabilization, acceptance testing and emission compliance testing. The emission compliance tests will be conducted using EPA Reference Methods 5 and 9 and will be conducted to demonstrate compliance with the permit conditions for particulate matter concentration of less than 0.03 gr/dscf. The acceptance test and compliance tests will be conducted on a one-time basis; therefore, the facility will only burn 510 TPD MSW during this test period.

March 19, 1987

If you have any questions regarding this request, please call me at (412) 636-5806 or Milton Kirkpatrick at (904) 785-7933. We look forward to hearing from you shortly.

Sincerely,

A handwritten signature in cursive script that reads "Mick Fomycela Sr".

D. S. Beachler, Manager
Environmental and Quality Engineering

cc: G. Pennington - Bay County
M. Kirkpatrick - Bay County

/kjd
0379MM-069E-2

ENG:DSB:87-130

-3-

March 19, 1987

bcc: W. G. Collins
J. J. Ludwig
J. J. Zebroski
William Thomas - Florida DER

DER

MAR 23 1987

BAQM

7/28/87
Pittsburgh, PA



Barry

Westinghouse
Electric Corporation

Resource Energy Systems
Division

Cost Building
2400 Ardmore Boulevard
Pittsburgh Pennsylvania 15221
(412) 636 5800
WIN 261 5800

ENG/MG:DSB:87-112

DER

OCT 1 1987

BAQM

September 28, 1987

Mr. William Thomas
Florida Department of Environmental Regulations
Twin Towers Office Building
2600 Blair Stone Road
Tallahassee, FL 32399

Dear Bill,

I am writing you regarding DER Construction Permits AC-03-84703 and AC-03-84704. As you may know, we have been talking to Barry Andrews, Pradeep Raval, and Tom Moody concerning both the construction and operating permits for the Bay County Resource Management Center located in Panama City, Florida.

As a result of our conversation on Friday, September 25, 1987, we would like to request an extension (as you suggested) to the construction permit expiration date January 30, 1988. We submitted a copy of the emission compliance test report to Tom Moody and to you on July 20, 1987. The report showed that the facility is in compliance with the conditions specified in the permits. We are completing the certificate of completion of construction, DER Form 17-1.202(3) and will be submitting them shortly to the District Office in Pensacola.

As you requested, we will submit a new application form to request permission to operate the facility at its maximum design rate to burn 510 TPD MSW.

If you have any questions, please call me at (412) 636-5806.

Sincerely,

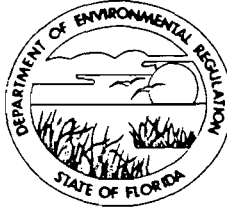
D. S. Beachler, Manager
Environmental and Quality Engineering

cc: Tom Moody, Florida DER, Pensacola
Clair Fancy, Florida DER, Tallahassee

Pradeep Raval } 10/2/87
Barry Andrews }
CHF/BT }

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION

NORTHWEST DISTRICT
160 GOVERNMENTAL CENTER
PENSACOLA, FLORIDA 32501-5794



BOB MARTINEZ
GOVERNOR
DALE TWACHTMANN
SECRETARY
ROBERT V. KRIEDEL
DISTRICT MANAGER

August 11, 1987

Mr. Gregg Pennington
Bay County Energy Systems, Inc.
7504 Hwy. 231 North
Panama City, Florida 32404

DER
AUG 12 1987
BAQM

Dear Mr. Pennington:

Department air and solid waste permits to build your resource-to-energy plant will expire October 30, 1987. Since the plant has been operational for six months and the required emissions testing has been accomplished, we urge you to resolve any remaining problems, certify the construction, and apply for the required operation permits.


Particularly, the issue of the tonnage capacity of the plant must first be resolved through our central air bureau permitting section in Tallahassee. If the construction permit is not modified appropriately before its expiration, your operation permit will limit the plant to 350 tons per day of municipal solid waste. Once that issue is resolved, you must apply for air and solid waste operation permits to this Pensacola office.

You are now in violation of Condition 10 in air construction permits AC03-84703 and 84704, which required that an application for operation permits be filed no later than August 1, 1987, and you are also in violation of Condition 20 in solid waste construction permit SC03-91036, which required a similar application within 75 days after completion of construction.

Mr. Gregg Pennington
Page two

If these matters are not resolved immediately, I may have no choice but to declare the plant out of compliance and institute enforcement action. If you have any questions, please call me at 436-8360.

Sincerely,



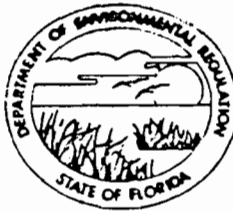
Thomas W. Moody, P.E.
Special Programs Supervisor

TWM/tmf

cc: Ms. Joy Bates, Bay County
Mr. Clair Fancy, BAQM

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL REGULATION

NORTHWEST DISTRICT
160 GOVERNMENTAL CENTER
PENSACOLA, FLORIDA 32501-5794



BOB MARTINEZ
GOVERNOR
DALE TWACHTMANN
SECRETARY
ROBERT V. KRIEDEL
DISTRICT MANAGER

August 11, 1987

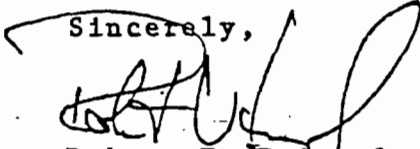
Honorable John Carter
Chairman, Washington County Board
of County Commissioners
Post Office Box 647
Chipley, Florida 32428-00647

Dear Chairman Carter:

Your Chipley and Mudhill Landfills do not meet the State's new requirements for the operation of landfills. We are concerned about your present status and the likelihood that you will not be able to meet these standards in the near future. However, there are alternatives. In example, Bay County's new Resource Recovery Facility (a refuse to energy plant) may be able to handle up to 510 tons per day of Municipal Solid Waste (MSW). At present, Bay County is providing only about 325 tons per day and thus could handle an additional 185 tons per day in out-of-county waste. The County is actively seeking additional waste sources; I understand the County estimates a tipping fee ranging from \$22 to \$28 per ton at present. Similarly, Timber Energy is examining the availability of MSW for a facility they are considering.

We suggest that you seriously consider any available alternatives. Some may be of mutual benefit to both parties, considering the rising costs of landfill operations. Your having given these alternatives serious consideration may also help you in the event the Department or other parties have to institute enforcement proceedings concerning your solid waste operations.

Sincerely,



Robert V. Kriegel
District Manager

RVK/rkf

cc: 



RESOURCE ENERGY SYSTEMS DIVISION

BAY COUNTY, FLORIDA WASTE-TO-ENERGY FACILITY

EMISSION COMPLIANCE TEST

**SUBMITTED TO THE FLORIDA
DEPARTMENT OF ENVIRONMENTAL RESOURCES**

JUNE 1987



**Westinghouse
Energy Services
A Westinghouse Company**

WCAP-11541

BAY COUNTY, FLORIDA WASTE-TO-ENERGY FACILITY

EMISSION COMPLIANCE TEST

**SUBMITTED TO THE FLORIDA
DEPARTMENT OF ENVIRONMENTAL RESOURCES**

JUNE 1987

BAY COUNTY, FLORIDA WASTE-TO-ENERGY FACILITY

EMISSION COMPLIANCE TEST

**SUBMITTED TO THE FLORIDA
DEPARTMENT OF ENVIRONMENTAL RESOURCES**

JUNE 1987

**Prepared by: David S. Beachler
Manager, Environmental and Quality Engineering**

Signature: David S Beachler

**Jill Weldon
Senior Engineer**

Signature: Jill Weldon

TABLE OF CONTENTS

<u>SECTION</u>		<u>PAGE</u>
1.0	Introduction	1
2.0	Facility Description	1
3.0	Plant Capacity	4
4.0	Test Program	5
5.0	Test Dates	5
6.0	Testing Personnel	6
7.0	Results	6
8.0	Conclusion	8

APPENDICES

A.	Emission Compliance Report for June 4 and 5 Compiled by ETS, Inc.	14
B.	Opacity Measurements for May 12, 13, and 14, 1987	15
C.	Computer Calculation Sheets for Additional Test Results Reported in Table 2.	16

LIST OF TABLES

	Page
Compliance Test Results for Bay County Energy Resources	10
Additional Test Results from Bay County Energy Resources	11

LIST OF FIGURES

	Page
Process Flow Diagram	2
Westinghouse/O'Connor Water Cooler Rotary Combustor/Boiler	4

**BAY COUNTY, FLORIDA WASTE-TO-ENERGY FACILITY
EMISSION COMPLIANCE TEST REPORT**

1.0 INTRODUCTION

In April 1987, Westinghouse contracted ETS, Inc. of Roanoke, Virginia to conduct stack sampling of the Bay County Resource Recovery Facility. These tests were conducted over a 1-1/2 month period during the start-up and equipment fine-tuning stages, the plant 72-hour acceptance test, and the Florida DER emission compliance tests. The DER offices were notified that tests were being conducted during the time periods of May 12-14, May 18-21, and June 4-5, 1987.

Westinghouse submits this test report to the Florida Department of Environmental Regulations as part of the Certificate of Completion of Construction.

2.0 FACILITY DESCRIPTION

The Bay County Resource Recovery Facility is located on Highway 231, 10 miles Northeast of Panama City, Florida. The facility processes 510 tons per day of municipal solid wastes (MSW) and waste wood. Heat generated by the combustion of waste in the combustor produces steam to drive a turbine generator. A process flow diagram of the Bay County facility is shown in Figure 1.

The plant consists of two combustor/boiler units, a turbine-generator, a truck scale, tipping floor, front end loaders, conveyors, air emission control equipment, a stack, ash handling equipment, a central control room, and all required ancillary equipment. The facility also includes administration offices, change rooms, parking areas, roadways, and security fencing.

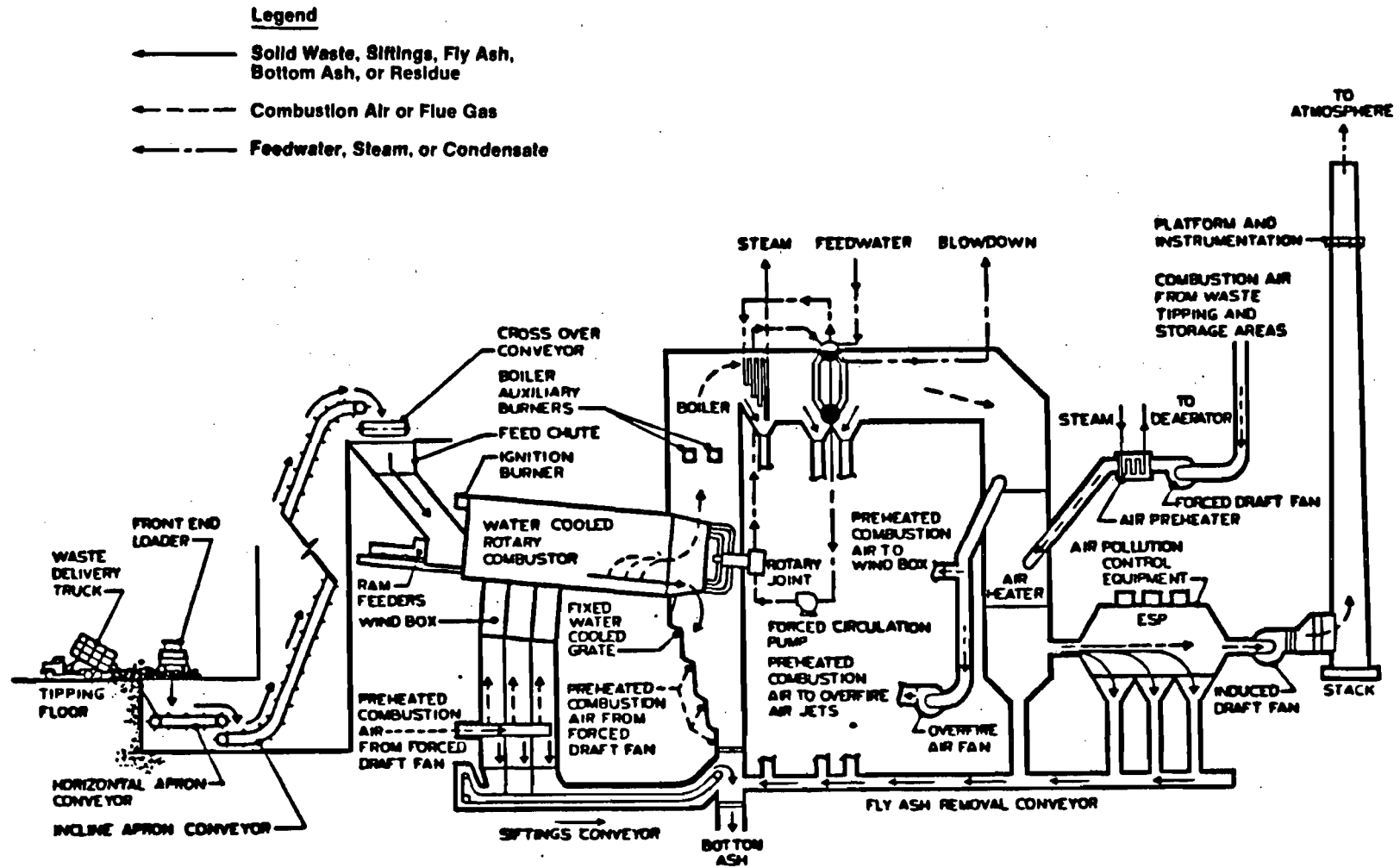
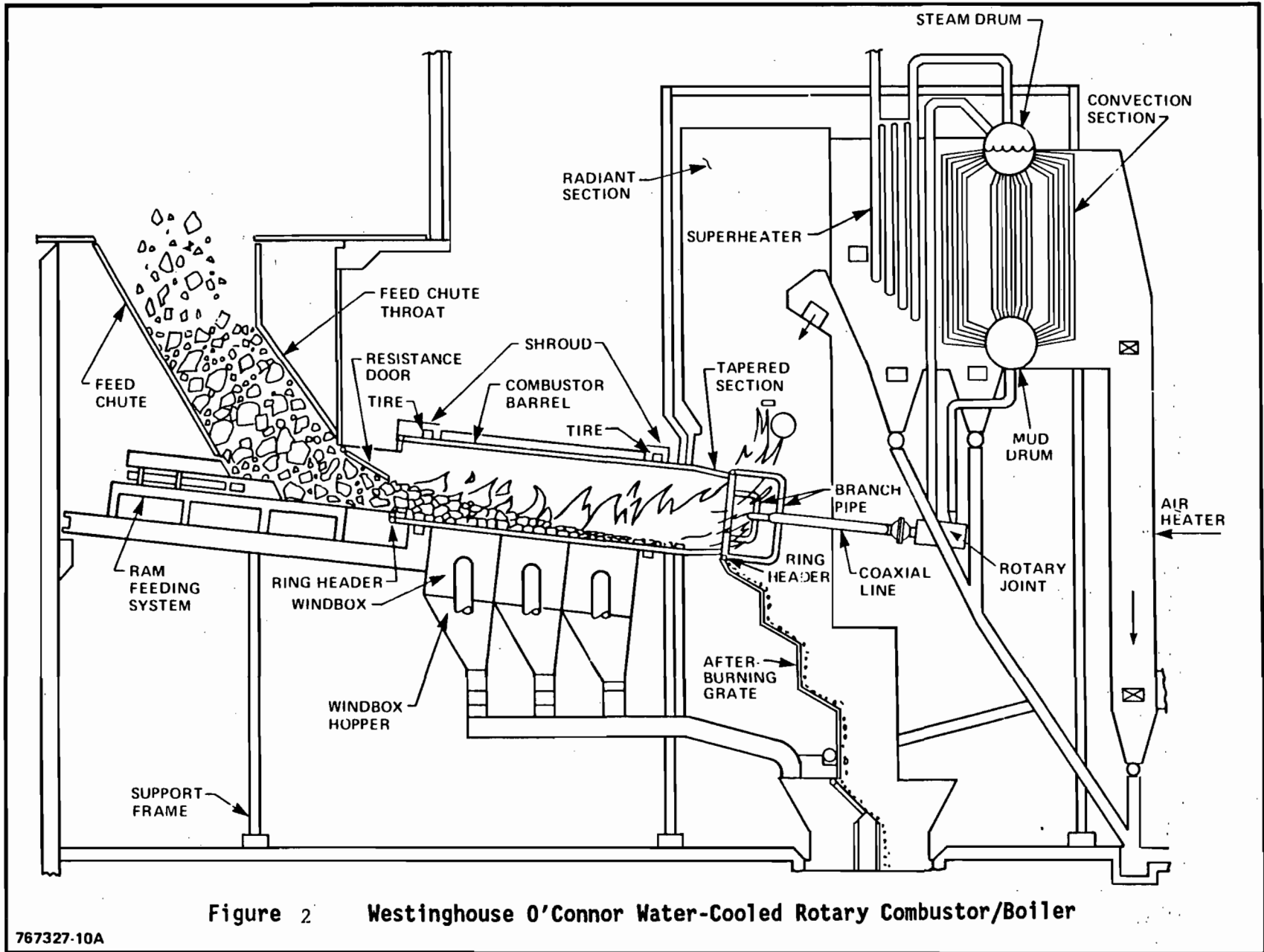


Figure 1 Simplified Process Flow Diagram, Gas Cycle

All municipal waste received at the plant enters through an automatic gate system and is unloaded on the tipping floor. Solid waste collection vehicles hauling the material to be processed are weighed at the scale prior to entering the plant and are then directed to a specific bay on the tipping floor. The weight is automatically entered into a computer system that records and files all pertinent data for each transaction. The vehicles enter the designated bay and discharge their load on the floor. The tipping floor accommodates approximately 1500 tons of waste while allowing room for maneuvering the incoming trucks and front end loaders.

A man-operated front-end loader disperses MSW on the tipping floor to separate large and unprocessable objects. Large items are separated from MSW; the combustible ones are processed through a shear shredder; the noncombustible items are removed and stored temporarily for landfill disposal. After sorting, the MSW is pushed onto the horizontal apron conveyor by the front-end loader. The horizontal apron conveyor transfers the MSW to the inclined apron conveyor and then into the combustor charging chute. The incline apron conveyor contains a weigh scale that continuously measures the weight of MSW sent for processing. When one line of apron feed conveyors is down for maintenance, a transfer conveyor at the charging hopper level feeds both combustors by changing the moving direction of the conveyor apron belt.

From the combustor charging chute, the MSW is pushed into the combustor by the hydraulic ram feeders (see Figure 2). The speed of the ram feeders and, consequently, the amount of solid waste fed to the combustor are controlled by the level of the solid waste in the chute. The feed throat of the combustor chute is provided with a water-cooled isolation door to prevent the flames in the rotary combustor barrel from reaching the solid waste in the charging chute. The combustion process begins when the MSW is pushed into the combustor. The slightly inclined combustor barrel rotates slowly, causing the waste to tumble and advance as combustion proceeds. A forced draft fan supplies air to the combustor. The air is preheated before entering the multiple zone windbox located beneath each combustor barrel.



The heat released from the combustion process is recovered through the rotary combustor walls, boiler water walls and tubes, primary and secondary superheater, and the air preheater. Hot gases, produced during the combustion process, flow from the combustor barrel through the boiler's radiant, superheater, and convection sections. To maximize energy recovery and expedite combustion of high-moisture waste, the combustion gases exiting the convection section pass through a heat exchanger which preheats the incoming combustion air. To prevent corrosion problems in the lower part of the boiler air heater, a steam preheater is located at the air heater inlet to increase the air temperature from ambient to 150°F.

The flue gases from the air heater enter the electrostatic precipitator (ESP). The ESP's were manufactured by Environmental Elements Corporation. They are arranged into three mechanical fields, each with its own electrical field and ash removal hopper. They are designed to remove 99.33% of the particulate matter in the flue gas at a gas flow rate of 56,000 ACFM (400°F) and at a particulate inlet loading of 3 gr/dscf.

The flue gas is drawn from the ESP by an induced draft fan which maintains a slightly negative pressure through the entire system. The flue gas from each unit discharges to the atmosphere through a separate flue in the common stack. The stack is made of precast concrete with two 4-ft, 6-in. diameter flues that are constructed of 4-in. thick acid resistant bricks. The stack is 125 feet tall and has air emissions monitoring ports located 60 feet from the stack base.

Three types of ash by-products are produced by the process: fly ash, siftings, and bottom ash. Fly ash is collected in hoppers under the convection, superheater, boiler/air heater, and ESP sections of each unit and is conveyed pneumatically to the bottom ash conveyor. Siftings are collected underneath the combustor by the siftings conveyor and are transferred by an ash drag system to the bottom ash conveyor. Bottom ash is collected on the ash burnout grate and is directed by a bifurcated chute into a water quench

basin. The fly ash, siftings, and bottom ash mixture are water quenched, dewatered, and removed by the bottom ash drag conveyor into trucks that are disposed of at a landfill.

Heat from the combustion of MSW is absorbed in the combustor barrel, boiler, and superheater to produce steam to drive the turbine-generator. Boiler feedwater moves through the boiler tubes by natural circulation as it is transformed into a mixture of saturated steam and water. Pumps circulate water through the rotary combustor by drawing water from the lower drum of the boiler through the rotary joint and into one of the combustor barrel's ring headers. The water passes through the combustor tubes and returns to the boiler steam drum as a mixture of saturated water and steam. Steam leaves the drum and passes through the primary and secondary tubes of the superheater section where the steam is heated to the design steam condition for the turbine (750°F).

The steam flows from the superheater to the turbine-generator where a portion of its energy is converted to electricity. The generator produces 3-phase, 60 Hz electrical power. Transformers provide power at reduced voltage for in-plant use, and at increased voltage for distribution to the utility grid.

3.0 PLANT CAPACITY

Plant capacity is based on the boiler steam flow rate. The facility is designed to process 510 tons per day of 4500 Btu/lb MSW in two units to produce a total of 136,000 lb/hr of steam at 600 psig and 750°F. The steam flow rate per ton of MSW is proportional to the heating value of the garbage. As the heating value fluctuates, the feed rate of MSW is adjusted to maintain a constant steam rate to the turbine. Because one cannot continuously predict the heating value of MSW, the measured steam flow rate is used to determine the capacity of each unit. During compliance testing, plant operators maintained the steam rate of each unit as close to the design condition of 68,000 lb/hr as practical.

4.0 TEST PROGRAM

In May and June 1987, Westinghouse conducted emissions testing of the Bay County Facility in accordance with Permit Numbers AC-03-84703 and AC-03-84704. Environmental testing consultants from ETS, Inc. performed particulate emissions testing in accordance with EPA Methods 1,2,3 and 5, and visible emissions testing in accordance with FAC Rule 17-2.700(6)(a)9., DER Method 9.

All stack sampling was performed through the sampling ports at the 60-ft elevation of the stack. Two sampling ports, positioned 90 degrees from each other, are located on each flue. Visible emissions were measured simultaneously with the stack sampling for particulate emissions.

Although the majority of the testing was conducted by ETS, a number of others also participated. A certified observer from Westinghouse measured visible emissions during the June testing program and environmental specialists from Roy F. Weston, Inc., employed by the Bay County Authority, witnessed at least three compliance test runs at each stack. Westinghouse provided individuals to monitor the plant operating conditions and to coordinate the test runs.

5.0 TEST DATES

Emission tests for determining particulate matter concentration were conducted from April 22 through June 5, 1987. The Florida DER was notified and invited to attend the DER emission compliance and plant acceptance tests as listed in the table below. Mr. Bert Lent of the Panama City Office of the Florida DER observed at least one day of testing on May 13, 1987. Mr. Lent indicated that he would try to attend testing during June 4-5, 1987.

<u>Date</u>	<u>Test</u>
5/12 - 5/14	DER Emission Compliance Test
5/18 - 5/21	Facility 72-hr Acceptance Test
6/4 - 6/5	DER Emission Compliance Test

Officials from Roy F. Weston observed various tests during these three testing periods. Roy F. Weston acted as an independent third-party engineer to observe the stack test for the Bay County Resource Recovery Authority.

6.0 TESTING PERSONNEL

Individuals from Westinghouse (RESO), ETS, Inc., and Roy F. Weston, Inc. participated during various phases of the 1-1/2-month test program. The following is a list of the participating individuals and their responsibility during the test program.

<u>Company</u>	<u>Name</u>	<u>Responsibility</u>
Westinghouse	David Beachler	Test Coordinator
	Jill Weldon	Test Coordinator
	Spencer Brady	Visual Emissions Observer, 6/4-6/5
ETS, Inc.	Jeff Smith	Stack Sampling Test Team Leader
	Jim Eckenrode	Stack Sampling
	Tony Underwood	Stack Sampling
	John Richardson	Visual Emissions Observer, 5/12-14
Roy F. Weston, Inc.	Denise Alston	Test Observer. 5/12-14
	Norman Getz	Test Observer, 5/20-21
	John Mills	Test Observer, 6/5

7.0 RESULTS

Westinghouse has collected a considerable amount of test data on the operation of Units 1 and 2 at the Bay County Resource Recovery Facility. Testing began in April to evaluate the performance of the ESP and associated

equipment during the start-up and break-in phases of the facility. Since initial compliance testing (May 12-14) was conducted at extremely high firing rates, additional testing was required in late May and June to evaluate the units at the more typical design conditions of the facility.

Scheduled compliance testing was conducted on May 12, 13, 14, and June 4 and 5. The testing was originally to be conducted entirely during the week of May 12. During the original testing period, however, there were a number of problems that disrupted the normal operation of Unit 1. Major leaks were discovered in the valve seals between the fly ash collection hoppers and the pneumatic conveying system. In addition, the water seal at the bottom of the combustor/boiler was not complete, allowing air to leak into a stagnant region where it could possibly re-entrain dust and increase the particulate loading to the ESP.

Compliance efforts were resumed in June to complete the test program. During the week of June 1, Unit 1 was shutdown and, after extensive troubleshooting efforts, major leaks in the fly ash removal system were repaired. The bifurcated chute flap door seal was also secured to provide the water seal at the bottom of the furnace. Plant operation was resumed and Unit 1 compliance testing was rescheduled for June 5. Unit 2 was retested on June 4 to confirm earlier results.

Table 1 contains the emission compliance test results for Units 1 and 2 conducted on June 4-5, 1987. The average outlet particulate emissions for Units 1 and 2 are 0.0193 gr/dscf and 0.0243 gr/dscf corrected to 12% CO₂, respectively. The average emissions for both units comply with the particulate limit of 0.03 gr/dscf corrected to 12% CO₂ set by the Florida DER. The complete testing report prepared by ETS, Inc. containing test results, sample calculations, raw data sheets, and test procedures is given as Appendix A to this report.

Particulate measurements made during the scheduled compliance tests on May 12-14 and May 20-21 are included in Table 2. An environmental specialist

TABLE 1 EMISSION COMPLIANCE TEST RESULTS FROM BAY COUNTY ENERGY RESOURCES

BAY COUNTY COMPLIANCE TEST RESULTS UNIT 1							
DATE	TIME	FLUE GAS FLOW KDSCFM	FLUE GAS FLOW KACFM	STACK TEMP DEG F	STEAM FLOW KLB/HR	PERCENT OF RATED CAPACITY	PARTICULATE GR/DSCF @12%CO2
6/5	959	25.8	52.4	425.0	71.1	104.5	0.0140
6/5	1140	27.9	55.1	429.0	66.5	97.8	0.0240
6/5	1307	25.8	52.8	427.0	65.0	95.6	0.0200
AVERAGE					67.5	99.3	0.0193

BAY COUNTY COMPLIANCE TEST RESULTS UNIT 2							
6/4	945	27.7	52.6	429.0	69.7	102.5	0.0250
6/4	1310	28.4	58.1	449.0	62.7	92.2	0.0190
6/4	1525	29.2	59.0	451.0	62.3	91.6	0.0290
AVERAGE					64.9	95.4	0.0243

TABLE 2 ADDITIONAL TEST RESULTS FROM BAY COUNTY ENERGY RESOURCES

BAY COUNTY ADDITIONAL TEST RESULTS UNIT 1							
DATE	TIME	FLUE GAS FLOW KDS CFM	FLUE GAS FLOW KACFM	STACK TEMP DEG F	STEAM FLOW KLB/HR	PERCENT OF CAPACITY	PARTICULATE GR/DSCF @12% CO2
4/22	1436	25.0	45.5	373.0	58.6	86.2	0.0176
4/22 (1)	1652	25.9	50.3	387.0	70.4	103.5	0.0279
4/27	1505	24.9	48.1	441.0	68.9	101.3	0.0265
4/29	1214	19.9	39.0	441.0	61.1	89.9	0.0252
5/20 (2)	1542	29.8	49.8	426.0	70.4	104	0.0256
6/1	1903	25.5	51.2	426.0	64.0	94.1	0.0177
6/1	2029	23.7	52.3	436.0	57.2	84.0	0.0195
AVERAGE					64.4	94.7	0.0229

BAY COUNTY ADDITIONAL TEST RESULTS UNIT 2							
4/23	925	28.9	56.8	422.0	64.0	94.1	0.0161
4/23	1148	24.3	48.2	422.0	65.6	96.5	0.0215
4/23	1356	23.4	45.4	405.0	62.6	92.1	0.0192
4/30	957	27.4	51.7	427.0	NO DATA	NO DATA	0.0167
5/12 (3)	1350	25.7	54.2	437.0	76.0	112	0.0246
5/13 (3,4)	1635	23.7	48.2	408.0	72.0	106	0.0355
5/14 (3)	826	25.3	51.3	421.0	80.0	118	0.0157
5/21	1016	34.1	57.3	431.0	72.6	107	0.0172
5/21 (5)	1705	30.6	50.2	411.0	69.9	103	0.0184
6/1	927	25.9	54.5	436.0	64.5	94.8	0.0164
6/1	1045	24.3	52.3	428.0	60.8	89.4	0.0173
6/1	1215	25.2	55.4	426.0	57.8	85.0	0.0177
6/3	1023	25.8	52.0	438.0	59.8	87.9	0.0191
AVERAGE					67.1	98.7	0.0196

- (1) TEST DISCONTINUED AFTER 1/2 HOUR DUE TO PLANT SHUTDOWN
- (2) WITNESSED BY CONSULTANT FROM ROY F. WESTON
- (3) DER EMISSION COMPLIANCE TEST WITNESSED BY WESTON CONSULTANT
- (4) FURNACE WENT POSITIVE FOR A FEW MINUTES WHILE CONDUCTING THIS TEST WHEN AN AIR ACTUATOR VALVE WAS BEING REPAIRED.
- (5) INCINERATOR WAS FIRED WITH MUNICIPAL WASTE AND WOOD CHIP MIXTURE.

from Weston witnessed these tests for Bay County. Additional Method 5 particulate testing was conducted before and between compliance runs to evaluate whether the ESP was meeting its performance guarantees and to assist in plant troubleshooting. Table 2 contains the results from some of those tests which show an average particulate emission level of 0.0229 gr/dscf at 12% CO₂ for Unit 1 and 0.0196 gr/dscf at 12% CO₂ for Unit 2. Appendix C of this report contains the computer calculation sheets for each of the test runs listed in Table 2. This data is supplied to reinforce the compliance data and demonstrate overall reliability of the particulate removal system.

The results of Method 9 opacity measurements which were conducted during the June 4-5 compliance tests are contained in the report in Appendix A. Additional opacity measurements for the May compliance tests are contained in Appendix B. Visual measurements of opacity were continuously between 5 and 10% and confirm the low particulate levels measured by Method 5. The measurements meet, in all cases, the Florida DER requirements of less than 10% opacity and no more than 20% opacity for up to three minutes.

8.0 CONCLUSION

The results of scheduled testing indicate that Units 1 and 2 of the Bay County Resource Recovery Facility are in compliance with the particulate and visual emission levels required by the State of Florida Department of Environmental Regulations. The Method 5 particulate measurements conducted on June 4-5 at the design capacity of 255 ton per day per unit averaged 0.0193 gr/dscf at 12% CO₂ for Unit 1 and 0.0243 gr/dscf at 12% CO₂ for Unit 2. Method 9 opacity measurements were consistently at or less than 10% for both units during the test runs. Additional testing, conducted at the plant for verification and troubleshooting purposes, confirmed the low emission levels measured during the compliance test runs with average Unit 1 emissions of 0.0229 gr/dscf at 12% CO₂ and Unit 2 emissions of 0.0196 gr/dscf at 12% CO₂. The performance of the plant from an air quality standpoint is clearly within the acceptable range of less than 0.03 gr/dscf

particulate and less than 10% opacity required by the State of Florida Department of Environmental Regulations. Westinghouse (RESO) submits this report for the Bay County Resource Recovery Facility to the Florida DER and to request the issuance of an operating permit to burn MSW at the maximum plant design capacity rate of 190×10^6 Btu/hr or an equivalent of 510 TPD MSW with a heating value of 4500 Btu/lb.

APPENDIX A

**EMISSION COMPLIANCE REPORT FOR
JUNE 4 AND 5 COMPILED BY
ETS, INC.**

STATIONARY SOURCE SAMPLING
FOR
WESTINGHOUSE ELECTRIC CORP.
BAY COUNTY RESOURCE RECOVERY FACILITY - PANAMA CITY, FL

JUNE, 1987



ETS, inc.

3140 Chaparral Dr., SW

Suite C-103

Roanoke, VA 24018

ETS CONTRACT NO: 87-718-T

CLIENT P.O. NO: RN-15467-H

*Pollution Control Consultants
Specializing In
Fabric Filtration and Sulfur Dioxide Removal*

REPORT CERTIFICATION

The sampling and analysis performed for this report were carried out under my direction and supervision.

DATE: 6/16/87

SIGNATURE: James P. Eckenrode
James Eckenrode

I have reviewed all testing details and results in this report and hereby certify that the test report is authentic and accurate.

DATE: 6/16/87

SIGNATURE: Gary P. Greiner
Gary P. Greiner
Vice President

TABLE OF CONTENTS

	<u>PAGE</u>
INTRODUCTION.....	1
SUMMARY OF RESULTS.....	1
FACILITY DESCRIPTION.....	3
FIELD TESTING.....	3
DISCUSSION.....	6
APPENDICES.....	8
A-1 SAMPLE CALCULATIONS AND COMPUTER CALCULATION SHEET	
B-1 EPA METHODS 1-5	
C-1 CLEAN-UP/FIELD DATA	
D-1 CALIBRATIONS	

LIST OF TABLES

TABLE I - EMISSION COMPLIANCE TEST RESULTS.....	2
---	---

LIST OF FIGURES

FIGURE 1 - METHOD 5 SAMPLING TRAIN.....	5
---	---

INTRODUCTION

Source sampling was performed on the outlet stack of Units 1 and 2 serving the Bay County Energy Resources MSW incinerators located in Panama City, FL. The purpose of these formal tests was to determine the particulate emissions of the waste incinerators. The test series consisted of EPA Methods 1-5. The tests were performed by Jim Eckenrode, Jeff Smith, and Tony Underwood of ETS, Inc. on June 4 and 5, 1987.

The following people took part or observed this test series:

PERSONNEL

COMPANY AFFILIATE

Jim Eckenrode

ETS, Inc.

Jeff Smith

ETS, Inc.

Tony Underwood

ETS, Inc.

Jill Welden

Westinghouse

Spencer Brady

Westinghouse

John Mills

Weston

SUMMARY

Particulate results are summarized in Table I. The State of Florida requires the incinerator emission concentrations be .03 gr/SCF corrected to 12% CO₂. The average emission rate for the three runs conducted on Unit #1 was 0.0190 gr/SCF corrected to 12% CO₂. The average emission rate for the three runs of Unit #2 was 0.0241 gr/SCF corrected to 12% CO₂. The average of each unit, as well as each test run, were well under the State of Florida standard. These tests were conducted while the system was operating

TABLE I

BAY COUNTY ENERGY RESOURCES EMISSION COMPLIANCE TESTS SUMMARY - JUNE 4-5, 1987

UNIT NO.	TEST REFERENCE NO.	1987 DATE	TIME	Q _s KDSCFM	Q _A KACFM	° F TEMP.	% MOISTURE	STEAM FLOW KLB/HR	% RATED CAPACITY	PARTICULATE CORRECTED 12% gr/SCF
1	1	6/5	0952	25.8	52.4	424.6	17.7	71.1	105	.0140
	2	6/5	1140	27.9	55.1	429.2	14.9	66.5	98	.0236
	3	6/5	1305	25.8	51.8	426.9	16.6	65.0	96	<u>.0195</u>
										AVERAGE =0.0190
<hr/>										
2	1	6/4	0945	26.7	52.6	428.8	14.6	69.7	103	.0248
	2	6/4	1311	28.4	58.1	448.8	16.0	62.7	92	.0188
	3	6/4	1527	29.2	59.0	451.3	14.7	62.3	92	<u>.0288</u>
										AVERAGE =0.0241

2

between 92% and 105% of rated capacity. Inlet feed conditions and operating data was monitored by Westinghouse personnel. John Mills of Weston observed the tests run on 6/5/87.

FACILITY DESCRIPTION

The facility uses two Westinghouse/O'Connor combustors to process 510 tons per day of municipal solid waste (MSW) and waste wood. Heat generated by the combustion of waste in the combustor produces steam to drive a turbine generator.

The flue gases leaving the boiler air heater pass through the pollution control equipment. The air pollution control equipment consists of an Electrostatic Precipitator (ESP), in which the small suspended particles (fly ash) contained in the flue gas, are collected on the collection plates. Periodic rapping of the plates causes fly ash to fall into hoppers located at the bottom of the Electrostatic Precipitator. From there the fly ash is then transferred by a pneumatic conveying system to the boiler bottom ash conveyor (submerged drag conveyor) where it is mixed with bottom ash. The flue gas is drawn from the ESP by the induced draft fan which maintains a controlled slightly negative pressure through the entire system. Using an induced draft fan for each combustion train, the flue gases from each unit discharge to the atmosphere through a separate flue in the common stack.

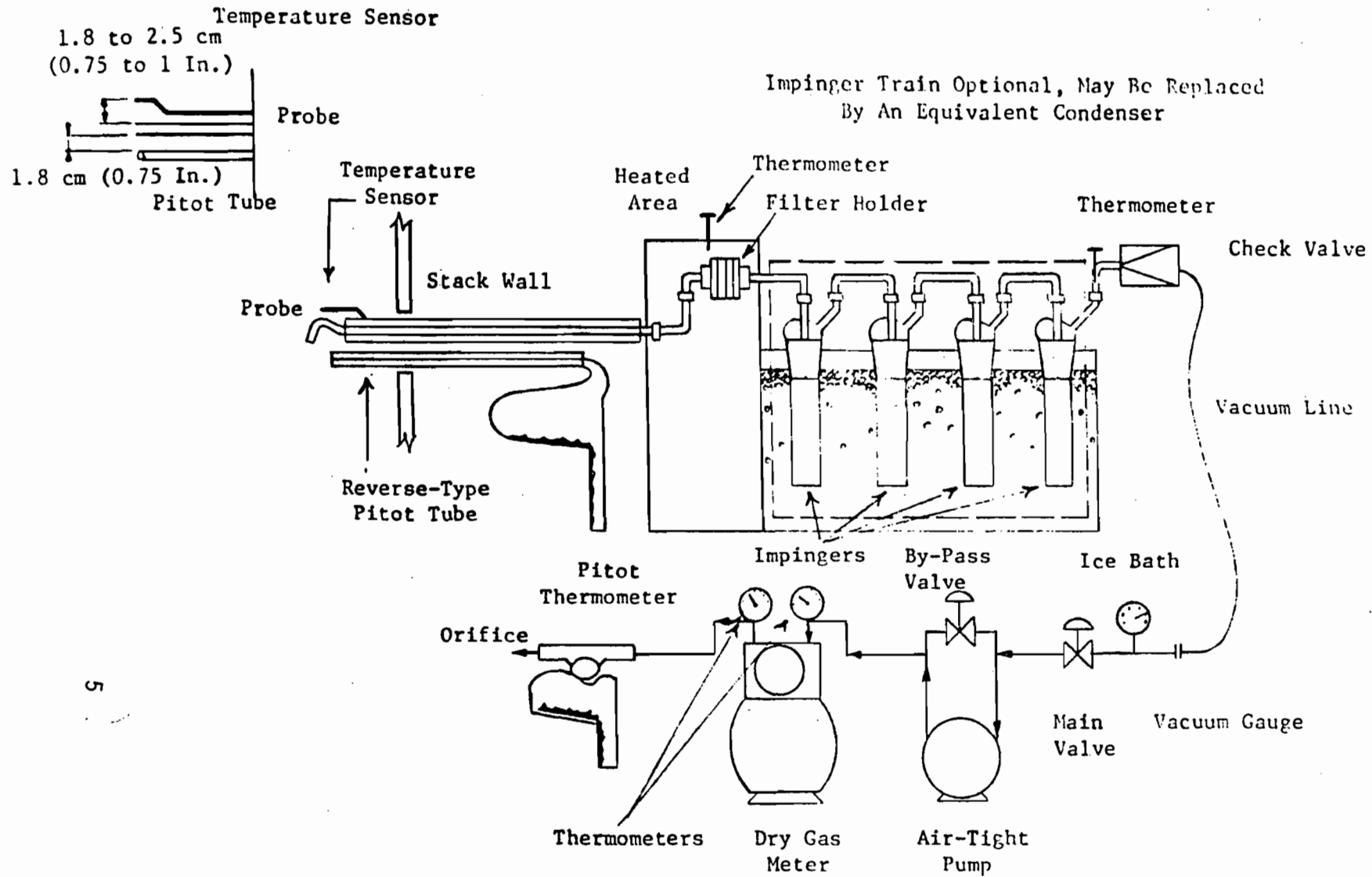
FIELD TESTING

Field testing was conducted on 6/4 and 6/5/87. All sampling

and analytical procedures used were those recommended by the United States Environmental Protection Agency. EPA Methods 1-5 were used in performing all of the compliance particulate tests.

Each test run was conducted to determine particulate mass emissions at the outlet. The sampling point number and locations were determined according to EPA Method 1 located in the Appendix. The round stack measured 53" I.D. and was traversed on two axes located 90° apart, with 6 traverse points per port. Each point was sampled for 5 minutes, resulting in a 60 minute particulate test excluding the time required to change ports. All tests used a .254" I.D. nozzle.

Testing was conducted utilizing a Method 5 sampling train and a Teflon lined stainless steel probe (see Figure 1). From the nozzle, the sample gas was pulled through the heated probe and then through a heated glass fiber filter which was maintained at a temperature sufficiently high to prevent the condensation of water ($248^{\circ} \pm 25^{\circ}$). Sample gas was subsequently passed through an impinger train consisting of four glass impingers immersed in an ice bath. The first and second impingers contained approximately 100 mls of water. The third impinger was initially dry and the fourth initially contained 200 gms of silica gel. Pre and post test impinger weights for the tests were taken on-site using a Mettler top loading balance. The heated stainless steel probe measured 84 inches in length.



Method 5 Sampling Train

FIGURE 1

Leak checks were performed on the sampling train before and after the tests were run. Post test leak checks were recorded on the field data sheets. All of the field data sheets can be found in Appendix C.

Velocity measurements were made according to Method 2, flue gas composition determinations and molecular weights according to Method 3, moisture determinations according to Method 4, and particulate emission determinations according to Method 5.

Spencer Brady of Westinghouse monitored the visible opacity readings on 6/4 and 6/5 according to EPA Method 9.

DISCUSSION

The tests on Unit #1 were conducted on 6/5 and the tests on Unit #2 were conducted on 6/4. The flows for Unit #1 were slightly less than the flows for Unit #2. In the first unit, gas flows averaged 53,100 ACFM whereas the gas flows for Unit #2 averaged 56,600 ACFM. Unit #1 also had a lower gas temperature. These results may be viewed on Table I. A complete list of calculated values can be seen in Appendix A.

The initial test period on 6/5/87 Unit #1 was aborted at 0839. Jill Welden informed the test team to void the test after sampling for 30 minutes. The test was interrupted because the system was not running at full capacity. The filter and probe wash was discarded and testing re-started when the system was back

to full capacity. Three complete tests were run following the aborted test period.

Weston personnel observed the stack testing and control room operations during the tests on 6/5. Westinghouse personnel were responsible for collecting operational data in the control room.

Lab Activity

Particulate weighings were performed at the ETS lab. Report sheets for each run are included with the computer printouts located in Appendix A.

Post test equipment calibrations were performed and data sheets are included in Appendix D.

Data Analysis

Particulate test calculations such as flue gas volume, flue gas velocity, mass emissions and percent isokinetics were performed by computer. Sample calculations and computer printouts of each run can be found in Appendix A.

APPENDIX A-1
SAMPLE CALCULATIONS
AND
COMPUTER CALCULATION SHEET

ETS

GENERAL

Name of firm	BAY RESOURCES
Location of plant	PANAMA CITY, FLA.
Type of unit	MSW INCINERATOR
Control equipment	ESP
Sampling location	OUTLET UNIT #1
Pollutants sampled	
Date test started	JUNE 5, 1987

REFERENCE METHOD #1

Area of stack, ft ²	A _u =	15.32
No. of equivalent dia. upstream	=	NA
No. of equivalent dia. downstream	=	NA
No. of traverse points	=	12
Total test time per run	t =	60

SPECIFIC
RUN No. 1
DATA AND CALCULATIONS

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ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

REFERENCE METHOD #2

Standard temperature

Degrees Fahrenheit

$t_{std} = 68$

Degrees Rankine

$$T_{std} = t_{std} + 459.67$$

$$T_{std} = 68 + 459.67$$

$T_{std} = 528$

Average stack temperature

Degrees Fahrenheit

$t_s = 425$

Degrees Rankine

$$T_s = t_s + 459.67$$

$$T_s = 425 + 459.67$$

$T_s = 884$

Barometric pressure, in. Hg

$P_b = 30.02$

Static pressure, in. H₂O

$P_{static} = -.55$

Stack absolute pressure, in. Hg

$$P_s = P_b + (P_{static} / 13.6)$$

$$P_s = 30.02 + (-.55 / 13.6)$$

$P_s = 29.98$

Average delta p, in. H₂O

$\Delta p = 0.608$

ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

REFERENCE METHOD #3

Gas analysis, % average

sample gas	1	2	3	avg
%CO ₂	11.5	11.5	11.5	11.5
%O ₂	7.3	7.2	7.2	7.2
%CO	0.0	0.0	0.0	0.0

%N₂ determination, % average

$$\%N_2 = 100 - (\%CO_2 + \%O_2 + \%CO)$$

$$\%N_2 = 100 - (11.5 + 7.2 + 0.0)$$

$$\%N_2 = 81.3$$

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ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

REFERENCE METHOD #4

Water vapor volume collected, SCF

Constants

$K_1 = \text{ft}^3/\text{ml}$ (english/metric units) @ 68 t _{std}	$K_1 =$	0.04707
$K_2 = \text{ft}^3/\text{ml}$ (english/metric units) @ 68 t _{std}	$K_2 =$	0.04715

Impingers

Volume initial, ml	$V_i =$	1147.0
Volume final, ml	$V_f =$	1303.0

$$V_{\text{water}} = K_1 (V_f - V_i)$$

$$V_{\text{water}} = 0.04707 * (1303.0 - 1147.0) \quad V_{\text{water}} = 7.34$$

Silica gel

Weight initial, g	$W_i =$	633.0
Weight final, g	$W_f =$	643.0

$$V_{\text{silica}} = K_2 (W_f - W_i)$$

$$V_{\text{silica}} = 0.04715 * (643.0 - 633.0) \quad V_{\text{silica}} = 0.47$$

ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

REFERENCE METHOD #5

Meter calibration factor	V... =	1.001
Delta H, in. H ₂ O average	ΔH =	1.537
Nozzel diameter, in.	D.. =	0.254
$A.. = \frac{\pi (D.. / 2)^2}{144}$		
$A.. = \frac{3.1416 * (0.254 / 2)^2}{144}$	A.. =	0.000352
Meter temp., average		
Degree fahrenheit	t... =	94
Degree rankine		
T... = t... + 459.67		
T... = 94 + 459.67	T... =	554
Volume metered, cf	V... =	37.846

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ETS
Test of
BAY RESOURCES

Date : 6/5/87 Run No. 1

CALCULATIONS

Dry gas volume corrected to
standard conditions

$$P_{std} = P_b + \Delta H / 13.6$$

$$P_{std} = 30.02 + 1.54 / 13.6$$

$$P_{std} = 30.13$$

SCF calculation

$$V_{std} = V_{meas} \cdot Y_{meas} \cdot \frac{T_{std}}{T_{meas}} \cdot \frac{P_{meas}}{P_{std}}$$

$$V_{std} = 37.85 * 1.001 * \frac{528}{554} * \frac{30.13}{29.92}$$

$$V_{std} = 36.367$$

Moisture content of stack gas

$$R_{meas} = \frac{V_{H_2O,meas} + V_{H_2O,std}}{V_{meas} + V_{H_2O,meas} + V_{H_2O,std}}$$

$$R_{meas} = \frac{7.34 + 0.47}{7.34 + 0.47 + 36.367}$$

$$R_{meas} = 0.177$$

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ETS
 Test of
 BAY RESOURCES
 Date : 6/5/87 Run No. 1

CALCULATIONS

Molecular weight of stack gas, lb/bl-mole

Dry molecular weight

$$M_d = .44 \%CO_2 + .32 \%O_2 + .28 \%CO + .28 \%N_2$$

$$M_d = .44 * 11.5 + .32 * 7.2 + .28 * 0.0 + .28 * 81.3$$

$$M_d = 30.13$$

Wet molecular weight

$$M_w = M_d (1 - R_{H_2O}) + 18 R_{H_2O}$$

$$M_w = 30.13 * (1 - .177) + 18 * .177$$

$$M_w = 27.98$$

Excess air, %

$$\%EA = \frac{\%O_2 - .5 \%CO}{.264 \%N_2 - (\%O_2 - .5 \%CO)} 100$$

$$\%EA = \frac{7.2 - .5 * 0.0}{.264 * 81.3 - (7.2 - .5 * 0.0)} 100$$

$$\%EA = 50.9$$

Average stack gas velocity, ft/sec

Constant

$$K_v = \text{ft/sec (english units)}$$

$$K_v = 85.49$$

$$v_s = K_v C_v (-\Delta p)_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$v_s = 85.49 * 0.84 * 0.774 * \sqrt{\frac{884}{.29.98 * 27.98}}$$

$$v_s = 57.04$$

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ETS
Test of
RAY RESOURCES

Date : 6/5/87 Run No. 1

CALCULATIONS

Average stack volumetric flow
rate, ACFM

$$Q_s = 60 v_s A_s$$

$$Q_s = 60 * 57.04 * 15.3$$

$$Q_s = 52430$$

Average stack volumetric flow
rate @ standard conditions, SCFM

$$Q_s = 60 v_s A_s (1 - B_{ws}) \frac{T_{std} P_s}{P_{std} T_s}$$

$$Q_s = 60 * 57.04 * 15.3 * (1 - .177) * \frac{528 * 29.98}{29.92 * 884}$$

$$Q_s = 25805$$

Isokinetic, %

$$\%I = \frac{T_s V_{std} P_{std}}{A_s \theta v_s P_s T_{std} 60 (1 - B_{ws})} 100$$

$$\%I = \frac{884 * 36.367 * 29.92}{.000352 * 60 * 57.04 * 29.98 * 528 * 60 * (1 - .177)} * 100$$

$$\%I = 102.3$$

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ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

EMISSION CALCULATIONS

Pollutant mass rate, lb/hr

Constant

$$K_D = \text{gm/lb-hr (english metric units)} \quad K_D = 0.132159$$

Dry - (Particulate catch before and including the filter.)

$$PMR_D = \frac{m_D}{V_{\text{measured}}} \cdot Q \cdot K_D$$

$$PMR_D = \frac{0.0315}{36.37} * 25805 * 0.132159 \quad PMR_D = 2.96$$

Wet - (Particulate catch including dry catch and solids condensed beyond the filter in glassware and impingers.)

$$PMR_W = \frac{m_W}{V_{\text{measured}}} \cdot Q \cdot K_D$$

$$PMR_W = \frac{0.0315}{36.37} * 25805 * 0.132159 \quad PMR_W = 2.96$$

Concentration, gr./SCF

Dry - (Particulate catch before and including the filter.)

$$C_{D,D} = 15.43 \frac{m_D}{V_{\text{measured}}}$$

$$C_{D,D} = 15.43 * \frac{0.0315}{36.37} \quad C_{D,D} = 0.0134$$

Wet - (Particulate catch including dry catch and solids condensed beyond the filter in glassware and impingers.)

$$C_{D,W} = 15.43 \frac{m_W}{V_{\text{measured}}}$$

$$C_{D,W} = 15.43 * \frac{0.0315}{36.37} \quad C_{D,W} = 0.0134$$

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ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

EMISSION CALCULATIONS

Concentration @ 50% EA, gr./SCF

Dry - (Particulate catch before and including the filter.)

$$C_{L500} = \frac{C_{LJ} (100 + \%EA)}{150}$$

$$C_{L500} = \frac{0.0134 * (100 + 50.9)}{150}$$

$$C_{L500} = 0.0135$$

Wet - (Particulate catch including dry catch and solids condensed beyond the filter in glassware and impingers.)

$$C_{L500} = \frac{C_{LW} (100 + \%EA)}{150}$$

$$C_{L500} = \frac{0.0134 * (100 + 50.9)}{150}$$

$$C_{L500} = 0.0135$$

Concentration @ 12% CO₂, gr./SCF

Dry - (Particulate catch before and including the filter.)

$$C_{L12CO2} = C_{LJ} \frac{12}{\%CO_2}$$

$$C_{L12CO2} = 0.0134 * \frac{12}{11.5}$$

$$C_{L12CO2} = 0.0140$$

Wet - (Particulate catch including dry catch and solids condensed beyond the filter in glassware and impingers.)

$$C_{L12CO2} = C_{LW} \frac{12}{\%CO_2}$$

$$C_{L12CO2} = 0.0134 * \frac{12}{11.5}$$

$$C_{L12CO2} = 0.0140$$

ETS
Test of
BAY RESOURCES
Date : 6/5/87 Run No. 1

EMISSION CALCULATIONS

Concentration @ 7% O₂, gr./SCF

Dry - (Particulate catch before and including the filter.)

$$C_{\text{dry}} = C_{\text{wet}} \frac{20.9 - 7}{20.9 - \%O_2}$$

$$C_{\text{dry}} = 0.0134 * \frac{20.9 - 7}{20.9 - 7.2}$$

$$C_{\text{dry}} = 0.0136$$

Wet - (Particulate catch including dry catch and solids condensed beyond the filter in glassware and impingers.)

$$C_{\text{wet}} = C_{\text{dry}} \frac{20.9 - 7}{20.9 - \%O_2}$$

$$C_{\text{wet}} = 0.0134 * \frac{20.9 - 7}{20.9 - 7.2}$$

$$C_{\text{wet}} = 0.0136$$

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ETS
 Test of
 BAY RESOURCES
 Date : 6/5/87 Run No. 1

SAMPLE RECOVERY SUMMARY

	Con 1	Con 2	Con 3	Con 4	Con 5
Tare	1.3361	79.1541		633.0	
Gross	1.3489	79.1729		643.0	
m_1	0.0128	m_2	NA		
	C_1	0.000		C_4	
	V_{1a}	100.000		V_{4a}	
	ρ_1	0.7850		ρ_4	
	W_1	0.00		W_4	0.00
	m_2	0.0187		m_5	NA

Tare and gross weights are in grams. Tare weight is the container without sample, gross weight is the container with sample. Containers 1 thru 5 used in weighing recovered sample are :

Container 1 - filter. (includes weight of filter)
 Filter identification : 87-304

Container 2 - washings from probe, cyclone, and front of filter holder.

Container 3 - solids trapped in impingers

Container 4 - silica gel. (method #4)

Container 5 - washings from fluid filled impingers and connecting glassware.

$$m_1 \text{ \& } m_2 = \text{Gross}_{\text{Con.}} - \text{Tare}_{\text{Con.}}$$

Additional inputs for containers 2 and 5 are required if container is used. The inputs are :

V_{1a} - volume of the acetone wash in milliliters.

C_1 - acetone blank residue concentration in milligrams per gram.

ρ_1 - density of acetone in grams per milliliter.

$$W_1 = C_1 V_{1a} \rho_1$$

ETS
 Test of
 BAY RESOURCES
 Date : 6/5/87 Run No. 1

SAMPLE RECOVERY SUMMARY

Dry* particulate mass, g

$$m_d = m_1 + m_2$$

$$m_d = 0.0128 + 0.0187$$

$$m_d = 0.0315$$

Wet* particulate mass, g

$$m_w = m_3 + m_4 + m_d$$

$$m_w = \text{NA} + \text{NA} + 0.0315$$

$$m_w = 0.0315$$

Fluid volumes for impingers in milliliters :

V_i - initial

V_f - final

	Imp 1	Imp 2	Imp 3	total
V_i	581.0	566.0	0.0	1147.0
V_f	712.0	591.0	0.0	1303.0

- * Dry - (Particulate catch before and including the filter.)
- * Wet - (Particulate catch including the dry catch and solids condensed beyond the filter in glassware and impingers.)

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ETC
PAY RESOURCES
TEST #4

FIELD DATA SUMMARY

Location: PANAMA CITY, FLA.	Meter calibration (V _m): 1.001	Sheet : 1 of 1
Operator: JIM ECKENRODE	Pitot tube (C _p): 0.84	Nozzle I.D.: .254
Date: 6/5/87	Probe heater setting: 250	Nozzle diameter, in.: 0.254
Run number: 1	Ambient temperature: 80	Thermometer I.D.: NONE
Stack diameter, in: 53.00	Bar. press. (P _b), in Hg: 30.02	Final leak rate, cfm: 0.000
	Assumed moisture, %: 15.0	Vac. @ leak check, in Hg: 6.0
Sample box I.D.: NUTECH	Static pres. (P _s), in H ₂ O: -.55	Filter I.D.: 87-304
Meter box I.D.: NUTECH #3	Reference (Δp): 0.650	Meter (ΔH ₀): 2.052

Point No.	Time (24h)	Stack orifice	Vol. flow, in. H ₂ O	Pitot tube	Gas sample volume, ft ³	Temp. in. Hg		FLT	EXT	Pvc	Lkc
						In	Out				
A 1	09:52:00	428	0.740	1.890	121.757	93	92	236	0	4.0	
A 2	09:57:00	428	0.790	2.000	125.200	91	92	246		4.0	
A 3	10:02:00	428	0.730	1.870	128.800	93	92	250		4.0	
A 4	10:07:00	429	0.680	1.720	132.300	92	92	252		4.0	
A 5	10:12:00	429	0.600	1.500	135.600	93	93	247		3.0	
A 6	10:17:00	396	0.400	1.000	138.600	93	93	252	0	2.0	
A	10:22:00	port	end		141.300						
B 1	10:24:00	428	0.610	1.550	141.300	95	94	249	0	3.0	
B 2	10:29:00	434	0.680	1.720	144.400	95	95	256		4.0	
B 3	10:34:00	435	0.690	1.750	147.700	96	95	257		4.0	
B 4	10:39:00	434	0.560	1.420	151.200	95	95	257		4.0	
B 5	10:44:00	433	0.500	1.250	154.400	96	96	263		3.0	
B 6	10:49:00	393	0.310	0.770	157.300	97	96	260	0	2.0	
B	10:54:00	port	end		159.603						
run	end							max	max		
		avg	avg	avg	sum	avg	avg	263	0	max	
		425	0.608	1.537	37.846	94	94	min	min	4.0	
								236	0		

FLT - Filter temp., °F
 EXT - Gas temp. from train exit, °F
 Pvc - Pump vacuum, in. Hg
 Lkc - Leak check leak rate, cfm

Best Available Copy

XDD2 avg.	11.5	pmrd	2.956468948082777
XQ2 avg.	7.233333333333333	pmrw	2.956468948082777
XCD avg.	0	ced	1.337677366069144D-02
As stack area	15.32071832265625	cew	1.337677366069144D-02
Vm avg. DGM	37.846	cew50%	1.345378000791836D-02
Dp avg. delta p	.6075	cew50%	1.345378000791836D-02
ts avg. temp	424.5833333333333	ced12	1.395837251550412D-02
DH avg. delta H	1.536666666666667	cew12	1.395837251550412D-02
tmi avg. temp	94.08333333333333	ced7	1.360515760761458D-02
tmo avg. temp	93.75	cew7	1.360515760761458D-02
XN2 avg.	81.26666666666667	saturation flag	-1
Md mole wt. dry	30.12933335105578	Bws saturated	.1768698385098112
Ms mole wt.	27.98402012002388	Vm corrected	37.846
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	93.91666666666667	Vf impingers	1303
Tm abs temp	553.5866666666667	Vi impingers	1147
Ts abs temp	884.2533333333333	sqr(Dp) avg.	.7736374035997174
Pm abs pre	30.13299019290914	Bws kick flag	0
Ps abs pre	29.97955882466376	Vm C kick flag	0
Vm(std)	36.36733572409156	not in use	NA
Vwc(std)	7.34292	Wa2	.0019625
Vwsg(std)	.4715	Wa5	0
Bws	.1768698385098112	K1 constant	.04707
vs velocity	57.03648374168055	K2 constant	.04715
Q vol rate	25804.82891193186	Tstd abs temp	527.67
time min.	60	container 1	.01279
XI isokinetic	102.2688935470582	container 2	.0187380374999061
Qa vol rate	52430.39409126303	container 3	NA
md mass dry	3.152803749990609D-02	container 4	10
mw mass wet	3.152803749990609D-02	container 5	NA
XEA	50.8635080757911	not in use	NA

* Variables defined in sample calculations.

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UNIT #1

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15:37:33

06-15-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

XCO2 avg.	11.066666666666667	pmrd	5.204822979421487
XO2 avg.	9.333333333333333	pmrw	5.204822979421487
XCO avg.	0	ced	2.179582086823444D-02
As stack area	15.32071832265625	cew	2.179582086823444D-02
Vm avg. DGM	40.201	ced50%	2.614065496509841D-02
Dp avg. delta p	.6725	cew50%	2.614065496509841D-02
ts avg. temp	429.25	ced12	2.363402262820604D-02
DH avg. delta H	1.710833333333333	cew12	2.363402262820604D-02
tmi avg. temp	99.75	ced7	2.619267248522116D-02
tmo avg. temp	98.58333333333333	cew7	2.619267248522116D-02
XN2 avg.	79.6	saturation flag	-1
Md mole wt. dry	30.1440000017484	Bws saturated	.1486586607112696
Ms mole wt.	28.33868922581083	Vm corrected	40.201
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	99.16666666666667	Vf impingers	1257
Tm abs temp	558.8366666666667	Vi impingers	1128
Te abs temp	888.92	sqr(Dp) avg.	.8154478297451962
Pm abs pres	30.14579656509895	Bws kick flag	0
Pe abs pres	29.97882353056673	Vm C kick flag	0
Vm(std)	38.28367482317118	not in use	NA
Vwc(std)	6.07203	Wa2	.0019625
Vwg(std)	.61295	Wa5	0
Bws	.1486586607112696	K1 constant	.04707
vs velocity	59.89974189443064	K2 constant	.04715
Qs vol rate	27881.22005939102	Tstd abs temp	527.67
time min.	60	container 1	2.918999999999999D-02
X1 isokinetic	99.64027100939907	container 2	2.488803749990775D-02
Qa vol rate	55062.42438986302	container 3	NA
md mass dry	5.407803749990775D-02	container 4	13
mw mass wet	5.407803749990775D-02	container 5	NA
XEA	79.90137964839974	not in use	NA

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UNIT #1
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15:37:54

06-15-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

XN2 avg. 11.4
 XO2 avg. 8.733333333333333
 XCO avg. 0
 Ae stack area 15.32071832265625
 Vm avg. DGM 37.669
 Dp avg. delta p .5933333333333333
 ts avg. temp 426.9166666666667
 DH avg. delta H 1.5058333333333333
 tmi avg. temp 99.25
 tmp avg. temp 97.75
 XN2 avg. 79.86666666666667
 Md mole wt. dry 30.173333333889643
 Me mole wt. 28.15001095268566
 An area nozzle 3.51880193415625D-04
 tm temp meter 98.5
 Tm abs temp 558.17
 Ts abs temp 886.5866666666667
 Pm abs pre 30.13072303610999
 Ps abs pre 29.97882353056673
 Vm(std) 35.89732158881965
 Vwg(std) 6.44859
 Vwg(std) .70725
 Bws .166209396381665
 vs velocity 56.34732929108269
 Qs vol rate 25754.60632677361
 time min. 60
 X isokinetic 101.1440140969227
 Qa vol rate 51796.89361815814
 md mass dry 4.310803749990768D-02
 mw mass wet 4.310803749990768D-02
 XEA 70.70685328152157

pmd 4.08731404274903
 pmrw 4.08731404274903
 ced 1.852943309371419D-02
 cew 1.852943309371419D-02
 ced50X 2.108734144345626D-02
 cew50X 2.108734144345626D-02
 ced12 1.950466641443599D-02
 cew12 1.950466641443599D-02
 ced7 .0211692428227097
 cew7 .0211692428227097
 saturation flag -1
 Bws saturated .166209396381665
 Vm corrected 37.669
 total # points 12
 Vf impingers 1298
 Vi impingers 1161
 sqr(Dp) avg. .765534402716032
 Bws kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 .0019625
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 3.023000000000001D-02
 container 2 1.287803749990768D-02
 container 3 NA
 container 4 15
 container 5 NA
 not in use NA

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unit #2

XCO2 avg.	11.366666666666667	pmrd	5.371203480508901
XO2 avg.	7.733333333333333	pmrw	5.371203480508901
XCO avg.	0	ced	2.347715933898708D-02
As stack area	15.320718322265625	caw	2.347715933898708D-02
Vm avg. DGM	38.487000000000001	ced50%	2.453542599677754D-02
Dp avg. delta p	.6225	caw50%	2.453542599677754D-02
ts avg. temp	428.8333333333333	ced12	2.478527085640865D-02
DH avg. delta H	1.5825	caw12	2.478527085640865D-02
tmi avg. temp	104.5	ced7	2.478474799828298D-02
tmo avg. temp	103.41666666666667	caw7	2.478474799828298D-02
XN2 avg.	80.9	saturation flag	-1
Md mole wt. dry	30.12800001402696	Bws saturated	.1461575121652505
Ms mole wt.	28.35540170443666	Vm corrected	38.487000000000001
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	103.9583333333333	Vf impingers	1249
Tm abs temp	563.6283333333333	Vi impingers	1132
Ts abs temp	888.5033333333333	sqr(Dp) avg.	.779277121052837
Pm abs pre	30.11636029085383	Bws kick flag	0
Ps abs pre	29.95808823644971	Vm C kick flag	0
Vm(std)	36.30434612283541	not in use	NA
Vwc(std)	5.50719	Wa2	.0019625
Vwsg(std)	.70725	Wa5	0
Bws	.1461575121652505	K1 constant	.04707
vs velocity	57.23228821468709	K2 constant	.04715
vol rate	26711.91958358461	Tstd abs temp	527.67
run time min.	60	container 1	3.038999999999997D-02
XI isokinetic	98.62490004518748	container 2	2.484803749990571D-02
Qa vol rate	52610.38600189799	container 3	NA
md mass dry	5.523803749990569D-02	container 4	15
mw mass wet	5.523803749990569D-02	container 5	NA
XEA	56.76146531939914	not in use	NA

unit #2

XCO2 avg.	10.4	pmrd	3.955833269966886
XO2 avg.	8.833333333333333	pmrw	3.955833269966886
XCO avg.	0	ced	1.625206685830415D-02
As stack area	15.320718322265625	cew	1.625206685830415D-02
Vm avg. DGM	41.792	ced50%	.0184979433588628
Dp avg. delta p	.7333333333333333	cew50%	.0184979433588628
ts avg. temp	448.8333333333333	ced12	1.875238483650479D-02
DH avg. delta H	1.851666666666667	cew12	1.875238483650479D-02
tmi avg. temp	101.8333333333333	ced7	1.872130361373205D-02
tmo avg. temp	99.91666666666667	cew7	1.872130361373205D-02
XN2 avg.	80.76666666666667	saturation flag	-1
Md mole wt. dry	30.01733334163825	Bws saturated	.1595899111659054
Ma mole wt.	28.09948818119513	Vm corrected	41.792
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	100.875	Vf impingers	1262
Tm abs temp	560.545	Vi impingers	1114
Te abs temp	908.5033333333333	egr(Dp) avg.	.8478833877467027
Pm abs pres	30.13615195696535	Bws kick flag	0
Pe abs pres	29.95955882466376	Vm C kick flag	0
Vm(std)	39.66480843599196	not in use	NA
Vwc(std)	6.96636	Wa2	.0019625
Vwsg(std)	.5658	Wa5	0
Bws	.1595899111659054	K1 constant	.04707
ve velocity	63.25241044705001	K2 constant	.04715
Q vol rate	28418.97560553189	Tstd abs temp	527.67
run time min.	60	container 1	.03128
XI isokinetic	101.2814703368752	container 2	1.049803749990718D-02
Qa vol rate	58144.34182129756	container 3	NA
md mass dry	4.177803749990719D-02	container 4	12
mw mass wet	4.177803749990719D-02	container 5	NA
YEA	70.72853120904214	not in use	NA

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06-15-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

XO2 avg.	11.633333333333333	pmrd	6.970056641344015
XO2 avg.	8.533333333333333	pmrw	6.970056641344015
XCO avg.	0	csd	2.789144751735134D-02
As stack area	15.320718322265625	csw	2.789144751735134D-02
Vm avg. DGM	41.904	csd50%	3.124482566401508D-02
Dp avg. delta p	.7616666666666667	csw50%	3.124482566401508D-02
ts avg. temp	451.3333333333333	ced12	2.877054758236815D-02
DH avg. delta H	1.916666666666667	csw12	2.877054758236815D-02
tmi avg. temp	90	csd7	3.134968639976257D-02
tmo avg. temp	89.5	csw7	3.134968639976257D-02
XN2 avg.	79.83333333333333	saturation flag	-1
Md mole wt. dry	30.20266667306423	Bws saturated	.147369669678075
Ms mole wt.	28.4043637162631	Vm corrected	41.904
An area nozzle	3.51880193415625D-04	total # points	12
ts temp meter	89.75	Vf impingers	1283
Tm abs temp	549.42	Vi impingers	1146
Ts abs temp	911.0033333333333	sqr(Dp) avg.	.863974970105123
Pm abs pres	30.140931368596	Bws kick flag	0
Ps abs pres	29.95147058959651	Vm D kick flag	0
Vm(std)	40.58285314591153	not in use	NA
Vwc(std)	6.44859	Wa2	.0019625
Vwag(std)	.5658	Wa5	0
Bws	.147369669678075	K1 constant	.04707
vs velocity	64.20282517269598	K2 constant	.04715
Qs vol rate	29177.2458076676	Tstd abs temp	527.67
time min.	60	container 1	4.991000000000001D-02
Xi isokinetic	100.9325714585557	container 2	.0234480374999072
Qa vol rate	59018.00399937716	container 3	NA
md mass dry	7.335803749990721D-02	container 4	12
mw mass wet	7.335803749990721D-02	container 5	NA
XEA	68.03444305594532	not in use	NA

APPENDIX B-1
EPA METHODS 1-5

METHOD 1—SAMPLE AND VELOCITY TRAVERSES FOR STATIONARY SOURCES

1. Principle and Applicability

1.1 Principle. To aid in the representative measurement of pollutant emissions and/or total volumetric flow rate from a stationary source, a measurement site where the effluent stream is flowing in a known direction is selected, and the cross-section of the stack is divided into a number of equal areas. A traverse point is then located within each of these equal areas.

1.2 Applicability. This method is applicable to flowing gas streams in ducts, stacks, and flues. The method cannot be used when: (1) flow is cyclonic or swirling (see Section 2.4), (2) a stack is smaller than about 0.30 meter (12 in.) in diameter, or 0.071 m² (1.13 in.²) cross-sectional area, or (3) the measurement site is less than two stack or duct diameters downstream or less than a half diameter upstream from a flow disturbance.

The requirements of this method must be considered before construction of a new facility from which emissions will be measured; failure to do so may require subsequent alterations to the stack or deviation from the standard procedure. Cases involving variants are subject to approval by the Administrator, U.S. Environmental Protection Agency.

2. Procedure

2.1 Selection of Measurement Site. Sampling or velocity measurement is performed at a site located at least eight stack or duct diameters downstream and two diameters upstream from any flow disturbance such as a bend, expansion, or contraction in the stack, or from a visible flame. If necessary, an alternative location may be selected, at a position at least two stack or duct diameters downstream and a half diameter upstream from any flow disturbance. For a rectangular cross section, an equivalent diameter (D_e) shall be calculated from the following equation, to determine the upstream and downstream distances:

$$D_e = \frac{2LW}{L+W}$$

where L=length and W=width.

2.2 Determining the Number of Traverse Points

2.2.1 Particulate Traverses. When the eight- and two-diameter criterion can be met, the minimum number of traverse points shall be: (1) twelve, for circular or rectangular stacks with diameters (or equivalent diameters) greater than 0.61 meter (24 in.); (2) eight, for circular stacks with diameters between 0.30 and 0.61 meter (12-24 in.); (3) nine, for rectangular stacks with equivalent diameters between 0.30 and 0.61 meter (12-24 in.).

When the eight- and two-diameter criterion cannot be met, the minimum number of traverse points is determined from Figure 1-1. Before referring to the figure, however, determine the distances from the chosen

measurement site to the nearest upstream and downstream disturbances, and divide each distance by the stack diameter or equivalent diameter, to determine the distance in terms of the number of duct diameters. Then, determine from Figure 1-1 the minimum number of traverse points that corresponds: (1) to the number of duct diameters upstream; and (2) to the number of diameters downstream. Select the higher of the two minimum numbers of traverse points, or a greater value, so that for circular stacks the number is a multiple of 4, and for rectangular stacks, the number is one of those shown in Table 1-1.

TABLE 1-1. CROSS-SECTION LAYOUT FOR RECTANGULAR STACKS

Number of traverse points	Stack shape
9	2x3
12	4x3
18	4x4
20	5x4
25	5x5
30	6x5
36	6x6
42	7x6
48	7x7

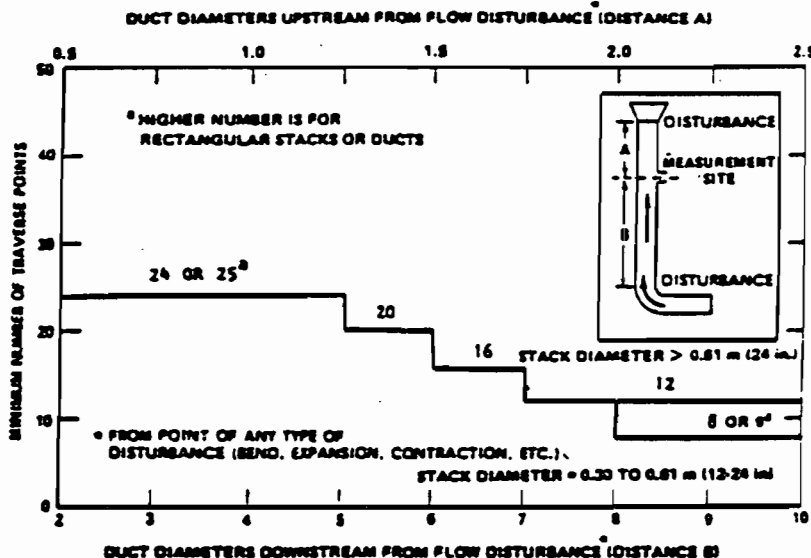


Figure 1-1. Minimum number of traverse points for particulate traverses.

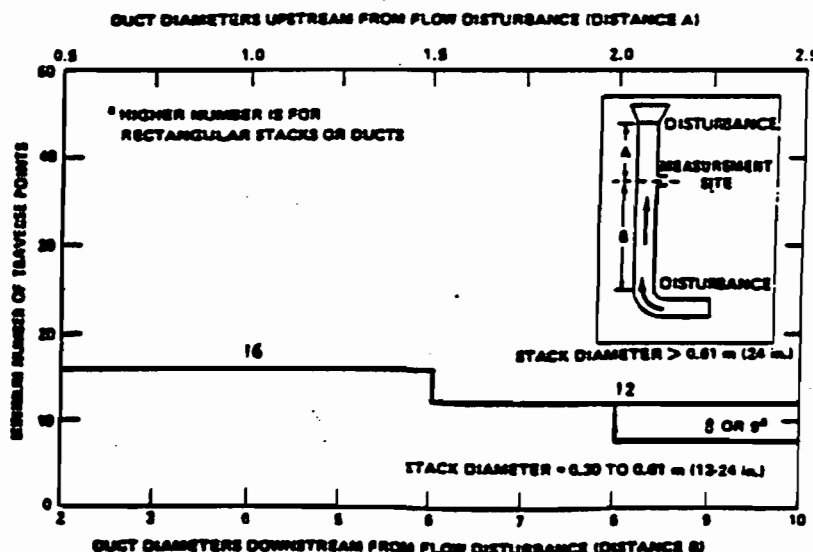


Figure 1-2. Minimum number of traverse points for velocity (nonparticulate) traverses.

2.2.2 Velocity (Non-Particulate) Traverses. When velocity or volumetric flow rate is to be determined (but not particulate matter), the same procedure as that for particulate traverses (Section 2.2.1) is followed, except that Figure 1-3 may be used instead of Figure 1-1.

2.3 Cross-sectional Layout and Location of Traverse Points.

2.3.1 Circular Stacks. Locate the traverse points on two perpendicular diameters according to Table 1-2 and the example shown in Figure 1-3. Any equation (for examples, see Citations 2 and 3 in the Bibliography) that gives the same values as those in Table 1-2 may be used in lieu of Table 1-2.

For particulate traverses, one of the diameters must be in a plane containing the greatest expected concentration variation, e.g., after bends, one diameter shall be in the plane of the bend. This requirement becomes less critical as the distance from the disturbance increases; therefore, other diameter locations may be used, subject to approval of the Administrator.

In addition for stacks having diameters greater than 0.61 m (24 in.) no traverse points shall be located within 2.5 centimeters (1.00 in.) of the stack walls; and for stack diameters equal to or less than 0.61 m (24 in.), no traverse points shall be located within 1.3 cm (0.50 in.) of the stack walls.

To meet these criteria, observe the procedures given below.

2.3.1.1 Stacks With Diameters Greater Than 0.61 m (24 in.). When any of the traverse points as located in Section 2.3.1 fall within 2.5 cm (1.00 in.) of the stack walls, relocate them away from the stack walls to:

(1) a distance of 2.5 cm (1.00 in.); or (2) a distance equal to the nozzle inside diameter, whichever is larger. These relocated traverse points (on each end of a diameter) shall be the "adjusted" traverse points.

Whenever two successive traverse points are combined to form a single adjusted traverse point, treat the adjusted point as two separate traverse points, both in the sampling (or velocity measurement) procedure, and in recording the data.

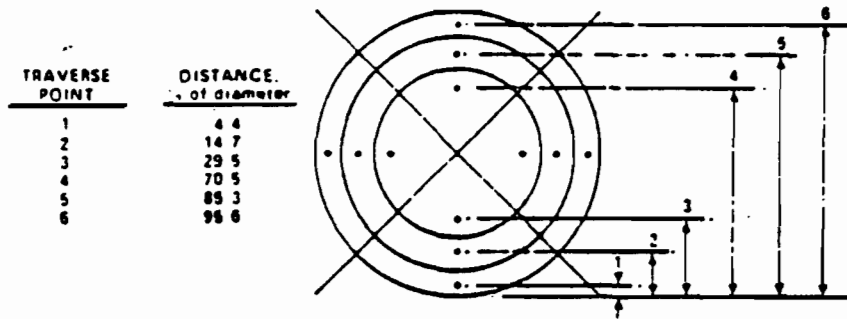


Figure 1-3. Example showing circular stack cross section divided into 12 equal areas, with location of traverse points indicated.

TABLE 1-2. LOCATION OF TRAVERSE POINTS IN CIRCULAR STACKS
(Percent of each diameter from stack wall to traverse point)

Traverse point number on a diameter	Number of traverse points on a diameter—											
	2	4	6	8	10	12	14	16	18	20	22	24
1	14.6	6.7	4.4	3.2	2.6	2.1	1.8	1.6	1.4	1.3	1.1	1.1
2	85.4	25.0	14.6	10.5	8.2	6.7	5.7	4.8	4.4	3.9	3.5	3.2
3		75.0	29.8	18.4	14.6	11.9	9.8	8.5	7.5	6.7	6.0	5.6
4		83.3	70.4	32.3	22.6	17.7	14.6	12.5	10.9	9.7	8.7	7.9
5			85.4	67.7	34.2	25.0	21.1	18.9	14.9	12.9	11.9	10.9
6			85.6	80.6	65.9	39.8	28.9	22.0	18.9	16.5	14.9	13.2
7				88.8	77.4	64.4	38.8	28.3	23.6	20.4	18.0	16.1
8				88.8	88.4	75.0	63.4	37.5	28.8	25.0	21.9	19.4
9					91.8	82.3	71.1	62.5	38.2	30.8	26.2	23.0
10					97.4	88.2	78.9	71.7	61.8	38.8	31.5	27.2
11						92.3	85.4	78.0	70.4	61.2	38.3	32.3
12						97.9	88.1	81.1	78.4	68.4	40.7	36.8
13							94.3	87.5	81.2	75.0	66.5	62.2
14							98.2	91.8	85.4	78.8	73.8	67.7
15								95.1	89.1	83.5	78.2	72.9
16								98.4	92.5	87.1	82.0	77.0
17									95.8	90.3	85.4	80.6
18									95.8	93.3	88.4	83.9
19										96.1	91.3	86.9
20										98.7	94.0	89.5
21											98.5	92.1
22											98.9	94.6
23												98.8
24												98.9

2.3.1.2 Stacks With Diameters Equal to or Less Than 0.61 m (24 in.). Follow the procedure in Section 2.3.1.1, noting only that any "adjusted" points should be relocated away from the stack walls to: (1) a distance of 1.3 cm (0.50 in.); or (2) a distance equal to the nozzle inside diameter, whichever is larger.

2.3.2 Rectangular Stacks. Determine the number of traverse points as explained in Sections 2.1 and 2.2 of this method. From Table 1-1, determine the grid configuration. Divide the stack cross-section into as many equal rectangular elemental areas as traverse points, and then locate a traverse point at the centroid of each equal area according to the example in Figure 1-4.

If the tester desires to use more than the minimum number of traverse points, expand the "minimum number of traverse points" matrix (see Table 1-1) by adding the extra traverse points along one or the other or both legs of the matrix; the final matrix need not be balanced. For example, if a 4x3 "minimum number of points" matrix were expanded to 36 points, the final matrix could be 9x4 or 12x3, and would not necessarily have to be 6x6. After constructing the final matrix, divide the stack cross-section into as many equal rectangular, elemental areas as traverse points, and locate a traverse point at the centroid of each equal area.

The situation of traverse points being too close to the stack walls is not expected to arise with rectangular stacks. If this problem should ever arise, the Administrator must be contacted for resolution of the matter.

2.4 Verification of Absence of Cyclonic Flow. In most stationary sources, the direction of stack gas flow is essentially parallel to the stack walls. However, cyclonic flow may exist (1) after such devices as cyclones and inertial demisters following venturi scrubbers, or (2) in stacks having tangential inlets or other duct configurations which tend to induce swirling; in these instances, the presence or absence of cyclonic flow at the sampling location must be determined. The following techniques are acceptable for this determination.

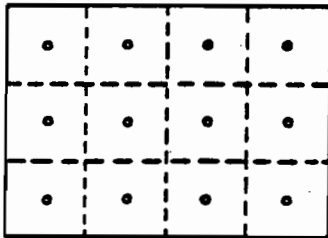


Figure 1-4. Example showing rectangular stack cross section divided into 12 equal areas, with a traverse point at centroid of each area.

Level and zero the manometer. Connect a Type S pitot tube to the manometer. Position the Type S pitot tube at each traverse point, in succession, so that the planes of the face openings of the pitot tube are perpendicular to the stack cross-sectional plane; when the Type S pitot tube is in this position, it is at "0° reference." Note the differential pressure (Δp) reading at each traverse point. If a null (zero) pitot reading is obtained at 0° reference at a given traverse point, an acceptable flow condition exists at that point. If the pitot reading is not zero at 0° reference, rotate the pitot tube (up to $\pm 90^\circ$ yaw angle), until a null reading is obtained. Carefully determine and record the value of the rotation angle (α) to the nearest degree. After the null technique has been applied at each traverse point, calculate the average of the absolute values of α ; assign a value of 0° to those points for which no rotation was required, and include these in the overall average. If the average value of α is greater than 10°, the overall flow condition in the stack is unacceptable and alternative methodology, subject to the approval of the Administrator, must be used to perform accurate sample and velocity traverses.

3. Bibliography

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2. Devorkin, Howard, et al. Air Pollution Source Testing Manual. Air Pollution Control District, Los Angeles, CA, November 1963.
3. Methods for Determination of Velocity, Volume, Dust and Mist Content of Gases. Western Precipitation Division of Joy Manufacturing Co. Los Angeles, CA. Bulletin WP-50, 1968.
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5. Hanson, H. A., et al. Particulate Sampling Strategies for Large Power Plants Including Nonuniform Flow. USEPA, ORD, ESRL, Research Triangle Park, N.C. EPA-600/2-76-170, June 1976.
6. Entropy Environmentalists, Inc. Determination of the Optimum Number of Sampling Points: An Analysis of Method 1 Criteria. Environmental Protection Agency, Research Triangle Park, N.C. EPA Contract No. 68-01-3172, Task 7.
7. Hanson, H.A., E.J. Davini, J.K. Morgan, and A.A. Iversen. Particulate Sampling Strategies for Large Power Plants Including Nonuniform Flow. U.S. Environmental Protection Agency, Research Triangle Park, N.C. Publication No. EPA-600/2-76-170, June 1976, 350 p.
8. Brooks, E.F., and R.L. Williams. Flow and Gas Sampling Manual. U.S. Environmental Protection Agency, Research Triangle Park, N.C. Publication No. EPA-600/2-76-203, July 1976, 83 p.
9. Entropy Environmentalists, Inc. Traverse Point Study. EPA Contract No. 68-02-3172, June 1977, 19 p.
10. Brown, J. and K. Yu. Test Report: Particulate Sampling Strategy in Circular Ducts. Emission Measurement Branch, Emission Standards and Engineering Division. U.S. Environmental Protection Agency, Research Triangle Park, N.C. 27711, July 31, 1980, 12 p.
11. Hawkaley, P.G.W., S. Badnoch, and J.H. Blackett. Measurement of Solids in Flue Gases. Leatherhead, England. The British Coal Utilisation Research Association, 1961, p. 129-133.
12. Knapp, K.T. The Number of Sampling Points Needed for Representative Source Sampling. In: Proceedings of the Fourth National Conference on Energy and the Environment, Theodore, L. et al. (ed.), Dayton.

SEE PROPOSED AMENDMENT NEXT PAGE

METHOD 2—DETERMINATION OF STACK GAS VELOCITY AND VOLUMETRIC FLOW RATE (TYPE S PITOT TUBE)

1. Principle and Applicability

1.1 Principle. The average gas velocity in a stack is determined from the gas density and from measurement of the average velocity head with a Type S (Stausscheibe or reverse type) pitot tube.

1.2 Applicability. This method is applicable for measurement of the average velocity of a gas stream and for quantifying gas flow.

This procedure is not applicable at measurement sites which fail to meet the criteria of Method 1, Section 2.1. Also, the method cannot be used for direct measurement in cyclonic or swirling gas streams. Section 2.4 of Method 1 shows how to determine cyclonic or swirling flow conditions. When unacceptable conditions exist, alternative procedures, subject to the approval of the Administrator, U.S. Environmental Protection Agency, must be employed to make accurate flow rate determinations; examples of such alternative procedures are: (1) to install straightening vanes; (2) to calculate the total volumetric flow rate stoichiometrically, or (3) to move to another measurement site at which the flow is acceptable.

2. Apparatus

Specifications for the apparatus are given below. Any other apparatus that has been demonstrated (subject to approval of the Administrator) to be capable of meeting the specifications will be considered acceptable.

2.1 Type S Pitot Tube. The Type S pitot tube (Figure 2-1) shall be made of metal tubing (e.g. stainless steel). It is recommended that the external tubing diameter (dimension D , Figure 2-2b) be between 0.48 and 0.95 centimeters ($\frac{1}{8}$ and $\frac{1}{4}$ inch). There shall be an equal distance from the base of each leg of the pitot tube to its face-opening plane (dimensions P_1 and P_2 , Figure 2-2b); it is recommended that this distance be between 1.05 and 1.50 times the external tubing diameter. The face openings of the pitot tube shall preferably, be aligned as shown in Figure 2-2; however, slight misalignments of the openings are permissible (see Figure 2-3).

The Type S pitot tube shall have a known coefficient, determined as outlined in Section 4, an identification number shall be assigned to the pitot tube; this number shall be permanently marked or engraved on the body of the tube.

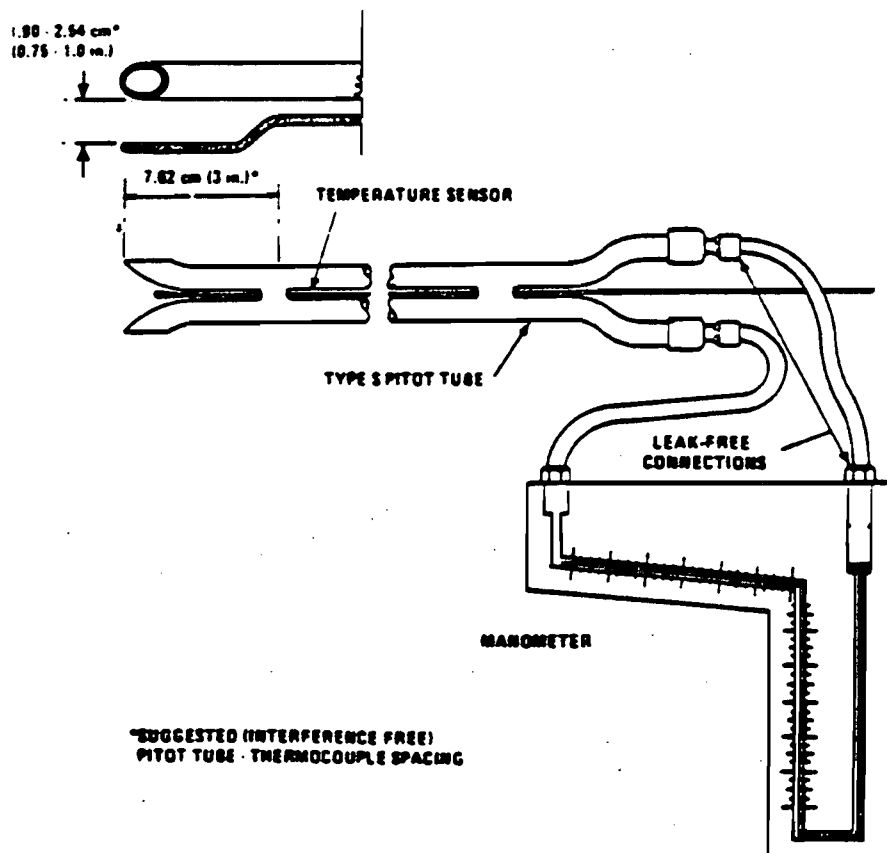


Figure 2-1. Type S pitot tube manometer assembly.

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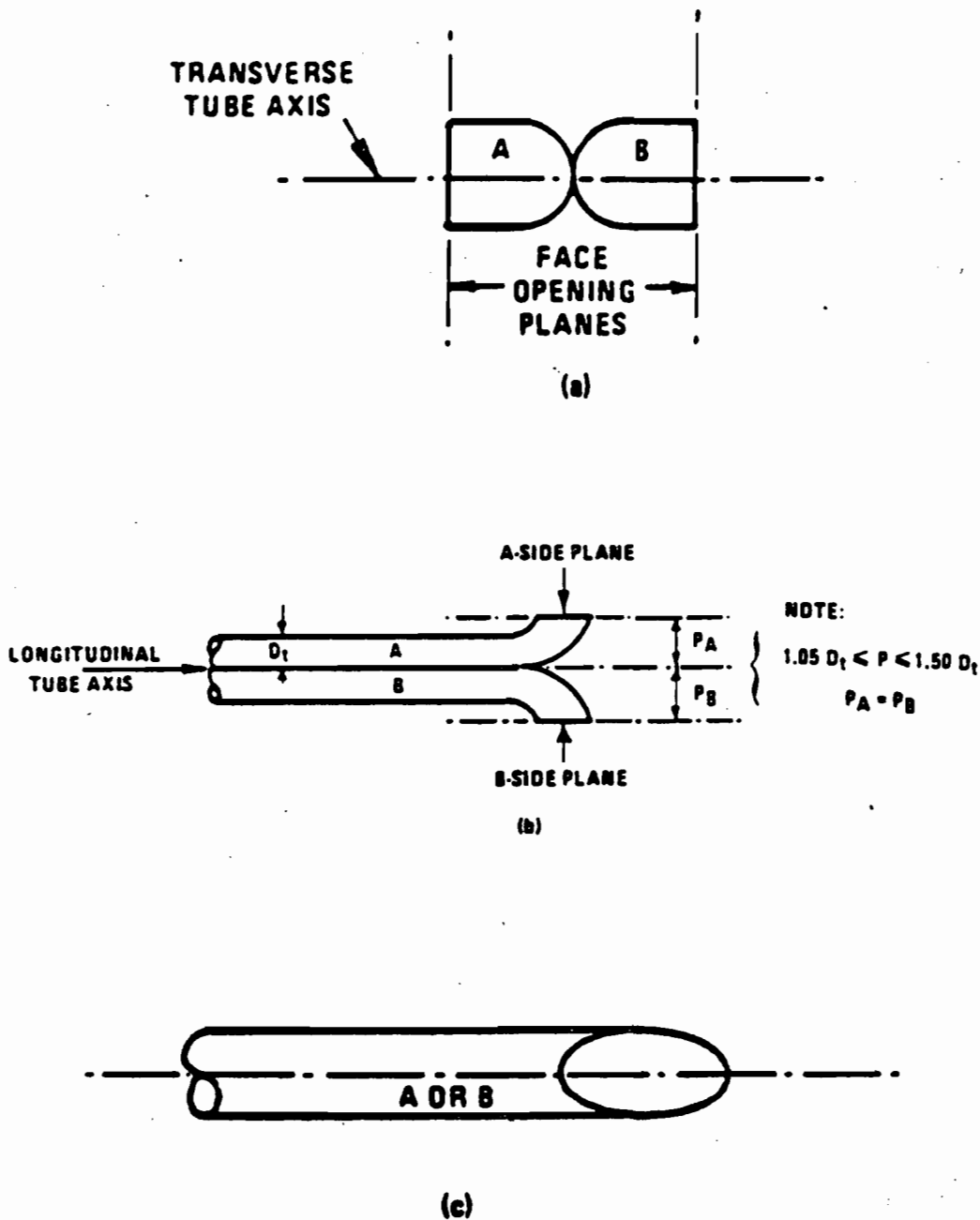


Figure 2-2. Properly constructed Type B pitot tube, shown in: (a) end view, face opening planes perpendicular to transverse axis; (b) top view, face opening planes parallel to longitudinal axis; (c) side view, both legs of equal length and centerlines coincident, when viewed from both sides. Baseline coefficient values of 0.84 may be assigned to pitot tubes constructed this way.

taken, as above, for the last two back purges at which suitably high Δp readings are observed.

2.2 Differential Pressure Gauge. An inclined manometer or equivalent device is used. Most sampling trains are equipped with a 10-in. (water column) inclined-vertical manometer, having 0.01-in. H₂O divisions on the 0-to 1-in. inclined scale, and 0.1-in. H₂O divisions on the 1- to 10-in. vertical scale. This type of manometer (or other gauge of equivalent sensitivity) is satisfactory for the measurement of Δp values as low as 1.3 mm (0.05 in.) H₂O. However, a differential pressure gauge of greater sensitivity shall be used (subject to the approval of the Administrator), if any of the following is found to be true: (1) the arithmetic average of all Δp readings at the traverse points in the stack is less than 1.3 mm (0.05 in.) H₂O; (2) for traverses of 12 or more points, more than 10 percent of the individual Δp readings are below 1.3 mm (0.05 in.) H₂O; (3) for traverses of fewer than 12 points, more than one Δp reading is below 1.3 mm (0.05 in.) H₂O. Citation 18 in Section 6 describes commercially available instrumentation for the measurement of low-range gas velocities.

As an alternative to criteria (1) through (3) above, the following calculation may be performed to determine the necessity of using a more sensitive differential pressure gauge:

$$T = \frac{\sum_{i=1}^n \sqrt{\Delta p_i + K}}{\sum_{i=1}^n \sqrt{\Delta p_i}}$$

where:

Δp_i = Individual velocity head reading at a traverse point, mm H₂O (in. H₂O).

n = Total number of traverse points.

K = 0.13 mm H₂O when metric units are used and 0.005 in. H₂O when English units are used.

If T is greater than 1.05, the velocity head data are unacceptable and a more sensitive differential pressure gauge must be used.

NOTE: If differential pressure gauges other than inclined manometers are used (e.g., magnetic gauges), their calibration must be checked after each test series. To check

the calibration of a differential pressure gauge, compare Δp readings of the gauge with those of a gauge-oil manometer at a minimum of three points, approximately representing the range of Δp values in the stack. If, at each point, the values of Δp read by the differential pressure gauge and gauge-oil manometer agree to within 5 percent, the differential pressure gauge shall be considered to be in proper calibration. Otherwise, the test series shall either be voided, or procedures to adjust the measured Δp values and final results shall be used subject to the approval of the Administrator.

2.3 Temperature Gauge. A thermocouple, liquid-filled bulb thermometer, bimetallic thermometer, mercury-in-glass thermometer, or other gauge, capable of measuring temperature to within 1.5 percent of the minimum absolute stack temperature shall be used. The temperature gauge shall be attached to the pitot tube such that the sensor tip does not touch any metal; the gauge shall be in an interference-free arrangement with respect to the pitot tube face openings (see Figure 2-1 and also Figure 2-7 in Section 4). Alternate positions may be used if the pitot tube-temperature gauge system is calibrated according to the procedure of Section 4. Provided that a difference of not more than 1 percent in the average velocity measurement is introduced, the temperature gauge need not be attached to the pitot tube; this alternative is subject to the approval of the Administrator.

2.4 Pressure Probe and Gauge. A piezometer tube and mercury- or water-filled U-tube manometer capable of measuring stack pressure to within 2.5 mm (0.1 in.) Hg is used. The static tap of a standard type pitot tube or one leg of a Type S pitot tube with the face opening planes positioned parallel to the gas flow may also be used as the pressure probe.

2.5 Barometer. A mercury, aneroid, or other barometer capable of measuring atmospheric pressure to within 2.5 mm Hg (0.1 in. Hg) may be used. In many cases, the barometric reading may be obtained from a nearby national weather service station, in which case the station value (which is the absolute barometric pressure) shall be requested and an adjustment (or elevation differences between the weather station and the sampling point) shall be applied at a rate of minus 2.5 mm (0.1 in.) Hg per 30-meter (100 feet) elevation increase or vice-versa for elevation decrease.

2.6 Gas Density Determination Equipment. Method 3 equipment, if needed (see Section 3.6), to determine the stack gas dry molecular weight, and Reference Method 4 or Method 5 equipment for moisture content determination; other methods may be used subject to approval of the Administrator.

2.7 Calibration Pitot Tube. When calibration of the Type S pitot tube is necessary (see Section 4), a standard pitot tube is used as a reference. The standard pitot tube shall, preferably, have a known coefficient, obtained either (1) directly from the National Bureau of Standards, Route 270, Quince Orchard Road, Gaithersburg, Maryland, or (2) by calibration against another standard pitot tube with an NBS-traceable coefficient. Alternatively, a standard pitot tube designed according to the criteria given in 2.7.1 through 2.7.5 below and illustrated in Figure 2-4 (see also Citations 7, 8, and 17 in Section 6) may be used. Pitot tubes designed according to these specifications will have baseline coefficients of about 0.99 ± 0.01 .

2.7.1 Hemispherical (shown in Figure 2-4), ellipsoidal, or conical tip.

2.7.2 A minimum of six diameters straight run (based upon D , the external diameter of the tube) between the tip and the static pressure holes.

2.7.3 A minimum of eight diameters straight run between the static pressure holes and the centerline of the external tube, following the 90 degree bend.

2.7.4 Static pressure holes of equal size (approximately 0.1 D), equally spaced in a piezometer ring configuration.

2.7.5 Ninety degree bend, with curved or mitered junction.

2.8 Differential Pressure Gauge for Type S Pitot Tube Calibration. An inclined manometer or equivalent is used. If the single-velocity calibration technique is employed (see Section 4.1.2.3), the calibration differential pressure gauge shall be readable to the nearest 0.13 mm H₂O (0.005 in. H₂O). For multiveLOCITY calibrations, the gauge shall be readable to the nearest 0.13 mm H₂O (0.005 in. H₂O) for Δp values between 1.3 and 25 mm H₂O (0.05 and 1.0 in. H₂O), and to the nearest 1.3 mm H₂O (0.05 in. H₂O) for Δp values above 25 mm H₂O (1.0 in. H₂O). A special, more sensitive gauge will be required to read Δp values below 1.3 mm H₂O (0.05 in. H₂O) (see Citation 18 in Section 6).

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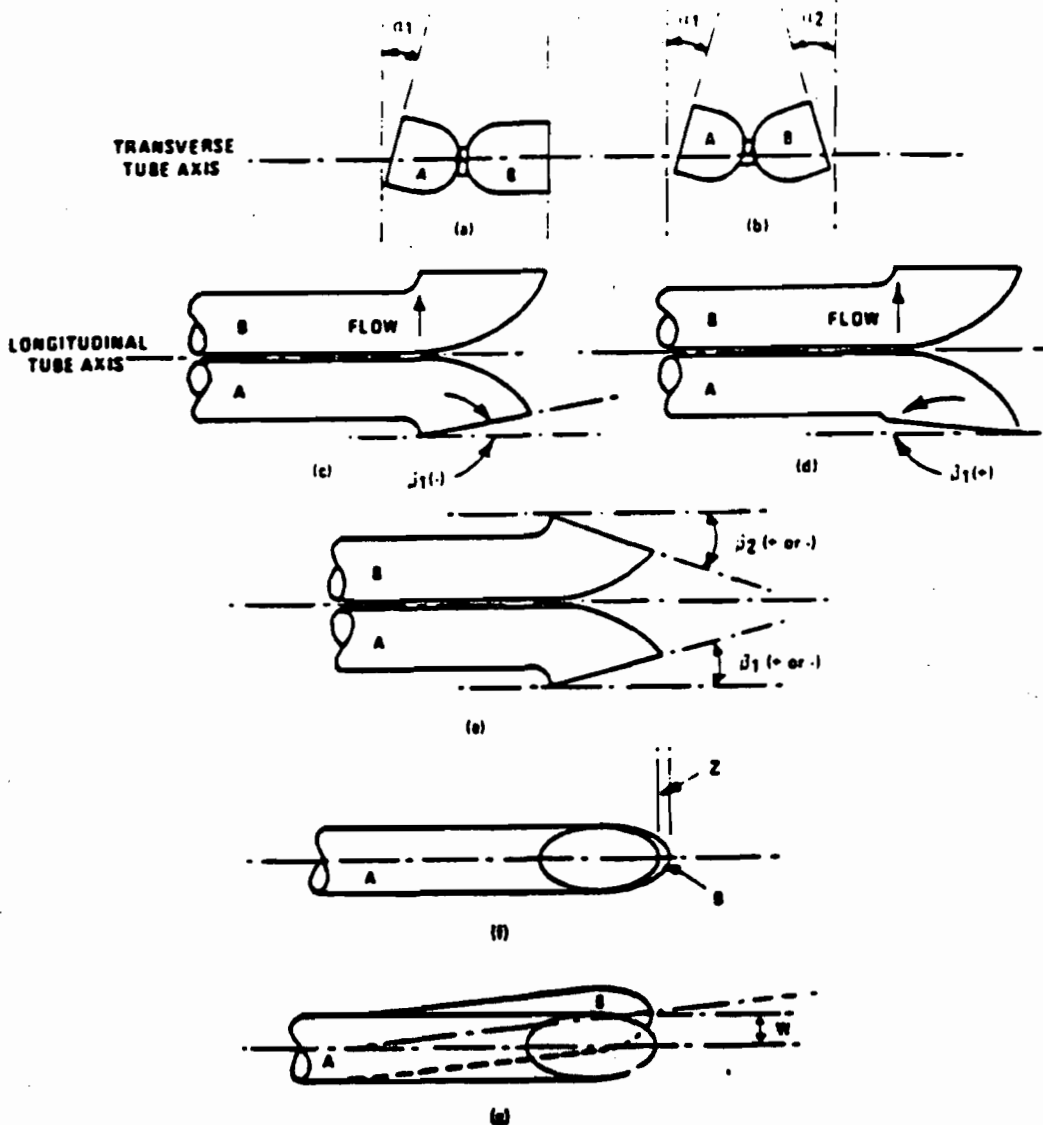


Figure 2-3. Types of face-opening misalignment that can result from field use or improper construction of Type S pitot tubes. These will not affect the baseline value of $Cp(s)$ so long as α_1 and $\alpha_2 \leq 10^\circ$, β_1 and $\beta_2 \leq 5^\circ$, $z \leq 0.32 \text{ cm}$ (1/8 in.) and $w \leq 0.08 \text{ cm}$ (1/32 in.) (caption 11 in Section 6).

A standard pitot tube may be used instead of a Type S, provided that it meets the specifications of Sections 2.7 and 4.2; note, however, that the static and impact pressure holes of standard pitot tubes are susceptible to plugging in particulate-laden gas streams. Therefore, whenever a standard pitot tube is used to perform a traverse, adequate proof must be furnished that the openings of the pitot tube have not plugged up during the traverse period; this can be done by taking a velocity head (Δp) reading at

the final traverse point, cleaning out the impact and static holes of the standard pitot tube by "back-purging" with pressurized air, and then taking another Δp reading. If the Δp readings made before and after the air purge are the same (± 5 percent), the traverse is acceptable. Otherwise, reject the run. Note that if Δp at the final traverse point is unsuitably low, another point may be selected. If "back-purging" at regular intervals is part of the procedure, then comparative Δp readings shall be

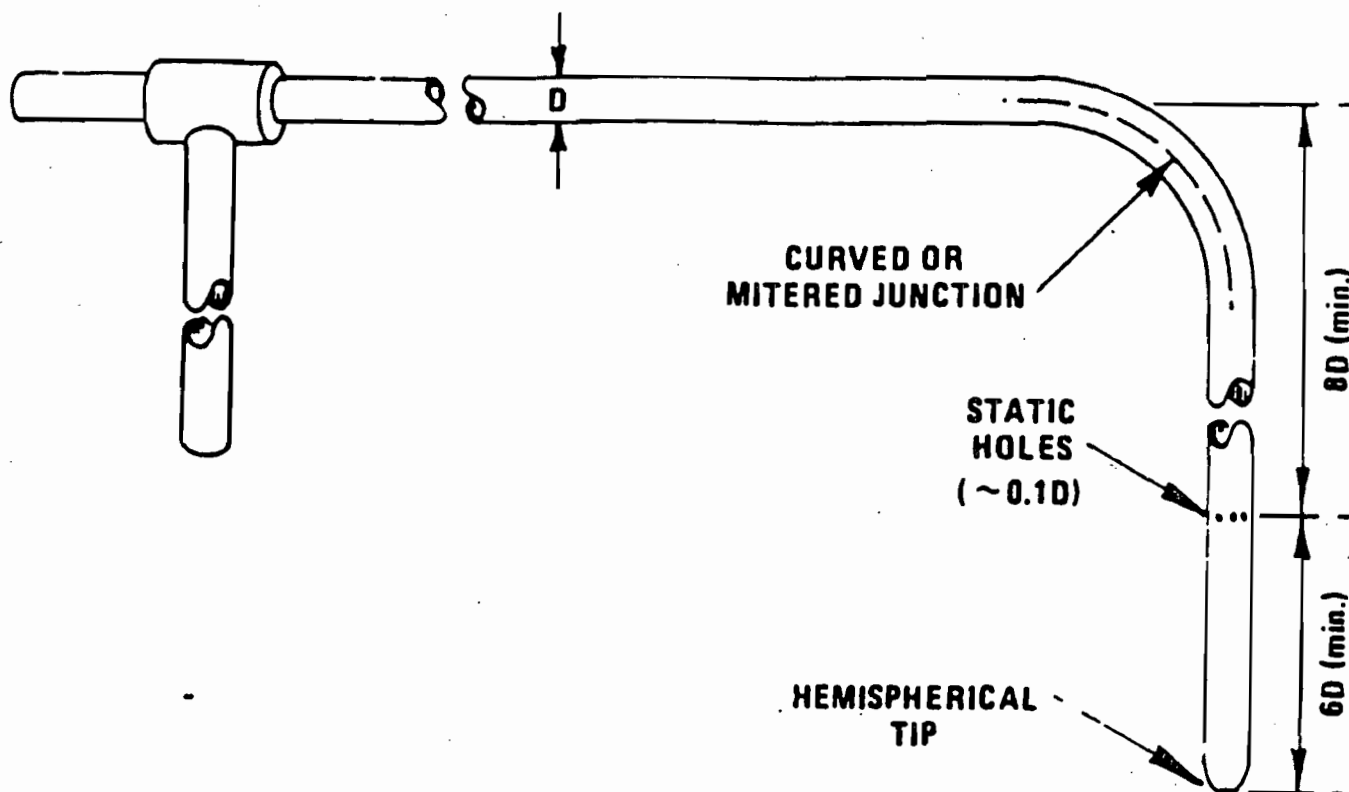


Figure 2-4. Standard pitot tube design specifications.

3. Procedure

3.1 Set up the apparatus as shown in Figure 2-1. Capillary tubing or surge tanks installed between the manometer and pitot tube may be used to dampen Δp fluctuations. It is recommended, but not required, that a pretest leak-check be conducted, as follows: (1) blow through the pitot impact opening until at least 7.6 cm (3 in.) H₂O velocity pressure registers on the manometer; then, close off the impact opening. The pressure shall remain stable for at least 15 seconds; (2) do the same for the static pressure side, except using suction to obtain the minimum of 7.6 cm (3 in.) H₂O. Other leak-check procedures, subject to the approval of the Administrator, may be used.

3.2 Level and zero the manometer. Because the manometer level and zero may

drift due to vibrations and temperature changes, make periodic checks during the traverse. Record all necessary data as shown in the example data sheet (Figure 2-5).

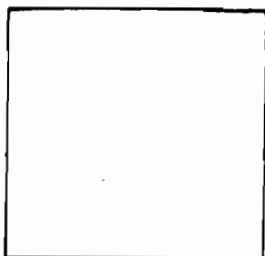
3.3 Measure the velocity head and temperature at the traverse points specified by Method 1. Ensure that the proper differential pressure gauge is being used for the range of Δp values encountered (see Section 2.2). If it is necessary to change to a more sensitive gauge, do so, and remeasure the Δp and temperature readings at each traverse point. Conduct a post-test leak-check (mandatory), as described in Section 3.1 above, to validate the traverse run.

3.4 Measure the static pressure in the stack. One reading is usually adequate.

3.5 Determine the atmospheric pressure.

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PLANT _____
 DATE _____ RUN NO. _____
 STACK DIAMETER OR DIMENSIONS, m(in.) _____
 BAROMETRIC PRESSURE, mm Hg (in. Hg) _____
 CROSS SECTIONAL AREA, m²(ft²) _____
 OPERATORS _____
 PITOT TUBE I.D. NO. _____
 AVG. COEFFICIENT, C_p = _____
 LAST DATE CALIBRATED _____



SCHEMATIC OF STACK
CROSS SECTION

Traverse Pt. No.	Vel. Hd., Δs mm (in.) H ₂ O	Stack Temperature		P _g mm Hg (in. Hg)	√Δs
		T _g , °C (°F)	T _g , °K (°R)		
Average					

Figure 2-5. Velocity traverse data.

3.6 Determine the stack gas dry molecular weight. For combustion processes or processes that emit essentially CO, O₂, CO, and N₂, use Method J. For processes emitting essentially air, an analysis need not be conducted; use a dry molecular weight of 29.0. For other processes, other methods, subject to the approval of the Administrator, must be used.

3.7 Obtain the moisture content from Reference Method 4 (or equivalent) or from Method 5.

3.8 Determine the cross-sectional area of the stack or duct at the sampling location. Whenever possible, physically measure the stack dimensions rather than using blueprints.

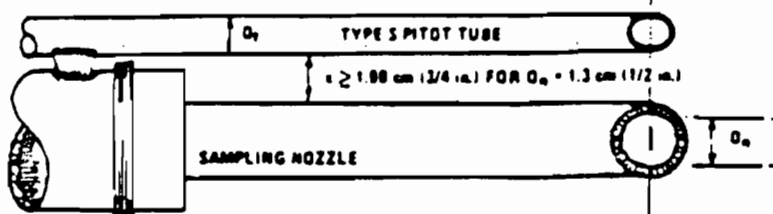
4. Calibration

4.1 Type S Pitot Tube. Before its initial use, carefully examine the Type S pitot tube in top, side, and end views to verify that the face openings of the tube are aligned within the specifications illustrated in Figure 2-3 or 2-3. The pitot tube shall not be used if it fails to meet these alignment specifications.

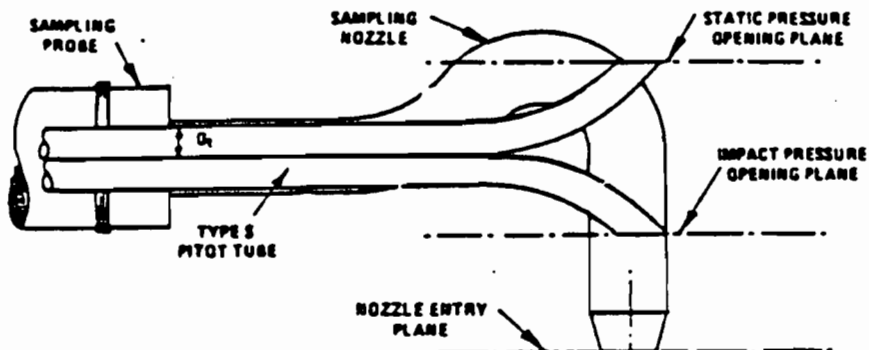
After verifying the face opening alignment, measure and record the following dimensions of the pitot tube: (a) the external tubing diameter (dimension D, Figure 2-2b); and (b) the base-to-opening plane distances (dimensions P₁ and P₂, Figure 2-2b). If D is between 0.48 and 0.88 cm (1/4 and 3/8 in.) and

if P₁ and P₂ are equal and between 1.03 and 1.50 D, there are two possible options: (1) the pitot tube may be calibrated according to the procedure outlined in Sections 4.1.2 through 4.1.5 below, or (2) a baseline (isolated tube) coefficient value of 0.84 may be assigned to the pitot tube. Note, however, that if the pitot tube is part of an assembly, calibration may still be required, despite knowledge of the baseline coefficient value (see Section 4.1.1).

If D, P₁, and P₂ are outside the specified limits, the pitot tube must be calibrated outlined in 4.1.2 through 4.1.5 below.



A. BOTTOM VIEW: SHOWING MINIMUM PITOT-NOZZLE SEPARATION.



B. SIDE VIEW: TO PREVENT PITOT TUBE FROM INTERFERING WITH GAS FLOW STREAMLINES APPROACHING THE NOZZLE, THE IMPACT PRESSURE OPENING PLANE OF THE PITOT TUBE SHALL BE EVEN WITH OR ABOVE THE NOZZLE ENTRY PLANE.

4.1.1 Type S Pitot Tube Assemblies. During sample and velocity traverses, the isolated Type S pitot tube is not always used; in many instances, the pitot tube is used in combination with other source-sampling components (thermocouple, sampling probe, nozzle) as part of an "assembly." The presence of other sampling components can sometimes affect the baseline value of the Type S pitot tube coefficient (Citation 9 in Section 6); therefore an assigned (or otherwise known) baseline coefficient value may or may not be valid for a given assembly. The baseline and assembly coefficient values will be identical only when the relative placement of the components in the assembly is such that aerodynamic interference effects are eliminated. Figures 2-6 through 2-8 illustrate interference-free component arrangements for Type S pitot tubes having external tubing diameters between 0.48 and 0.95 cm (3/16 and 3/8 in.). Type S pitot tube assemblies that fail to meet any or all of the specifications of Figures 2-6 through 2-8 shall be calibrated according to the procedure outlined in Sections 4.1.2 through 4.1.5 below, and prior to calibration, the values of the intercomponent spacings (pitot-nozzle, pitot-thermocouple, pitot-probe sheath) shall be measured and recorded.

NOTE: Do not use any Type S pitot tube assembly which is constructed such that the impact pressure opening plane of the pitot tube is below the entry plane of the nozzle (see Figure 2-6b).

4.1.2 Calibration Setup. If the Type S pitot tube is to be calibrated, one leg of the tube shall be permanently marked A, and the other, B. Calibration shall be done in a flow system having the following essential design features:

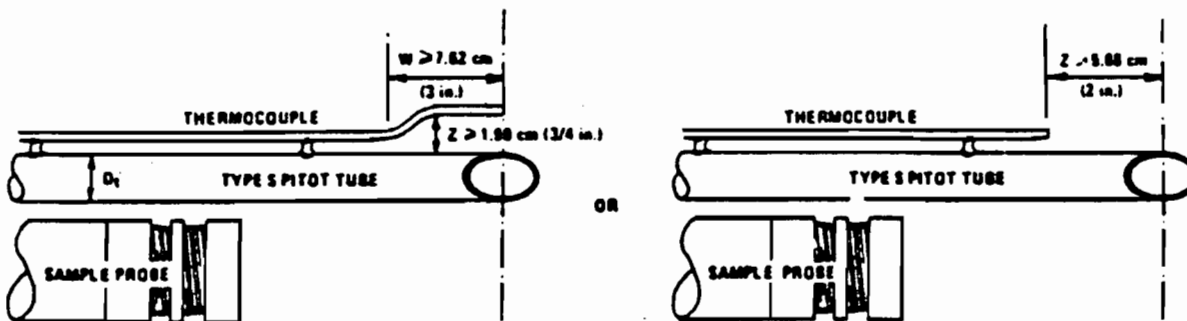


Figure 2-7. Proper thermocouple placement to prevent interference; D_t between 0.48 and 0.95 cm (3/16 and 3/8 in.).

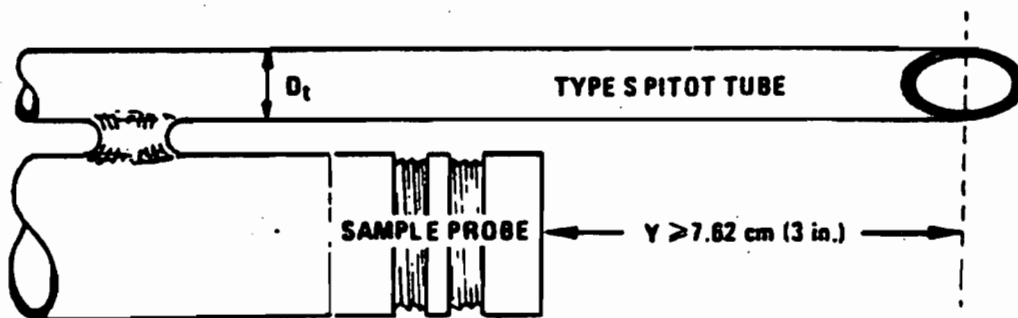


Figure 2-8. Minimum pitot-sample probe separation needed to prevent interference; D_t between 0.48 and 0.95 cm (3/16 and 3/8 in.).

4.1.2.1 The flowing gas stream must be confined to a duct of definite cross-sectional area, either circular or rectangular. For circular cross-sections, the minimum duct diameter shall be 30.5 cm (12 in.); for rectangular cross-sections, the width (shorter side) shall be at least 25.4 cm (10 in.).

4.1.2.2 The cross-sectional area of the calibration duct must be constant over a distance of 10 or more duct diameters. For a rectangular cross-section, use an equivalent diameter, calculated from the following equation, to determine the number of duct diameters:

$$D = \frac{2LW}{(L+W)}$$

Equation 2-1

where:
 D = Equivalent diameter
 L = Length
 W = Width

To ensure the presence of stable, fully developed flow patterns at the calibration site, or "test section," the site must be located at least eight diameters downstream and two diameters upstream from the nearest disturbances.

NOTE: The eight- and two-diameter criteria are not absolute; other test section locations may be used (subject to approval of the Administrator), provided that the flow at the test site is stable and demonstrably parallel to the duct axis.

4.1.2.3 The flow system shall have the capacity to generate a test-section velocity around 915 m/min (3,000 ft/min). This velocity must be constant with time to guarantee steady flow during calibration. Note that Type S pitot tube coefficients obtained by single-velocity calibration at 915 m/min (3,000 ft/min) will generally be valid to within ± 3 percent for the measurement of velocities above 305 m/min (1,000 ft/min) and to within ± 5 to 6 percent for the measurement of velocities between 180 and 305 m/min (600 and 1,000 ft/min). If a more precise correlation between C_p and velocity is desired, the flow system shall have the capacity to generate at least four distinct, time-invariant test-section velocities covering the velocity range from 180 to 1,525 m/min (600 to 5,000 ft/min), and calibration data shall be taken at regular velocity intervals over this range (see Citations 9 and 14 in Section 6 for details).

4.1.2.4 Two entry ports, one each for the standard and Type S pitot tubes, shall be cut in the test section; the standard pitot entry port shall be located slightly downstream of the Type S port, so that the standard and Type S impact openings will lie in the same cross-sectional plane during calibration. To facilitate alignment of the pitot tubes during calibration, it is advisable that the test section be constructed of plexiglas or some other transparent material.

4.1.3 Calibration Procedure. Note that this procedure is a general one and must not be used without first referring to the special considerations presented in Section 4.1.5. Note also that this procedure applies only to single-velocity calibration. To obtain calibration data for the A and B sides of the Type S pitot tube, proceed as follows:

4.1.3.1 Make sure that the manometer is properly filled and that the oil is free from contamination and is of the proper density. Inspect and leak-check all pitot lines; repair or replace if necessary.

PITOT TUBE IDENTIFICATION NUMBER: _____ DATE: _____
 CALIBRATED BY: _____

"A" SIDE CALIBRATION				
RUN NO.	ΔP_{std} cm H ₂ O (in. H ₂ O)	$\Delta P(s)$ cm H ₂ O (in. H ₂ O)	$C_p(s)$	DEVIATION $C_p(s) - \bar{C}_p(A)$
1				
2				
3				
			\bar{C}_p (SIDE A)	

"B" SIDE CALIBRATION				
RUN NO.	ΔP_{std} cm H ₂ O (in. H ₂ O)	$\Delta P(s)$ cm H ₂ O (in. H ₂ O)	$C_p(s)$	DEVIATION $C_p(s) - \bar{C}_p(B)$
1				
2				
3				
			\bar{C}_p (SIDE B)	

$$\text{AVERAGE DEVIATION} = \frac{1}{3} \sum |C_p(s) - \bar{C}_p(A \text{ OR } B)| \quad \leftarrow \text{MUST BE } < 0.01$$

$$|\bar{C}_p(\text{SIDE A}) - \bar{C}_p(\text{SIDE B})| \quad \leftarrow \text{MUST BE } < 0.01$$

Figure 2-9. Pitot tube calibration data.

4.1.3.2 Level and zero the manometer. Turn on the fan and allow the flow to stabilize. Seal the Type S entry port.

4.1.3.3 Ensure that the manometer is level and zeroed. Position the standard pitot tube at the calibration point (determined as outlined in Section 4.1.3.1), and align the tube so that its tip is pointed directly into the flow. Particular care should be taken in aligning the tube to avoid yaw and pitch angles. Make sure that the entry port surrounding the tube is properly sealed.

4.1.3.4 Read ΔP_{std} and record its value in a data table similar to the one shown in Figure 2-9. Remove the standard pitot tube from the duct and disconnect it from the manometer. Seal the standard entry port.

4.1.3.5 Connect the Type S pitot tube to the manometer. Open the Type S entry port. Check the manometer level and zero. Insert and align the Type S pitot tube so that its A side impact opening is at the same

point as was the standard pitot tube and is pointed directly into the flow. Make sure that the entry port surrounding the tube is properly sealed.

4.1.3.6 Read Δp , and enter its value in the data table. Remove the Type S pitot tube from the duct and disconnect it from the manometer.

4.1.3.7 Repeat steps 4.1.3.3 through 4.1.3.6 above until three pairs of Δp readings have been obtained.

4.1.3.8 Repeat steps 4.1.3.3 through 4.1.3.7 above for the B side of the Type S pitot tube.

4.1.3.9 Perform calculations, as described in Section 4.1.4 below.

4.1.4 Calculations.

4.1.4.1 For each of the six pairs of Δp readings (i.e., three from side A and three from side B) obtained in Section 4.1.3 above calculate the value of the Type S pitot tube coefficient as follows:

$$C_{p_{std}} = C_{p_{meas}} \sqrt{\frac{\Delta p_{std}}{\Delta p_{meas}}}$$

Equation 2-2

where:

$C_{p_{std}}$ = Type S pitot tube coefficient

$C_{p_{meas}}$ = Standard pitot tube coefficient; use 0.99 if the coefficient is unknown and the tube is designed according to the criteria of Sections 2.7.1 to 2.7.5 of this method.

Δp_{std} = Velocity head measured by the standard pitot tube, cm H₂O (in. H₂O)

Δp_{meas} = Velocity head measured by the Type S pitot tube, cm H₂O (in. H₂O)

4.1.4.2 Calculate C_p (side A), the mean A-side coefficient, and C_p (side B), the mean B-side coefficient; calculate the difference between these two average values.

4.1.4.3 Calculate the deviation of each of the three A-side values of $C_{p_{meas}}$ from C_p (side A), and the deviation of each B-side value of $C_{p_{meas}}$ from C_p (side B). Use the following equation:

$$\text{Deviation} = C_{p_{meas}} - \bar{C}_p(A \text{ or } B)$$

Equation 2-3

4.1.4.4 Calculate δ , the average deviation from the mean, for both the A and B sides of the pitot tube. Use the following equation:

$$\delta \text{ (side A or B)} = \frac{\sum_{j=1}^3 |C_{p_{meas}} - \bar{C}_p(A \text{ or } B)|}{3}$$

Equation 2-4

4.1.4.5 Use the Type S pitot tube only if the values of δ (side A) and δ (side B) are less than or equal to 0.01 and if the absolute value of the difference between C_p (A) and C_p (B) is 0.01 or less.

4.1.5 Special considerations.

4.1.5.1 Selection of calibration point.

4.1.5.1.1 When an isolated Type S pitot tube is calibrated, select a calibration point at or near the center of the duct, and follow the procedures outlined in Sections 4.1.3 and 4.1.4 above. The Type S pitot coefficients so obtained, i.e., C_p (side A) and C_p (side B), will be valid, so long as either: (1) the isolated pitot tube is used; or (2) the pitot tube is used with other components (nozzle, thermocouple, sample probe) in an arrangement that is free from aerodynamic interference effects (see Figures 2-6 through 2-8).

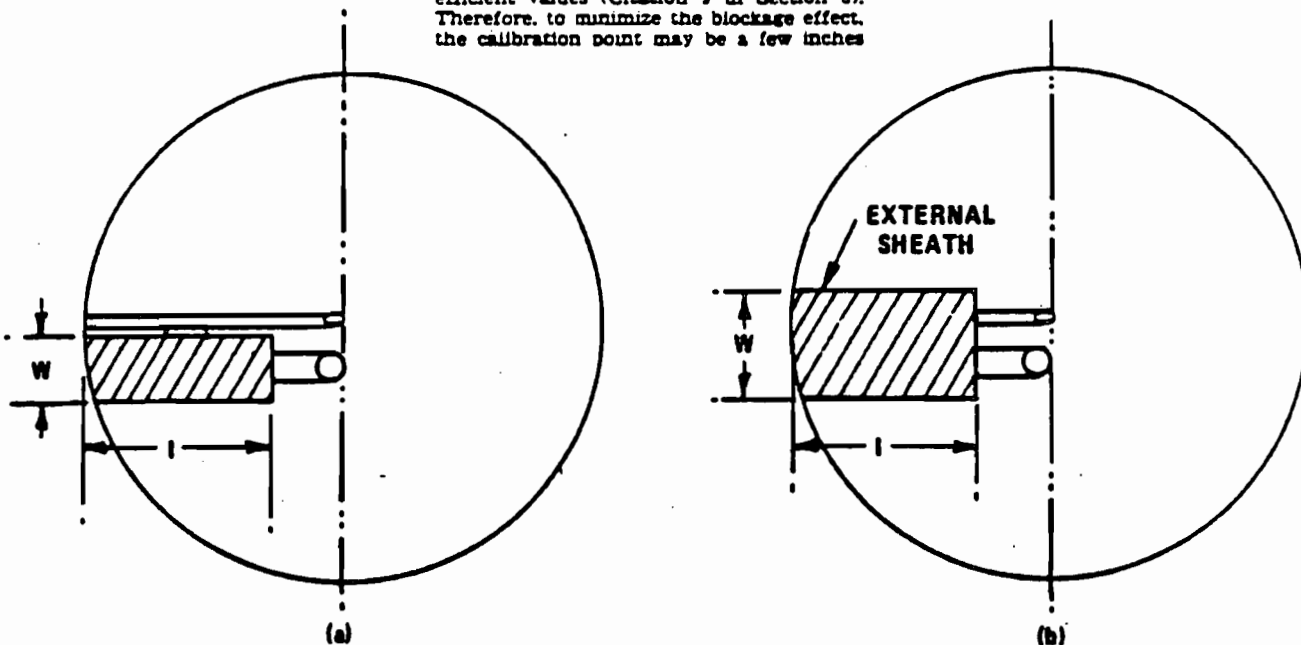
4.1.5.1.2 For Type S pitot tube-thermocouple combinations (without sample probe), select a calibration point at or near the center of the duct, and follow the procedures outlined in Sections 4.1.3 and 4.1.4 above. The coefficients so obtained will be valid so long as the pitot tube-thermocouple combination is used by itself or with other components in an interference-free arrangement (Figures 2-6, and 2-8).

4.1.5.1.3 For assemblies with sample probes, the calibration point should be located at or near the center of the duct; however, insertion of a probe sheath into a small duct may cause significant cross-sectional area blockage and yield incorrect coefficient values (Citation 9 in Section 6). Therefore, to minimize the blockage effect, the calibration point may be a few inches

off-center if necessary. The actual blockage effect will be negligible when the theoretical blockage, as determined by a projected-area model of the probe sheath, is 2 percent or less of the duct cross-sectional area for assemblies without external sheaths (Figure 2-10a), and 3 percent or less for assemblies with external sheaths (Figure 2-10b).

4.1.5.2 For those probe assemblies in which pitot tube-nozzle interference is a factor (i.e., those in which the pitot-nozzle separation distance fails to meet the specification illustrated in Figure 2-4a), the value of $C_{p_{meas}}$ depends upon the amount of free-space between the tube and nozzle, and therefore is a function of nozzle size. In these instances, separate calibrations shall be performed with each of the commonly used nozzle sizes in place. Note that the single-velocity calibration technique is acceptable for this purpose, even though the larger nozzle sizes (>0.635 cm or 1/4 in.) are not ordinarily used for isokinetic sampling at velocities around 915 m/min (3,000 ft/min), which is the calibration velocity; note also that it is not necessary to draw an isokinetic sample during calibration (see Citation 19 in Section 6).

4.1.5.3 For a probe assembly constructed such that its pitot tube is always used in the same orientation, only one side of the pitot tube need be calibrated (the side which will face the flow). The pitot tube must still meet the alignment specifications of Figure 2-2 or 2-3, however, and must have an average deviation (δ) value of 0.01 or less (see Section 4.1.4.4).



$$\text{ESTIMATED SHEATH BLOCKAGE (\%)} = \left[\frac{l \times W}{\text{DUCT AREA}} \right] \times 100$$

Figure 2-10. Projected-area models for typical pitot tube assemblies.

Figure 2-10. Projected-area models for typical pitot tube assemblies.

4.1.6 Field Use and Recalibration.

4.1.6.1 Field Use.

4.1.6.1.1 When a Type S pitot tube (isolated tube or assembly) is used in the field, the appropriate coefficient value (whether assigned or obtained by calibration) shall be used to perform velocity calculations. For calibrated Type S pitot tubes, the A side coefficient shall be used when the A side of the tube faces the flow, and the B side coefficient shall be used when the B side faces the flow; alternatively, the arithmetic average of the A and B side coefficient values may be used, irrespective of which side faces the flow.

4.1.6.1.2 When a probe assembly is used to sample a small duct (12 to 36 in. in diameter), the probe sheath sometimes blocks a significant part of the duct cross-section, causing a reduction in the effective value of C_p . Consult Citation 9 in Section 6 for details. Conventional pitot-sampling probe assemblies are not recommended for use in ducts having inside diameters smaller than 12 inches (Citation 16 in Section 6).

4.1.6.2 Recalibration.

4.1.6.2.1 Isolated Pitot Tubes. After each field use, the pitot tube shall be carefully reexamined in top, side, and end views. If the pitot face openings are still aligned within the specifications illustrated in Figure 2-2 or 2-3, it can be assumed that the baseline coefficient of the pitot tube has not changed. If, however, the tube has been damaged to the extent that it no longer meets the specifications of Figure 2-2 or 2-3, the damage shall either be repaired to restore proper alignment of the face openings or the tube shall be discarded.

4.1.6.2.2 Pitot Tube Assemblies. After each field use, check the face opening alignment of the pitot tube, as in Section 1.8.2.1; also, remeasure the intercomponent spacings of the assembly. If the intercomponent spacings have not changed and the face opening alignment is acceptable, it can be assumed that the coefficient of the assembly has not changed. If the face opening alignment is no longer within the specifications of Figures 2-2 or 2-3, either repair the damage or replace the pitot tube (calibrating the new assembly, if necessary). If the intercomponent spacings have changed, restore the original spacings or recalibrate the assembly.

4.2 Standard pitot tube (if applicable). If a standard pitot tube is used for the velocity traverse, the tube shall be constructed according to the criteria of Section 2.7 and shall be assigned a baseline coefficient value of 0.99. If the standard pitot tube is used as part of an assembly, the tube shall be in an interference-free arrangement (subject to the approval of the Administrator).

4.3 Temperature Gauges. After each field use, calibrate dial thermometers, liquid-filled bulb thermometers, thermocouple-potentiometer systems, and other gauges at a temperature within 10 percent of the average absolute stack temperature. For temperatures up to 405° C (761° F), use an ASTM mercury-in-glass reference thermometer, or equivalent, as a reference; alternatively, either a reference thermocouple and potentiometer (calibrated by NBS) or thermometric fixed points, e.g., ice bath and boiling water (corrected for barometric pressure) may be used. For temperatures above 405° C (761° F), use an NBS-calibrated reference thermocouple-potentiometer system or an alternate reference, subject to the approval of the Administrator.

If, during calibration, the absolute temperatures measured with the gauge being calibrated and the reference gauge agree within 1.5 percent, the temperature data taken in the field shall be considered valid. Otherwise, the pollutant emission test shall either be considered invalid or adjustments (if appropriate) of the test results shall be made, subject to the approval of the Administrator.

4.4 Barometer. Calibrate the barometer used against a mercury barometer.

5. Calculations

Carry out calculations, retaining at least one extra decimal figure beyond that of the acquired data. Round off figures after final calculation.

5.1 Nomenclature.

- A = Cross-sectional area of stack, m² (ft²).
- B_w = Water vapor in the gas stream (from Method 5 or Reference Method 4), proportion by volume.
- C_p = Pitot tube coefficient, dimensionless.
- K_p = Pitot tube constant.

$$34.97 \frac{m}{sec} \left[\frac{(g/g \cdot mole)(mm \text{ Hg})}{(^\circ K)(mm \text{ H}_2\text{O})} \right]^{1/2}$$

for the metric system and

$$35.43 \frac{ft}{sec} \left[\frac{(lb \text{ lb-mole}^{-1})(in. \text{ Hg})}{(^\circ R)(in. \text{ H}_2\text{O})} \right]^{1/2}$$

for the English system.

- M_s = Molecular weight of stack gas, dry basis (see Section 3.8) g/g-mole (lb/lb-mole).
- M_w = Molecular weight of stack gas, wet basis, g/g-mole (lb/lb-mole).
- = M_s (1 - B_w) - 18.0 B_w

Equation 2-3

- P_{atm} = Barometric pressure at measurement site, mm Hg (in. Hg).
- P_s = Stack static pressure, mm Hg (in. Hg).

- P_a = Absolute stack gas pressure, mm Hg (in. Hg).
- = P_{atm} + P_s

Equation 2-6

- P_{std} = Standard absolute pressure, 760 mm Hg (29.92 in. Hg).
- Q_{sc} = Dry volumetric stack gas flow rate corrected to standard conditions, dscm/hr (dscf/hr).
- t = Stack temperature, °C (°F).
- T_a = Absolute stack temperature, °K (°R).
- = 273 + t for metric

Equation 2-7

- = 460 + t for English

Equation 2-8

- T_{std} = Standard absolute temperature, 293 °K (528° R)
- v = Average stack gas velocity, m/sec (ft/sec).
- h₀ = Velocity head of stack gas, mm H₂O (in. H₂O).
- 3,600 = Conversion factor, sec/hr.
- 18.0 = Molecular weight of water, g/g-mole (lb/lb-mole).

5.2 Average stack gas velocity.

$$v = K_p C_p (\sqrt{\Delta p})_{0.5} \sqrt{\frac{T_{std}}{P_a M}}$$

Equation 2-9

5.3 Average stack gas dry volumetric flow rate.

$$Q_{sc} = 3,600 (1 - B_w) v A \left(\frac{T_{std}}{T_a} \right) \left(\frac{P_a}{P_{std}} \right)$$

Equation 2-10

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METHOD 3

METHOD 3—GAS ANALYSIS FOR CARBON DIOXIDE, OXYGEN, EXCESS AIR, AND DRY MOLECULAR WEIGHT

1. Principle and Applicability

1.1 Principle. A gas sample is extracted from a stack, by one of the following methods: (1) single-point, grab sampling; (2) single-point, integrated sampling; or (3) multi-point, integrated sampling. The gas sample is analyzed for percent carbon dioxide (CO₂), percent oxygen (O₂), and, if necessary, percent carbon monoxide (CO). If a dry molecular weight determination is to be made, either an Orsat or a Pyrite analyzer may be used for the analysis; for excess air or emission rate correction factor determination, an Orsat analyzer must be used.

1.2 Applicability. This method is applicable for determining CO₂ and O₂ concentrations, excess air, and dry molecular weight of a sample from a gas stream of a fossil-fuel combustion process. The method may also be applicable to other processes where it has been determined that compounds other than CO₂, O₂, CO, and nitrogen (N₂) are not present in concentrations sufficient to affect the results.

Other methods, as well as modifications to the procedure described herein, are also applicable for some or all of the above determinations. Examples of specific methods and modifications include: (1) a multi-point sampling method using an Orsat analyzer to analyze individual grab samples obtained at each point; (2) a method using CO₂ or O₂ and stoichiometric calculations to determine dry molecular weight and excess air; (3) assigning a value of 30.0 for dry molecular weight, in lieu of actual measurements, for processes burning natural gas, coal, or oil. These methods and modifications may be used, but are subject to the approval of the Administrator, U.S. Environmental Protection Agency.

2. Apparatus

As an alternative to the sampling apparatus and systems described herein, other sampling systems (e.g., liquid displacement) may be used provided such systems are capable of obtaining a representative sample and maintaining a constant sampling rate, and are otherwise capable of yielding acceptable results. Use of such systems is subject to the approval of the Administrator.

2.1 Grab Sampling (Figure 3-1).

2.1.1 Probe. The probe should be made of stainless steel or borosilicate glass tubing and should be equipped with an in-stack or out-stack filter to remove particulate matter (a plug of glass wool is satisfactory for this purpose). Any other materials inert to O₂, CO₂, CO, and N₂, and resistant to temperature at sampling conditions may be used for the probe; examples of such material are aluminum, copper, quartz glass and Teflon.

2.1.2 Pump. A one-way squeeze bulb, or equivalent, is used to transport the gas sample to the analyzer.

2.2 Integrated Sampling (Figure 3-2).

2.2.1 Probe. A probe such as that described in Section 2.1.1 is suitable.

2.2.2 Condenser. An air-cooled or water-cooled condenser, or other condenser that will not remove O₂, CO₂, CO, and N₂, may be used to remove excess moisture which would interfere with the operation of the pump and flow meter.

2.2.3 Valve. A needle valve is used to adjust sample gas flow rate.

2.2.4 Pump. A leak-free, diaphragm-type pump, or equivalent, is used to transport sample gas to the flexible bag. Install a small surge tank between the pump and rate meter to eliminate the pulsation effect of the diaphragm pump on the rotameter.

2.2.5 Rate Meter. The rotameter, or equivalent rate meter, used should be capable of measuring flow rate to within ± 2 percent of the selected flow rate. A flow rate range of 500 to 1000 cm³/min is suggested.

2.2.6 Flexible Bag. Any leak-free plastic (e.g., Tedlar, Mylar, Teflon) or plastic-coated aluminum (e.g., aluminized Mylar) bag, or equivalent, having a capacity consistent with the selected flow rate and time

length of the test run, may be used. A capacity in the range of 55 to 90 liters is suggested.

To leak-check the bag, connect it to a water manometer and pressurize the bag to 5 to 10 cm H₂O (2 to 4 in. H₂O). Allow to stand for 10 minutes. Any displacement in the water manometer indicates a leak. An alternative leak-check method is to pressurize the bag to 5 to 10 cm H₂O (2 to 4 in. H₂O) and allow to stand overnight. A deflated bag indicates a leak.

2.2.7 Pressure Gauge. A water-filled U-tube manometer, or equivalent, of about 28 cm (12 in.) is used for the flexible bag leak-check.

2.2.8 Vacuum Gauge. A mercury manometer, or equivalent, of at least 760 mm Hg (30 in. Hg) is used for the sampling train leak-check.

2.3 Analysis. For Orsat and Pyrite analyzer maintenance and operation procedures, follow the instructions recommended by the manufacturer, unless otherwise specified herein.

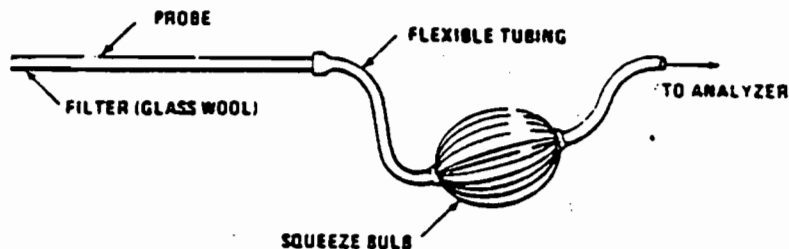


Figure 3-1. Grab sampling train.

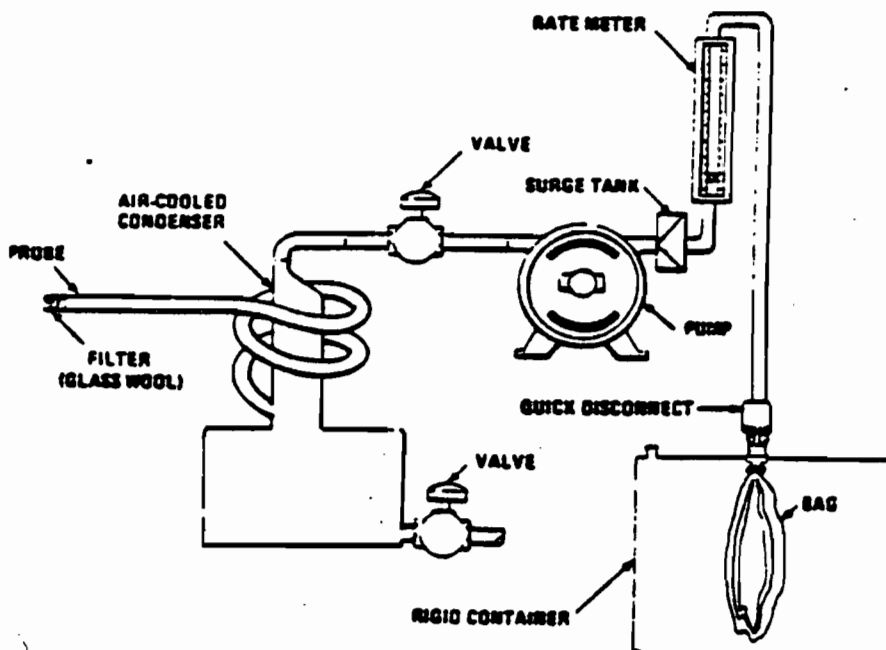


Figure 3-2. Integrated gas-sampling train.

Mention of trade names or specific products does not constitute endorsement by the Environmental Protection Agency.

2.3.1 Dry Molecular Weight Determination. An Orsat analyzer or Pyrite type combustion gas analyzer may be used.

2.3.2 Emission Rate Correction Factor or Excess Air Determination. An Orsat analyzer must be used. For low CO₂ (less than 4.0 percent) or high O₂ (greater than 13.0 percent) concentrations, the measuring burette of the Orsat must have at least 0.1 percent subdivisions.

3. Dry Molecular Weight Determination

Any of the three sampling and analytical procedures described below may be used for determining the dry molecular weight.

3.1 Single-Point, Grab Sampling and Analytical Procedure.

3.1.1 The sampling point in the duct shall either be at the centroid of the cross section or at a point no closer to the walls than 1.00 m (3.3 ft), unless otherwise specified by the Administrator.

3.1.2 Set up the equipment as shown in Figure 3-1, making sure all connections ahead of the analyzer are tight and leak-free. If an Orsat analyzer is used, it is recommended that the analyzer be leak-checked by following the procedure in Section 5; however, the leak-check is optional.

3.1.3 Place the probe in the stack, with the tip of the probe positioned at the sampling point; purge the sampling line. Draw a sample into the analyzer and immediately analyze it for percent CO₂ and percent O₂. Determine the percentage of the gas that is N₂ and CO by subtracting the sum of the percent CO₂ and percent O₂ from 100 percent. Calculate the dry molecular weight as indicated in Section 6.3.

3.1.4 Repeat the sampling, analysis, and calculation procedures, until the dry molecular weights of any three grab samples differ from their mean by no more than 0.3 g/g-mole (0.3 lb/lb-mole). Average these three molecular weights, and report the results to the nearest 0.1 g/g-mole (lb/lb-mole).

3.2 Single-Point, Integrated Sampling and Analytical Procedure.

3.2.1 The sampling point in the duct shall be located as specified in Section 3.1.1.

3.2.2 Leak-check (optional) the flexible bag as in Section 2.2.6. Set up the equipment as shown in Figure 3-2. Just prior to sampling, leak-check (optional) the train by placing a vacuum gauge at the condenser inlet, pulling a vacuum of at least 250 mm Hg (10 in. Hg), plugging the outlet at the quick disconnect, and then turning off the pump. The vacuum should remain stable for at least 0.5 minute. Evacuate the flexible bag. Connect the probe and place it in the stack, with the tip of the probe positioned at the sampling point; purge the sampling line. Next, connect the bag and make sure that all connections are tight and leak free.

3.2.3 Sample at a constant rate. The sampling run should be simultaneous with, and for the same total length of time as, the pollutant emission rate determination. Collection of at least 30 liters (1.00 ft³) of sample gas is recommended; however, smaller volumes may be collected, if desired.

3.2.4 Obtain one integrated flue gas sample during each pollutant emission rate determination. Within 8 hours after the sample is taken, analyze it for percent CO₂ and percent O₂, using either an Orsat analyzer or a Pyrite-type combustion gas analyzer. If an Orsat analyzer is used, it is recommended that the Orsat leak-check described in Section 5 be performed before this determination; however, the check is optional.

Determine the percentage of the gas that is N₂ and CO by subtracting the sum of the percent CO₂ and percent O₂ from 100 percent. Calculate the dry molecular weight as indicated in Section 6.3.

3.2.5 Repeat the analysis and calculation procedures until the individual dry molecular weights for any three analyses differ from their mean by no more than 0.3 g/g-mole (0.3 lb/lb-mole). Average these three molecular weights, and report the results to the nearest 0.1 g/g-mole (0.1 lb/lb-mole).

3.3 Multi-Point, Integrated Sampling and Analytical Procedure.

3.3.1 Unless otherwise specified by the Administrator, a minimum of eight traverse points shall be used for circular stacks having diameters less than 0.61 m (24 in.), a minimum of nine shall be used for rectangular stacks having equivalent diameters less than 0.61 m (24 in.), and a minimum of twelve traverse points shall be used for all other cases. The traverse points shall be located according to Method 1. The use of fewer points is subject to approval of the Administrator.

3.3.2 Follow the procedures outlined in sections 3.2.2 through 3.2.5, except for the following: traverse all sampling points and sample at each point for an equal length of time. Record sampling data as shown in Figure 3-3.

4. Emission Rate Correction Factor or Excess Air Determination

Note: A Pyrite-type combustion gas analyzer is not acceptable for excess air or emission rate correction factor determination, unless approved by the Administrator. If both percent CO₂ and percent O₂ are measured, the analytical results of any of the three procedures given below may also be used for calculating the dry molecular weight.

Each of the three procedures below shall be used only when specified in an applicable subpart of the standards. The use of these procedures for other purposes must have specific prior approval of the Administrator.

4.1 Single-Point, Grab Sampling and Analytical Procedure.

4.1.1 The sampling point in the duct shall either be at the centroid of the cross-section or at a point no closer to the walls than 1.00 m (3.3 ft), unless otherwise specified by the Administrator.

4.1.2 Set up the equipment as shown in Figure 3-1, making sure all connections ahead of the analyzer are tight and leak-free. Leak-check the Orsat analyzer according to the procedure described in Section 5. This leak-check is mandatory.

FIGURE 3-3—SAMPLING RATE DATA

Time	Traverse pt.	O ₂ (%)	CO (%)
Average			

* % Dev. = $(O_2 - \bar{O}_2) / \bar{O}_2 \times 100$ (shall be <10%)

4.1.3 Place the probe in the stack, with the tip of the probe positioned at the sampling point; purge the sampling line. Draw a sample into the analyzer. For emission rate correction factor determination, immediately analyze the sample, as outlined in Sections 4.1.4 and 4.1.5, for percent CO₂ or percent O₂. If excess air is desired, proceed as follows: (1) immediately analyze the sample, as in Sections 4.1.4 and 4.1.5, for percent CO₂, O₂, and CO; (2) determine the percentage of the gas that is N₂ by subtracting the sum of the percent CO₂, percent O₂, and percent CO from 100 percent; and (3) calculate percent excess air as outlined in Section 6.2.

4.1.4 To insure complete absorption of the CO₂, O₂, or if applicable, CO, make repeated passes through each absorbing solution until two consecutive readings are the same. Several passes (three or four) should be made between readings. (If constant readings cannot be obtained after three consecutive readings, replace the absorbing solution.)

4.1.5 After the analysis is completed, leak-check (mandatory) the Orsat analyzer once again, as described in Section 5. For the results of the analysis to be valid, the Orsat analyzer must pass this leak test before and after the analysis.

Note: Since this single-point, grab sampling and analytical procedure in normally conducted in conjunction with a single-point, grab sampling and analytical procedure for a pollutant, only one analysis is ordinarily conducted. Therefore, great care must be taken to obtain a valid sample and analysis. Although in most cases only CO₂ or O₂ is required, it is recommended that both CO₂ and O₂ be measured, and that Section 4.4 be used to validate the analytical data.

4.2 Single-Point, Integrated Sampling and Analytical Procedure.

4.2.1 The sampling point in the duct shall be located as specified in Section 4.1.1.

4.2.2 Leak-check (mandatory) the flexible bag as in Section 2.2.6. Set up the equipment as shown in Figure 3-2. Just prior to sampling, leak-check (mandatory) the train by placing a vacuum gauge at the condenser inlet, pulling a vacuum of at least 250 mm Hg (10 in. Hg), plugging the outlet at the quick disconnect, and then turning off the pump. The vacuum shall remain stable for at least 0.5 minute. Evacuate the flexible bag. Connect the probe and place it in the stack, with the tip of the probe positioned at the sampling point; purge the sampling line. Next, connect the bag and make sure that all connections are tight and leak free.

4.2.3 Sample at a constant rate, or as specified by the Administrator. The sampling run must be simultaneous with, and for the same total length of time as, the pollutant emission rate determination. Collect at least 30 liters (1.00 ft³) of sample gas. Smaller volumes may be collected, subject to approval of the Administrator.

4.2.4 Obtain one integrated flue gas sample during each pollutant emission rate determination. For emission rate correction factor determination, analyze the sample within 4 hours after it is taken for percent CO₂ or percent O₂ (as outlined in Sections 4.2.5 through 4.2.7). The Orsat analyzer must be leak-checked (see Section 5) before the analysis. If excess air is desired, proceed as follows: (1) within 4 hours after the sample is taken, analyze it (as in Sections 4.2.5 through 4.2.7) for percent CO₂, O₂, and CO; (2) determine the percentage of the gas that is N₂ by subtracting the sum of the percent CO₂, percent O₂, and percent CO from 100 percent; (3) calculate percent excess air, as outlined in Section 6.2.

4.2.5 To insure complete absorption of the CO₂, O₂, or if applicable, CO, make repeated passes through each absorbing solution until two consecutive readings are the same. Several passes (three of four) should be made between readings. (If constant readings cannot be obtained after three consecutive readings, replace the absorbing solution.)

4.2.6 Repeat the analysis until the following criteria are met:

4.2.6.1 For percent CO₂, repeat the analytical procedure until the results of any three analyses differ by no more than (a) 0.3 percent by volume when CO₂ is greater than 4.0 percent or (b) 0.2 percent by volume when CO₂ is less than or equal to 4.0 percent. Average the three acceptable values of percent CO₂ and report the results to the nearest 0.1 percent.

4.2.6.2 For percent O₂, repeat the analytical procedure until the results of any three analyses differ by no more than (a) 0.3 percent by volume when O₂ is less than 15.0 percent or (b) 0.2 percent by volume when O₂ is greater than or equal to 15.0 percent. Average the three acceptable values of percent O₂ and report the results to the nearest 0.1 percent.

4.2.6.3 For percent CO, repeat the analytical procedure until the results of any three analyses differ by no more than 0.3 percent. Average the three acceptable values of percent CO and report the results to the nearest 0.1 percent.

4.2.7 After the analysis is completed, leak-check (mandatory) the Orsat analyzer once again, as described in Section 5. For the results of the analysis to be valid, the Orsat analyzer must pass this leak test before an after the analysis.

NOTE: Although in most instances only CO₂ or O₂ is required, it is recommended that both CO₂ and O₂ be measured, and that Citation 5 in the Bibliography be used to validate the analytical data.

4.3 Multi-Point, Integrated Sampling and Analytical Procedure.

4.3.1 Both the minimum number of sampling points and the sampling point location shall be as specified in Section 3.3.1 of this method. The use of fewer points than specified is subject to the approval of the Administrator.

4.3.2 Follow the procedures outlined in Sections 4.2.2 through 4.2.7, except for the following: Traverse all sampling points and sample at each point for an equal length of time. Record sampling data as shown in Figure 3-3.

4.4 Quality Control Procedures.

4.4.1 Data Validation When Both CO₂ and O₂ Are Measured. Although in most instances, only CO₂ or O₂ measurement is required, it is recommended that both CO₂ and O₂ be measured to provide a check on the quality of the data. The following quality control procedure is suggested.

NOTE: Since the method for validating the CO₂ and O₂ analysis is based on combustion of organic and fossil fuels and dilution of the gas stream with air, this method does not apply to sources that (1) remove CO₂ or O₂ through processes other than combustion, (2) add O₂ (e.g., oxygen enrichment) and N₂ in proportions different from that of air, (3) add CO₂ (e.g., cement or lime kilns), or (4) have no fuel factor, F_u, values obtainable (e.g., extremely variable waste mixtures). This method validates the measured proportions of CO₂ and O₂ for the fuel type, but the method does not detect sample dilution resulting from leaks during or after sample collection. The method is applicable

for samples collected downstream of most lime or limestone flue-gas desulfurization units as the CO₂ added or removed from the gas stream is not significant in relation to the total CO₂ concentration. The CO₂ concentrations from other types of scrubbers using only water or basic slurry can be significantly affected and would render the F_u check minimally useful.

4.4.1.1 Calculate a fuel factor, F_u, using the following equation:

$$F_u = \frac{20.9 - \%O_2}{\%CO_2}$$

Eq. 3-3

Where:
 %O₂ = Percent O₂ by volume (dry basis).
 %CO₂ = Percent CO₂ by volume (dry basis).
 20.9 = Percent O₂ by volume in ambient air.
 If CO is present in quantities measurable by this method, adjust the O₂ and CO₂ values before performing the calculation for F_u, as follows:

$$\%CO_2(\text{adj}) = \%CO_2 - \%CO$$

$$\%O_2(\text{adj}) = \%O_2 - 0.5 \%CO$$

Where: %CO = Percent CO by volume (dry basis).

4.4.1.2 Compare the calculated F_u factor with the expected F_u values. The following table may be used in establishing acceptable ranges for the expected F_u, if the fuel being burned is known. When fuels are burned in combination, calculate the combined fuel F_u and F_u factors (as defined in Method 19) according to the procedure in Method 19 Section 5.2.3. Then calculate the F_u factor as follows:

$$F_u = \frac{0.209 F_u}{F_u}$$

Eq. 3-4

Fuel type	F _u range
Coal	Anthracite and lignite..... 1.016-1.130
	Subbituminous..... 1.003-1.230
Oil	Diesel..... 1.280-1.413
	Residual..... 1.210-1.370
Gas	Natural..... 1.030-1.836
	Propane..... 1.434-1.588
	Butane..... 1.405-1.953
Wood 1.000-1.120
Wood bark 1.003-1.130

Calculated F_u values beyond the acceptable ranges shown in this table should be investigated before accepting the test results. For example, the strength of the solutions in the gas analyzer and the analyzing technique should be checked by sampling and analyzing a known concentration, such as air; the fuel factor should be reviewed and verified. An acceptability range of ±12 percent is appropriate for the F_u factor of mixed fuels with variable fuel ratios. The level of the emission rate relative to the compliance level should be considered in determining if a retest is appropriate. I.e., if the measured emissions are much lower or much greater than the compliance limit, repetition of the test would not significantly change the compliance status of the source and would be unnecessarily time-consuming and costly.

5. Leak-Check Procedure for Orsat Analyzers

Moving an Orsat analyzer (frequently causes it to leak. Therefore, an Orsat analyzer should be thoroughly leak-checked on site before the flue gas sample is introduced into it. The procedure for leak-checking an Orsat analyzer is:

5.1.1 Bring the liquid level in each pipette up to the reference mark on the capillary tubing and then close the pipette stopcock.

5.1.2 Raise the leveling bulb sufficiently to bring the confining liquid meniscus onto the graduated portion of the burette and then close the manifold stopcock.

5.1.3 Record the meniscus position.

5.1.4 Observe the meniscus in the burette and the liquid level in the pipette for movement over the next 4 minutes.

5.1.5 For the Orsat analyzer to pass the leak-check, two conditions must be met.

5.1.5.1 The liquid level in each pipette must not fall below the bottom of the capillary tubing during this 4-minute interval.

5.1.5.2 The meniscus in the burette must not change by more than 0.2 ml during this 4-minute interval.

5.1.6 If the analyzer fails the leak-check procedure, all rubber connections and stopcocks should be checked until the cause of the leak is identified. Leaking stopcocks must be disassembled, cleaned, and re-greased. Leaking rubber connections must be replaced. After the analyzer is reassembled, the leak-check procedure must be repeated.

6. Calculations

6.1 Nomenclature.

6.1 Nomenclature.

M_w = Dry molecular weight, g/g-mole (lb/lb-mole).

%EA = Percent excess air.

%CO₂ = Percent CO₂ by volume (dry basis).

%O₂ = Percent O₂ by volume (dry basis).

%CO = Percent CO by volume (dry basis).

%N₂ = Percent N₂ by volume (dry basis).

0.284 = Ratio of O₂ to N₂ in air, v/v.

0.280 = Molecular weight of N₂ or CO, divided by 100.

0.320 = Molecular weight of O₂ divided by 100.

0.440 = Molecular weight of CO, divided by 100.

6.2 Percent Excess Air. Calculate the percent excess air (if applicable), by substituting the appropriate values of percent O₂, CO, and N₂ (obtained from Section 4.1.3 or 4.2.4) into Equation 3-1.

% EA =

$$\frac{\%O_2 - 0.5 \%CO}{0.284 \%N_2 - \%O_2 - 0.5 \%CO} \times 100$$

$$\times 100$$

$$\times 100$$

$$\times 100$$

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$$\times 100$$

$$\times 100$$

NOTE: The above equation does not consider argon in air (about 0.9 percent, molecular weight of 39.9). A negative error of about 0.4 percent is introduced. The tester may opt to include argon in the analysis using procedures subject to approval of the Administrator.

7. Bibliography

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4. Mitchell, W. J. and M. R. Midgett. Field Reliability of the Orsat Analyzer. *Journal of Air Pollution Control Association* 26:491-495. May 1976.

5. Shigehara, R. T., R. M. Neullcht, and W. S. Smith. Validating Orsat Analysis Data from Fossil Fuel-Fired Units. *Stack Sampling News*. 6(2):21-26. August, 1976.

METHOD 4—DETERMINATION OF MOISTURE CONTENT IN STACK GASES

1. Principle and Applicability

1.1 Principle. A gas sample is extracted at a constant rate from the source; moisture is removed from the sample stream and determined either volumetrically or gravimetrically.

1.2 Applicability. This method is applicable for determining the moisture content of stack gas.

Two procedures are given. The first is a reference method, for accurate determinations of moisture content (such as are needed to calculate emission data). The second is an approximation method, which provides estimates of percent moisture to aid in setting isokinetic sampling rates prior to a pollutant emission measurement run. The approximation method described herein is only a suggested approach; alternative means for approximating the moisture content, e.g., drying tubes, wet bulb-dry bulb techniques, condensation techniques, stoichiometric calculations, previous experience, etc., are also acceptable.

The reference method is often conducted simultaneously with a pollutant emission measurement run; when it is, calculation of percent isokinetic, pollutant emission rate, etc., for the run shall be based upon the results of the reference method or its equivalent; these calculations shall not be based upon the results of the approximation method, unless the approximation method is shown, to the satisfaction of the Administrator, U.S. Environmental Protection Agency, to be capable of yielding results within 1 percent H₂O of the reference method.

NOTE: The reference method may yield questionable results when applied to saturated gas streams or to streams that contain water droplets. Therefore, when these conditions exist or are suspected, a second determination of the moisture content shall be made simultaneously with the reference method, as follows: Assume that the gas stream is saturated. Attach a temperature sensor (capable of measuring to $\pm 1^\circ\text{C}$ (2°F)) to the reference method probe. Measure the stack gas temperature at each traverse point (see Section 2.2.1) during the reference method traverse; calculate the average stack gas temperature. Next, determine the moisture percentage, either by: (1) using a psychrometric chart and making appropriate corrections if stack pressure is different from that of the chart, or (2) using saturation vapor pressure tables. In cases where the psychrometric chart or the saturation vapor pressure tables are not applicable (based on evaluation of the process), alternate methods, subject to the approval of the Administrator, shall be used.

2. Reference Method

The procedure described in Method 5 for determining moisture content is acceptable as a reference method.

2.1 Apparatus. A schematic of the sampling train used in this reference method is shown in Figure 4-1. All components shall be maintained and calibrated according to the procedure outlined in Method 5.

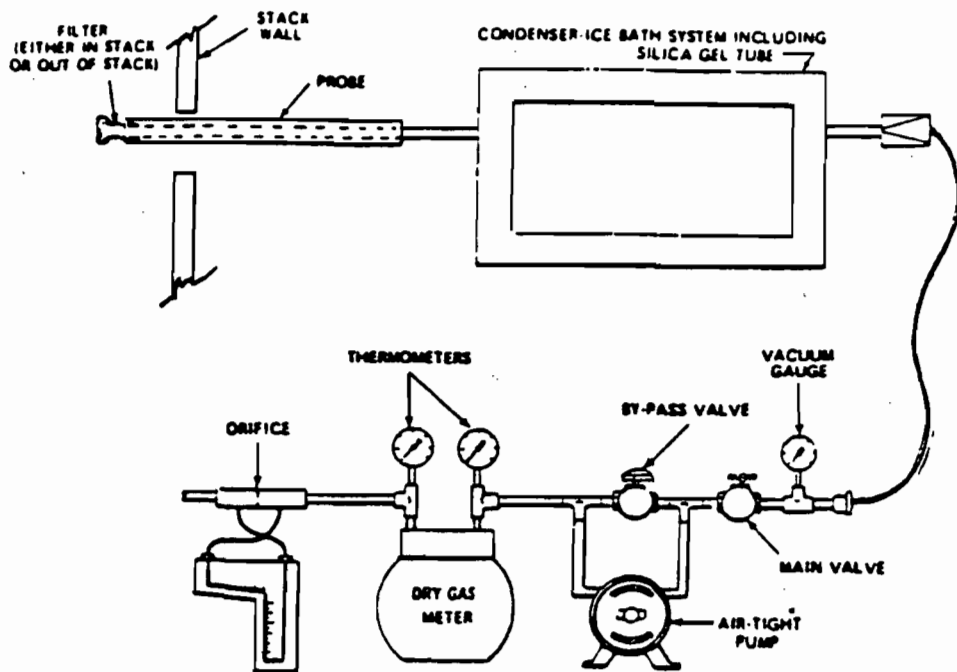


Figure 4-1. Moisture sampling train-reference method.

2.1.1 Probe. The probe is constructed of stainless steel or glass tubing, sufficiently heated to prevent water condensation, and is equipped with a filter, either in-stack (e.g., a plug of glass wool inserted into the end of the probe) or heated out-stack (e.g., as described in Method 5), to remove particulate matter.

When stack conditions permit, other metals or plastic tubing may be used for the probe, subject to the approval of the Administrator.

2.1.2 Condenser. The condenser consists of four impingers connected in series with ground glass, leak-free fittings or any similarly leak-free non-contaminating fittings. The first, third, and fourth impingers shall be of the Greenburg-Smith design, modified by replacing the tip with a 1.3 centimeter ($\frac{1}{2}$ inch) ID glass tube extending to about 1.3 cm ($\frac{1}{2}$ in.) from the bottom of the flask. The second impinger shall be of the Greenburg-Smith design with the standard tip. Modifications (e.g., using flexible connections between the impingers, using materials other than glass, or using flexible vacuum lines to connect the filter holder to the condenser) may be used, subject to the approval of the Administrator.

The first two impingers shall contain known volumes of water, the third shall be empty, and the fourth shall contain a known weight of 6- to 16-mesh indicating type silica gel, or equivalent desiccant. If the silica gel has been previously used, dry at 175°C (350°F) for 2 hours. New silica gel may be used as received. A thermometer, capable of measuring temperature to within 1°C (2°F), shall be placed at the outlet of the fourth impinger, for monitoring purposes.

Alternatively, any system may be used (subject to the approval of the Administrator) that cools the sample gas stream and allows measurement of both the water that has been condensed and the moisture leaving the condenser, each to within 1 ml or 1 g. Acceptable means are to measure the condensed water, either gravimetrically or volumetrically, and to measure the moisture leaving the condenser by: (1) monitoring the temperature and pressure at the exit of the condenser and using Dalton's law of partial pressures, or (2) passing the sample gas stream through a tared silica gel (or equivalent desiccant) trap, with exit gases kept below 20°C (65°F), and determining the weight gain.

If means other than silica gel are used to determine the amount of moisture leaving the condenser, it is recommended that silica gel (or equivalent) still be used between the condenser system and pump, to prevent moisture condensation in the pump and metering devices and to avoid the need to make corrections for moisture in the metered volume.

2.1.3 **Cooling System.** An ice bath container and crushed ice (or equivalent) are used to aid in condensing moisture.

2.1.4 **Metering System.** This system includes a vacuum gauge, leak-free pump, thermometers capable of measuring temperature to within 3° C (5.4° F), dry gas meter capable of measuring volume to within 2 percent, and related equipment as shown in Figure 4-1. Other metering systems, capable of maintaining a constant sampling rate and determining sample gas volume, may be used, subject to the approval of the Administrator.

2.1.5 **Barometer.** Mercury, aneroid, or other barometer capable of measuring atmospheric pressure to within 2.5 mm Hg (0.1 in. Hg) may be used. In many cases, the barometric reading may be obtained from a nearby national weather service station, in which case the station value (which is the absolute barometric pressure) shall be requested and an adjustment for elevation differences between the weather station and the sampling point shall be applied at a rate of minus 2.5 mm Hg (0.1 in. Hg) per 30 m (100 ft) elevation increase or vice versa for elevation decrease.

2.1.6 **Graduated Cylinder and/or Balance.** These items are used to measure condensed water and moisture caught in the silica gel to within 1 ml or 0.5 g. Graduated cylinders shall have subdivisions no greater than 2 ml. Most laboratory balances are capable of weighing to the nearest 0.5 g or less. These balances are suitable for use here.

2.2 **Procedure.** The following procedure is written for a condenser system (such as the impinger system described in Section 2.1.2) incorporating volumetric analysis to measure the condensed moisture, and silica gel and gravimetric analysis to measure the moisture leaving the condenser.

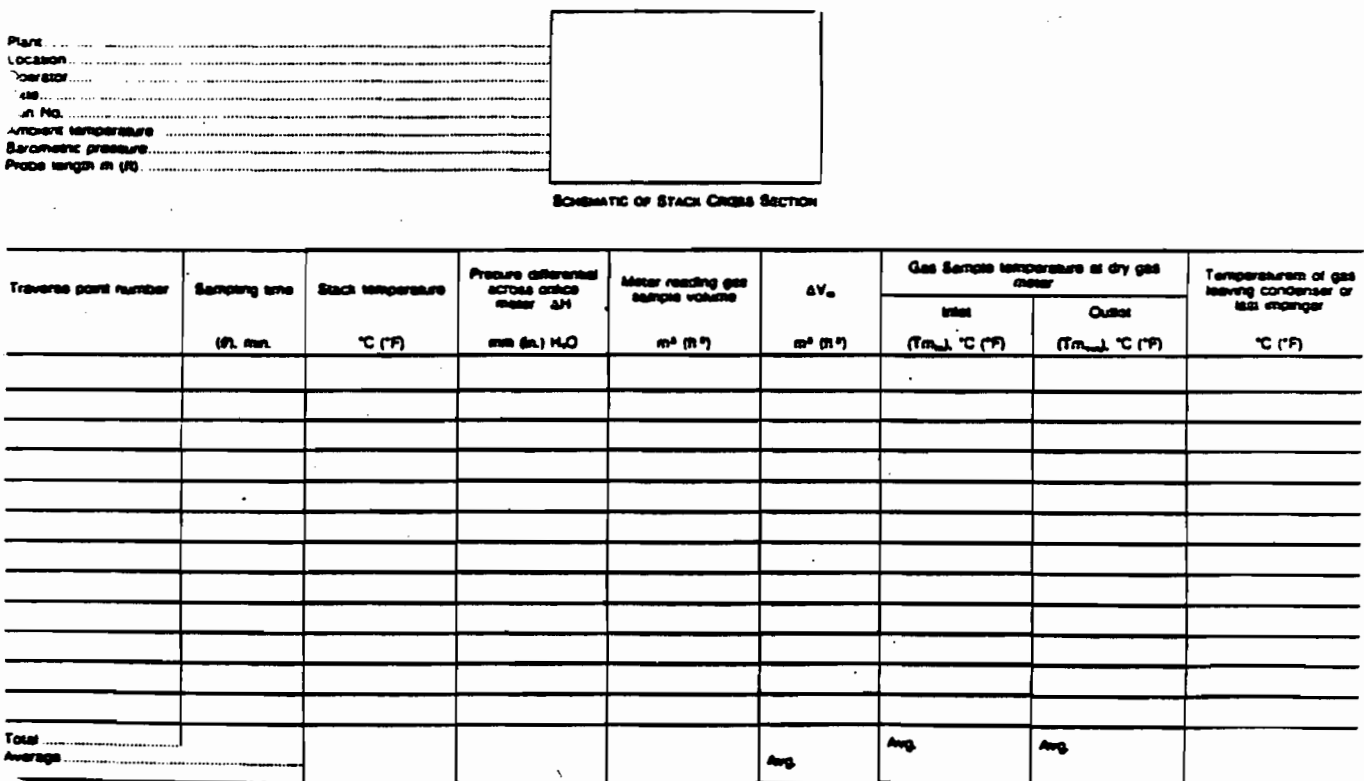
2.2.1 Unless otherwise specified by the Administrator, a minimum of eight traverse points shall be used for circular stacks having diameters less than 0.61 m (24 in.), a minimum of nine points shall be used for rectangular stacks having equivalent diameters less than 0.61 m (24 in.), and a minimum of twelve traverse points shall be used in all other cases. The traverse points shall be located according to Method 1. The use of fewer points is subject to the approval of the Administrator. Select a suitable probe and probe length such that all traverse points can be sampled. Consider sampling from opposite sides of the stack (four total sampling ports) for large stacks, to permit use of shorter probe lengths. Mark the probe with heat resistant tape or by some other method to denote the proper distance into the stack or duct for each sampling point. Place known volumes of water in the first two impingers. Weigh and record the weight of the silica gel to the nearest 0.5 g, and transfer the silica gel to the fourth impinger; alternatively, the silica gel may first be transferred to the impinger, and the weight of the silica gel plus impinger recorded.

2.2.2 Select a total sampling time such that a minimum total gas volume of 0.80 scm (21 scf) will be collected, at a rate no greater than 0.021 m³/min (0.75 cfm). When both moisture content and pollutant emission rate are to be determined, the moisture determination shall be simultaneous with, and for the same total length of time as, the pollutant emission rate run, unless otherwise specified in an applicable subpart of the standards.

2.2.3 Set up the sampling train as shown in Figure 4-1. Turn on the probe heater and (if applicable) the filter heating system to temperatures of about 120° C (248° F), to prevent water condensation ahead of the condenser; allow time for the temperatures to stabilize. Place crushed ice in the ice bath container. It is recommended, but not required, that a leak check be done, as follows: Disconnect the probe from the first impinger or (if applicable) from the filter holder. Plug the inlet to the first impinger (or filter holder) and pull a 380 mm (15 in.) Hg vacuum; a lower vacuum may be used, provided that it is not exceeded during the test. A leakage rate in excess of 4 percent of the average sampling rate or 0.00057 m³/min (0.02 cfm), whichever is less, is unacceptable. Following the leak check, reconnect the probe to the sampling train.

2.2.4 During the sampling run, maintain a sampling rate within 10 percent of constant rate, or as specified by the Administrator. For each run, record the data re-

FIGURE 4-2—FIELD MOISTURE DETERMINATION REFERENCE METHOD



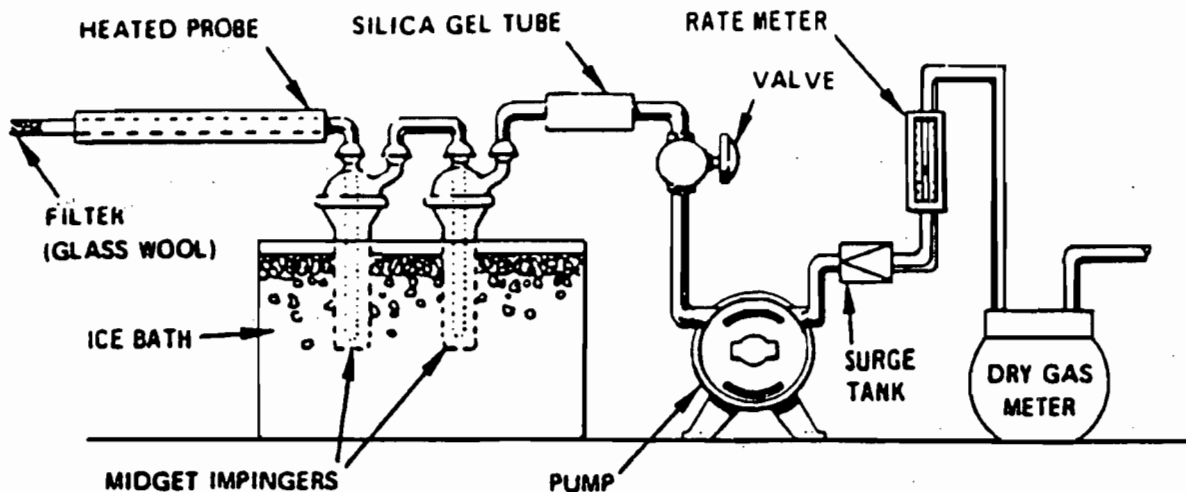


Figure 4-4. Moisture-sampling train - approximation method.

FIGURE 4-5—FIELD MOISTURE DETERMINATION—APPROXIMATION METHOD

Location _____ Comments _____
 Test _____
 Date _____
 Operator _____
 Barometric pressure _____

Clock time	Gas volume through meter, (V _{or}), m ³ (ft ³)	Rate meter reading (m ³ /min) (ft ³ /min)	Meter temperature, C (°F)

3.2.2 Connect the probe, insert it into the stack, and sample at a constant rate of 2 lpm (0.071 cfm). Continue sampling until the dry gas meter registers about 30 liters (1.1 ft³) or until visible liquid droplets are carried over from the first impinger to the second. Record temperature, pressure, and dry gas meter readings as required by Figure 4-5.

3.2.3 After collecting the sample, combine the contents of the two impingers and measure the volume to the nearest 0.5 ml.

3.3 Calculations. The calculation method presented is designed to estimate the moisture in the stack gas; therefore, other data, which are only necessary for accurate moisture determinations, are not collected. The following equations adequately estimate the moisture content, for the purpose of determining isokinetic sampling rate settings.

3.3.1 Nomenclature.

- B_w = Approximate proportion, by volume, of water vapor in the gas stream leaving the second impinger, 0.025.
- B_w = Water vapor in the gas stream, proportion by volume.
- M_w = Molecular weight of water, 18.0 g/g-mole (18.0 lb/lb-mole).

- P_a = Absolute pressure (for this method, same as barometric pressure) at the dry gas meter.
- P_{std} = Standard absolute pressure, 760 mm Hg (29.92 in. Hg).
- R = Ideal gas constant, 0.06236 (mm Hg) (m³)/(g-mole) (°K) for metric units and 21.85 (in. Hg) (ft³)/(lb-mole) (°R) for English units.
- T_a = Absolute temperature at meter, °K (°R).
- T_{std} = Standard absolute temperature, 293° K (528° R).
- V_r = Final volume of impinger contents, ml.
- V_i = Initial volume of impinger contents, ml.

- V_d = Dry gas volume measured by dry gas meter, dcm (dsf).
- V_{d, std} = Dry gas volume measured by dry gas meter, corrected to standard conditions, dscm (dsf).
- V_{w, std} = Volume of water vapor condensed, corrected to standard conditions, scm (scf).
- ρ_w = Density of water, 0.9982 g/ml (0.00220) lb/ml).
- Y = Dry gas meter calibration factor.
- 3.3.2 Volume of water vapor collected, where:

$$V_{w, std} = \frac{(V_r - V_i) M_w RT_{std}}{P_{std} M_w} = K_1 (V_r - V_i)$$

Equation 4-3
 K₁ = 0.001333 m³/ml for metric units
 = 0.04707 ft³/ml for English units.

3.3.3 Gas volume.

$$V_{d, std} = V_d \left(\frac{P_a}{P_{std}} \right) \left(\frac{T_{std}}{T_a} \right) = K_2 \frac{V_d P_a}{T_a}$$

Equation 4-4

where:
 K₁ = 0.3858 °K/mm Hg for metric units
 = 17.64 °R/in. Hg for English units
 3.3.4 Approximate moisture content.

$$B_{w, a} = \frac{V_{w, std}}{V_{d, std} + V_{w, std}} + B_{w, s} = \frac{V_{w, std}}{V_{d, std} + V_{w, std}} + (0.025)$$

Equation 4-7

4. Calibration

4.1 For the reference method, calibrate equipment as specified in the following sections of Method 5: Section 3.3 (metering system); Section 5.3 (temperature gauges); and Section 5.7 (barometer). The recommended leak check of the metering system (Section 5.6 of Method 5) also applies to the reference method. For the approximation method, use the procedures outlined in Section 5.1.1 of Method 6 to calibrate the metering system, and the procedure of Method 5, Section 5.7 to calibrate the barometer.

5. Bibliography

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2. Devortin, Howard, et al. Air Pollution Source Testing Manual. Air Pollution Control District, Los Angeles, Calif. November, 1963.
3. Methods for Determination of Velocity, Volume Dust and Mist Content of Gases. Western Precipitation Division of Joy Manufacturing Co., Los Angeles, Calif. Bulletin WP-50, 1968.

quired on the example data sheet shown in Figure 4-2. Be sure to record the dry gas meter reading at the beginning and end of each sampling time increment and whenever sampling is halted. Take other appropriate readings at each sample point, at least once during each time increment.

2.2.5 To begin sampling, position the probe tip at the first traverse point. Immediately start the pump and adjust the flow to the desired rate. Traverse the cross section, sampling at each traverse point for an equal length of time. Add more ice and, if necessary, salt to maintain a temperature of less 20° C (68° F) at the silica gel outlet.

2.2.6 After collecting the sample, disconnect the probe from the filter holder (or from the first impinger) and conduct a leak check (mandatory) as described in Section 2.2.3. Record the leak rate. If the leakage rate exceeds the allowable rate, the tester shall either reject the test results or shall correct the sample volume as in Section 6.3 of Method 5. Next, measure the volume of the moisture condensed to the nearest ml. Determine the increase in weight of the silica gel (or silica gel plus impinger) to the nearest 0.5 g. Record this information (see example data sheet, Figure 4-3) and calculate the moisture percentage, as described in 2.3 below.

2.2.7 A quality control check of the volume metering system at the field site is suggested before collecting the sample following the procedure in Method 5, Section 4.4.

2.3 Calculations. Carry out the following calculations, retaining at least one extra decimal figure beyond that of the acquired data. Round off figures after final calculation.

FIGURE 4-3—ANALYTICAL DATA—REFERENCE METHOD

	Impinger volume, ml	Silica gel weight, g
Final		
Initial		
Difference		

2.3.1 Nomenclature.

B_v = Proportion of water vapor, by volume, in the gas stream.

M_w = Molecular weight of water, 18.0 g/mole (18.0 lb/lb-mole).

P_a = Absolute pressure (for this method, same as barometric pressure) at the dry gas meter, mm Hg (in. Hg).

P_s = Standard absolute pressure, 760 mm Hg (29.92 in. Hg).

R = Ideal gas constant, 0.08206 (mm Hg) (m³)/(g-mole) (°K) for metric units and 21.85 (in. Hg) (ft³)/(lb-mole) (°R) for English units.

T_a = Absolute temperature at meter, °K (°R).

T_s = Standard absolute temperature, 293° K (528°R).

V_d = Dry gas volume measured by dry gas meter, dcm (dcf).

ΔV_d = Incremental dry gas volume measured by dry gas meter at each traverse point, dcm (dcf).

$V_{d,scm}$ = Dry gas volume measured by the dry gas meter, corrected to standard conditions, dscm (dcscf).

$V_{w,scm}$ = Volume of water vapor condensed corrected to standard conditions, scm (scf).

$V_{w,scm}$ = Volume of water vapor collected in silica gel corrected to standard conditions, scm (scf).

V_f = Final volume of condenser water, ml.

V_i = Initial volume, if any, of condenser water, ml.

W_f = Final weight of silica gel or silica gel plus impinger, g.

W_i = Initial weight of silica gel or silica gel plus impinger, g.

Y = Dry gas meter calibration factor.

ρ_w = Density of water, 0.9982 g/ml (0.002201 lb/ml).

2.3.2 Volume of water vapor condensed.

$$V_{w,scm} = \frac{(V_f - V_i)\rho_w RT_{sc}}{P_{sc} M_w}$$

$$= K_1 (V_f - V_i)$$

Equation 4-1

where:

$K_1 = 0.001333 \text{ m}^3/\text{ml}$ for metric units

$= 0.04707 \text{ ft}^3/\text{ml}$ for English units

2.3.3 Volume of water vapor collected in silica gel.

$$V_{w,scm} = \frac{(W_f - W_i) RT_{sc}}{P_{sc} M_w}$$

$$= K_2 (W_f - W_i)$$

Equation 4-2

where:

$K_2 = 0.001335 \text{ m}^3/\text{g}$ for metric units

$= 0.04715 \text{ ft}^3/\text{g}$ for English units

2.3.4 Sample gas volume.

$$V_{d,scm} = V_d Y \frac{(P_a)(T_{sc})}{(P_{sc})(T_a)}$$

$$= K_3 Y \frac{V_d P_a}{T_a}$$

Equation 4-3

where:

$K_3 = 0.3858 \text{ } ^\circ\text{K}/\text{mm Hg}$ for metric units

$= 17.64 \text{ } ^\circ\text{R}/\text{in. Hg}$ for English units

NOTE: If the post-test leak rate (Section 2.2.8) exceeds the allowable rate, correct the value of V_d in Equation 4-3, as described in Section 6.3 of Method 5.

2.3.5 Moisture Content.

$$B_v = \frac{V_{w,scm} + V_{d,scm}}{V_{w,scm} + V_{d,scm} + V_{d,scm}}$$

Equation 4-4

NOTE: In saturated or moisture droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one using a value based upon the saturated conditions (see Section 1.2), and

another based upon the results of the impinger analysis. The lower of these two values of B_v shall be considered correct.

2.3.6 Verification of constant sampling rate. For each time increment, determine the ΔV_d . Calculate the average. If the value for any time increment differs from the average by more than 10 percent, reject the results and repeat the run.

3. Approximation Method

The approximation method described below is presented only as a suggested method (see Section 1.2).

3.1 Apparatus.

3.1.1 Probe. Stainless steel glass tubing, sufficiently heated to prevent water condensation and equipped with a filter (either in-stack or heated out-stack) to remove particulate matter. A plug of glass wool inserted into the end of the probe, is a satisfactory filter.

3.1.2 Impingers. Two midjet impingers, each with 30 ml capacity, or equivalent.

3.1.3 Ice Bath. Container and ice, to aid in condensing moisture in impingers.

3.1.4 Drying Tube. Tube packed with new or regenerated 6- to 16-mesh indicating-type silica gel (or equivalent desiccant), to dry the sample gas and to protect the meter and pump.

3.1.5 Valve. Needle valve, to regulate the sample gas flow rate.

3.1.6 Pump. Leak-free, diaphragm type, or equivalent, to pull the gas sample through the train.

3.1.7 Volume Meter. Dry gas meter, sufficiently accurate to measure the sample volume within 2%, and calibrated over the range of flow rates and conditions actually encountered during sampling.

3.1.8 Rate Meter. Rotameter, to measure the flow range from 0 to 3 lpm (0 to 0.11 cfm).

3.1.9 Graduated Cylinder. 25 ml.

3.1.10 Barometer. Mercury, aneroid, or other barometer, as described in Section 2.1.5 above.

3.1.11 Vacuum Gauge. At least 760 mm Hg (30 in. Hg) gauge, to be used for the sampling leak check.

3.2 Procedure.

3.2.1 Place exactly 5 ml distilled water in each impinger.

Leak check the sampling train as follows: Temporarily insert a vacuum gauge at or near the probe inlet; then, plug the probe inlet and pull a vacuum of at least 250 mm Hg (10 in. Hg). Note the time rate of change of the dry gas meter dial; alternatively, a rotameter (0-60 cc/min) may be temporarily attached to the dry gas meter outlet to determine the leakage rate. A leak rate not in excess of 2 percent of the average sampling rate is acceptable.

NOTE: Carefully release the probe inlet plug before turning off the pump.

LOCATION _____ COMMENTS _____
 TEST _____
 DATE _____
 OPERATOR _____
 BAROMETRIC PRESSURE _____

CLOCK TIME	GAS VOLUME THROUGH METER, (Vm), m ³ (ft ³)	RATE METER SETTING m ³ /min. (ft ³ /min.)	METER TEMPERATURE, °C (°F)

Figure 4-5. Field moisture determination - approximation method.

3.2.2 Connect the probe, insert it into the stack, and sample at a constant rate of 2 lpm (0.071 cfm). Continue sampling until the dry gas meter registers about 30 liters (1.1 ft³) or until visible liquid droplets are carried over from the first impinger to the second. Record temperature, pressure, and dry gas meter readings as required by Figure 4-5.

3.2.3 After collecting the sample, combine the contents of the two impingers and measure the volume to the nearest 0.5 ml.

3.3 Calculations. The calculation method presented is designed to estimate the moisture in the stack gas; therefore, other data, which are only necessary for accurate moisture determinations, are not collected. The following equations adequately estimate the moisture content, for the purpose of determining isokinetic sampling rate settings.

3.3.1 Nomenclature.

B_{vol} - Approximate proportion, by volume, of water vapor in the gas stream leaving the second impinger, 0.025.

B_{wt} - Water vapor in the gas stream, proportion by volume.

M_w - Molecular weight of water, 18.0 g/mole (18.0 lb/lb-mole).

P_a - Absolute pressure (for this method, same as barometric pressure) at the dry gas meter.

P_{std} - Standard absolute pressure, 760 mm Hg (29.92 in. Hg).

R - Ideal gas constant, 0.08206 (mm Hg) (m³)/(g-mole) (°K) for metric units and 21.85 (in. Hg) (ft³)/(lb-mole) (°R) for English units.

T_m - Absolute temperature at meter, °K (°R).

T_{std} - Standard absolute temperature, 293°K (528°R).

V_f - Final volume of impinger contents, ml.

V_i - Initial volume of impinger contents, ml.

V_{dm} - Dry gas volume measured by dry gas meter, dcm (def).

V_{dscm} - Dry gas volume measured by dry gas meter, corrected to standard conditions, dscm (dscf).

V_{wscm} - Volume of water vapor condensed, corrected to standard conditions, scm (scf).

ρ_w - Density of water, 0.9982 g/ml (0.002201 lb/ml).

Y - Dry gas meter calibration factor.

3.3.2 Volume of water vapor collected, where:

$$V_{wscm} = \frac{(V_f - V_i) \rho_w R T_{std}}{P_a M_w} = K_1 (V_f - V_i)$$

Equation 4-5

$K_1 = 0.001333$ m³/ml for metric units

$= 0.04707$ ft³/ml for English units.

3.3.3 Gas volume.

$$V_{dscm} = V_{dm} \left(\frac{P_a}{P_{std}} \right) \left(\frac{T_{std}}{T_m} \right) = K_2 \frac{V_{dm} P_a}{T_m}$$

Equation 4-4

where:

$K_2 = 0.3858$ °K/mm Hg for metric units

$= 17.64$ °R/in. Hg for English units

3.3.4 Approximate moisture content.

$$B_{wt} = \frac{V_{wscm}}{V_{dscm} + V_{wscm}} + B_{vol} = \frac{V_{wscm}}{V_{dscm} + V_{wscm}} + (0.025)$$

Equation 4-

4. Calibration

4.1 For the reference method, calibrate equipment as specified in the following sections of Method 5: Section 5.3 (metering system); Section 5.5 (temperature gauges); and Section 5.7 (barometer). The recommended leak check of the metering system (Section 5.6 of Method 5) also applies to the reference method. For the approximation method, use the procedures outlined in Section 5.1.1 of Method 6 to calibrate the metering system, and the procedure of Method 5, Section 5.7 to calibrate the barometer.

5. Bibliography

1. Air Pollution Engineering Manual (Second Edition). Danielson, J. A. (ed.). U.S. Environmental Protection Agency, Office of Air Quality Planning and Standards, Research Triangle Park, N.C. Publication No. AP-40, 1973.

2. Devorkin, Howard, et al. Air Pollution Source Testing Manual. Air Pollution Control District, Los Angeles, Calif. November, 1963.

3. Methods for Determination of Velocity, Volume Dust and Mist Content of Gases, Western Precipitation Division of Joy Manufacturing Co., Los Angeles, Calif. Bulletin WP-50, 1968.

48 FR 55670 12/14/83

METHOD 5—DETERMINATION OF PARTICULATE EMISSIONS FROM STATIONARY SOURCES

1. Principle and Applicability

1.1 Principle. Particulate matter is withdrawn isokinetically from the source and collected on a glass fiber filter maintained at a temperature in the range of 120±14° C (248±25° F) or such other temperature as specified by an applicable subpart of the standards or approved by Administrator.

U.S. Environmental Protection Agency, for a particular application. The particulate mass, which includes any material that condenses at or above the filtration temperature, is determined gravimetrically after removal of uncombined water.

1.2 Applicability. This method is applicable for the determination of particulate emissions from stationary sources.

2. Apparatus

2.1 Sampling Train. A schematic of the sampling train used in this method is shown in Figure 5-1. Complete construction details are given in APTD-0581 (Citation 2 in Bibliography); commercial models of this train are also available. For changes from APTD-0581 and for allowable modifications of the train shown in Figure 5-1, see the following subsections.

The operating and maintenance procedures for the sampling train are described in APTD-0576 (Citation 3 in Bibliography). Since correct usage is important in obtaining valid results, all users should read APTD-0576 and adopt the operating and maintenance procedures outlined in it, unless otherwise specified herein. The sampling train consists of the following components:

2.1.1 Probe Nozzle. Stainless steel (316) or glass with sharp, tapered leading edge. The angle of taper shall be 30° and the taper shall be on the outside to preserve a constant internal diameter. The probe nozzle shall be of the button-hook or elbow design, unless otherwise specified by the Administrator. If made of stainless steel, the nozzle shall be constructed from seamless tubing; other materials of construction may be used, subject to the approval of the Administrator.

A range of nozzle sizes suitable for isokinetic sampling should be available, e.g., 0.32 to 1.27 cm (1/8 to 1/2 in.)—or larger if higher volume sampling trains are used—inside diameter (ID) nozzles in increments of 0.16 cm (5/16 in.). Each nozzle shall be calibrated according to the procedures outlined in Section 5.

2.1.2 Probe Liner. Borosilicate or quartz glass tubing with a heating system capable of maintaining a gas temperature at the exit end during sampling of 120±14° C (248±25° F), or such other temperature as specified by an applicable subpart of the standards or approved by the Administrator for a particular application. (The tester may opt to operate the equipment at a temperature lower than that specified.) Since the actual temperature at the outlet of the probe is not usually monitored during sampling, probes constructed according to APTD-0581 and utilizing the calibration curves of APTD-0576 (or calibrated according to the procedure outlined in APTD-0576) will be considered acceptable.

Either borosilicate or quartz glass probe liners may be used for stack temperatures up to about 480° C (900° F); quartz liner shall be used for temperatures between 48 and 900° C (900 and 1,650° F). Both types of liners may be used at higher temperatures than specified for short periods of time, subject to the approval of the Administrator. The softening temperature for borosilicate is 820° C (1,508° F), and for quartz it is 1,500° C (2,732° F).

Whenever practical, every effort should be made to use borosilicate or quartz glass probe liners. Alternatively, metal liners (e.g., 316 stainless steel, Incoloy 825, or other corrosion resistant metals) made of seamless tubing may be used, subject to the approval of the Administrator.

2.1.3 Pitot Tube. Type S, as described in Section 2.1 of Method 2, or other device approved by the Administrator. The pitot tube shall be attached to the probe (as shown in Figure 5-1) to allow constant monitoring of the stack gas velocity. The impact (high pressure) opening plane of the pitot tube shall be even with or above the nozzle entry plane (see Method 2, Figure 2-6b) during sampling. The Type S pitot tube assembly shall have a known coefficient, determined as outlined in Section 4 of Method 2.

*Mention of trade names or specific product does not constitute endorsement by the Environmental Protection Agency.

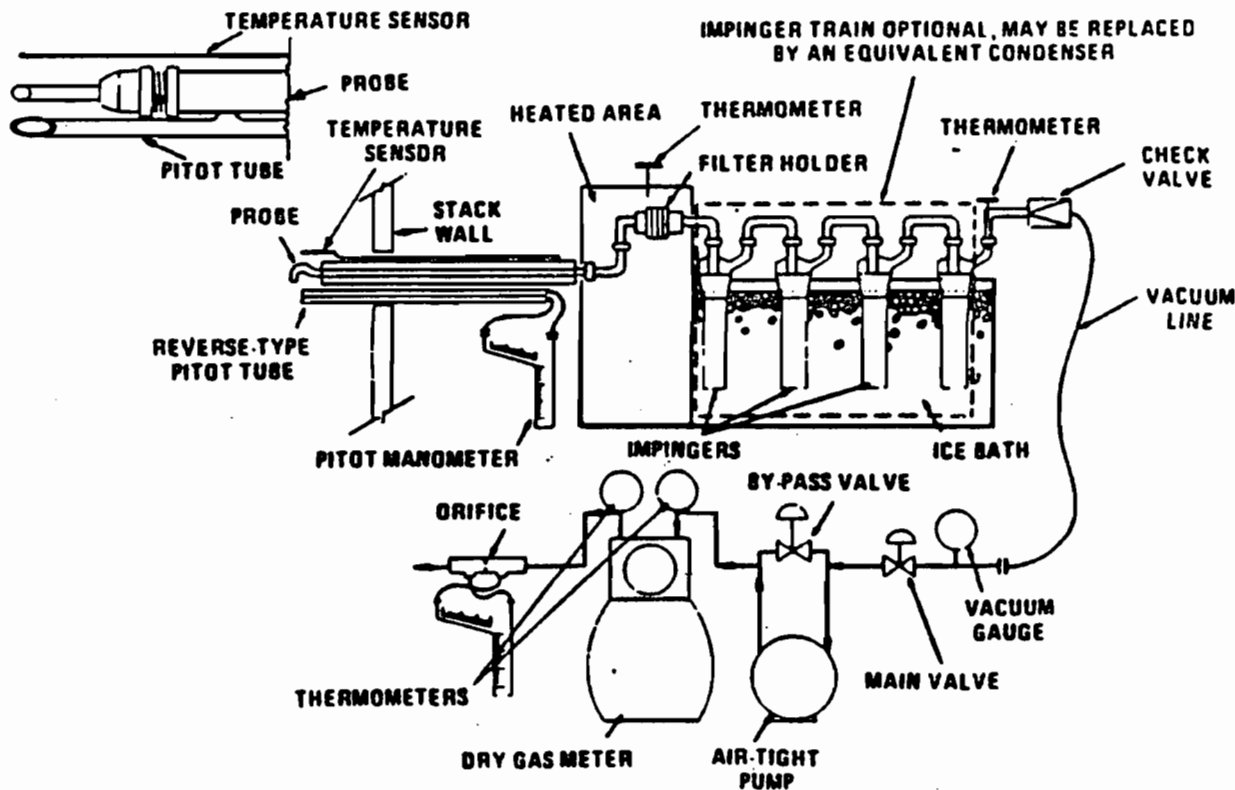


Figure 5-1. Particulate sampling train.

2.1.4 **Differential Pressure Gauge.** Inclined manometer or equivalent device (two), as described in Section 2.2 of Method 2. One manometer shall be used for velocity head (Δp) readings, and the other, for orifice differential pressure readings.

2.1.5 **Filter Holder.** Borosilicate glass, with a glass frit filter support and a silicone rubber gasket. Other materials of construction (e.g., stainless steel, Teflon, Viton) may be used, subject to approval of the Administrator. The holder design shall provide a positive seal against leakage from the outside or around the filter. The holder shall be attached immediately at the outlet of the probe (or cyclone, if used).

2.1.6 **Filter Heating System.** Any heating system capable of maintaining a temperature around the filter holder during sampling of $120 \pm 14^\circ \text{C}$ ($248 \pm 25^\circ \text{F}$), or such other temperature as specified by an applicable subpart of the standards or approved by the Administrator for a particular application. Alternatively, the tester may opt to operate the equipment at a temperature lower than that specified. A temperature gauge capable of measuring temperature to within 3°C (5.4°F) shall be installed so that the temperature around the filter holder can be regulated and monitored during sampling. Heating systems other than the one shown in APTD-0581 may be used.

2.1.7 **Condenser.** The following system shall be used to determine the stack gas moisture content: Four impingers connected in series with leak-free ground glass fittings or any similar leak-free non-contaminating fittings. The first, third, and fourth impingers shall be of the Greenburg-Smith design, modified by replacing the tip with 1.3 cm ($\frac{1}{2}$ in.) ID glass tube extending to about 1.3 cm ($\frac{1}{2}$ in.) from the bottom of the flask. The second impinger shall be of the Greenburg-Smith design with the standard tip. Modifications (e.g., using flexible connections between the impingers, using materials other than glass, or using flexible vacuum lines to connect the filter holder to the condenser) may be used, subject to the approval of the Administrator. The first and second impingers shall contain known quantities of water (Section 4.1.3), the third shall be empty, and the fourth shall contain a known weight of silica gel, or equivalent desiccant. A thermometer, capable of measuring temperature to within 1°C (2°F) shall be placed at the outlet of the fourth impinger for monitoring purposes.

Alternatively, any system that cools the sample gas stream and allows measurement of the water condensed and moisture leaving the condenser, each to within 1 ml or 1 g may be used, subject to the approval of the Administrator. Acceptable means are to measure the condensed water either gravimetrically or volumetrically and to measure the moisture leaving the condenser by: (1) monitoring the temperature and pressure at the exit of the condenser and using Dalton's law of partial pressures; or (2) passing the sample gas stream through a tared silica gel (or equivalent desiccant) trap with exit gases kept below 20°C (68°F) and determining the weight gain.

If means other than silica gel are used to determine the amount of moisture leaving the condenser, it is recommended that silica gel (or equivalent) still be used between the condenser system and pump to prevent moisture condensation in the pump and metering devices and to avoid the need to make corrections for moisture in the metered volume.

NOTE If a determination of the particulate matter collected in the impingers is desired in addition to moisture content, the impinger system described above shall be used, without modification. Individual States or control agencies requiring this information shall be contacted as to the sample recovery and analysis of the impinger contents.

2.1.8 **Metering System.** Vacuum gauge, leak-free pump, thermometers capable of measuring temperature to within 3°C (5.4°F), dry gas meter capable of measuring volume to within 2 percent, and related equipment, as shown in Figure 5-1. Other metering systems capable of maintaining sampling rates within 10 percent of isokinetic and of determining sample volumes to within 2 percent may be used, subject to the approval of the Administrator. When the metering system is used in conjunction with a pitot tube, the system shall enable checks of isokinetic rates.

Sampling trains utilizing metering systems designed for higher flow rates than that described in APTD-0581 or APDT-0578 may be used provided that the specifications of this method are met.

2.1.9 **Barometer.** Mercury aneroid, or other barometer capable of measuring atmospheric pressure to within 2.5 mm Hg (0.1 in. Hg). In many cases the barometric reading may be obtained from a nearby national weather service station, in which case the station value (which is the absolute barometric pressure) shall be requested and an adjustment for elevation differences between the weather station and sampling point shall be applied at a rate of minus 2.5 mm Hg (0.1 in. Hg) per 30 m (100 ft) elevation increase or vice versa for elevation decrease.

2.1.10 **Gas Density Determination Equipment.** Temperature sensor and pressure gauge, as described in Sections 2.3 and 2.4 of Method 2, and gas analyzer, if necessary, as described in Method 3. The temperature sensor shall, preferably, be permanently attached to the pitot tube or sampling probe in a fixed configuration, such that the tip of the sensor extends beyond the leading edge of the probe sheath and does not touch any metal. Alternatively, the sensor may be attached just prior to use in the field. Note, however, that if the temperature sensor is attached in the field, the sensor must be placed in an interference-free arrangement with respect to the Type S pitot tube openings (see Method 2, Figure 2-7). As a second alternative, if a difference of not more than 1 percent in the average velocity measurement is to be introduced, the temperature gauge need not be attached to the probe or pitot tube. (This alternative is subject to the approval of the Administrator.)

2.2 **Sample Recovery.** The following items are needed.

2.2.1 **Probe-Liner and Probe-Nomadic Brushes.** Nylon bristle brushes with stainless steel wire handles. The probe brush shall have extensions (at least as long as the probe) of stainless steel, Nylon, Teflon, or similarly inert material. The brushes shall be properly sized and shaped to brush out the probe liner and nozzle.

2.2.2 **Wash Bottles—Two.** Glass wash bottles are recommended; polyethylene wash bottles may be used at the option of the tester. It is recommended that acetone not be stored in polyethylene bottles for longer than a month.

2.2.3 **Glass Sample Storage Containers.** Chemically resistant, borosilicate glass bottles, for acetone washes, 500 ml or 1000 ml. Screw cap liners shall either be rubber-backed Teflon or shall be constructed so as to be leak-free and resistant to chemical attack by acetone. (Narrow mouth glass bottles have been found to be less prone to leakage.) Alternatively, polyethylene bottles may be used.

2.2.4 **Petri Dishes.** For filter samples, glass or polyethylene, unless otherwise specified by the Administrator.

2.2.5 **Graduated Cylinder and/or Balance.** To measure condensed water to within 1 ml or 1 g. Graduated cylinders shall have subdivisions no greater than 2 ml. Most laboratory balances are capable of weighing to the nearest 0.5 g or less. Any of these balances is suitable or use here and in Section 2.3.4.

2.2.6 **Plastic Storage Containers.** Airtight containers to store silica gel.

2.2.7 **Funnel and Rubber Policeman.** To aid in transfer of silica gel to container; not

2.2.8 **Funnel, Glass or polyethylene.** To aid in sample recovery.

2.2 **Analysis.** For analysis, the following equipment is needed.

2.3.1 **Glass Weighing Dishes.**

2.3.2 **Desiccator.**

2.3.3 **Analytical Balance.** To measure to within 0.1 mg.

2.3.4 **Balance.** To measure to within 0.5 g.

2.3.5 **Beakers.** 250 ml.

2.3.6 **Hygrometer.** To measure the relative humidity of the laboratory environment.

2.3.7 **Temperature Gauge.** To measure the temperature of the laboratory environment.

3. Reagents

3.1 **Sampling.** The reagents used in sampling are as follows:

3.1.1 **Filters.** Glass fiber filters, without organic binder, exhibiting at least 99.95 percent efficiency (<0.05 percent penetration) on 0.3-micron dioctyl phthalate smoke particles. The filter efficiency test shall be conducted in accordance with ASTM standard method D2906-71 (Reapproved 1978) (incorporated by reference—see § 60.17). Test data from the supplier's quality control program are sufficient for this purpose. In sources containing SO_2 or SO_3 , the filter material must be of a type that is unreactive to SO_2 or SO_3 . Citation 10 in Section 7 Bibliography, may be used to select the appropriate filter.

3.1.2 **Silica Gel.** Indicating type, 6 to 16 mesh. If previously used, dry at 175°C (350°F) for 2 hours. New silica gel may be used as received. Alternatively, other types of desiccants (equivalent or better) may be used, subject to the approval of the Administrator.

3.1.3 **Water.** When analysis of the material caught in the impingers is required, distilled water shall be used. Run blanks prior to field use to eliminate a high blank on test samples.

3.1.4 **Crushed Ice.**

3.1.5 **Stopcock Grease.** Acetone-insoluble, heat-stable silicone grease. This is not necessary if screw-on connectors with Teflon sleeves, or similar, are used. Alternatively, other types of stopcock grease may be used, subject to the approval of the Administrator.

3.2 Sample Recovery. Acetone-reagent grade, <0.001 percent residue, in glass bottles—is required. Acetone from metal containers generally has a high residue blank and should not be used. Sometimes, suppliers transfer acetone to glass bottles from metal containers; thus, acetone blanks shall be run prior to field use and only acetone with low blank values (<0.001 percent) shall be used. In no case shall a blank value of greater than 0.001 percent of the weight of acetone used be subtracted from the sample weight.

3.3 Analysis. Two reagents are required for the analysis:

3.3.1 Acetone. Same as 3.2.

3.3.2 Desiccant. Anhydrous calcium sulfate, indicating type. Alternatively, other types of desiccants may be used, subject to the approval of the Administrator.

4. Procedure

4.1 Sampling. The complexity of this method is such that, in order to obtain reliable results, testers should be trained and experienced with the test procedures.

4.1.1 Pretest Preparation. It is suggested that sampling equipment be maintained according to the procedures described in APTD-0576.

Weigh several 200 to 300 g portions of silica gel in air-tight containers to the nearest 0.5 g. Record the total weight of the silica gel plus container, on each container. As an alternative, the silica gel need not be preweighed, but may be weighed directly in the impinger or sampling holder just prior to train assembly.

Check filters visually against light for irregularities and flaws or pinhole leaks. Label filters of the proper diameter on the back side near the edge using numbering machine ink. As an alternative, label the shipping containers (glass or plastic petri dishes) and keep the filters in these containers at all times except during sampling and weighing.

Desiccate the filters at $20 \pm 1.6^\circ \text{C}$ ($68 \pm 10^\circ \text{F}$) and ambient pressure for at least 24 hours and weigh at intervals of at least 6 hours to a constant weight, i.e., 0.5 mg change from previous weighing; record results to the nearest 0.1 mg. During each weighing the filter must not be exposed to the laboratory atmosphere for a period greater than 2 minutes and a relative humidity above 50 percent. Alternatively (unless otherwise specified by the Administrator), the filters may be oven dried at 105°C (220°F) for 2 to 3 hours, desiccated for 2 hours, and weighed. Procedures other than those described, which account for relative humidity effects, may be used, subject to the approval of the Administrator.

4.1.2 Preliminary Determinations. Select the sampling site and the minimum number of sampling points according to Method 1 or as specified by the Administrator. Determine the stack pressure, temperature, and the range of velocity heads using Method 2; it is recommended that a leak-check of the pitot lines (see Method 2, Section 3.1) be performed. Determine the moisture content using Approximation Method 4 or its alternatives for the purpose of making isokinetic sampling rate settings. Determine the stack

gas dry molecular weight, as described in Method 2, Section 3.6; if integrated Method 3 sampling is used for molecular weight determination, the integrated bag sample shall be taken simultaneously with, and for the same total length of time as, the particulate sample run.

Select a nozzle size based on the range of velocity heads, such that it is not necessary to change the nozzle size in order to maintain isokinetic sampling rates. During the run, do not change the nozzle size. Ensure that the proper differential pressure gauge is chosen for the range of velocity heads encountered (see Section 2.2 of Method 2).

Select a suitable probe liner and probe length such that all traverse points can be sampled. For large stacks, consider sampling from opposite sides of the stack to reduce the length of probes.

Select a total sampling time greater than or equal to the minimum total sampling time specified in the test procedures for the specific industry such that (1) the sampling time per point is not less than 2 min (or some greater time interval as specified by the Administrator), and (2) the sample volume taken (corrected to standard conditions) will exceed the required minimum total gas sample volume. The latter is based on an approximate average sampling rate.

It is recommended that the number of minutes sampled at each point be an integer or an integer plus one-half minute, in order to avoid timekeeping errors. The sampling time at each point shall be the same.

In some circumstances, e.g., batch cycles, it may be necessary to sample for shorter times at the traverse points and to obtain smaller gas sample volumes. In these cases, the Administrator's approval must first be obtained.

4.1.3 Preparation of Collection Train. During preparation and assembly of the sampling train, keep all openings where contamination can occur covered until just prior to assembly or until sampling is about to begin.

Place 100 ml of water in each of the first two impingers, leave the third impinger empty, and transfer approximately 200 to 300 g of preweighed silica gel from its container to the fourth impinger. More silica gel may be used, but care should be taken to ensure that it is not entrained and carried out from the impinger during sampling. Place the container in a clean place for later use in the sample recovery. Alternatively, the weight of the silica gel plus impinger may be determined to the nearest 0.5 g and recorded.

Using a tweezer or clean disposable surgical gloves, place a labeled (identified) and weighed filter in the filter holder. Be sure that the filter is properly centered and the gasket properly placed so as to prevent the sample gas stream from circumventing the filter. Check the filter for tears after assembly is completed.

When glass liners are used, install the selected nozzle using a Viton A O-ring when stack temperatures are less than 260°C (500°F) and an asbestos string gasket when temperatures are higher. See APTD-0576 for details. Other connecting systems using either 316 stainless steel or Teflon ferrules may be used. When metal liners are used, install the nozzle as above or by a leak-free direct mechanical connection. Mark the probe with heat resistant tape or by some other method to denote the proper distance into the stack or duct for each sampling point.

Set up the train as in Figure 5-1, using (if necessary) a very light coat of silicone grease on all ground glass joints, greasing only the outer portion (see APTD-0576) to avoid possibility of contamination by the silicone grease. Subject to the approval of the Administrator, a glass cyclone may be used between the probe and filter holder when the total particulate catch is expected to exceed 100 mg or when water droplets are present in the stack gas.

Place crushed ice around the impingers.

4.1.4 Leak-Check Procedures.

4.1.4.1 Pretest Leak-Check. A pretest leak-check is recommended, but not required. If the tester opts to conduct the pretest leak-check, the following procedure shall be used.

After the sampling train has been assembled, turn on and set the filter and probe heating systems at the desired operating temperatures. Allow time for the temperatures to stabilize. If a Viton A O-ring or other leak-free connection is used in assembling the probe nozzle to the probe liner, leak-check the train at the sampling site by plugging the nozzle and pulling a 380 mm Hg (15 in. Hg) vacuum.

NOTE: A lower vacuum may be used, provided that it is not exceeded during the test.

If an asbestos string is used, do not connect the probe to the train during the leak-check. Instead, leak-check the train by first plugging the inlet to the filter holder (cyclone, if applicable) and pulling a 380 mm Hg (15 in. Hg) vacuum (see Note immediately above). Then connect the probe to the train and leak-check at about 25 mm Hg (1 in. Hg) vacuum; alternatively, the probe may be leak-checked with the rest of the sampling train, in one step, at 380 mm Hg (15 in. Hg) vacuum. Leakage rates in excess of 4 percent of the average sampling rate or $0.00057 \text{ m}^3/\text{min}$ (0.02 cfm), whichever is less, are unacceptable.

The following leak-check instructions for the sampling train described in APTD-0576 and APTD-0581 may be helpful. Start the pump with bypass valve fully open and coarse adjust valve, completely closed. Partially open the coarse adjust valve and slowly close the bypass valve until the desired vacuum is reached. Do not reverse direction of bypass valve; this will cause water to back up into the filter holder. If the desired vacuum is exceeded, either leak-check at this higher vacuum or end the leak-check as shown below and start over.

When the leak-check is completed, first slowly remove the plug from the inlet to the probe, filter holder, or cyclone (if applicable) and immediately turn off the vacuum pump. This prevents the water in the impingers from being forced backward into the filter holder and silica gel from being entrained backward into the third impinger.

4.1.4.2 Leak-Checks During Sample Run. If, during the sampling run, a component (e.g., filter assembly or impinger) change becomes unnecessary, a leak-check shall be conducted immediately before the change is made. The leak-check shall be done according to the procedure outlined in Section 4.1.4.1 above, except that it shall be done at a vacuum equal to or greater than the maximum value recorded up to that point in the test. If the leakage rate is found to be no greater than $0.00057 \text{ m}^3/\text{min}$ (0.02 cfm) or 4 percent of the average sampling rate (whichever is less), the results are acceptable, and no correction will need to be applied to the total volume of dry gas metered:

if, however, a higher leakage rate is obtained, the tester shall either record the leakage rate and plan to correct the sample volume as shown in Section 6.3 of this method, or shall void the sampling run.

Immediately after component changes, leak-checks are optional; if such leak-checks are done, the procedure outlined in Section 4.1.4.1 above shall be used.

4.1.4.3 Post-test Leak-Check. A leak-check is mandatory at the conclusion of each sampling run. The leakcheck shall be done in accordance with the procedures outlined in Section 4.1.4.1, except: that it shall be conducted at a vacuum equal to or greater than the maximum value reached during the sampling run. If the leakage rate is found to be no greater than 0.00057 m³/min (0.02 cfm) or 4 percent of the average sampling rate (whichever is less), the results are acceptable, and no correction need be applied to the total volume of dry gas metered.

If, however, a higher leakage rate is obtained, the tester shall either record the leakage rate and correct the sample volume as shown in Section 6.3 of this method, or shall void the sampling run.

4.1.5 Particulate Train Operation. During the sampling run, maintain an isokinetic sampling rate (within 10 percent of true isokinetic unless otherwise specified by the Administrator) and a temperature around the filter of 120±14° C (248±25° F), or such other temperature as specified by an applicable subpart of the standards or approved by the Administrator.

For each run, record the data required on a data sheet such as the one shown in Figure 5-2. Be sure to record the initial dry gas meter reading. Record the dry gas meter readings at the beginning and end of each sampling time increment, when changes in flow rates are made, before and after each leak-check, and when sampling is halted.

Take other readings required by Figure 5-2 at least once at each sample point during each time increment and additional readings when significant changes (20 percent variation in velocity head readings) necessitate additional adjustments in flow rate. Level and zero the manometer. Because the manometer level and zero may drift due to vibrations and temperature changes, make periodic checks during the traverse.

Clean the portholes prior to the test run to minimize the chance of sampling deposited material. To begin sampling, remove the nozzle cap, verify that the filter and probe heating systems are up to temperature, and that the pitot tube and probe are properly positioned. Position the nozzle at the first traverse point with the tip pointing directly into the gas stream. Immediately start the pump and adjust the flow to isokinetic conditions. Nomographs are available, which aid in the rapid adjustment of the isokinetic sampling rate without excessive computations. These nomographs are designed for use when the Type S pitot tube coefficient is 0.85±0.02, and the stack gas equivalent density (dry molecular weight) is equal to 29±4. APTD-0376 details the procedure for using the nomographs. If C_p and M_g are outside the above stated ranges do not use the nomographs unless appropriate steps (see Citation 7 in Bibliography) are taken to compensate for the deviations.

When the stack is under significant negative pressure (height of impinger stem), take care to close the coarse adjust valve before inserting the probe into the stack to prevent water from backing into the filter holder. If necessary, the pump may be turned on with the coarse adjust valve closed.

When the probe is in position, block off the openings around the probe and porthole to prevent unrepresentative dilution of the gas stream.

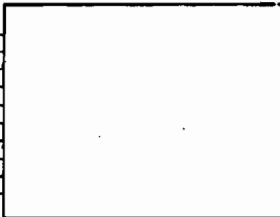
Traverse the stack cross-section, as required by Method 1 or as specified by the Administrator, being careful not to bump the probe nozzle into the stack walls when sampling near the walls or when removing or inserting the probe through the portholes; this minimizes the chance of extracting deposited material.

During the test run, make periodic adjustments to keep the temperature around the filter holder at the proper level; add more ice and, if necessary, salt to maintain a temperature of less than 20° C (68° F) at the condenser/silica gel outlet. Also, periodically check the level and zero of the manometer.

If the pressure drop across the filter becomes too high, making isokinetic sampling difficult to maintain, the filter may be replaced in the midst of a sample run. It is recommended that another complete filter assembly be used rather than attempting to change the filter itself. Before a new filter assembly is installed, conduct a leak-check (see Section 4.1.4.2). The total particulate weight shall include the summation of all filter assembly catches.

A single train shall be used for the entire sample run, except in cases where simultaneous sampling is required in two or more separate ducts or at two or more different locations within the same duct, or, in cases where equipment failure necessitates a change of trains. In all other situations, the use of two or more trains will be subject to the approval of the Administrator.

FIGURE 5-2—PARTICULATE FIELD DATA

Plant _____ Location _____ Operator _____ Date _____ Run No. _____ Sample box No. _____ Meter box No. _____ Meter Size _____ C factor _____ Pitot tube coefficient, C_p _____	 <p style="text-align: center; margin-top: 5px;">SCHEMATIC OF STACK CROSS SECTION</p>	Ambient temperature _____ Barometric pressure _____ Assumed moisture, % _____ Probe length, ft (m) _____ Nozzle identification No. _____ Average calibrated nozzle diameter, cm (in) _____ Probe heater setting _____ Leak rate, m ³ /min (cfm) _____ Probe liner material _____ Stack pressure, mm Hg (in. Hg) _____ Filter No. _____
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Traverse point number	Sampling time <small>(min. sec.)</small>	Vacuum <small>mm Hg (in. Hg)</small>	Stack temperature <small>(T_g °C (°F))</small>	Velocity head <small>(A.P.) mm (in.) H₂O</small>	Pressure differential across orifice meter <small>mm H₂O (in. H₂O)</small>	Gas sample volume <small>m³ (ft³)</small>	Gas sample temperature at dry gas meter		Filter holder temperature <small>°C (°F)</small>	Temperature of gas leaving condenser or last impinger <small>°C (°F)</small>
							Inlet <small>°C (°F)</small>	Outlet <small>°C (°F)</small>		
Total							Avg.	Avg.		
Average							Avg.	Avg.		

Note that when two or more trains are used, separate analyses of the front-half and (if applicable) impinger catches from each train shall be performed, unless identical nozzle sizes were used on all trains, in which case, the front-half catches from the individual trains may be combined (as may the impinger catches) and one analysis of front-half catch and one analysis of impinger catch may be performed. Consult with the Administrator for details concerning the calculation of results when two or more trains are used.

At the end of the sample run, turn off the coarse adjust valve, remove the probe and nozzle from the stack, turn off the pump, record the final dry gas meter reading, and conduct a post-test leak-check, as outlined in Section 4.1.4.3. Also, leak-check the pitot lines as described in Method 2, Section 3.1; the lines must pass this leak-check, in order to validate the velocity head data.

4.1.6 Calculation of Percent Isokinetic. Calculate percent isokinetic (see Calculations, Section 6) to determine whether the run was valid or another test run should be made. If there was difficulty in maintaining isokinetic rates due to source conditions, consult with the Administrator for possible variance on the isokinetic rates.

4.2 Sample Recovery. Proper cleanup procedure begins as soon as the probe is removed from the stack at the end of the sampling period. Allow the probe to cool.

When the probe can be safely handled, wipe off all external particulate matter near the tip of the probe nozzle and place a cap over it to prevent losing or gaining particulate matter. Do not cap off the probe tip tightly while the sampling train is cooling down as this would create a vacuum in the filter holder, thus drawing water from the impingers into the filter holder.

Before moving the sample train to the cleanup site, remove the probe from the sample train, wipe off the silicone grease, and cap the open outlet of the probe. Be careful not to lose any condensate that might be present. Wipe off the silicone grease from the filter inlet where the probe was fastened and cap it. Remove the umbilical cord from the last impinger and cap the impinger. If a flexible line is used between the first impinger or condenser and the filter holder, disconnect the line at the filter holder and let any condensed water or liquid drain into the impingers or condenser. After wiping off the silicone grease, cap off the filter holder outlet and impinger inlet. Either ground-glass stoppers, plastic caps, or serum caps may be used to close these openings.

Transfer the probe and filter-impinger assembly to the cleanup area. This area should be clean and protected from the wind so that the chances of contaminating or losing the sample will be minimized.

Save a portion of the acetone used for cleanup as a blank. Take 300 ml of this acetone directly from the wash bottle being used and place it in a glass sample container labeled "acetone blank."

Inspect the train prior to and during disassembly and note any abnormal conditions. Treat the samples as follows:

Container No. 1. Carefully remove the filter from the filter holder and place it in its identified petri dish container. Use a pair of tweezers and/or clean disposable surgical gloves to handle the filter. If it is necessary to fold the filter, do so such that the particulate cake is inside the fold. Carefully transfer to the petri dish any particulate matter and/or filter fibers which adhere to the filter holder gasket, by using a dry Nylon

bristle brush and/or a sharp-edged blade. Seal the container.

Container No. 2. Taking care to see that dust on the outside of the probe or other exterior surfaces does not get into the sample, quantitatively recover particulate matter or any condensate from the probe nozzle, probe fitting, probe liner, and front half of the filter holder by washing these components with acetone and placing the wash in a glass container. Distilled water may be used instead of acetone when approved by the Administrator and shall be used when specified by the Administrator; in these cases, save a water blank and follow the Administrator's directions on analysis. Perform the acetone rinses as follows:

Carefully remove the probe nozzle and clean the inside surface by rinsing with acetone from a wash bottle and brushing with a Nylon bristle brush. Brush until the acetone rinse shows no visible particles, after which make a final rinse of the inside surface with acetone.

Brush and rinse the inside parts of the Swagelok fitting with acetone in a similar way until no visible particles remain.

Rinse the probe liner with acetone by tilting and rotating the probe while squirting acetone into its upper end so that all inside surfaces will be wetted with acetone. Let the acetone drain from the lower end into the sample container. A funnel (glass or polyethylene) may be used to aid on transferring liquid washes to the container. Follow the acetone rinse with a probe brush. Hold the probe in an inclined position, squirt acetone into the upper end as the probe brush is being pushed with a twisting action through the probe; hold a sample container underneath the lower end of the probe, and catch any acetone and particulate matter which is brushed from the probe. Run the brush through the probe three times or more until no visible particulate matter is carried out with the acetone or until none remains in the probe liner on visual inspection. With stainless steel or other metal probes, run the brush through in the above prescribed manner at least six times since metal probes have small crevices in which particulate matter can be entrapped. Rinse the brush with acetone, and quantitatively collect these washings in the sample container. After the brushing, make a final acetone rinse of the probe as described above.

It is recommended that two people be used to clean the probe to minimize sample losses. Between sampling runs, keep brushes clean and protected from contaminations.

After ensuring that all joints have been wiped clean of silicone grease, clean the inside of the front half of the filter holder by rubbing the surfaces with a Nylon bristle brush and rinsing with acetone. Rinse each surface three times or more if needed to remove visible particulate. Make a final rinse of the brush and filter holder. Carefully rinse out the glass cyclone, also (if applicable). After all acetone washings and particulate matter have been collected in the sample container, tighten the lid on the sample container so that acetone will not leak out when it is shipped to the laboratory. Mark the height of the fluid level to determine whether or not leakage occurred during transport. Label the container to clearly identify its contents.

Container No. 3. Note the color of the indicating silica gel to determine if it has been completely spent and make a notation of its condition. Transfer the silica gel from the fourth impinger to its original container and seal. A funnel may make it easier to pour the silica gel without spilling. A rubber policeman may be used as an aid in removing the silica gel from the impinger; it is not necessary to remove the small amount of dust particles that may adhere to the impinger wall and are difficult to remove. Since the gain in weight is to be used for moisture calculations, do not use any water or other liquids to transfer the silica gel. If a balance is available in the field, follow the procedure for container No. 3 in Section 4.3.

Impinger Water. Treat the impingers as follows: Make a notation of any color or film in the liquid catch. Measure the liquid which is in the first three impingers to within ±1 ml by using a graduated cylinder or by weighing it to within ±0.5 g by using a balance (if one is available). Record the volume or weight of liquid present. This information is required to calculate the moisture content of the effluent gas.

Discard the liquid after measuring and recording the volume or weight, unless analysis of the impinger catch is required (see Note, Section 2.1.7).

If a different type of condenser is used, measure the amount of moisture condensed either volumetrically or gravimetrically.

Whenever possible, containers should be shipped in such a way that they remain upright at all times.

4.3 Analysis. Record the data required on a sheet such as the one shown in Figure 3-3. Handle each sample container as follows:

FIGURE 3-3—ANALYTICAL DATA

Plant _____
 Date _____
 Run No. _____
 Filter No. _____
 Amount liquid lost during transport _____
 Acetone blank volume, ml _____
 Acetone wash volume, ml _____
 Acetone blank concentration, mg/mg (equation 5-4) _____
 Acetone wash blank, mg (equation 5-5) _____

Container Number	Weight of particulate collected, mg		
	First weight	Tap weight	Weight gain
1			
2			
Total _____			
Loss acetone blank _____			
Weight of particulate matter _____			

	Volume of liquid water collected	
	Impinger volume, ml	Silica gel weight, g
First _____		
Initial _____		
Liquid collected _____		
Total volume collected _____		

*Convert weight of water to volume by dividing total weight measured by density of water (1 g/ml).

$$\frac{\text{Increase, g}}{(1 \text{ g/ml})} = \text{Volume water, ml}$$

Container No. 1. Leave the contents in the shipping container or transfer the filter and any loose particulate from the sample container to a tared glass weighing dish. Desiccates for 24 hours in a desiccator containing anhydrous calcium sulfate. Weigh to a constant weight and report the results to the nearest 0.1 mg. For purposes of this Section, 4.3, the term "constant weight" means a difference of no more than 0.5 mg or 1 percent of total weight less tare weight, whichever is greater, between two consecutive weighings, with no less than 6 hours of desiccation time between weighings.

Alternatively, the sample may be oven dried at 105° C (220° F) for 2 to 3 hours, cooled in the desiccator, and weighed to a constant weight, unless otherwise specified by the Administrator. The tester may also opt to oven dry the sample at 105° C (220° F) for 2 to 3 hours, weigh the sample, and use this weight as a final weight.

Container No. 2. Note the level of liquid in the container and confirm on the analysis sheet whether or not leakage occurred during transport. If a noticeable amount of leakage has occurred, either void the sample or use methods, subject to the approval of the Administrator, to correct the final results. Measure the liquid in this container either volumetrically to ±1 ml or gravimetrically to ±0.5 g. Transfer the contents to a tared 250-ml beaker and evaporate to dryness at ambient temperature and pressure. Desiccates for 24 hours and weigh to a constant weight. Report the results to the nearest 0.1 mg.

Container No. 3. Weigh the spent silica gel (or silica gel plus impinger) to the nearest 0.5 g using a balance. This step may be conducted in the field.

"Acetone Blank" Container. Measure acetone in this container either volumetrically or gravimetrically. Transfer the acetone to a tared 250-ml beaker and evaporate to dryness at ambient temperature and pressure. Desiccates for 24 hours and weigh to a constant weight. Report the results to the nearest 0.1 mg.

NOTE: At the option of the tester, the contents of Container No. 2 as well as the acetone blank container may be evaporated at temperatures higher than ambient. If evaporation is done at an elevated temperature, the temperature must be below the boiling point of the solvent; also, to prevent "bumping," the evaporation process must be closely supervised, and the contents of the beaker must be swirled occasionally to maintain an even temperature. Use extreme care, as acetone is highly flammable and has a low flash point.

4.4 Quality Control Procedures. The following quality control procedures are suggested to check the volume metering system calibration values at the field test site prior to sample collection. These procedures are optional for the tester.

4.4.1 Meter Orifice Check. Using the calibration data obtained during the calibration procedure described in Section 3.1, determine the ΔH₀ for the metering system orifice. The ΔH₀ is the orifice pressure differential in units of in. H₂O that correlates to 0.75 cfm of air at 52°R and 29.92 in. Hg. The ΔH₀ is calculated as follows:

$$\Delta H_0 = 0.0319 \Delta H \frac{T_a}{P_m} \frac{\Theta^3}{Y^3}$$

Eq. 3-6

Where:

ΔH = Average pressure differential across the orifice meter, in. H₂O.

T_a = Absolute average dry gas meter temperature, °R.

P_m = Barometric pressure, in. Hg.

Θ = Total sampling time, min.

Y = Dry gas meter calibration factor, dimensionless.

V_a = Volume of gas sample as measured by dry gas meter, dcf.

0.0319 = (0.0567 in. Hg/°R) x (0.75 cfm)³

Before beginning the field test (a set of three runs usually constitutes a field test), operate the metering system (i.e., pump, volume meter, and orifice) at the ΔH₀ pressure differential for 10 minutes. Record the volume collected, the dry gas meter temperature, and the barometric pressure. Calculate a dry gas meter calibration check value, Y_c, as follows:

$$Y_c = \frac{10}{V_a} \left[\frac{0.0319 T_a}{P_m} \right] \Theta$$

Eq. 5-10

Where:

Y_c = Dry gas meter calibration check value, dimensionless.

10 = 10 minutes of run time.

Compare the Y_c value with the dry gas meter calibration factor Y to determine that:

0.97Y < Y_c < 1.03Y

If the Y_c value is not within this range, the volume metering system should be investigated before beginning the test.

4.4.2 Calibrated Critical Orifice. A calibrated critical orifice, calibrated against a wet test meter or spirometer and designed to be inserted at the inlet of the sampling meter box, may be used as a quality control check, such procedure being subject to approval by the Administrator.

5. Calibration

Maintain a laboratory log of all calibrations.

5.1 Probe Nozzle. Probe nozzles shall be calibrated before their initial use in the field. Using a micrometer, measure the inside diameter of the nozzle to the nearest 0.025 mm (0.001 in.). Make three separate measurements using different diameters each time, and obtain the average of the measurements. The difference between the high and low numbers shall not exceed 0.1 mm (0.004 in.). When nozzles become nicked, dented, or corroded, they shall be reshaped, sharpened, and recalibrated before use. Each nozzle shall be permanently and uniquely identified.

5.2 Pitot Tube. The Type S pitot tube assembly shall be calibrated according to the procedure outlined in Section 4 of Method 2.

5.3 Metering System. Before its initial use in the field, the metering system shall be calibrated according to the procedure outlined in APTD-0576. Instead of physically adjusting the dry gas meter dial readings to correspond to the wet test meter readings, calibration factors may be used to mathematically correct the gas meter dial readings to the proper values. Before calibrating the metering system, it is suggested that a leak-check be conducted. For metering systems having diaphragm pumps, the normal leak-check procedure will not detect leakages within the pump, for these cases

the following leak-check procedure is suggested: make a 10-minute calibration run at 0.0057 m³/min (0.02 cfm); at the end of the run, take the difference of the measured wet test meter and dry gas meter volumes divide the difference by 10, to get the leak rate. The leak rate should not exceed 0.00037 m³/min (0.02 cfm).

After each field use, the calibration of the metering system shall be checked by performing three calibration runs at a single intermediate orifice setting (based on the previous field test). With the vacuum set at the maximum value reached during the test series. To adjust the vacuum, insert a valve between the wet test meter and the inlet of the metering system. Calculate the average value of the calibration factor. If the calibration has changed by more than 5 percent, recalibrate the meter over the full range of orifice settings, as outlined in APTD-0576.

Alternative procedures, e.g., using the orifice meter coefficients, may be used, subject to the approval of the Administrator.

NOTE: If the dry gas meter coefficient values obtained before and after a test series differ by more than 5 percent, the test series shall either be voided, or calculations for test series shall be performed using whichever meter coefficient value (i.e., before or after) gives the lower value of total sample volume.

5.3.1 Calibration Prior to Use. Before its initial use in the field, the metering system shall be calibrated as follows: Connect the metering system inlet to the outlet of a wet test meter that is accurate to within 1 percent. Refer to Figure 5.5. The wet test meter should have a capacity of 30 liters/rev (1 ft³/rev). A spirometer of 600 liters (14 ft³) or more capacity, or equivalent, may be used for this calibration, although a wet test meter is usually more practical. The wet test meter should be periodically calibrated with a spirometer or a liquid displacement meter to ensure the accuracy of the wet test meter. Spirometers or wet test meters of other sizes may be used, provided that the specified accuracies of the procedure are maintained. Run the metering system pump for about 15 minutes with the orifice manometer indicating a median reading as expected in field use to allow the pump to warm up and to permit the interior surface of the wet test meter to be thoroughly wetted. Then, at each of a minimum of three orifice manometer settings, pass an exact quantity of gas through the wet test meter and note the gas volume indicated by the dry gas meter. Also note the barometric pressure, and the temperatures of the wet test meter, the inlet of the dry gas meter, and the outlet of the dry gas meter. Select the highest and lowest orifice settings to bracket the expected field operating range of the orifice. Use a minimum volume of 0.15 m³ (5 cf) at all orifice settings. Record all the data on a form similar to Figure 5.6, and calculate Y, the dry gas meter calibration factor, and ΔH₀, the orifice calibration factor, at each orifice setting as shown on Figure 5.6. Allowable tolerances for individual Y and ΔH₀ values are given in Figure 5.6. Use the average of the Y values in the calculations in Section 6.

ing procedure is suggested (see Figure 5-4): Close the main valve on the meter box. Insert a one-hole rubber stopper with rubber tubing attached into the orifice exhaust pipe. Disconnect and vent the low side of the orifice manometer. Close off the low side orifice tap. Pressurize the system to 13 to 18 cm (5 to 7 in.) water column by blowing into the rubber tubing. Pinch off the tubing and observe the manometer for one minute. A loss of pressure on the manometer indicates a leak in the meter box; leaks, if present, must be corrected.

5.7 Barometer. Calibrate against a mercury barometer.

6. Calculations

Carry out calculations, retaining at least one extra decimal figure beyond that of the acquired data. Round off figures after the final calculation. Other forms of the equations may be used as long as they give equivalent results.

6.1 Nomenclature

- A_n = Cross-sectional area of nozzle, m^2 (ft^2).
- S_w = Water vapor in the gas stream, proportion by volume.
- C_w = Acetone blank residue concentration, mg/g .
- C = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, $g/dscm$ ($g/dscf$).
- I = Percent of isokinetic sampling.
- L_m = Maximum acceptable leakage rate for either a pretest leak check or for a leak check following a component change: equal to $0.0057 m^3/min$ ($0.02 cfm$) or 4 percent of the average sampling rate, whichever is less.
- L_i = Individual leakage rate observed during the leak check conducted prior to the " i^{th} " component change ($i=1, 2, 3, \dots, n$), m^3/min (cfm).
- L_p = Leakage rate observed during the post-test leak check, m^3/min (cfm).
- m_c = Total amount of particulate matter collected, mg .
- M_w = Molecular weight of water, $18.0 g/g\text{-mole}$ ($18.0 lb/lb\text{-mole}$).
- m_a = Mass of residue of acetone after evaporation, mg .
- P_{at} = Barometric pressure at the sampling site, $mm Hg$ ($in. Hg$).

- P_s = Absolute stack gas pressure, $mm Hg$ ($in. Hg$).
- P_{std} = Standard absolute pressure, $760 mm Hg$ ($29.92 in. Hg$).
- R = Ideal gas constant, $0.08216 mm Hg\text{-}m^3/K\text{-}g\text{-mole}$ ($21.85 in. Hg\text{-}ft^3/R\text{-}lb\text{-mole}$).
- T_a = Absolute average dry gas meter temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
- T_s = Absolute average stack gas temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
- T_{std} = Standard absolute temperature, $293^{\circ} K$ ($528^{\circ} R$).
- V_w = Volume of acetone blank, ml .
- V_{wa} = Volume of acetone used in wash, ml .
- V_{liq} = Total volume of liquid collected in impingers and silica gel (see Figure 5-3), ml .
- V_{dm} = Volume of gas sample as measured by dry gas meter, $dscm$ ($dscf$).
- V_{dscm} = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, $dscm$ ($dscf$).
- V_{scm} = Volume of water vapor in the gas sample, corrected to standard conditions, scm (scf).
- v_s = Stack gas velocity, calculated by Method 2, Equation 2-9, using data obtained from Method 5, m/sec (ft/sec).
- W_a = Weight of residue in acetone wash, mg .
- Y = Dry gas meter calibration factor.
- ΔH = Average pressure differential across the orifice meter (see Figure 5-2), $mm H_2O$ ($in. H_2O$).
- ρ_a = Density of acetone, mg/ml (see label on bottle).

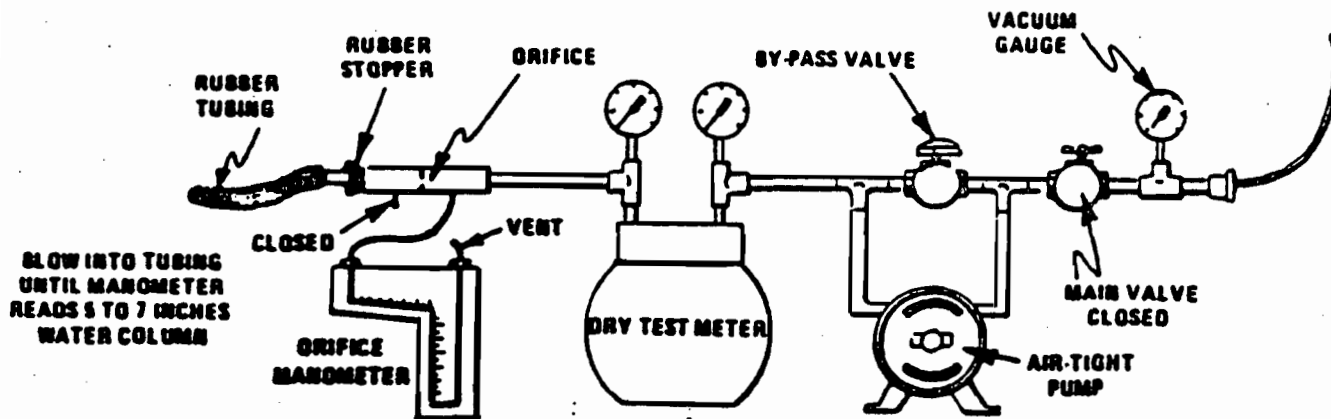


Figure 6-4. Leak check of meter box.

- ρ_w = Density of water. 9.982 g/ml (0.002201 lb/ml).
- θ = Total sampling time, min.
- θ_1 = Sampling time interval, from the beginning of a run until the first component change, min.
- θ_2 = Sampling time interval, between two successive component changes, beginning with the interval between the first and second changes, min.
- θ_n = Sampling time interval, from the final (n^{th}) component change until the end of the sampling run, min.
- 13.6 = Specific gravity of mercury.
- 60 = Sec/min.
- 100 = Conversion to percent.

6.2 Average dry gas meter temperature and average orifice pressure drop. See data sheet (Figure 5-2).

6.3 Dry Gas Volume. Correct the sample volume measured by the dry gas meter to standard conditions (20° C, 760 mm Hg or 68° F, 29.92 in. Hg) by using Equation 5-1.

$$V_{s(\text{std})} = V_s Y \left(\frac{T_{\text{std}}}{T_s} \right) \left[\frac{P_{\text{std}} + \frac{\Delta H}{13.6}}{P_{\text{std}}} \right]$$

$$= K_1 V_s Y \frac{P_{\text{std}} + (\Delta H/13.6)}{T_s}$$

Equation 5-1

where:

- $K_1 = 0.3858 \text{ } ^\circ\text{K/mm Hg for metric units}$
- $= 17.64 \text{ } ^\circ\text{R/in. Hg for English units}$

NOTE: Equation 5-1 can be used as written unless the leakage rate observed during any of the mandatory leak checks (i.e., the post-test leak check or leak checks conducted prior to component changes) exceeds L_1 . If L_1 or L_2 exceeds L_1 , Equation 5-1 must be modified as follows:

(a) Case I. No component changes made during sampling run. In this case, replace V_s in Equation 5-1 with the expression:

$$V_s = (L_1 - L_2)\theta$$

(b) Case II. One or more component changes made during the sampling run. In this case, replace V_s in Equation 5-1 by the expression:

$$V_s = (L_1 - L_2)\theta_1 - \sum_{i=2}^n (L_i - L_{i-1})\theta_i - (L_n - L_{n-1})\theta_n$$

and substitute only for those leakage rates (L_i or L_{i-1}) which exceed L_1 .

6.4 Volume of water vapor.

$$V_{w(\text{std})} = V_w \left(\frac{P_w}{P_{\text{std}}} \right) \left(\frac{RT_{\text{std}}}{P_{\text{std}}} \right) = K_2 V_w$$

Equation 5-2

where:

- $K_2 = 0.001333 \text{ m}^3/\text{ml for metric units}$
- $= 0.04707 \text{ (ft}^3/\text{ml for English units)}$

6.5 Moisture Content.

$$B_w = \frac{V_{w(\text{std})}}{V_{s(\text{std})} + V_{w(\text{std})}}$$

Equation 5-3

NOTE: In saturated or water droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one from the impinger analysis (Equation 5-3), and a second from the assumption of saturated conditions. The lower of the two values of B_w shall be considered correct. The procedure for determining the moisture content based upon assumption of saturated conditions is given in the Note of Section 1.2 of Method 4. For the purposes of this method, the average stack gas temperature from Figure 5-2 may be used to make this determination, provided that the accuracy of the in-stack temperature sensor is $\pm 1^\circ \text{C}$ (2°F).

6.6 Acetone Blank Concentration.

Equation 5-4

$$C_a = \frac{m_a}{V_{\text{std}}}$$

6.7 Acetone Wash Blank.

$$W_a = C_a V_{\text{std}} \quad \text{Equation 5-5}$$

6.8 Total Particulate Weight. Determine the total particulate catch from the sum of the weights obtained from containers 1 and 2 less the acetone blank (see Figure 5-3).

NOTE: Refer to Section 4.1.5 to assist in calculation of results involving two or more filter assemblies or two or more sampling trains.

6.9 Particulate Concentration.

$$C_p = (0.001 \text{ g/mg}) (m_p / V_{s(\text{std})}) \quad \text{Equation 5-6}$$

6.10 Conversion Factors:

From	To	Multiply by
cc	m ³	0.02832
g/m ³	g/R ³	16.41
g/R ³	g/m ³	2.205 x 10 ⁻³
g/R ³	g/m ³	35.31

6.11 Isokinetic Variation.

6.11.1 Calculation From Raw Data.

$$I = \frac{100 T_s K_3 V_{s(\text{std})} (P_{\text{std}} / T_{\text{std}} \times P_{\text{std}} + \Delta H / 13.6)}{60 \rho_w P_s A}$$

Equation 5-7

where:

- $K_3 = 0.003484 \text{ mm Hg} \cdot \text{m}^3/\text{ml} \cdot \text{K for metric units}$
- $= 0.002669 \text{ in. Hg} \cdot \text{ft}^3/\text{ml} \cdot \text{R for English units}$

6.11.2 Calculation From Intermediate Values.

$$I = \frac{T_s V_{s(\text{std})} P_{\text{std}} (100)}{T_{\text{std}} V_w \theta A_s P_s 60 (1 - B_w)}$$

$$= K_4 \frac{T_s V_{s(\text{std})}}{P_s V_w \theta (1 - B_w)}$$

Equation 5-8

where:

- $K_4 = 4.320 \text{ for metric units}$
- $= 0.09450 \text{ for English units}$

6.12 Acceptable Results. If 90 percent $< I < 110$ percent, the results are acceptable. If the particulate results are low in comparison to the standard, and I is over 110 percent or less than 90 percent, the Administrator may accept the results. Citation 4 in the bibliography section can be used to make acceptability judgments. If I is judged to be unacceptable, reject the particulate results and repeat the test.

7. Alternative Procedures

7.1 Dry Gas Meter as a Calibration Standard. A dry gas meter may be used as a calibration standard for volume measurements in place of the wet test meter specified in Section 5.3, provided that it is calibrated initially and recalibrated periodically as follows:

7.1.1 Standard Dry Gas Meter Calibration.

7.1.1.1 The dry gas meter to be calibrated and used as a secondary reference meter should be of high quality and have an appropriately sized capacity, e.g., 3 liters/rev (0.1 ft³/rev). A spirometer (400 liters or more capacity), or equivalent, may be used for this calibration, although a wet test meter is usually more practical. The wet test meter should have a capacity of 30 liters/rev (1 ft³/rev) and capable of measuring volume to within ± 1.0 percent; wet test meters should be checked against a spirometer or a liquid displacement meter to ensure the accuracy of the wet test meter. Spirometers or wet test meters of other sizes may be used, provided that the specified accuracies of the procedure are maintained.

7.1.1.2 Set up the components as shown in Figure 5.7. A spirometer, or equivalent, may be used in place of the wet test meter in the system. Run the pump for at least 5 minutes at a flow rate of about 10 liters/min (0.35 cfm) to condition the interior surface of the wet test meter. The pressure drop indicated by the manometer at the inlet side of the dry gas meter should be minimized (no greater than 100 mm H₂O (4 in. H₂O) at a flow rate of 30 liters/min (1 cfm)). This can be accomplished by using large diameter tubing connections and straight pipe fittings.

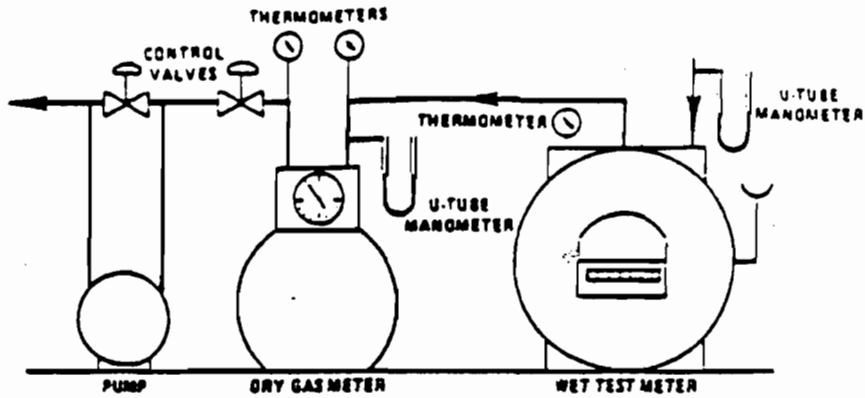


Figure 5.7. Equipment arrangement for dry-gas meter calibration.

7.1.1.3 Collect the data as shown in the example data sheet (see Figure 5-8). Make triplicate runs at each of the flow rates and at no less than five different flow rates. The

range of flow rates should be between 10 and 34 liters/min (0.35 and 1.2 cfm) or over the expected operating range.

DATE: _____
 DRY GAS METER IDENTIFICATION: _____
 BAROMETRIC PRESSURE (P_b): _____ in. Hg

APPROXIMATE FLOW RATE (Q̄) cfm	SPIROMETER (WET METER) GAS VOLUME (V _w) ft ³	DRY GAS METER VOLUME (V _{dm}) ft ³	TEMPERATURES				DRY GAS METER PRESSURE (ΔP) in. H ₂ O	TIME (Θ) min.	FLOW RATE (Q) cfm	METER COEFFICIENT (V _{dm})	AVERAGE METER COEFFICIENT (V̄ _{dm})
			SPIROMETER (WET METER) (t _w) °F	DRY GAS METER							
				INLET (t ₁) °F	OUTLET (t ₂) °F	AVERAGE (t ₃) °F					
0.40											
0.60											
0.80											
1.00											
1.20											

$$Q = 17.29 \cdot \frac{V_w}{\Theta} \cdot \frac{P_b}{P_b + \Delta P}$$

$$V_{dm} = \frac{V_w}{V_{dm}} \cdot \frac{P_b + \Delta P}{P_b} \cdot \frac{P_b}{P_b + \Delta P}$$

Figure 5.8. Example data sheet for calibration of a standard dry gas meter for method B sampling equipment (English units).

7.1.1.4 Calculate flow rate, Q , for each run using the wet test meter gas volume, V_w , and the run time, θ . Calculate the dry gas meter coefficient, Y_m , for each run. These calculations are as follows:

$$Q = K \frac{P_w V_w}{t_w + t_m \theta}$$

$$Y_m = \frac{V_w (t_w + t_m) P_w}{V_m (t_w + t_m) \left(P_w + \frac{\Delta P}{13.6} \right)}$$

Where:

K = 0.3538 for international system of units (SI); 17.64 for English units.

V_w = Wet test meter volume, liters (ft³).

V_m = Dry gas meter volume, liters (ft³).

t_w = Average dry gas meter temperature, °C (°F).

t_m = 273° C for SI units; 460° F for English units.

t_a = Average wet test meter temperature, °C (°F).

P_w = Barometric pressure, mm Hg (in. Hg).

ΔP = Dry gas meter inlet differential pressure, mm H₂O (in. H₂O).

θ = Run time, min.

7.1.1.5 Compare the three Y_m values at each of the flow rates and determine the maximum and minimum values. The difference between the maximum and minimum values at each flow rate should be no greater than 0.030. Extra sets of triplicate runs may be made in order to complete this requirement. In addition, the meter coefficients should be between 0.95 and 1.05. If these specifications cannot be met in three sets of successive triplicate runs, the meter is not suitable as a calibration standard and should not be used as such. If these specifications are met, average the three Y_m values at each flow rate resulting in five average meter coefficients, Y_m .

7.1.1.6 Prepare a curve of meter coefficient, Y_m , versus flow rate, Q , for the dry gas meter. This curve shall be used as a reference when the meter is used to calibrate other dry gas meters and to determine whether recalibration is required.

7.1.2 Standard Dry Gas Meter Recalibration.

7.1.2.1 Recalibrate the standard dry gas meter against a wet test meter or spirometer annually or after every 200 hours of operation, whichever comes first. This require-

ment is valid provided the standard dry gas meter is kept in a laboratory and, if transported, cared for as any other laboratory instrument. Abuse to the standard meter may cause a change in the calibration and will require more frequent recalibrations.

7.1.2.2 As an alternative to full recalibration, a two-point calibration check may be made. Follow the same procedure and equipment arrangement as for a full recalibration, but run the meter at only two flow rates (suggested rates are 14 and 28 liters/min (0.5 and 1.0 cfm)). Calculate the meter coefficients for these two points, and compare the values with the meter calibration curve. If the two coefficients are within ±1.5 percent of the calibration curve values at the same flow rates, the meter need not be recalibrated until the next date for a recalibration check.

8. Bibliography

1. Addendum to Specifications for Incinerator Testing at Federal Facilities. PHS, NCAPC, Dec. 6, 1967.
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APPENDIX C-1
CLEAN-UP/FIELD DATA

METHOD 5 OR 17 FIELD DATA SHEET

ETS, INC.

PROJECT Westinghouse BAY RESOURCES
 LOCATION PANAMA City, FLA.
 PROCESS MSW
 CONTROL EQUIPMENT ESP
 SAMPLING POINT OUTLET
 POLLUTANTS _____

DATE 6/4-6/5/87 STD TEMP 68 °F
 RUN # 1-6 METER BOX 3
 OPERATOR Jim Eckenrode GAS METER 2.052 ΔHe
 AMBIENT TEMP 80-90 GAS METER Y 1.001

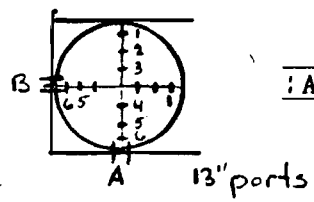
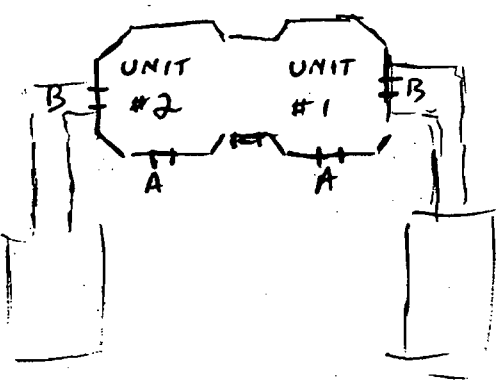
	P _b
611	30.04
612	30.03
613	30.10
614	30.00
615	30.02

Cp .84 PITOT TUBE COEFFICIENT
 P BAR _____ "HG" LOCAL BAROMETRIC PRESSURE
 DN .254 "NOZZLE" DIAMETER
 AS 15.32072 ft² STACK AREA
 ASSUMED Bws 15% % MOISTURE

STACK SCHEMATIC
 UPSTREAM DISTURBANCE >8
 DOWNSTREAM DISTURBANCE >2

TRAVERSE POINTS		13" ports	
NUMBER	INCHES	NUMBER	INCHES
A 1	50.67		63.67
2	45.26		58.26
3	37.31		50.31
4	15.69		28.69
5	7.74		20.74
6	2.33		15.33
B 1	50.67		
2	45.26		
3	37.31		
4	15.69		
5	7.74		
6	2.33		

ORSAT MEASUREMENT			
CO ₂	O ₂	CO	N ₂
: AVG :			



cmthd5

PLANT

Run 1 unit #2

PLANT Westinghouse

DATE June 4, 1987

RUN # unit #2

START TIME: 0945

P #	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft ³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXIT	VAC. in Hg
B 1	0			430	.64	1.56	981.064	101	101	250	<70	2
2	5			433	.71	1.90	984.2	101	102	245	<70	3
3	10		-58	437	.79	1.99	987.7	102	102	255	<70	3
4	15			437	.75	1.90	991.3	104	103	254	<70	3
5	20			439	.65	1.65	994.9	105	103	255	<70	3
6	25			424	.46	1.15	998.4	106	104	256	<70	2
							1001.2					

A 1	30		-57	434	.66	1.69	1001.2	106	106	250	<70	3
2	35			439	.77	1.98	1004.5	103	103	260	<70	3
3	40			441	.83	2.10	1008.1	103	102	259	<70	3
4	45			437	.50	1.28	1011.9	106	104	255	<70	3
5	50			435	.51	1.29	1014.8	108	105	261	<70	3
6	55			360	.20	.50	1017.6	108	106	265	<70	2
	60						1019.551					

AVG:

TIME: 1048

LEAK CHECK @ 4 " Hg

FLA 2

Run 2 unit #2

PLANT Westinghouse

DATE June 4, 1957

RUN # unit #2

START TIME: 1311

P #	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft ³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXT	VAC. IN Hg
1	0			438	.49	1.24	19.56	101	99	250	<70	3
2	5		-.47	437	.46	1.15	22.6	99	99	270	<70	3
3	10			437	.45	1.12	25.4	99	99	251	<70	3
4	15			444	.71	1.80	28.2	100	99	254	<70	4
5	20			448	.72	1.82	31.7	102	100	253	<70	4
6	25			431	.57	1.44	35.2	103	100	251	<70	3
							38.3					
1	30			447	.90	2.29	38.3	102	100	251	<70	5
2	35			459	1.00	2.50	42.2	103	101	257	<70	5
3	40		-.69	462	1.05	2.65	46.3	104	101	257	<70	6
4	45			463	.87	2.20	50.4	102	100	253	<70	5
5	50			465	.90	2.30	54.3	103	100	256	<70	5
	55			455	.68	1.71	58.1	104	101	256	<70	4
	60						61.448					

AVG:

TIME: 1413

LEAK CHECK .01 @ 10" Hg

cmthd5

PLA2

Run 3 unit #2

PLANT Westinghouse

DATE June 4 / 1987

RUN # cmtd H 2

START TIME: 1527

P #	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts	STACK Δ P	METER Δ h	Vm ft ³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXT	VAC ID H
1	0			446	.97	2.47	61.6	86	92	92	251	<70 : 5
2	5		-73	462	.93	2.30	65.6		93	93	254	<70 : 5
3	10			465	.99	2.50	69.6		94	93	257	<70 : 5
4	15			465	.98	2.48	73.6		94	92	244	<70 : 5
5	20			465	.85	2.12	77.7		92	91	241	<70 : 5
6	25			440	.31	.78	81.4		90	91	254	<70 : 2
							83.7					
1	30		-60	444	.68	1.68	83.7		88	88	246	<70 : 4
2	35			451	.73	1.85	86.9		88	88	242	<70 : 4
3	40			451	.77	2.00	90.3		87	87	262	<70 : 4
4	45			453	.79	1.98	93.9		88	87	256	<70 : 4
5	50			456	.60	1.50	97.4		87	86	244	<70 : 3
6	55			418	.54	1.34	100.5		87	86	240	<70 : 3
	60						103.5					

AVG:

END TIME: 1630

LEAK CHECK .01 @ 6 " Hg

cmthd5

VOIDED

PLANT Westinghouse

DATE JUNE 5 1987

RUN #

START TIME: 0809

103.819

unit #1

P	SAMPLE #	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXT	VAC. in Hg
A	1	0		423	.53	1.31	103.819	85	85	250	<70	3
	2	5		423	.53	1.31	106.7	85	85	255	<70	3
	3	10	-0.52	422	.56	1.42	109.6	86	85	255	<70	3
	4	15		420	.57	1.45	112.6	86	86	255	<70	3
	5	20		422	.50	1.27	115.7	86	86	256	<70	3
	6	25		345	.35	.90	118.6	88	87	258	<70	2
							121.1					
							0839					
B	1	30										
	2	35										
	3	40										
	4	45										
	5	50										
	6	55										
		60										
AVG:												

ND TIME:

LEAK CHECK

" Hg

cmthd5

VOIDED @ 0839

FLA1

Run #1 Unit #1

PLANT Westinghouse

DATE JUNE 5, 1987

RUN # _____

Unit # 1

START TIME: 0952

P #	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXIT	VAC. in Hg
B 1	0			428	.74	1.89	121.757	93	92	236	<70	4
2	5			428	.79	2.00	125.2	91	92	248	<70	4
3	10			428	.73	1.87	128.8	93	92	250	<70	4
4	15		-.55	429	.68	1.72	132.3	92	92	252	<70	4
5	20			429	.60	1.50	135.6	93	93	247	<70	3
6	25			396	.40	1.00	138.6	93	93	252	<70	2
							141.3					
A 1	30			428	.61	1.55	141.3	95	94	249	<70	3
2	35			434	.68	1.72	144.4	95	95	256	<70	4
3	40			435	.69	1.75	147.7	96	95	257	<70	4
4	45			434	.56	1.42	151.2	95	95	257	<70	4
5	50			433	.50	1.25	154.4	96	96	263	<70	3
6	55			393	.31	.77	157.3	97	96	260	<70	2
	60						159.603					

AVG:

END TIME: 1054

LEAK CHECK 004 @ 6 " Hg

FLA1

Run ~~2~~ Unit #1

PLANT Westinghouse

DATE June 5, 1987

RUN # unit #1

START TIME: 1140

#	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXT ID	VAC Hg
1	0			433	.66	1.70	159.753	97	96	251	270	3
2	5			433	.70	1.79	162.8	97	97	254	270	4
3	10		-0.56	433	.77	1.96	166.2	97	97	251	270	4
4	15			434	.65	1.70	169.7	98	96	256	270	4
5	20			434	.66	1.70	173.1	99	98	256	270	4
6	25			399	.38	0.95	176.5	101	99	255	270	2.5
					81 <u>20</u>		179.0					
1	30			434	.81	2.0	179.0	101	99	255	270	4.5
2	35			435	.83	2.10	182.7	102	100	256	270	5.0
3	40			435	.82	2.09	180.3	102	100	254	270	5.0
4	45			434	.71	1.82	190.0	101	101	253	270	5.0
5	50			435	.63	1.59	193.5	101	100	253	270	4.0
6	55			412	.45	1.13	196.8	101	100	256	270	3
	60						199.954					

AVG:
 END TIME: 1244

LEAK CHECK .008 @ 7 " Hg

cmthd5

FLA1

Run 3 unit #1

PLANT Westinghouse

DATE June 5, 1987

RUN # unit #1

START TIME: 1305

P #	SAMPLE TIME	CLOCK TIME	STATIC PRESS.	Ts °F	STACK Δ P	METER Δ h	Vm ft ³	Tm INLET	Tm OUTLET	OVEN TEMP.	TEXT	VAC. in Hg
A1	0			431	.67	1.70	200.05	100	99	246	<70	5
	2	5		434	.70	1.80	203.3	98	97	260	<70	5
	3	10	-56	435	.70	1.80	206.6	98	97	257	<70	5
	4	15		435	.66	1.68	210.1	101	97	257	<70	5
	5	20		435	.58	1.48	213.4	100	98	262	<70	5
	6	25		370	.31	.78	216.4	101	98	267	<70	3
							219.1					
B1	30			429	.66	1.68	219.1	100	99	257	<70	5
	2	35		431	.67	1.70	222.3	99	98	261	<70	5
	3	40		434	.65	1.65	225.6	98	97	262	<70	5
	4	45		435	.61	1.52	228.9	98	97	262	<70	5
		50		434	.52	1.30	232.1	99	98	263	<70	4
		55		420	.39	.98	235.1	99	98	262	<70	4
		60					237.20					

AVG:
E. TIME:

LEAK CHECK .001 @ 5" Hg

METHOD 5 LAB ANALYSIS

CLIENT Westinghouse

DATE OF TEST June 5, 1987 DATE OF ANALYSIS June 5, 1987 / June 8, 1987

SAMPLE BOX # Nutech #3 ΔH @ 2.052 Y 1.001

BAROMETRIC PRESSURE 30.02 inches Hg

	* RUN 1 unit #1	RUN 1 RUN 2 unit #1	RUN 2 RUN 3 unit #1	RUN 3 RUN 4 unit #1
PROBE WASH				
BEAKER	TARE _____ mg	TARE <u>79.15413</u>	TARE <u>81.31562</u>	TARE <u>77.18925</u>
	FINAL _____ mg	FINAL <u>79.1787</u>	FINAL <u>81.34051</u>	FINAL <u>77.20213</u>
FILTER	# <u>87-305</u>	# <u>87-304</u>	# <u>86-262</u>	# <u>86-263</u>
	TARE _____ mg	TARE <u>1.33607</u>	TARE <u>1.4115</u>	TARE <u>1.22594</u>
	FINAL _____ mg	FINAL <u>1.34886</u>	FINAL <u>1.17034</u>	FINAL <u>1.25617</u>
IMPINGER	#1 <u>544</u> / _____ ml	#1 <u>581</u> / <u>712</u>	#1 <u>575</u> / <u>678</u>	#1 <u>591</u> / <u>702</u>
	#2 <u>551</u> / _____ ml	#2 <u>566</u> / <u>591</u>	#2 <u>553</u> / <u>579</u>	#2 <u>570</u> / <u>596</u>
	#3 _____ ml	#3 <u>—</u>	#3 <u>—</u>	#3 <u>—</u>
SILICA GEL	TARE <u>623</u> mg	TARE <u>633</u>	TARE <u>616</u>	TARE <u>639</u>
	FINAL _____ mg	FINAL <u>643</u>	FINAL <u>629</u>	FINAL <u>654</u>
ORSAT	CO ₂ _____ %	CO ₂ <u>11.5 11.5</u> %	CO ₂ <u>11.0 11.1 11.1</u> %	CO ₂ <u>11.4 11.4</u> %
	O ₂ _____ %	O ₂ <u>7.3 7.2 7.2</u> %	O ₂ <u>9.4 9.3 9.3</u> %	O ₂ <u>8.6 8.8 8.8</u> %

* VOIDED - STEAM FLOW ONLY ABOUT 60% OF MAX.

METHOD 5 LAB ANALYSIS

CLIENT Westinghouse

DATE OF TEST June 4, 1987 DATE OF ANALYSIS June 4, 1987 / June 8, 1987

SAMPLE BOX # Nutech #3 ΔH @ 2.052 Y 1.001

BAROMETRIC PRESSURE 30.0 inches Hg

	RUN 1 Unit #2	RUN 2 Unit #2	RUN 3 Unit #2	RUN 4
PROBE WASH				
BEAKER	TARE <u>77.38123</u> mg FINAL <u>77.40608</u> mg	TARE <u>78.94502</u> FINAL <u>78.9552</u>	TARE <u>77.88238</u> FINAL <u>77.90583</u>	BLANK TARE <u>82.13472</u> FINAL <u>82.13657</u>
FILTER	# <u>86-275</u> TARE <u>1.18793</u> mg FINAL <u>1.21832</u> mg	# <u>86-261</u> TARE <u>1.20971</u> FINAL <u>1.24099</u>	# <u>86-274</u> TARE <u>1.14692</u> FINAL <u>1.19683</u>	# <u>86-233</u> TARE <u>1.34697</u> FINAL <u>1.34698</u>
IMPINGER	#1 <u>1567/637</u> ml #2 <u>565/612</u> ml #3 <u>—</u> ml	#1 <u>555/673</u> #2 <u>559/589</u> #3 <u>—</u>	#1 <u>574/683</u> #2 <u>572/600</u> #3 <u>—</u>	#1 <u>—</u> #2 <u>—</u> #3 <u>—</u>
SILICA GEL	TARE <u>617</u> mg FINAL <u>632</u> mg	TARE <u>612</u> FINAL <u>624</u>	TARE <u>635</u> FINAL <u>647</u>	TARE <u>—</u> FINAL <u>—</u>
ORSAT	CO ₂ <u>11.3 11.4 11.4</u> % O ₂ <u>7.6 7.8 7.8</u> %	CO ₂ <u>10.4 10.4</u> % O ₂ <u>8.9 8.8 8.8</u> %	CO ₂ <u>11.5 11.7 11.7</u> % O ₂ <u>8.6 8.5 8.5</u> %	

Acetone Blank

$$C_a = \frac{m_a}{V_a \cdot P_a}$$

$$C_a = .000025$$

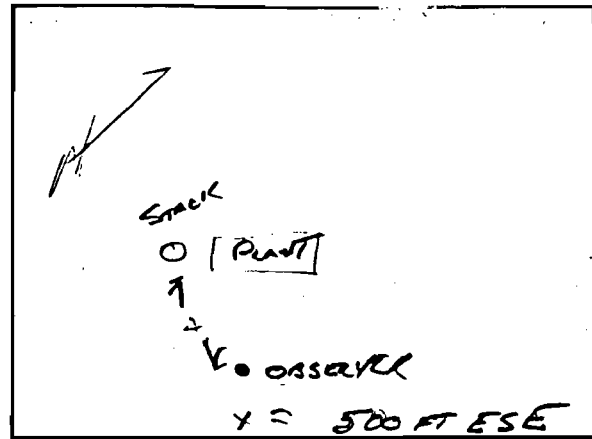
$$m_a = 82.13472 - 82.13657 = .001937$$

$$V_a = 100 \text{ ml}$$

$$P_a = .785$$

USEPA METHOD 9
 OPACITY OBSERVATION
 BAY COUNTY
 RESOURCE RECOVERY FACILITY
 PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 1
 DATE: 9.53 6/5/87
 TIME STARTED: 9:53
 TIME ENDED: 10:53
 OBSERVER: S.J. BEARD
 WIND SPEED/DIRECTION: 3-5 E
 SKY CONDITIONS: Clear

SEC 15 30 45 60

15 30 45 60

N 1	10	5	5	10
N 2	5	5	5	5
N 3	5	5	5	5
N 4	5	5	5	5
N 5	5	5	5	5
N 6	5	5	5	5
N 7	5	5	5	5
N 8	5	5	5	5
N 9	5	5	5	5
N 10	5	5	5	5
N 11	5	5	5	5
N 12	5	5	5	5
N 13	5	5	5	5
N 14	5	5	5	5
N 15	5	5	5	5
N 16	5	5	5	5
N 17	5	5	5	5
N 18	5	5	5	5
N 19	5	5	5	5
N 20	5	5	5	5
N 21	5	5	5	5
N 22	5	5	5	5
N 23	5	10	10	10
N 24	10	10	5	5
N 25	5	5	5	5
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N 27	10	10	10	5
N 28	10	5	5	5
N 29	5	5	5	5
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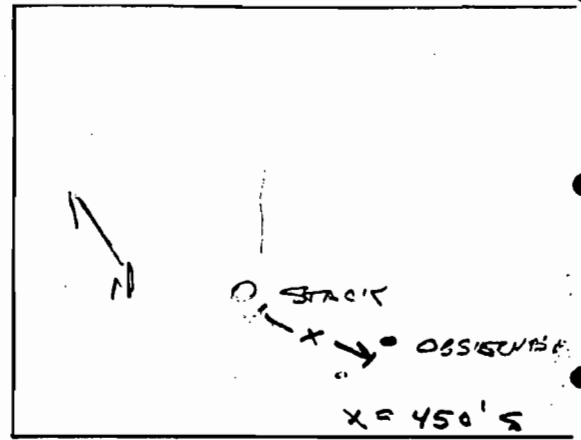
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37	5	5	5	10
38	5	5	10	10
39	5	5	10	10
40	10	10	10	15
41	10	10	10	10
42	10	10	10	10
43	10	5	10	10
44	5	5	5	5
45	5	5	5	5
46	5	5	5	5
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55	5	5	5	5
56	5	5	5	5
57	5	5	5	5
58	5	5	10	5
59	5	5	5	5
60	10	10	10	10

1 on unit # 100
 # 1 on unit # 100
 # 1 on unit # 100

USEPA METHOD 9 OPACITY OBSERVATION

BAY COUNTY
RESOURCE RECOVERY FACILITY
PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 1
 DATE: 6/5/87
 TIME STARTED: 11:40
 TIME ENDED: 12:40
 OBSERVER: S.J. BERRY
 WIND SPEED/DIRECTION: 2-3 N
 SKY CONDITIONS: PARTLY CLOUDY

Ref. = on chart

SEC 15 30 45 60

15 30 45 60

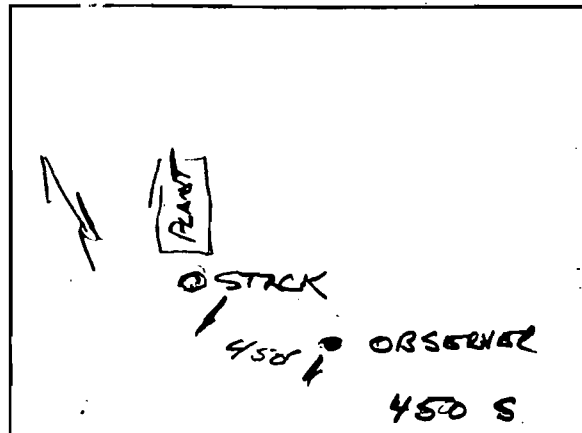
MIN	15	30	45	60
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3	5	5	5	5
4	5	5	5	5
5	5	5	10	5
6	5	5	5	5
7	5	5	5	5
8	5	5	5	5
9	5	5	5	5
10	5	5	5	5
11	5	5	5	5
12	5	5	5	5
13	5	5	5	5
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56	5	5	5	5
57	5	5	5	5
58	5	10	10	10
59	5	5	5	5
60	5	5	5	5

USEPA METHOD 9
OPACITY OBSERVATION

BAY COUNTY
RESOURCE RECOVERY FACILITY
PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 1
DATE: 6/5/87
TIME STARTED: 1:10
TIME ENDED: 2:10
OBSERVER: S. J. BRADY
WIND SPEED/DIRECTION: 2 - W
SKY CONDITIONS: CLEAR

SEC 15 30 45 60

15 30 45 60

MIN 1	5	5	5	5
2	5	5	5	5
3	5	5	5	5
4	5	5	5	5
5	5	5	5	5
6	5	5	5	5
7	5	5	5	5
8	5	5	5	5
9	5	5	5	5
10	5	5	5	5
11	5	5	5	5
12	5	5	5	5
13	5	5	5	5
14	5	5	5	5
15	5	5	5	5
16	5	5	5	5
17	5	5	5	5
18	5	5	5	5
19	5	5	5	5
20	5	5	5	5
21	5	5	5	5
22	5	5	5	5
23	5	5	5	5
24	5	5	5	5
25	5	5	5	5
26	5	5	5	5
27	5	5	5	5
28	5	5	5	5
29	5	5	5	5
30	5	5	5	5

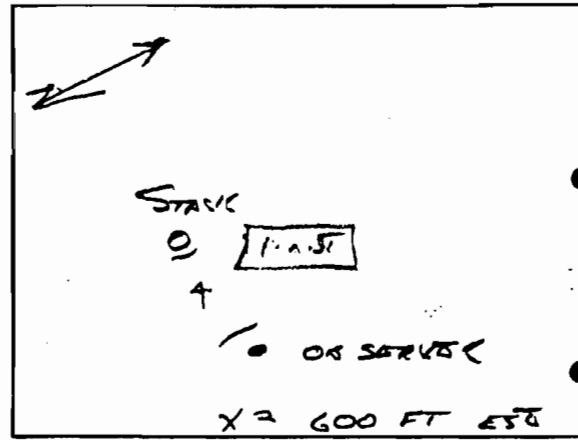
31	5	5	5	5
32	5	5	5	5
33	5	5	5	5
34	5	5	5	5
35	5	5	5	5
36	5	5	5	5
37	5	5	5	5
38	5	5	5	5
39	5	5	5	5
40	5	5	5	5
41	5	5	5	5
42	5	5	5	5
43	5	5	5	5
44	5	5	5	5
45	5	5	5	5
46	5	5	5	5
47	5	5	5	5
48	5	5	5	5
49	5	5	5	5
50	5	5	5	5
51	5	5	5	5
52	5	5	5	5
53	5	5	5	5
54	5	5	5	5
55	5	5	5	5
56	5	5	5	5
57	5	5	5	5
58	5	5	5	5
59	5	5	5	5
60	5	5	5	5

KUN # 3 ON UNIT # 1

USEPA METHOD 9
OPACITY OBSERVATION

BAY COUNTY
RESOURCE RECOVERY FACILITY
PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 2
DATE: 6/14/87
TIME STARTED: 10:10 AM
TIME ENDED: 10:47
OBSERVER: S. J. BRADY
WIND SPEED/DIRECTION: < 2 S
SKY CONDITIONS: SCATTERED CLOUDS

SEC 15 30 45 60

15 30 45 60

MIN	15	30	45	60
1	5	5	5	5
2	5	5	5	5
3	5	5	5	5
4	5	5	5	5
5	5	5	5	5
6	5	5	5	5
7	5	5	5	5
8	5	5	5	5
9	5	5	5	5
10	5	5	5	5
11	5	5	5	5
12	5	5	5	5
13	5	5	5	5
14	5	5	5	5
15	5	5	5	5
16	5	5	5	5
17	5	5	5	5
18	5	5	5	5
19	5	5	5	5
20	5	5	5	5
21	5	5	5	5
22	5	5	5	5
23	5	5	5	5
24	5	5	5	5
25	5	5	5	5
26	5	5	5	6
27	5	5	5	5
28	5	5	5	5
29	5	5	5	5
30	5	5	5	10

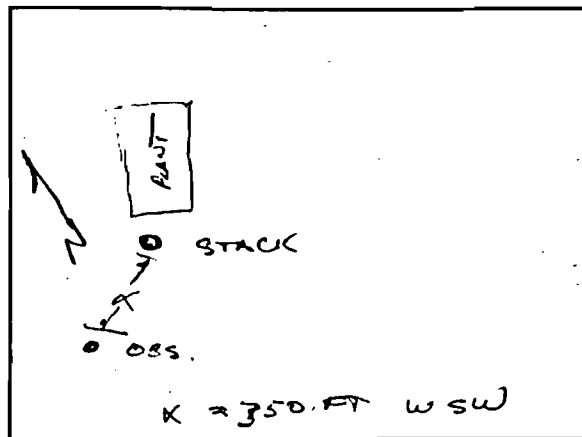
	15	30	45	60
31	10	10	5	5
32	10	10	10	10
33	10	10	5	5
34	5	5	10	5
35	5	5	5	5
36	5	5	5	5
37	5	5	5	5
38	END OF TEST			
39				
40				
41				
42				
43				
44				
45				
46				
47				
48				
49				
50				
51				
52				
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55				
56				
57				
58				
59				
60				

Run # 2 in # 2

USEPA METHOD 9
OPACITY OBSERVATION

BAY COUNTY
RESOURCE RECOVERY FACILITY
PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 2
DATE: 6/4/87
TIME STARTED: 1:10 PM
TIME ENDED: 2:10 PM
OBSERVER: S.J. Beary
WIND SPEED/DIRECTION: 2-4 S
SKY CONDITIONS: Partly Cloudy

SEC 15 30 45 60

15 30 45 60

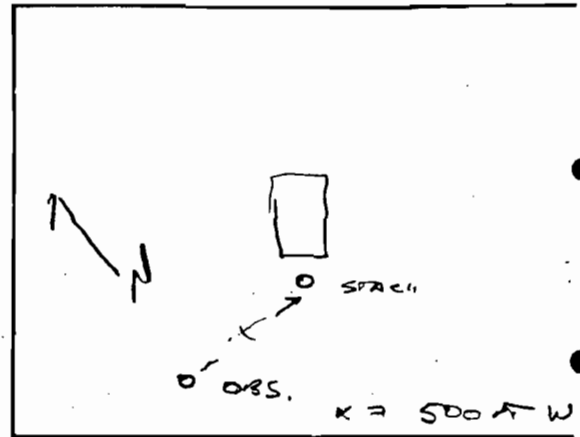
MIN	15	30	45	60
1	5	5	5	5
2	5	5	5	5
3	5	5	5	5
4	5	5	5	5
5	5	5	5	5
6	5	5	5	5
7	5	5	5	5
8	5	5	5	5
9	5	5	5	5
10	5	5	5	5
11	5	5	5	5
12	5	5	5	10
13	10	10	10	10
14	10	10	10	5
15	10	5	5	10
16	5	5	5	5
17	5	5	5	5
18	5	5	5	5
19	5	5	5	5
20	5	5	5	5
21	5	5	5	5
22	5	5	5	5
23	5	5	5	5
24	5	5	5	5
25	10	10	10	10
26	10	5	10	5
27	5	5	5	5
28	5	5	5	5
29	5	5	5	5
30	5	5	5	5

31	5	5	5	5
32	5	5	5	5
33	5	5	5	5
34	5	5	5	5
35	5	5	5	5
36	5	5	5	5
37	5	5	5	5
38	5	5	5	5
39	5	5	5	5
40	5	5	5	5
41	5	5	5	5
42	5	5	5	5
43	5	5	5	5
44	5	5	5	5
45	5	5	5	5
46	5	10	10	10
47	5	10	5	5
48	5	5	5	5
49	5	5	5	5
50	5	5	5	5
51	5	5	5	5
52	5	5	5	5
53	5	5	5	5
54	5	5	5	5
55	5	5	5	5
56	5	5	5	5
57	5	5	5	5
58	5	5	5	5
59	5	5	5	5
60	5	5	5	5

Run # 2 on Unit # 2

USEPA METHOD 9
 OPACITY OBSERVATION
 BAY COUNTY
 RESOURCE RECOVERY FACILITY
 PANAMA CITY, FLORIDA

OBSERVER POSITION



UNIT: 2
 DATE: 6/4/87
 TIME STARTED: 3:25
 TIME ENDED: 4:25
 OBSERVER: S. J. Bepko
 WIND SPEED/DIRECTION: 3-4 SW
 SKY CONDITIONS: P. Cloudy

SEC 15 30 45 60

15 30 45 60

MIN	1	2	3	4
1	5	5	5	5
2	5	5	5	5
3	5	5	5	5
4	5	5	5	5
5	5	5	5	5
6	5	5	5	5
7	5	5	5	5
8	5	5	5	5
9	5	5	5	5
10	5	5	5	5
11	10	10	5	5
12	10	10	10	5
13	5	5	5	5
14	5	5	5	5
15	10	10	10	5
16	5	5	5	10
17	5	5	5	5
18	5	5	5	5
19	5	5	5	5
20	5	5	5	5
21	5	5	5	5
22	5	5	5	5
23	5	5	5	5
24	5	5	5	6
25	5	5	5	5
26	5	5	5	5
27	5	5	5	6
28	5	5	5	5
29	5	5	5	5
30	5	5	5	5

	15	30	45	60
31	5	5	5	5
32	5	5	5	5
33	5	5	5	5
34	5	10	5	5
35	5	5	5	5
36	5	5	5	5
37	5	5	5	5
38	5	5	5	5
39	10	10	10	10
40	10	10	10	10
41	10	10	10	5
42	10	10	5	5
43	5	5	10	5
44	5	5	5	5
45	5	5	5	5
46	5	5	5	5
47	5	5	5	5
48	5	5	5	5
49	10	5	5	5
50	5	5	5	5
51	10	5	5	5
52	5	10	10	5
53	10	10	5	5
54	10	5	5	5
55	10	10	5	10
56	5	5	5	5
57	5	5	5	5
58	5	5	10	5
59	10	5	5	5
60	20	20	10	10

RUN # 3 ON UNIT # 2

CLOUDS
 WIND
 5-7 mph
 WIND 8-12 mph
 WIND 12-15

K
 COLM
 2-4

APPENDIX D-1
CALIBRATIONS

ETS, INC.

Pitot tube assembly level? ✓ yes no

Pitot tube openings damaged? yes (explain below) ✓ no

$a_1 = \underline{2.5}^\circ (<10^\circ)$, $a_2 = \underline{1.5}^\circ (<10^\circ)$, $B_1 = \underline{1}^\circ (<5^\circ)$,

$B_2 = \underline{1}^\circ (<5^\circ)$

$\gamma = \underline{4.0}$ $\theta = \underline{0.0}$

$Z = A \sin \gamma = \underline{0.07}$ cm (in.); < 0.32 cm ($< 1/8$ in.),

$W = A \sin \theta = \underline{0.0}$ cm (in.); < 0.08 cm ($< 1/32$ in.)

$P_A \underline{.58}$ cm (in.) $P_B \underline{.58}$ cm (in.)

$D_t = \underline{.313}$ cm (in.)

Comments: Pitot tube meets all baseline criteria \therefore γ

C_p of .84 may be assumed

JS

Calibration required? No

THERMOCOUPLE CALIBRATION DATA SHEET

Date: 1-12-87 Thermocouple No.: NUTECH 3
 Ambient Temperature: 69°F °F Barometric pressure: 30.00 "Hg
 Calibrator: J. SMITH Reference: Mercury-in-glass:
 Other: _____

Reference point No. ^a	Source* (specify)	Reference thermometer temperature, °F	Thermocouple potentiometer temperature, °F	Difference, % ^b
32°	ICE BATH	32	32	0.0
100	HOT OIL	100	98	2.0
200	↓	200	199	1.0
300		300	300	0.0
400		400	400	0.0
			Average % difference	0.60

^a Every 100°F for each reference point when using furnace up to 500°F.
 * Source: 1) Ice bath
 2) Furnace
^b Percent difference ≤ 1.5%

$$\frac{\text{Ref. temp. } ^\circ\text{F} - \text{thermocouple temp. } ^\circ\text{F}}{\text{Ref. temp. } ^\circ\text{F}} \times 100$$

NOZZLE CALIBRATION

Date 6/4/87Calibrated by Jim Eckenrode

Nozzle identification number.	D_1 , in.	D_2 , in.	D_3 , in.	ΔD , in.	D_{avg}
(1/4")	.255	.254	.254	.001	.254

where:

$D_{1,2,3}$ = nozzle diameter measured of a different diameter, in.
Tolerance = measure within 0.001 in.

ΔD = maximum difference in any two measurements, in.
Tolerance = 0.004 in.

D_{avg} = average of D_1 , D_2 , and D_3 .

ETS, INC.

DRY GAS METER AND ORIFICE
POST TEST CALIBRATION

Date 6/16/87

Box No. #3

Barometric pressure, $P_b = \overset{29.95}{\text{in. Hg}}$

Dry gas meter No. 710917

$\Delta H = \underline{1.67}$ in. H₂O

Vac. = 10" Hg

Orifice manometer setting, ΔH , in. H ₂ O	Gas volume wet test meter V_w , ft ³	Gas volume dry gas meter V_d , ft ³	Temperature				Time θ , min	γ	$\Delta H\theta$
			Wet test		Dry gas meter				
			Meter t_w , °F	Inlet t_{di} , °F	Outlet t_{do} , °F	Average t_d , °F			
1.67	13.883	14.396	70	92	94	93	21	1.002	2.054
1.67	12.619	13.086	70	93	94	93.5	19	1.003	2.034
1.67	12.624	13.090	70	94	95	94.5	19	1.005	2.028
Average								1.003	2.039

Calculations

ΔH	$\frac{\Delta H}{13.6}$	γ	$\Delta H\theta$
		$\frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \theta}{V_w} \right]^2$
1.67	.1228	1.002	2.054
1.67	.1228	1.003	2.034
1.67	.1228	1.005	2.028

γ = Ratio of accuracy of wet test meter to dry test meter. Tolerance = ± 0.01

$\Delta H\theta$ = Orifice pressure differential that gives 0.75 cfm of air at 70 F and 29.92 inches of mercury, in. H₂O. Tolerance = ± 0.15

JE.

APPENDIX B

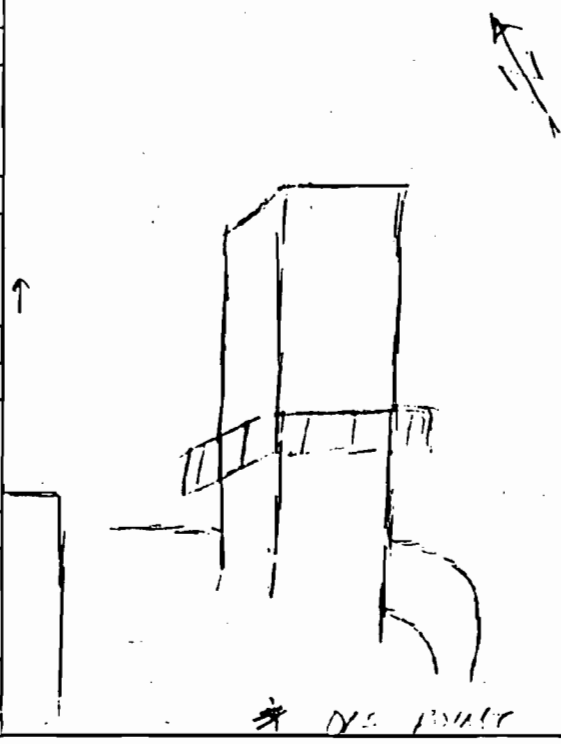
**OPACITY MEASUREMENTS FOR
MAY 12, 13, AND 14, 1987
COMPLIANCE TESTS**

BEST AVAILABLE COPY

INITIAL FINAL

OBSERVER LOCATION		
DISTANCE TO DISCHARGE	350'	
DIRECTION TO DISCHARGE	N-NW	
HEIGHT OF OBSERVATION POINT	6.5 FT	
BACKGROUND DESCRIPTION		
	LIGHT BLUE	
WEATHER CONDITIONS		
WIND DIRECTION	S-SE	
WIND SPEED	0-5 MPH	
AMBIENT TEMPERATURE	75 OF	
SKY CONDITIONS		
	PARTLY CLOUDY	
PLUME DESCRIPTION		
COLOR	Light Gray w/ some	
DISTANCE VISIBLE	6.5 FT	

DIAGRAM OF OBSERVER AND EMISSION POINT



COMMENTS

PARTLY CLOUDY - PLUME NOW APPEARS LIGHT GRAY w/ SOME BROWN.

OBSERVER SIGNATURE

John Richardson

ETS, INC.

Date: 5/12/87
 COMPANY Bay County Resource REGISTRATION NO. _____
 LOCATION PANAMA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT = 150'
 OBSERVER JOHN RICHARDSON CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 2 : 15 A.M. (P.M.) FINAL 2 : 35 A.M. (P.M.)
Run 1, Unit 2

VISIBLE EMISSION READINGS

HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT
	0							
	1							
	2							
	3							
	4							
	5							
	6							
	7							
	8							
	9							
	10							
	11							
	12							
	13							
	14							
<u>2</u>	<u>15</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>16</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>17</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>18</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>19</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>20</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>21</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>22</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>23</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>24</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>25</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>26</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>27</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>28</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>29</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			

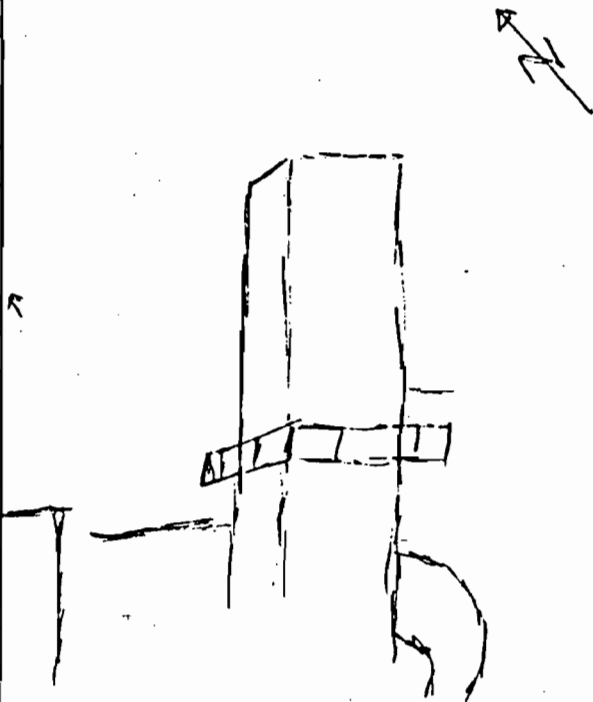
HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT
	<u>30</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>31</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>32</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>33</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>34</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>35</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	<u>36</u>							
	<u>37</u>							
	<u>38</u>							
	<u>39</u>							
	<u>40</u>							
	<u>41</u>							
	<u>42</u>							
	<u>43</u>							
	<u>44</u>							
	<u>45</u>							
	<u>46</u>							
	<u>47</u>							
	<u>48</u>							
	<u>49</u>							
	<u>50</u>							
	<u>51</u>							
	<u>52</u>							
	<u>53</u>							
	<u>54</u>							
	<u>55</u>							
	<u>56</u>							
	<u>57</u>							
	<u>58</u>							
	<u>59</u>							

BEST AVAILABLE COPY

INITIAL FINAL

OBSERVER LOCATION		
DISTANCE TO DISCHARGE	750'	
DIRECTION TO DISCHARGE	N. NE	
HEIGHT OF OBSERVATION POINT	GROUND	
BACKGROUND DESCRIPTION		
	GRAY WHITE	
	CLOUDS	
WEATHER CONDITIONS		
WIND DIRECTION	SOUTH	
WIND SPEED	0-5 MPH	
AMBIENT TEMPERATURE	85°F	
SKY CONDITIONS		
	CLOUDY	
PLUME DESCRIPTION		
COLOR	FAINT GRAY	
DISTANCE VISIBLE	5-10'	

DIAGRAM OF OBSERVER AND EMISSION POINT



* OBSERVATION PT.

COMMENTS

HAS BECOME COMPLETELY CLOUDY NOW. BACKGROUND IS STILL SOMEWHAT LIGHT.

OBSERVER SIGNATURE

John Richard

ETS, INC.

DATE 5/13/87
 COMPANY BAV Co. Resources REGISTRATION NO. _____
 LOCATION Prattville City, AL
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT ≈ 150'
 OBSERVER JOHN RICHARDSON CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 5 : 10 A.M./P.M. FINAL 5 : 35 A.M./P.M.

VISIBLE EMISSION READINGS

RUN 2 BOILER 2

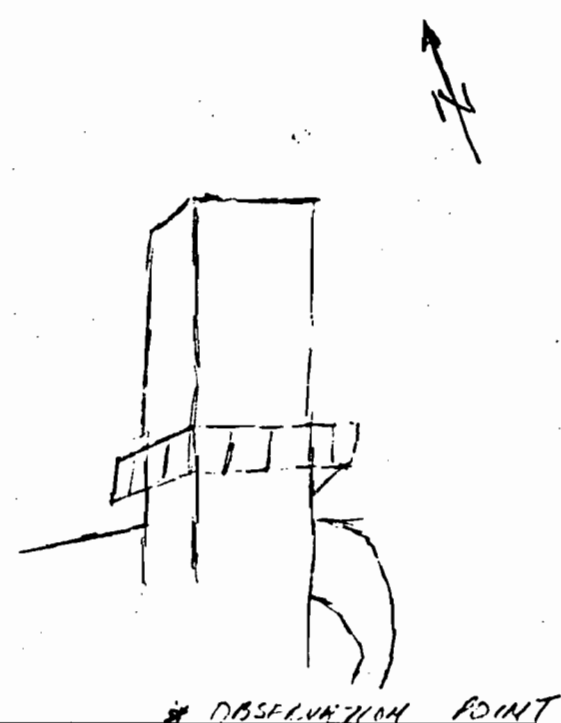
HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT			0	15	30	45	DET.	ATT.	COMMENT
	0									30	5	5	5	5			
	1									31	5	5	5	5			
	2									32	5	5	5	5			
	3									33	5	5	5	5			
	4									34	5	5	5	5			
	5									35	5	5	5	5			
	6									36							
	7									37							
	8									38							
	9									39							
5	10	5	10	5	10					40							
	11	5	5	5	5					41							
	12	5	5	5	5					42							
	13	5	5	5	5					43							
	14	5	5	5	5					44							
	15	5	5	5	5					45							
	16	5	5	5	5					46							
	17	5	5	5	5					47							
	18	5	5	5	5					48							
	19	5	5	5	5					49							
	20	5	5	5	5					50							
	21	5	5	5	5					51							
	22	5	5	5	5					52							
	23	5	5	5	5					53							
	24	5	5	5	5					54							
	25	5	5	5	5					55							
	26	5	5	5	5					56							
	27	5	5	5	5					57							
	28	5	5	5	5					58							
	29	5	5	5	5					59							

BEST AVAILABLE COPY


INITIAL FINAL

OBSERVER LOCATION		
DISTANCE TO DISCHARGE	350'	
DIRECTION TO DISCHARGE	N-NE	
HEIGHT OF OBSERVATION POINT	GROUND	
BACKGROUND DESCRIPTION		
	BLUE-GRAY SKY	
WEATHER CONDITIONS		
WIND DIRECTION	SOUTH	
WIND SPEED	0-5 MPH	
AMBIENT TEMPERATURE	75°F	
SKY CONDITIONS		
	MODERATELY CLOUDY	
PLUME DESCRIPTION		
COLOR	LIGHT GRAY	
DISTANCE VISIBLE		

DIAGRAM OF OBSERVER AND EMISSION POINT



COMMENTS


 READINGS AGAIN VERY LOW W/ EXCEPTION OF @ 7:29 PM
 WHEN OPACITY ROSE TO ≈ 10%. BACKGROUND IMPROVED
 OVER RUN 2 @ 1:50 PM.
 RUN 3 ↑ Boiler @ 9:10

OBSERVER SIGNATURE

John Richards

ETS, INC.

D. 5/12/87
 COMPANY BAY COUNTY RESOURCES REGISTRATION NO. _____
 LOCATION PANAMA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT 150'
 OBSERVER JOHN RICHARDSON CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 7 : 10 A.M. (P.M.) FINAL 7 : 30 A.M. (P.M.)
RUN 3 BOILER 2

VISIBLE EMISSION READINGS

HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT
	0							
	1							
	2							
	3							
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	8							
	9							
<u>7</u>	10	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	11	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	12	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	13	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	14	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	15	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	16	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	17	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	18	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	19	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	20	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	21	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	22	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	23	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	24	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	25	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	26	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	27	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>			
	28	<u>5</u>	<u>5</u>	<u>5</u>	<u>10</u>			
	29	<u>10</u>	<u>10</u>	<u>10</u>	<u>5</u>			


HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT
	30							
	31							
	32							
	33							
	34							
	35							
	36							
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BEST AVAILABLE COPY

	INITIAL	FINAL	
OBSERVER LOCATION			<p>DIAGRAM OF OBSERVER AND EMISSION POINT</p>
● DISTANCE TO DISCHARGE	500'		
● DIRECTION TO DISCHARGE	N-NW		
● HEIGHT OF OBSERVATION POINT	GROUND		
BACKGROUND DESCRIPTION			
	BLUE SKY w/		
	SOME CLOUDS		
WEATHER CONDITIONS			
● WIND DIRECTION	EAST		
● WIND SPEED	0-5 MPH		
● AMBIENT TEMPERATURE	75°F		
SKY CONDITIONS			
	PARTLY CLOUDY		
PLUME DESCRIPTION			
● COLOR	FAINT WHITE		
● DISTANCE VISIBLE	15-20'		

* OBS. POINT

COMMENTS


 MOSTLY SUNNY TODAY. PLUME IS LIGHTER COLOR
 NOW - MORE WHITE BUT STILL VERY LIGHT NOT
 EXCEEDING 5%.

OBSERVER SIGNATURE

John Richardson

Best Available Copy

ETS, INC.

Date: 5/14/87
 COMPANY BAY COUNTY RESOURCES REGISTRATION NO. _____
 LOCATION PANAMA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT ≈ 150'
 OBSERVER JOHN RICHMOND CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 8 : 35 A.M./P.M. FINAL 9 : 00 A.M./P.M.

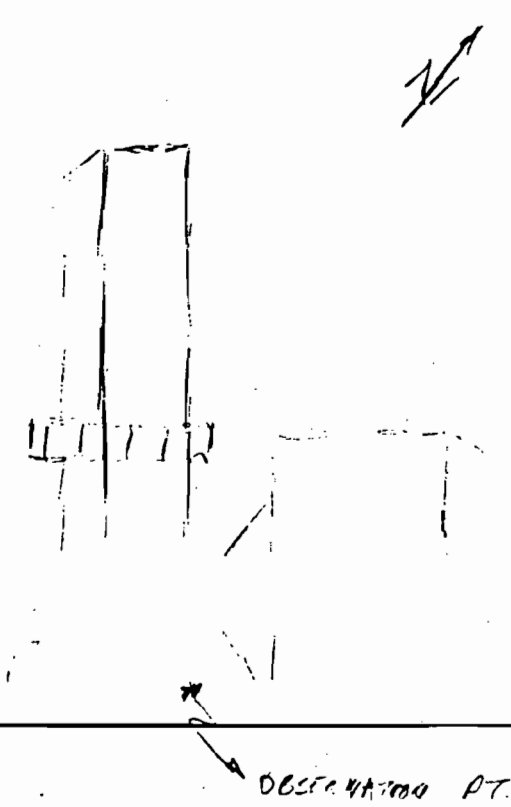
VISIBLE EMISSION READINGS
RUN 4 BOILER 2

HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE		
		0	15	30	45	DET.	ATT.	COMMENT			0	15	30	45	DET.	ATT.	COMMENT
	0									30							
	1									31							
	2									32							
	3									33							
	4									34							
	5								8	35	5	5	5	5			
	6									36	5	5	5	5			
	7									37	5	5	5	5			
	8									38	5	5	5	5			
	9									39	5	5	5	5			
	10									40	5	5	5	5			
	11									41	5	5	5	5			
	12									42	5	5	5	5			
	13									43	5	5	5	5			
	14									44	5	5	5	5			
	15									45	5	5	5	5			
	16									46	5	5	5	5			
	17									47	5	5	5	5			
	18									48	5	5	5	5			
	19									49	5	5	5	5			
	20									50	5	5	5	5			
	21									51	5	5	5	5			
	22									52	5	5	5	5			
	23									53	5	5	5	5			
	24									54	5	5	5	5			
	25									55	5	5	5	5			
	26									56	5	5	5	5			
	27									57	5	5	5	5			
	28									58	5	5	5	5			
	29									59	5	5	5	5			


BEST AVAILABLE COPY

	INITIAL	FINAL
OBSERVER LOCATION		
DISTANCE TO DISCHARGE	500'	
DIRECTION TO DISCHARGE	N-ALLW	
HEIGHT OF OBSERVATION POINT	GROUND	
BACKGROUND DESCRIPTION		
	BLUE SKY	
WEATHER CONDITIONS		
WIND DIRECTION	S-SE	
WIND SPEED	5-10 MPH	
AMBIENT TEMPERATURE	75°F	
SKY CONDITIONS		
	PARTLY CLOUDY	
PLUME DESCRIPTION		
COLOR	GRAY	
DISTANCE VISIBLE	5-10'	

DIAGRAM OF OBSERVER AND EMISSION POINT



COMMENTS

EMISSIONS ARE QUITE LOW. 

BACKGROUND PRETTY GOOD, MOSTLY CLEAR SKY.

OBSERVER SIGNATURE *John Richardson*

Best Available Copy

ETS, INC.

Date 5/13/87
 COMPANY BAY COUNTY REFINERS REGISTRATION NO. _____
 LOCATION MARIETTA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT 2150'
 OBSERVER JOHN RICHMOND CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 9 : 35 A.M./P.M. FINAL 10 : 00 A.M./P.M.
RUN 1 BOILER 1

VISIBLE EMISSION READINGS

HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			
		0	15	30	45	DET.	ATT.	COMMENT			0	15	30	45	DET.	ATT.	COMMENT	
	0																	
	1																	
	2																	
	3																	
	4																	
	5																	
	6																	
	7																	
	8																	
	9								9	35	5	5	5	5				
	10									36	5	5	5	5				
	11									37	5	5	5	5				
	12									38	5	5	5	5				
	13									39	5	5	5	5				
	14									40	5	5	5	5				
	15									41	5	5	5	5				
	16									42	5	5	5	5				
	17									43	5	5	5	5				
	18									44	5	5	5	5				
	19									45	5	5	5	5				
	20									46	5	5	5	5				
	21									47	5	5	5	5				
	22									48	5	5	5	5				
	23									49	5	5	5	5				
	24									50	5	5	5	5				
	25									51	5	5	5	5				
	26									52	5	5	5	5				
	27									53	5	5	5	5				
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	29									55	5	5	5	5				
										56	5	5	5	5				
										57	5	5	5	5				
										58	5	5	5	5				
										59	5	5	5	5				

BEST AVAILABLE COPY

	INITIAL	FINAL	
OBSERVER LOCATION			<p align="center">DIAGRAM OF OBSERVER AND EMISSION POINT</p>
STANCE TO DISCHARGE	500'		
DIRECTION TO DISCHARGE	N-11W		
HEIGHT OF OBSERVATION POINT	GROUND		
BACKGROUND DESCRIPTION	WHITE CLOUDS		
WEATHER CONDITIONS			
WIND DIRECTION	SE		
WIND SPEED	0-5 MPH		
AMBIENT TEMPERATURE	80°F		
SKY CONDITIONS	MOSTLY CLOUDY		
PLUME DESCRIPTION			
COLOR	FAINT GRAY		
DISTANCE VISIBLE	5-10'		

COMMENTS

HAS BECOME MORE CLOUDY THAN RUN 1, IS NOW MOSTLY CLOUDY. EMISSIONS AGAIN VERY LOW + NEARLY NONDETECTABLE!

OBSERVER SIGNATURE

John Richardson

ETS, INC.

DATE 5/13/87
 COMPANY BAY COUNTY RESOURCES REGISTRATION NO. _____
 LOCATION PRINNA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT 150'
 OBSERVER JOHN RICHARDSON CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 11 : 45 A.M./P.M. FINAL 12 : 10 A.M./P.M.

RUN 2. BOLLER 1

VISIBLE EMISSION READINGS

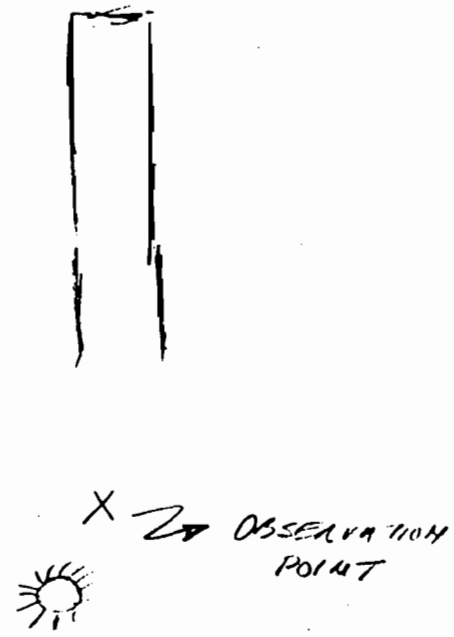
HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE					
		0	15	30	45	DET.	ATT.	COMMENT			0	15	30	45	DET.	ATT.	COMMENT			
<u>12</u>	<u>0</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>1</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>2</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>3</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>4</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>6</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>7</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>8</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>9</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>10</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>															
	<u>11</u>																			
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									<u>11</u>	<u>45</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>46</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>47</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>48</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>49</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>50</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>51</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>52</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>53</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>54</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>55</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>56</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>57</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						
										<u>58</u>	<u>5</u>	<u>10</u>	<u>5</u>	<u>5</u>						
										<u>59</u>	<u>5</u>	<u>5</u>	<u>5</u>	<u>5</u>						

BEST AVAILABLE COPY

INITIAL FINAL

OBSERVER LOCATION		
DISTANCE TO DISCHARGE	400'	
DIRECTION TO DISCHARGE		
HEIGHT OF OBSERVATION POINT	GROUND	
BACKGROUND DESCRIPTION		
	WHITE CLOUDS	
WEATHER CONDITIONS		
WIND DIRECTION		
WIND SPEED	10-15 MPH	
AMBIENT TEMPERATURE	85°	
SKY CONDITIONS		
	Mostly Cloudy CLOUDY	
PLUME DESCRIPTION		
COLOR	HARDLY	
DISTANCE VISIBLE	DISCERNABLE	

DIAGRAM OF OBSERVER AND EMISSION POINT



COMMENTS

EMISSIONS ARE VERY LOW, JUST BARELY DETECTABLE
 BACKGROUND WAS FAIRLY GOOD AT DUSK OF READING BUT
 CHANGED TO MEDIUM GRAY CLOUDS. FAINT GRAY EMISSIONS
 OBSERVED THROUGHOUT READINGS.

OBSERVER SIGNATURE

John Richardson

ETS, INC.

D. 5/13/87
 COMPANY BAY COUNTY RESOURCES REGISTRATION NO. _____
 LOCATION PANAMA CITY, FL.
 EMISSION POINT NAME _____ HEIGHT TO DISCHARGE POINT ≈ 150'
 OBSERVER JOAN RICHARDSON CERTIFICATION EXPIRATION DATE 10/87

CLOCK TIME: INITIAL 1 : 50 A.M. (P.M.) FINAL 2 : 20 A.M. (P.M.)
RUN 3 BOILER 1

VISIBLE EMISSION READINGS

HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			HR.	MIN.	SECONDS				STEAM PLUME CHECK IF APPLICABLE			
		0	15	30	45	DET.	ATT.	COMMENT			0	15	30	45	DET.	ATT.	COMMENT	
2	0	5	5	5	5													
	1	5	5	5	5													
	2	5	5	5	5													
	3	5	5	5	5													
	4	5	5	5	5													
	5	5	5	5	5													
	6	5	5	5	5													
	7	5	5	5	5													
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	9	5	5	5	5													
	10	5	5	5	5													
	11	5	5	5	5													
	12	5	5	5	5													
	13	5	5	5	5													
	14	5	5	5	5													
	15	5	5	5	5													
	16	5	5	5	5													
	17	5	5	5	5													
	18	5	5	5	5													
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										59	5	5	5	5				

APPENDIX C

**COMPUTER CALCULATION SHEETS FOR
ADDITIONAL TEST RESULTS
REPORTED IN TABLE 2**

%CO2 avg.	13	pmrd	4.075032751833441
%O2 avg.	6	pmrw	4.075032751833441
%CO avg.	0	csd	1.906570629048159D-02
As stack area	15.32071832265625	csw	1.906570629048159D-02
Vm avg. DGM	47.149	csd50%	1.766775282446131D-02
Dp avg. delta p	.4908333333333333	csw50%	1.766775282446131D-02
ts avg. temp	373.75	csd12	1.759911349890608D-02
DH avg. delta H	2.389166666666667	csw12	1.759911349890608D-02
tmi avg. temp	92.41666666666667	csd7	1.778612865433382D-02
tmo avg. temp	91.08333333333333	csw7	1.778612865433382D-02
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1329112443417778
ts mole wt.	28.68253348934865	Vm corrected	47.149
area nozzle	4.97440652865625D-04	total # points	12
cm temp meter	91.75	Vf impingers	326
Tm abs temp	551.42	Vi impingers	200
Ts abs temp	833.42	sqr(Dp) avg.	.6988695723679337
Pm abs prs	30.11567401468032	Bws kick flag	0
Ps abs prs	29.9105882361191	Vm C kick flag	0
Vm(std)	45.45874602257414	not in use	NA
Vwc(std)	5.93082	Wa2	0
Vwsg(std)	1.0373	Wa5	0
Bws	.1329112443417778	K1 constant	.04707
vs velocity	49.46539289227205	K2 constant	.04715
vol rate	24954.98114192209	Tstd abs temp	527.67
run time min.	60	container 1	1.140000000000002D-02
%I isokinetic	93.50757805391747	container 2	4.476999999999975D-02
Qa vol rate	45470.72107332136	container 3	NA
md mass dry	5.616999999999978D-02	container 4	22
mw mass wet	5.616999999999978D-02	container 5	NA
	39.00156035616007	not in use	NA

* Definitions & units of the variables are given in the sample calculations

%CO2 avg.	13	omrd	6.683764211788342
%H2O avg.	6	omrw	6.683764211788342
%O2 avg.	0	csd	3.0207266809406780-02
As stack area	15.32071832265625	csw	3.0207266809406780-02
Vm avg. DGM	25.383000000000001	csd50%	2.7992381470662580-02
Dp avg. delta p	.58	csw50%	2.7992381470662580-02
ts avg. temp	387.33333333333333	csd12	2.7883630900990880-02
DH avg. delta H	2.7783333333333333	csw12	2.7883630900990880-02
tmi avg. temp	89.33333333333333	csd7	.0281799334145898
tmo avg. temp	87	csw7	.0281799334145898
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1748840164916596
Ms mole wt.	28.16542893551144	Vm corrected	25.383000000000001
An area nozzle	4.97440652865625D-04	total # points	6
tm temp meter	88.16666666666667	Vf impingers	296
Tm abs temp	547.8366666666667	Vi impingers	200
Ts abs temp	847.00333333333333	sqr(Dp) avg.	.7595053951150426
Tm abs prs	30.14428920995612	Bws kick flag	0
Ts abs prs	29.91132353021612	Vm C kick flag	0
Tstd	24.65652072063845	not in use	NA
Vwc(std)	4.51872	Wa2	0
Vwsg(std)	.70725	Wa5	0
Bws	.1748840164916596	K1 constant	.04707
vs velocity	54.68799812415877	K2 constant	.04715
Qs vol rate	25833.82598045883	Tstd abs temp	527.67
run time min.	30	container 1	1.071999999999998D-02
%I isokinetic	97.98501469721418	container 2	3.7549999999999953D-02
Qd vol rate	50271.56489341139	container 3	NA
md mass dry	4.8269999999999951D-02	container 4	15
mw mass wet	4.8269999999999951D-02	container 5	NA
WEA	39.00156035616007	not in use	NA

4/27 1505 #1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

%CO2 avg.	13	pmrd	6.123921122589878
%O2 avg.	6	pmrw	6.123921122589878
%CO avg.	0	csd	2.867596632509704D-02
As stack area	15.320718322265625	csw	2.867596632509704D-02
Vm avg. DGM	47.399	csd50%	2.657336042606127D-02
Dp avg. delta p	.5158333333333333	csw50%	2.657336042606127D-02
ts avg. temp	440.6666666666667	csd12	2.647012276162804D-02
DH avg. delta H	2.324166666666667	csw12	2.647012276162804D-02
tmi avg. temp	85.5	csd7	2.675140477749577D-02
tmo avg. temp	82.9166666666667	csw7	2.675140477749577D-02
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1234471419635653
Ms mole wt.	28.79913123086259	Vm corrected	47.399
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	84.20833333333333	Vf impingers	393
abs temp	543.8783333333333	Vi impingers	300
abs temp	900.3366666666667	sqr(Dp) avg.	.7157758413700033
Pm abs prs	30.43089460304967	Bws kick flag	0
Ps abs prs	30.21367647188758	Vm C kick flag	0
Vm(std)	46.81845015367414	not in use	NA
Vwc(std)	4.37751	Wa2	0
Vwsg(std)	2.21605	Wa5	0
Bws	.1234471419635653	K1 constant	.04707
vs velocity	52.28566395485316	K2 constant	.04715
vol rate	24933.92032259212	Tstd abs temp	527.67
run time min.	60	container 1	3.85299999999998D-02
%I isokinetic	96.38580271252002	container 2	4.847999999999963D-02
vol rate	48063.23578592198	container 3	NA
mass dry	8.700999999999962D-02	container 4	47
mv mass wet	8.700999999999962D-02	container 5	NA
%EA	39.00156035616007	not in use	NA

%O2 avg.	13	pmrd	4.670828915775861
%O2 avg.	6	pmrw	4.670828915775861
%CO avg.	0	csd	2.729277513196588D-02
As stack area	15.32071832265625	cew	2.729277513196588D-02
Vm avg. DGM	36.99799999999999	csd50%	2.529158886528706D-02
Dp avg. delta p	.3533333333333333	esw50%	2.529158886528706D-02
ts avg. temp	418.25	csd12	2.519333089104542D-02
DH avg. delta H	1.628333333333333	cew12	2.519333089104542D-02
tmi avg. temp	94.5	csd7	2.546104521044135D-02
tmo avg. temp	91.75	cew7	2.546104521044135D-02
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1462431164718465
Ms mole wt.	28.51828482440424	Vm corrected	36.99799999999999
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	93.125	Vf impingers	383
Tn abs temp	552.795	Vi impingers	300
T abs temp	877.92	sqr(Dp) avg.	.5819842002140952
F abs prs	30.01973038879852	Bws kick flag	0
Ps abs prs	29.87352941250719	Vm C kick flag	0
Vm(std)	35.46956164476588	not in use	NA
Vwc(std)	3.90681	Wa2	0
Vwsg(std)	2.1689	Wa5	0
Bws	.1462431164718465	K1 constant	.04707
vs velocity	42.42563001861055	K2 constant	.04715
Qs vol rate	19981.37193655277	Tstd abs temp	527.67
rt time min.	60	container 1	3.245000000000001D-02
%r isokinetic	91.1207152800595	container 2	3.028899999999979D-02
Qa vol rate	38999.4676305817	container 3	NA
md mass dry	6.273899999999979D-02	container 4	46
mw mass wet	6.273899999999979D-02	container 5	NA
%EA	39.00156035616007	not in use	NA

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5/20 1542 #1

X002 avg.	12.4	pmrd	5.7850844460056904
X02 avg.	8	pmrw	5.7850844460056904
X00 avg.	0	ced	2.642635204269229D-02
As stack area	15.32071832265625	cew	2.642635204269229D-02
Vm avg. DGM	35.478000000000001	ced50%	2.844715260099988D-02
Dp avg. delta p	.5533333333333333	cew50%	2.844715260099988D-02
ts avg. temp	425.6666666666667	ced12	2.557388907357319D-02
DH avg. delta H	1.341666666666667	cew12	2.557388907357319D-02
tmi avg. temp	93.33333333333333	ced7	2.847490652518484D-02
tmo avg. temp	92.25	cew7	2.847490652518484D-02
XN2 avg.	79.6	saturation flag	-1
Md mole wt. dry	30.30400000810623	Bws saturated	.140012550570485
Ms mole wt.	28.58128558475201	Vm corrected	35.478000000000001
An area nozzle	3.74394940740625D-04	total # points	12
ts temp meter	92.79166666666667	Vf impingers	304
Ts abs temp	552.4616666666667	Vi impingers	200
Ts abs temp	885.3366666666667	egr(Dp) avg.	.7417901993534804
ts abs pre	30.0986519580172	Bws kick flag	0
ts abs pre	29.9705882361191	Vm C kick flag	0
ts abs pre	34.12234872763515	not in use	NA
ts abs pre	4.89528	Wa2	0
Vmag(std)	.6601	Wa5	0
Bws	.140012550570485	K1 constant	.04707
Vs velocity	54.15335493600144	K2 constant	.04715
vs vol rate	25559.45142488106	Tstd abs temp	527.67
run time min.	60	container 1	5.843999999999999D-02
XI isokinetic	91.05112052753139	container 2	NA
qs vol rate	49781.93631827699	container 3	NA
ms mass dry	5.843999999999999D-02	container 4	14
mw mass wet	5.843999999999999D-02	container 5	NA
XEA	61.47037181887387	not in use	NA

XO2 avg.	11.7	pmrd	4.160724421385723
XO2 avg.	8.2	pmrw	4.160724421385723
XCO avg.	0	ced	1.903243669898595D-01
As stack area	15.320718322265625	cew	1.903243669898595D-01
Vm avg. DGM	37.977999999999999	ced50%	2.072481002660821D-01
Dp avg. delta p	.60083333333333333	cew50%	2.072481002660821D-01
ts avg. temp	436.16666666666667	ced12	1.952044789639585D-01
DH avg. delta H	1.4633333333333333	cew12	1.952044789639585D-01
tmi avg. temp	83.083333333333333	ced7	2.083077722849742D-01
tmo avg. temp	82.25	cew7	2.083077722849742D-01
XN2 avg.	80.1	saturation flag -1	
Md mole wt. dry	30.20000000089407	Bws saturated	.1734456583746612
Me mole wt.	28.08396297521911	Vm corrected	37.977999999999999
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	82.666666666666667	Vf impingers	1276
Tm abs temp	542.33666666666667	Vi impingers	1133
ts abs temp	895.83666666666667	egr(Dp) avg.	.7683717843300268
ts abs pres	30.12759803619764	Bws kick flag	0
ts abs pres	29.98176470695482	Vm C kick flag	0
Vm(std)	37.24453212224598	not in use	NA
Vwc(std)	6.73101	Wa2	0
Vwg(std)	1.08445	Wa5	0
Bws	.1734456583746612	K1 constant	.04707
vs velocity	56.914462292332026	K2 constant	.04715
Qs vol rate	25524.28579381761	Tstd abs temp	527.67
run time min.	60	container 1	2.462999999999999D-02
isokinetic	105.8868380398462	container 2	2.130999999999997D-02
Da vol rate	52318.22671596475	container 3	NA
md mass dry	.04593999999999997	container 4	23
mw mass wet	.04593999999999997	container 5	NA
Wsg	63.33807137563551	not in use	NA

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6/11 2029 #1

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06-16-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

Q2 avg. 10.3
 Q02 avg. 9.6
 Q00 avg. 0
 Ae stack area 15.32071832265625
 Vm avg. DGM 35.371000000000001
 Dp avg. delta p .52916666666666667
 ts avg. temp 442.5
 DH avg. delta H 1.2741666666666667
 tmi avg. temp 85
 tmp avg. temp 82.416666666666667
 XN2 avg. 80.1
 Md mole wt. dry 30.032000000226498
 Ms mole wt. 27.95913401979631
 An area nozzle 3.51880193415625D-04
 tm temp meter 83.70833333333333
 t abs temp 543.37833333333333
 t abs temp 902.17
 Pm abs pre 30.1136887228623
 Ps abs pre 29.98470588334291
 Vm(std) 34.60539969155991
 Vwc(std) 6.63687
 Vweg(std) .5658
 Bws .1722794200530634
 vs velocity 53.16074185230925
 Qs vol rate 23709.22679019665
 run time min. 60
 isokinetic 105.9155033195273
 Qa vol rate 48867.6451045604
 md mass dry 4.763999999999882D-02
 mw mass wet 4.763999999999882D-02
 TQ 83.14279858023556

pmrd 4.31352340952869
 pmrw 4.31352340952869
 ced 2.124192197032377D-02
 cew 2.124192197032377D-02
 ced50% 2.593536691245391D-02
 cew50% 2.593536691245391D-02
 ced12 2.474787025668789D-02
 cew12 2.474787025668789D-02
 ced7 2.612944400459664D-02
 cew7 2.612944400459664D-02
 saturation flag -1
 Bws saturated .1722794200530634
 Vm corrected 35.371000000000001
 total # points 12
 Vf impingers 1222
 Vi impingers 1081
 sqr(Dp) avg. .7136150383847175
 Bws kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 2.579000000000001D-02
 container 2 2.184999999999881D-02
 container 3 NA
 container 4 12
 container 5 NA
 not in use NA

%CO2 avg.	13	pmsd	4.303796045924327
%O2 avg.	6	pmsw	4.303796045924327
%CO avg.	0	csd	1.740908012526307D-02
As stack area	15.32071832265625	csw	1.740908012526307D-02
Vm avg. DGM	53.721	csd50%	1.613259534517987D-02
Dp avg. delta p	.725	csw50%	1.613259534517987D-02
ts avg. temp	422.3333333333333	csd12	1.606992011562744D-02
DH avg. delta H	3.325833333333333	csw12	1.606992011562744D-02
tmi avg. temp	90.25	csd7	1.624068545607043D-02
tmo avg. temp	85	csw7	1.624068545607043D-02
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1515834990417485
Ms mole wt.	28.45249131102209	Vm corrected	53.721
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	87.625	Vf impingers	362
Tm abs temp	547.295	Vi impingers	200
Ts abs temp	882.0033333333333	sqr(Dp) avg.	.8455823430303823
Fm abs prs	30.2645465617681	Bws kick flag	0
Ps abs prs	29.97220589369353	Vm C kick flag	0
Vm(std)	52.44350037054014	not in use	NA
(std)	7.62534	Wa2	0
sg(std)	1.74455	Wa5	0
Bws	.1515834990417485	K1 constant	.04707
velocity	61.75414579682566	K2 constant	.04715
vol rate	28863.89253254145	Tstd abs temp	527.67
run time min.	60	container 1	2.523999999999998D-02
%I isokinetic	93.26600518257677	container 2	3.392999999999997D-02
Qa vol rate	56767.07238056474	container 3	NA
md mass dry	5.9169999999999978D-02	container 4	37
mw mass wet	5.9169999999999978D-02	container 5	NA
%EA	39.00156035616007	not in use	NA

4/23 1148 #2

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

%CO2 avg. 13
 %O2 avg. 6
 %CO avg. 0
 As stack area 15.32071832265625
 Vm avg. DGM 46.967
 Dp avg. delta p .5208333333333333
 ts avg. temp 422.3333333333333
 DH avg. delta H 2.395
 tmi avg. temp 90.09090909090909
 tmo avg. temp 90.54545454545455
 %N2 avg. 81
 Md mole wt. dry 30.32000002264977
 Ms mole wt. 28.37725967539609
 area nozzle 4.97440652865625D-04
 temp meter 90.31818181818182
 Tm abs temp 549.9881818181818
 Ts abs temp 882.0033333333333
 Pm abs prs 30.19610293623692
 Ps abs prs 29.9905882361191
 Vm(std) 45.52241066555498
 Vwc(std) 6.96636
 Vwsg(std) 1.55595
 Bws .1576899629612044
 velocity 52.40819890147679
 vol rate 24334.2049187623
 run time min. 60
 %I isokinetic 96.02729476090306
 Qa vol rate 48175.87519003611
 md mass dry 6.8819999999999949D-02
 mw mass wet 6.8819999999999949D-02
 39.00156035616007

pmrd 4.861759395706837
 pmrw 4.861759395706837
 csd .0233268094653758
 csw .0233268094653758
 csd50% 2.161641942545387D-02
 csw50% 2.161641942545387D-02
 csd12 2.153243950650074D-02
 csw12 2.153243950650074D-02
 csd7 2.176125174305441D-02
 csw7 2.176125174305441D-02
 saturation flag -1
 Bws saturated .1576899629612044
 Vm corrected 46.967
 total # points 12
 Vf impingers 348
 Vi impingers 200
 sqr(Dp) avg. .7168812763440377
 Bws kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 .02165
 container 2 4.7169999999999949D-02
 container 3 NA
 container 4 33
 container 5 NA
 not in use NA

INLET

%CO2 avg.	13	pmrd	4.183341112101452
%O2 avg.	6	pmrw	4.183341112101452
%CO avg.	0	csd	2.084698151429925D-02
As stack area	15.32071832265625	csd	2.084698151429925D-02
Vmf ² avg. DGM	45.075999999999999	csd50%	1.931841972802413D-02
Dp avg. delta p	.47083333333333333	csd50%	1.931841972802413D-02
ts avg. temp	405.25	csd12	1.924336755166084D-02
DH avg. delta H	2.205	csd12	1.924336755166084D-02
tmi avg. temp	90	csd7	1.944785520235136D-02
tmo avg. temp	87.666666666666667	csd7	1.944785520235136D-02
%N2 avg.	81	csd7	1.944785520235136D-02
Md mole wt. dry	30.32000002264977	saturation flag	-1
Ms mole wt.	28.40658250598094	Bws saturated	.155309863080446
an area nozzle	4.97440652865625D-04	Vm corrected	45.075999999999999
temp meter	88.833333333333333	total # points	12
bs temp	548.50333333333333	Vf impingers	345
abs temp	864.92	Vi impingers	200
Pm abs prs	30.18213234839349	sqr(Dp) avg.	.6819317183956953
Ps abs prs	29.9905882361191	Bws kick flag	0
Vm(std)	43.78757660306248	Vm C kick flag	0
Vwc(std)	6.82515	not in use	NA
Vwsg(std)	1.2259	Wa2	0
Bws	.155309863080446	Wa5	0
vr velocity	49.34253726153965	K1 constant	.04707
vol rate	23429.2905497124	K2 constant	.04715
run time min.	60	Tstd abs temp	527.67
%I isokinetic	95.93528605455597	container 1	1.313999999999999D-02
Qa vol rate	45357.78688255315	container 2	4.6020000000000039D-02
mass dry	5.9160000000000038D-02	container 3	NA
mass wet	5.9160000000000038D-02	container 4	26
	39.00156035616007	container 5	NA
		not in use	NA

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4/30 0957 #2

%CO2 avg.	13	pmsd	4.240091853977272
%O2 avg.	6	pmsw	4.240091853977272
%CO avg.	0	csd	1.809104990973947D-02
As stack area	15.32071832265625	csw	1.809104990973947D-02
Vm avg. DGM	49.375	csd50%	1.676456110623303D-02
Dp avg. delta p	.6008333333333333	csw50%	1.676456110623303D-02
ts avg. temp	426.8333333333333	csd12	1.669943068591336D-02
DH avg. delta H	2.746666666666667	csw12	1.669943068591336D-02
tmi avg. temp	92.08333333333333	csd7	1.687688545518199D-02
tmo avg. temp	88.08333333333333	csw7	1.687688545518199D-02
%N2 avg.	81	saturation flag	-1
Md mole wt. dry	30.32000002264977	Bws saturated	.1090675305335957
Ms mole wt.	28.97628804400551	Vm corrected	49.375
an area nozzle	4.97440652865625D-04	total # points	12
an temp meter	90.08333333333333	Vf impingers	372
Tm abs temp	549.7533333333333	Vi impingers	300
Ts abs temp	886.5033333333333	sqr(Dp) avg.	.7738064284392416
Pm abs prs	30.09196077864888	Bws kick flag	0
Ps abs prs	29.8605882361191	Vm C kick flag	0
Vm(std)	47.71166982051774	not in use	NA
Vwc(std)	3.38904	Wa2	0
Vwsg(std)	2.4518	Wa5	0
Bws	.1090675305335957	K1 constant	.04707
velocity	56.2466379644777	K2 constant	.04715
vol rate	27364.69008279531	Tstd abs temp	527.67
time min.	60	container 1	.03489
%i isokinetic	89.49951570796815	container 2	2.104999999999979D-02
Qa vol rate	51704.33381101117	container 3	NA
md mass dry	.0559399999999998	container 4	52
mw mass wet	.0559399999999998	container 5	NA
%EA	39.00156035616007	not in use	NA

5/12 1353 UNIT #2

%CO2 avg.	10.233333333333333	pmrd	4.615307284361401
%O2 avg.	9.7	pmrw	4.615307284361401
%CO avg.	0	csd	2.096669303877474D-02
As stack area	15.32071832265625	csw	2.096669303877474D-02
Vm avg. DGM	48.067	csd50%	2.583208440074983D-02
Dp avg. delta p	.6566666666666667	csw50%	2.583208440074983D-02
ts avg. temp	436.8333333333333	csd12	2.458635014319862D-02
DH avg. delta H	2.669166666666667	csw12	2.458635014319862D-02
tmi avg. temp	90.16666666666667	csd7	2.602116385420505D-02
tmo avg. temp	90.75	csw7	2.602116385420505D-02
%N2 avg.	80.06666666666667	saturation flag	-1
Md mole wt. dry	30.025333333500226	Bws saturated	.1958879733752114
Ms mole wt.	27.6697151588473	Vm corrected	48.067
an area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	90.45833333333333	Vf impingers	408
Tm abs temp	550.1283333333333	Vi impingers	200
Ts abs temp	896.5033333333333	sqr(Dp) avg.	.790002785072928
Pm abs prs	30.19626224939695	Bws kick flag	0
Ps abs prs	29.95808823646971	Vm C kick flag	0
Vm(std)	46.57695413358648	not in use	NA
Vwc(std)	9.79056	Wa2	0
Vwsg(std)	1.55595	Wa5	0
Ews	.1958879733752114	K1 constant	.04707
vs velocity	58.99833938977261	K2 constant	.04715
Qr vol rate	25700.98245786192	Tstd abs temp	527.67
t time min.	60	container 1	3.518000000000002D-02
isokinetic	93.02677438327585	container 2	2.810999999999986D-02
qa vol rate	54233.81635771087	container 3	NA
md mass dry	6.328999999999987D-02	container 4	33
mw mass wet	6.328999999999987D-02	container 5	NA
%EA	84.80800252794236	not in use	NA

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5/13 1632 UNIT #2

%CO2 avg.	10.4	pmrd	6.258070821786027
%O2 avg.	9.6	pmrw	6.258070821786027
%CO avg.	.15	csd	3.080595149363257D-02
As stack area	15.32071832265625	csw	3.080595149363257D-02
Vm avg. DGM	47.911	csd50%	3.746599181721248D-02
Dp avg. delta p	.5258333333333333	csw50%	3.746599181721248D-02
ts avg. temp	408.8333333333333	csd12	3.554532864649912D-02
DH avg. delta H	2.455833333333333	csw12	3.554532864649912D-02
tmi avg. temp	100.5833333333333	csd7	3.789404676684836D-02
tmo avg. temp	98.0833333333333	csw7	3.789404676684836D-02
%N2 avg.	79.85	saturation flag	-1
Md mole wt. dry	30.04800000190735	Bws saturated	.1916804268881441
Ms mole wt.	27.73863421839339	Vm corrected	47.911
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	99.3333333333333	Vf impingers	397
Tm abs temp	559.0033333333333	Vi impingers	200
Ts abs temp	868.5033333333333	sqr(Dp) avg.	.714703505575787
Pm abs prs	30.18057597532714	Bws kick flag	0
Ps abs prs	29.9610294128578	Vm C kick flag	0
Vm(std)	45.6649780900539	not in use	NA
Vwc(std)	9.27279	Wa2	0
Vwsg(std)	1.55595	Wa5	0
Bws	.1916804268881441	K1 constant	.04707
vs velocity	52.46689434732531	K2 constant	.04715
Rs vol rate	23718.37259849367	Tstd abs temp	527.67
n time min.	60	container 1	.01693
%I isokinetic	98.8291293231339	container 2	7.423999999999964D-02
vol rate	48229.83057359619	container 3	NA
mass dry	9.116999999999964D-02	container 4	33
mw mass wet	9.116999999999964D-02	container 5	NA
ZEA	82.42899505128014	not in use	NA

5/14 0825 UNIT #2

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05-17-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

XCO2 avg.	10.2	pmrd	2.892550182725949
XO2 avg.	9.7	pmrw	2.892550182725949
XCO avg.	0	ced	1.334679936634712D-02
As stack area	15.320718322265625	cew	1.334679936634712D-02
Vm avg. DGM	49.185	ced50%	1.643816742478097D-02
Dp avg. delta p	.5875	cew50%	1.643816742478097D-02
ts avg. temp	421.16666666666667	ced12	1.570211690158485D-02
DH avg. delta H	2.4566666666666667	cew12	1.570211690158485D-02
tmi avg. temp	97.66666666666667	ced7	1.656433146603715D-02
tmg avg. temp	91.16666666666667	cew7	1.656433146603715D-02
XN2 avg.	80.1	saturation flag	-1
Md mole wt. dry	30.02000000178814	Rws saturated	.1783223462470211
Me mole wt.	27.87656539958008	Vm corrected	49.185
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	94.41666666666667	Vf impingers	382
Tm abs temp	554.08666666666667	Vi impingers	200
Te abs temp	880.83666666666667	sqr(Dp) avg.	.7571383251240413
Fm abs pres	30.18063724983522	Rws kick flag	0
Fe abs pres	29.9705882361191	Vm C kick flag	0
Vm(std)	47.29533146288439	not in use	NA
Vwc(std)	8.56674	Wa2	0
Vwg(std)	1.6974	Wa5	0
Rws	.1783223462470211	K1 constant	.04707
vs velocity	55.82776733136092	K2 constant	.04715
Qv vol rate	25303.62897807033	Tstd abs temp	527.67
Qv time min.	60	container 1	2.4679999999999998D-02
Qv isokinetic	95.94493775269678	container 2	1.6230000000000019D-02
Qa vol rate	51319.28987199428	container 3	NA
md mass dry	4.091000000000017D-02	container 4	36
mw mass wet	4.091000000000017D-02	container 5	NA
XEA	84.74280207839728	not in use	NA

5/21 1016 #2

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05-28-1987

Page 1

ENVIRO-PRO processor direct variable dump V1.0 English/Metric units

XCO2 avg. 10.9
 XO2 avg. 8.7
 XCO avg. 0
 As stack area 15.320718322265625
 Vm avg. DGM 30.06
 Dp avg. delta p .5666666666666667
 ts avg. temp 411.5
 DI avg. delta H 1.1175
 () avg. temp 111.25
 () avg. temp 111.25
 .N2 avg. 80.4
 Md mole wt. dry 30.092000000762939
 Ms mole wt. 28.05536619073703
 An area nozzle 3.74394940740625D-04
 tm temp meter 111.25
 Tm abs temp 570.92
 Ts abs temp 871.17
 Pm abs pres 30.14216911534228
 Ps abs pres 30.0305882361191
 Vm(std) 29.05269261874298
 Vwc(std) 5.50719
 Vwsg(std) .3772
 Ems .1684282017538337
 V velocity 54.60235474606763
 Qa vol rate 25374.83606135349
 run time min. 48
 XI isokinetic 97.60926967189145
 Qa vol rate 50192.83780909529
 md mass dry 3.1450000000000001D-02
 mw mass wet 3.1450000000000001D-02
 XEA 69.45775116469892

pmd 3.63014849707071
 pmrw 3.63014849707071
 ced 1.670321943539691D-02
 csw 1.670321943539691D-02
 ced50% 1.886993335155235D-02
 csw50% 1.886993335155235D-02
 ced12 1.838886543346449D-02
 csw12 1.838886543346449D-02
 ced7 .0190307172983514
 csw7 .0190307172983514
 saturation flag -1
 Bws saturated .1684282017538337
 Vm corrected 30.06
 total # points 12
 Vf impingers 317
 Vi impingers 200
 sqr(Dp) avg. .7477479030408388
 Bws kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 3.1450000000000001D-02
 container 2 NA
 container 3 NA
 container 4 B
 container 5 NA
 not in use NA

5/21 1705 #2

%CO2 avg.	11.5	pard	4.330703635654533
XO2 avg.	9	parw	4.330703635654533
XCO avg.	0	cad	1.641596898375574D-02
As stack area	15.32071832265625	caw	1.641596898375574D-02
Vm avg. DGM	40.885	cad50%	1.916018009301828D-02
Dp avg. delta p	.7441666666666667	caw50%	1.916018009301828D-02
ts avg. temp	431	cad12	1.712970676565816D-02
DH avg. delta H	2.650833333333333	caw12	1.712970676565816D-02
tmi avg. temp	106.25	cad7	1.917495545602307D-02
tp avg. temp	106.25	caw7	1.917495545602307D-02
YN2 avg.	79.5	saturation flag	-1
Md mole wt. dry	30.20000000298023	Bws saturated	9.481966548862889D-02
Ms mole wt.	29.043200008373638	Vm corrected	40.885
An area nozzle	4.97440652865625D-04	total # points	12
tm temp meter	106.25	Vf impingers	261
abs temp	565.92	Vi impingers	200
abs temp	890.67	egr(Dp) avg.	.8578449567575251
Pm abs pre	30.25491421021908	Bws kick flag	0
Ps abs pre	30.0305882361191	Vm C kick flag	0
Vm(etd)	40.01317867071964	not in use	NA
Vwc(etd)	2.87127	Ma2	0
Vweg(etd)	1.3202	Ma5	0
Bws	9.481966548862889D-02	K1 constant	.04707
vs velocity	62.25262439380102	K2 constant	.04715
Qs vol rate	30801.43728733193	Tstd abs temp	527.67
run time min.	48	container 1	1.846999999999999D-02
XI isokinetic	83.35443015154688	container 2	2.410000000000068D-02
Qa vol rate	57225.29539101268	container 3	NA
md mass dry	4.257000000000066D-02	container 4	28
mw mass wet	4.257000000000066D-02	container 5	NA
%EA	75.07507578728418	not in use	NA

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Page

ENVIRO-PRI: process direct variable dump V1.0 English/Metric units

Q2 avg. 10.6
 X02 avg. 8.8
 X00 avg. 0
 As stack area 15.32071832265625
 Vm avg. DGM 35.675
 Dp avg. delta p .6383333333333333
 ts avg. temp 436.58333333333333
 DH avg. delta H 1.6058333333333333
 tmi avg. temp 88.41666666666667
 tmo avg. temp 87.66666666666667
 XN2 avg. 80.6
 Md mole wt. dry 30.04800000786781
 Me mole wt. 27.70263862031667
 An area nozzle 3.51880193415625D-04
 tm temp meter 88.04166666666667
 Tm abs temp 547.71166666666667
 Ts abs temp 896.25333333333333
 Tm abs pre 30.14807597708021
 Ts abs pre 29.98955882466376
 Vm(std) 34.66621953131728
 Vmc(std) 7.67241
 Vweg(std) .70725
 Bwe .1946681097293766
 vs velocity 59.27024533877442
 Qs vol rate 25892.98428569535
 run time min. 60
 T isokinetic 97.15327179850558
 Wa vol rate 54483.76402500555
 md mass dry .03245999999999991
 mw mass wet .03245999999999991
 XEA 70.52186242876619

parr 3.204136620378747
 parrw 3.204136620378747
 ced 1.444800750619818D-02
 csw 1.444800750619818D-02
 ced50% 1.642467432227805D-02
 csw50% 1.642467432227805D-02
 ced12 1.635623491267719D-02
 csw12 1.635623491267719D-02
 ced7 1.659729794678049D-02
 csw7 1.659729794678049D-02
 saturation flag -1
 Bwe saturated .1946681097293766
 Vm corrected 35.675
 total # points 12
 Vf impingers 1345
 Vi impingers 1182
 spr(Dp) avg. .7946435024367213
 Bwe kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 1.835999999999999D-02
 container 2 1.4099999999999991D-02
 container 3 NA
 container 4 15
 container 5 NA
 not in use NA

Q2 avg. 11.2
 Q22 avg. 7.7
 QCD avg. 0
 Ae stack area 15.32071832265625
 Vm avg. DGM 36.765
 Dp avg. delta p .5483333333333333
 te avg. temp 428.41666666666667
 DH avg. delta H 1.3475
 tmi avg. temp 92.83333333333333
 tmp avg. temp 91.25
 QN2 avg. 81.1
 Md mole wt. dry 30.10000001490116
 Me mole wt. 27.86051347254392
 An area nozzle 3.51880193415625D-04
 Qm temp meter 92.04166666666667
 Qa abs temp 551.71166666666667
 Qs abs temp 888.08666666666667
 Qm abs pre 30.1290808795738
 Qs abs pre 29.9910294128578
 Vm(std) 35.44403670963133
 Vwc(std) 7.48413
 Vweg(std) .5658
 Rwe .1850815321982904
 vs velocity 54.51748525187301
 Qs vol rate 24323.00283671308
 run time min. 60
 isokinetic 105.7448055184965
 Qa vol rate 50114.82211221077
 md mass dry 3.716000000000086D-02
 mass wet 3.716000000000086D-02
 A 56.16174630321331

pmd 3.370052009806408
 pmrw 3.370052009806408
 ccd 1.617701744012152D-02
 csw 1.617701744012152D-02
 ccd50% 1.684154195617942D-02
 csw50% 1.684154195617942D-02
 ccd12 1.733251868584448D-02
 csw12 1.733251868584448D-02
 ccd7 1.703488960188949D-02
 csw7 1.703488960188949D-02
 saturation flag -1
 Rwe saturated .1850815321982904
 Vm corrected 36.765
 total # points 12
 Vf impingers 1324
 Vi impingers 1165
 sgr(Dp) avg. .7363830463136791
 Rwe kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 1.466000000000001D-02
 container 2 2.250000000000085D-02
 container 3 NA
 container 4 12
 container 5 NA
 not in use NA

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Page 11

ENVIRON-PND processor direct variable dump V1.0 English/Metric units

CO2 avg. 10.8
 XO2 avg. 8.3
 XCO avg. 0
 As stack area 15.32071832265625
 Vm avg. DGM 35.629
 Dp avg. delta p .5725
 ts avg. temp 426.25
 DH avg. delta H 1.41
 tmi avg. temp 92.33333333333333
 tmo avg. temp 92
 XN2 avg. 80.9
 Md mole wt. dry 30.06000001132488
 Ms mole wt. 27.95814955443721
 An area nozzle 3.51880193415625D-04
 tm temp meter 92.16666666666667
 Tm abs temp 551.8366666666667
 Ts abs temp 885.92
 Pa abs pre 30.13367646768019
 Ps abs pre 29.99250000105185
 vm(std) 34.34631055999344
 VmC(std) 6.87222
 Vmag(std) .3772
 Bws .1742827906230467
 vs velocity 55.64896050896046
 Qs vol rate 25219.57401876082
 run time min. 60
 I isokinetic 98.82695205734169
 wa vol rate 51154.92293438428
 md mass dry 3.536000000000034D-02
 mw mass wet 3.536000000000034D-02
 BEA 63.56451471604416

pmd 3.431285032015661
 pmw 3.431285032015661
 ced 1.588539761925769D-02
 csw 1.588539761925769D-02
 ced50% 1.732191568443525D-02
 csw50% 1.732191568443525D-02
 ced12 1.765044179917521D-02
 csw12 1.765044179917521D-02
 ced7 1.752436726451581D-02
 csw7 1.752436726451581D-02
 saturation flag -1
 Bws saturated .1742827906230467
 Vm corrected 35.629
 total # points 12
 Vf impingers 1250
 Vi impingers 1104
 sgr(Dp) avg. .7539208355672653
 Bws kick flag 0
 Vm C kick flag 0
 not in use NA
 Wa2 0
 Wa5 0
 K1 constant .04707
 K2 constant .04715
 Tstd abs temp 527.67
 container 1 .01538
 container 2 1.998000000000033D-02
 container 3 NA
 container 4 8
 container 5 NA
 not in use NA

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Q2 avg.	12	pmrd	4.23278376386771
Q22 avg.	7.8	pmrw	4.23278376386771
QCO avg.	0	csd	1.911786248478093D-02
As stack area	15.32071832265625	csw	1.911786248478093D-02
Vm avg. DGM	37.265	csd50%	2.017920356230977D-02
Dp avg. delta p	.5966666666666667	csw50%	2.017920356230977D-02
ts avg. temp	438.5	csd12	1.911786248478093D-02
DH avg. delta H	1.4575	csw12	1.911786248478093D-02
tmi avg. temp	96	csd7	2.028536557128412D-02
tmo avg. temp	95.25	csw7	2.028536557128412D-02
ZN2 avg.	80.2	saturation flag	-1
Md mole wt. dry	30.23200001120567	Bws saturated	.157390402281733
Ms mole wt.	28.30680060873185	Vm corrected	37.265
An area nozzle	3.51880193415625D-04	total # points	12
tm temp meter	95.625	Vf impingers	1263
Tm abs temp	555.295	Vi impingers	1129
ts abs temp	898.17	sqr(Dp) avg.	.7663568122153158
Fm abs pre	30.20714911464105	Bws kick flag	0
Pe abs pre	30.061764470695482	Vm C kick flag	0
Vm(std)	35.78675181624649	not in use	NA
Vmc(std)	6.30738	Wa2	0
Vwsg(std)	.3772	Wa5	0
Bws	.157390402281733	K1 constant	.04707
vs velocity	56.53953981358393	K2 constant	.04715
Qs vol rate	25850.31205268073	Tstd abs temp	527.67
run time min.	60	container 1	.02653
isokinetic	100.459161047966	container 2	1.780999999999999D-02
Wa vol rate	51973.58181459167	container 3	NA
md mass dry	4.433999999999999D-02	container 4	8
mass wet	4.433999999999999D-02	container 5	NA
SA	58.32735154131695	not in use	NA



Westinghouse Electric Corporation
Resource Energy Systems Division

SIERRA CLUB



The Florida Chapter

Northwest Florida Group
Sierra Club
4649 Soundside Drive
Gulf Breeze, Fl. 32561
11-4-86

Bay County Commission
644 Mulberry Ave.
Panama City, Fl. 32401

Dear Sirs,

The Northwest Florida Sierra Club an active though belated interest in the new Bay County incinerator. We would like a meeting with you, Weston Engineering your consultant, Westinghouse Corp. the builder and operator, Ford Motor Credit Corp. the owner, the Florida Department of Environmental Regulation, and the Environmental Protection Agency Region IV.

The following details particularly concern us.

- 1) We would like to review the baseline air and water quality studies of point sources and general background used to site this plant and determine its proposed operating emissions.
- 2) We are concerned about the lack of front end separation of plastics, glass, and metals from the combustible material on the front end of this plant and would like to review Westinghouse test burn data on similar plants.
- 3) The lack of acid gas(SO₂,HCl,NO_x) sensors in the output gas stream makes it impossible to know what is being emitted and set discharge limits. There are already out of DER limits total reduced sulfur emissions in the immediate area and this plant may well exceed those quantities with no way to measure the nature and quantity.
- 4) Birth defect up to 700% above the national norms were reported by the Congenital Defects Surveillance and Referral Program report for 1979-1980. We would like to know what has been done since then to update this study and how the data compares with other areas

DER

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BAQM

such as Brunswick, Georgia which has paper mills and tall oil plants, and Port St. Joe, Pensacola, and Palatka which have only paper mills. 5) We wish to review the output gas, water, and ash streame from similar plant and the disposal sites for ash and water for the new plant.

The plant drawings will be reviewed in Pensacola or Tallahassee before we have a conference with you. Thank you for your consideration.

Sincerely,

A handwritten signature in cursive script that reads "Richard D. Radford".

Richard D. Radford

Chair

Northwest Florida Group

Sierra Club

CC attached

CC

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